

Neutronics Exercises

5. 2

5.1 2

5.2 4

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5. LWR Plants

5.1 Pressurizer in PWR

Exercise description:

Using the definition of isobaric thermal expansion and isothermal compressibility, explain why in a system completely filled with an incompressible fluid (e.g. water) subject to temperature variation there is a need for a pressurizer? Please use https://link.springer.com/chapter/10.1007/978-3-662-53219-5_5 for the numerical data.

(a) A vessel having a cross-section A is filled with water at $T_0=323K$. The vessel is open. By means of a heater, the water temperature is increased by 5K. Estimate the variation of the water level in the vessel.

(b) A closed vessel is filled with water at an initial pressure of $p_0=1\text{bar}$ and an initial temperature $T_0=323K$. By means of a heater, the water temperature is increased by 5K. Estimate the variation of pressure in the vessel.

(c) A closed vessel is filled with a mixture of steam and water at an initial pressure of $p_0=1\text{bar}$ and an initial temperature $T_0=423K$. By means of a heater, the water temperature is increased by 5K. Estimate the variation of the pressure in the vessel.

Expected results: (a) 0.00229 (b) $\Delta P = 52\text{bars}$ (c) $\Delta P = 0.012\text{bars}$

Exercise solution:

Derivative properties: their value depends on the derivative of the properties characterizing the state of a fluid

$$\beta = \frac{1}{v} \left. \frac{\partial v}{\partial T} \right|_p = - \frac{1}{p} \left. \frac{\partial p}{\partial T} \right|_v \quad \begin{matrix} \text{Isobaric} \\ \text{thermal expansion} \end{matrix}$$

$$v = \text{specific volume} = \frac{V}{m} = \frac{1}{p} \quad [\text{m}^3/\text{kg}]$$

$$k = - \frac{1}{v} \left. \frac{\partial v}{\partial p} \right|_T = \frac{1}{p} \left. \frac{\partial p}{\partial v} \right|_T \quad \begin{matrix} \text{isothermal} \\ \text{compressibility} \end{matrix}$$

Isobaric thermal expansion: measure of volume change with temperature at constant pressure. Unit of $[\text{K}^{-1}]$

Isobaric compressibility: measure of density change with pressure at constant temperature. Unit of $[\text{Pa}^{-1}]$

(a) $\beta = 4.57 \times 10^{-4} \text{ K}^{-1}$ at 1bar and 50C for water

See p 284 of https://link.springer.com/chapter/10.1007/978-3-662-53219-5_5

Then $\Delta V = V_0 \beta \Delta T$ and $\Delta h = \frac{\Delta V}{A} \beta \Delta T$ and finally $\Delta h = \frac{\Delta h}{h_0} = \beta \Delta T = 0.00285$

(b) $\kappa = 0.44e - 6 \text{ Pa}^{-1}$ at 1bar and 50C for water

$$d\nu = \left. \frac{\partial \nu}{\partial T} \right|_P dT + \left. \frac{\partial \nu}{\partial p} \right|_T dp = \nu \beta \Delta T - \nu \kappa \Delta p$$

V is constant so $\Delta \nu = V_0 (\beta \Delta T - \kappa \Delta p) = 0$

$$\Delta p = \beta \frac{\Delta T}{\kappa} = 52 \text{ bars!}$$

(c) for steam at 150 C, $\beta = 2452.4e - 6 \text{ K}^{-1}$ and $\kappa = 10086e - 6 \text{ Pa}^{-1}$

$$\Delta p = \beta \frac{\Delta T}{\kappa} = 0.012 \text{ bars}$$

To avoid large pressure changes in the system, a pressurizer, e.g. a closed volume connected to the primary coolant loop, is needed.

5.2 Power of a PWR main circulation pump

Exercise description:

Consider a typical 1300 MWe PWR at 155 bar, with thermal power of 3817 MWth, with temperature in cold leg and hot leg of TCL = 290 °C and THL = 320 °C respectively. Knowing that there are 4 main loops and therefore 4 main circulation pumps and considering that the pressure drop across a loop is of 8 bar, compute the power needed to operate one main circulation pump. Assume a pump efficiency of 80%.

Exercise solution:

Q: how to calculate the power we need to operate the main circulation pumps?

$$\begin{aligned} P_{th} &= 3817 \text{ MW} & T_{in} &= 290^\circ \text{C} & \text{cold leg} \\ P_{el} &= 1300 \text{ MW} & T_{out} &= 320^\circ \text{C} & \text{hot leg} \\ N_c &= 4 \text{ loops} & P &= 155 \text{ bar} \end{aligned}$$

$\Delta p = 8 \text{ bar}$ friction losses along primary circuit

$$\dot{m}_{core} = \frac{P_{th}}{C_p \Delta T} = 31000 \text{ kg/s}$$

$$\dot{m}_{loop} = \frac{\dot{m}_{core}}{N_c} = 7758 \text{ kg/s}$$

$$\eta \frac{P_p}{\dot{m}} = \frac{\dot{m}}{\rho} \left[\frac{g \Delta H}{\eta} + \frac{\Delta p}{\rho} \right]$$

closed loop

$$P_{pump} = \frac{\dot{m}_{loop} \times \Delta p_{losses}}{\eta \cdot \rho} = \frac{7758 \times 8 \times 10^5}{0.8 \times 1000} = 7,76 \text{ MW}$$

$$4 \text{ loops} \Rightarrow 4 \times 7.76 = 31 \text{ MW}$$

→ In a PWR about 2% of the electric power is needed for the main circulation pumps

5.3 Energy balance on condenser

Exercise description:

Consider a 900 MWe reactor with a Rankine cycle efficiency of about 30%. Knowing that, for environmental reasons, the temperature increase on the secondary side of the condenser cannot exceed 10 K, compute the mass flow-rate needed for the secondary side of the condenser.

Exercise solution:

The outlet water temperature in a condenser $T_{c,out}$ is limited by environmental regulations. Generally

$$(T_{c,out} - T_{c,in}) \approx 10 \text{ K}$$

Because $\eta = 30\%$ \Rightarrow 70% of the thermal power produced by the reactor is released to the condenser coolant (river, sea)

$$70\% P_{th} \approx 2100 \text{ MW} = P_c$$

Energy balance on condenser

$$\dot{m}_c (h_{c,out} - h_{c,in}) = P_c$$
$$\Rightarrow \dot{m}_c c_p \Delta T_c = P_c \quad \uparrow 2100 \text{ MW} \Rightarrow \boxed{\dot{m}_c \approx 50.000 \text{ kg/s}}$$

$\uparrow \quad \uparrow$
 4180 J/kgK 10°C
 $(50 \text{ m}^3/\text{s}!)$