

## 7.1 RE energy: Bi(111) vs Au(111)

The RE band splitting has been measured on Bi(111) as reported in the figure below.

- 1) Evaluate the RE energy  $E_R$
  - 2) Compare the Bi(111) surface ( $[Xe] 4f^{14} 5d^{10} 6s^2 6p^3$ ) with the case of Au(111) shown in the lecture ( $([Xe] 5d^{10} 6s^1)$ ).
- Are the splitting the same? Is there a correlation between  $E_R$  and the work function (i.e. the potential one needs to overcome to extract an electron from a given material)? Comment

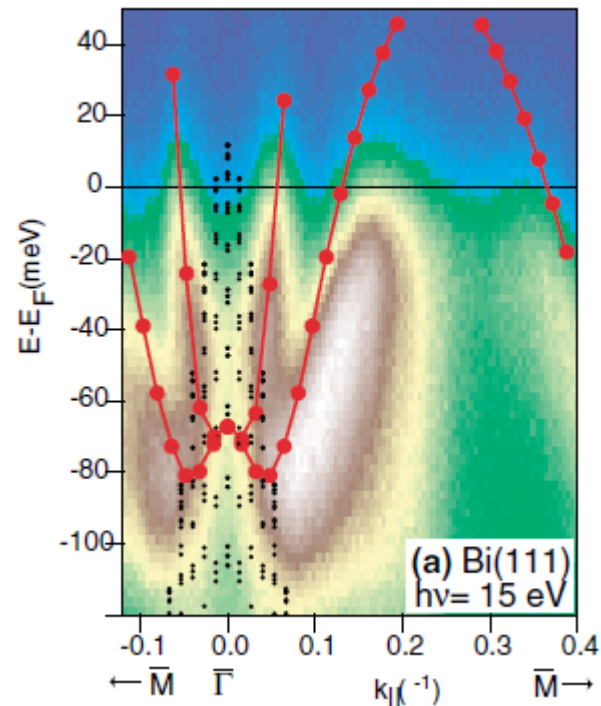
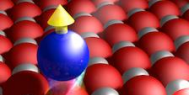


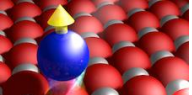
Table 1. The periodic table of elements\*

<sup>1</sup> H 13.598 0.7542																	<sup>2</sup> He 24.58 -0.22				
<sup>3</sup> Li 5.39 2.9 0.6182	<sup>4</sup> Be 9.32 4.98 -0.19															<sup>5</sup> B 8.30 4.45 0.227	<sup>6</sup> C 11.26 5.0 1.2629	<sup>7</sup> N 14.54 7.46 -0.07	<sup>8</sup> O 13.61 13.61 1.462	<sup>9</sup> F 17.42 17.42 3.399	<sup>10</sup> Ne 21.56 21.56 -0.3
<sup>11</sup> Na 5.14 2.75 0.5479	<sup>12</sup> Mg 7.64 3.66 -0.22															<sup>13</sup> Al 5.98 4.28 0.442	<sup>14</sup> Si 8.15 4.85 1.385	<sup>15</sup> P 10.55 10.55 0.7464	<sup>16</sup> S 10.36 10.36 2.077	<sup>17</sup> Cl 13.01 13.01 3.615	<sup>18</sup> Ar 15.76 15.76 -0.36
<sup>19</sup> K 4.34 2.30 0.5015	<sup>20</sup> Ca 6.11 2.87 0.0215	<sup>21</sup> Sc 6.56 3.4 0.189	<sup>22</sup> Ti 6.83 4.33 0.080	<sup>23</sup> V 6.74 4.3 0.526	<sup>24</sup> Cr 6.76 4.5 0.667	<sup>25</sup> Mn 7.43 4.1 <0.005	<sup>26</sup> Fe 7.90 4.5 0.164	<sup>27</sup> Co 7.86 5.0 0.662	<sup>28</sup> Ni 7.63 5.15 1.157	<sup>29</sup> Cu 7.72 4.51 1.228	<sup>30</sup> Zn 9.39 4.33 0.093	<sup>31</sup> Ga 6.00 4.15 0.31	<sup>32</sup> Ge 7.89 5.0 1.23	<sup>33</sup> As 9.81 4.77 0.81	<sup>34</sup> Se 9.75 5.9 2.0208	<sup>35</sup> Br 11.84 9.01 3.364	<sup>36</sup> Kr 14.00 14.00 <0				
<sup>37</sup> Rb 4.18 2.16 0.4859	<sup>38</sup> Sr 5.69 2.59 0.11	<sup>39</sup> Y 6.38 3.1 0.308	<sup>40</sup> Zr 6.95 4.05 0.427	<sup>41</sup> Nb 6.88 4.3 0.894	<sup>42</sup> Mo 7.18 4.6 0.747	<sup>43</sup> Tc 7.28 4.9 0.55	<sup>44</sup> Ru 7.36 4.71 1.05	<sup>45</sup> Rh 7.46 4.98 1.138	<sup>46</sup> Pd 8.343 5.12 0.558	<sup>47</sup> Ag 7.57 4.26 1.303	<sup>48</sup> Cd 7.57 4.22 0.260	<sup>49</sup> In 8.99 4.12 0.30	<sup>50</sup> Sn 8.89 4.42 1.15	<sup>51</sup> Sb 9.78 4.55 1.07	<sup>52</sup> Te 9.75 4.95 1.9708	<sup>53</sup> I 10.45 10.45 3.059	<sup>54</sup> Xe 12.13 12.13 <0				
<sup>55</sup> Cs 3.893 2.14 0.4716	<sup>56</sup> Ba 5.21 2.7 0.17	<sup>57</sup> La 5.57 2.96 0.518	<sup>72</sup> Hf 7.00 3.9 ≥0.1	<sup>73</sup> Ta 7.89 4.25 0.323	<sup>74</sup> W 7.98 4.55 0.816	<sup>75</sup> Re 7.87 4.87 0.12	<sup>76</sup> Os 8.70 4.83 1.12	<sup>77</sup> Ir 9.0 5.27 1.566	<sup>78</sup> Pt 8.96 5.65 2.128	<sup>79</sup> Au 9.22 5.1 2.309	<sup>80</sup> Hg 10.43 4.49 0.186	<sup>81</sup> Tl 6.11 4.1 0.3	<sup>82</sup> Pb 7.41 4.25 0.364	<sup>83</sup> Bi 7.29 4.22 0.946	<sup>84</sup> Po 8.43 5.0 1.9	<sup>85</sup> At 9.64 2.8 2.8	<sup>86</sup> Rn 10.74 10.74 <0				
<sup>87</sup> Fr 3.98 2.1	<sup>88</sup> Ra 5.28 2.8 0.17	<sup>89</sup> Ac 5.17 3.2																			
			<sup>58</sup> Ce 5.466 2.97 0.518	<sup>59</sup> Pr 5.42 2.96 ≥0.1	<sup>60</sup> Nd 5.49 3.2 ≥0.05	<sup>61</sup> Pm 5.54 3.1	<sup>62</sup> Sm 5.6 2.85 ≥0.05	<sup>63</sup> Eu 5.67 2.5 ≥0.05	<sup>64</sup> Gd 6.14 3.17 ≥0.1	<sup>65</sup> Tb 5.85 3.15 ≥0.1	<sup>66</sup> Dy 5.93 3.25 0.15	<sup>67</sup> Ho 6.02 3.22 ≥0.005	<sup>68</sup> Er 6.10 3.25 ≥0.005	<sup>69</sup> Tm 6.18 3.1 0.035	<sup>90</sup> Yb 6.25 3.0 0.010						
			<sup>90</sup> Th 6.08 3.4 ≥0.05	<sup>91</sup> Pa 5.89 3.7 ≥0.05	<sup>92</sup> U 6.05 3.63 ≥0.05	<sup>93</sup> Np 6.19 3.9	<sup>94</sup> Pu 6.06 3.6 ≥0.05	<sup>95</sup> Am 6.00 3.7	<sup>96</sup> Cm 6.02 3.9	<sup>97</sup> Bk 6.23 3.8	<sup>98</sup> Cf 6.30 4.0	<sup>99</sup> Es 6.42 3.3	<sup>100</sup> Fm	<sup>101</sup> Md	<sup>102</sup> No						

\*The numbers under the symbols of elements are the ionization energies (upper rows), work functions (middle rows) and electron affinities (lower rows), given in eV/atom.



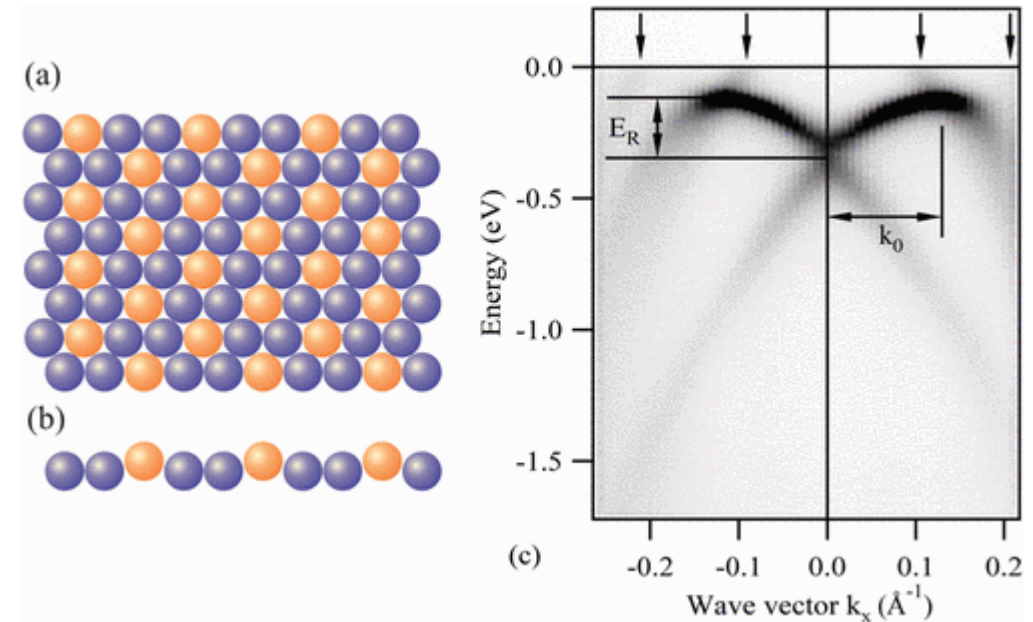
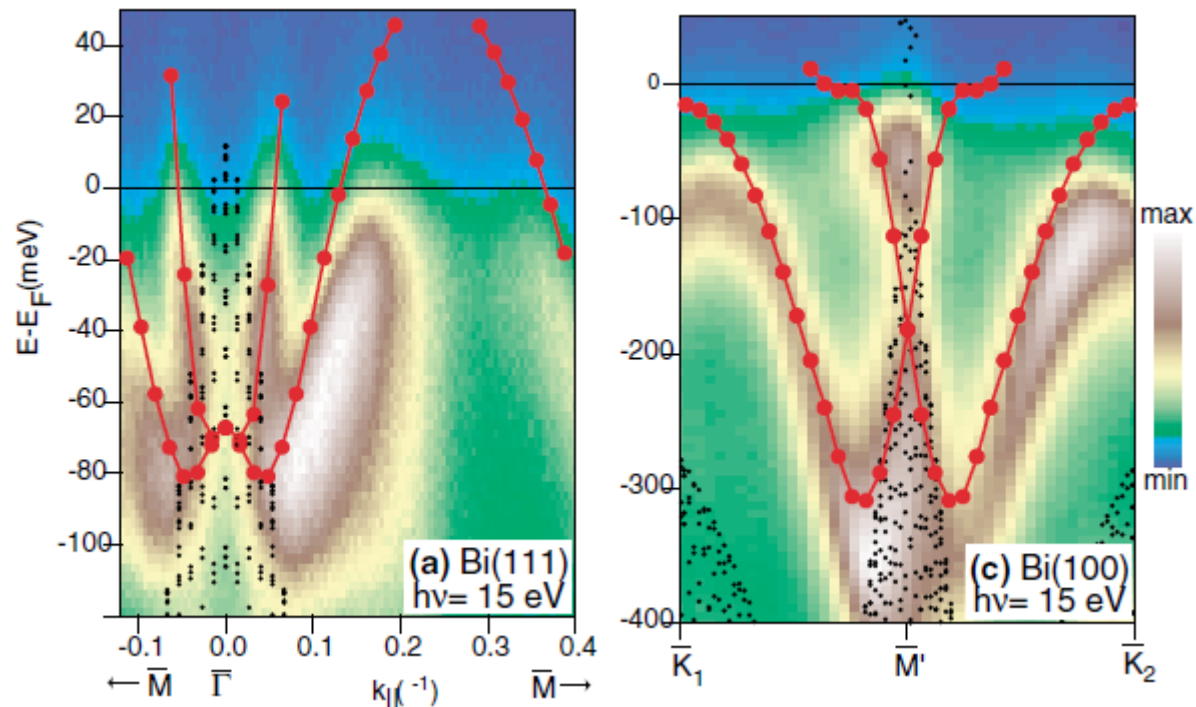
- 1)  $E_R$  is about 17 meV for Bi(111)
- 2) In the case of Au(111)  $E_R$  is about 2 meV i.e about 10 times smaller. The work function of Bi (4.22 eV) is smaller than the one of Au (5.1 eV). On the other end both Bi (Z=83) and Au (Z=79) are heavy metals with similar SOC. This big difference suggests that the Rashba effect is very sensitive to the detail of the local atomic potential

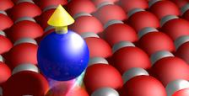


## 7.2 RE energy on Bi surfaces

The RE band splitting has been measured on different Bi surfaces as reported in the figures below.

- 1) Evaluate the RE energy  $E_R$  for the different samples
- 2) Compare the different Bi surfaces: are the splitting the same? comment





- 1)  $E_R$  is 17 meV for Bi(111), 130 meV for Bi(100) and 250 meV for the AgBi alloy
- 2) The big difference observed between the three Bi surface demonstrates that the Rashba effect is very sensitive to potential gradient in z direction but also in the x-y plane