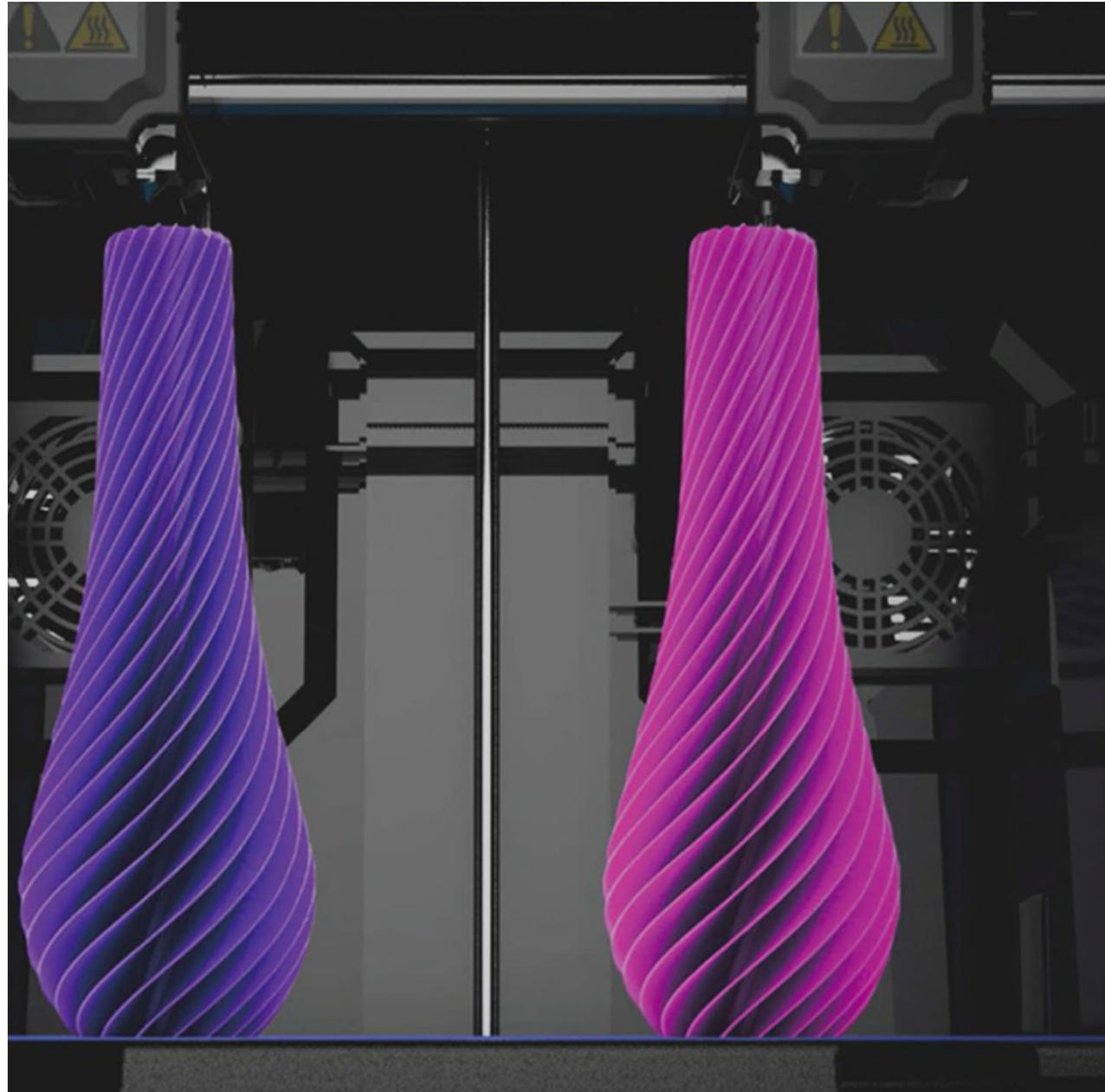


# EPFL

Advanced Ceramic Technologies

## Shaping

M. Stuer



*What's on the*  
**- MENU -**  
*today ?*



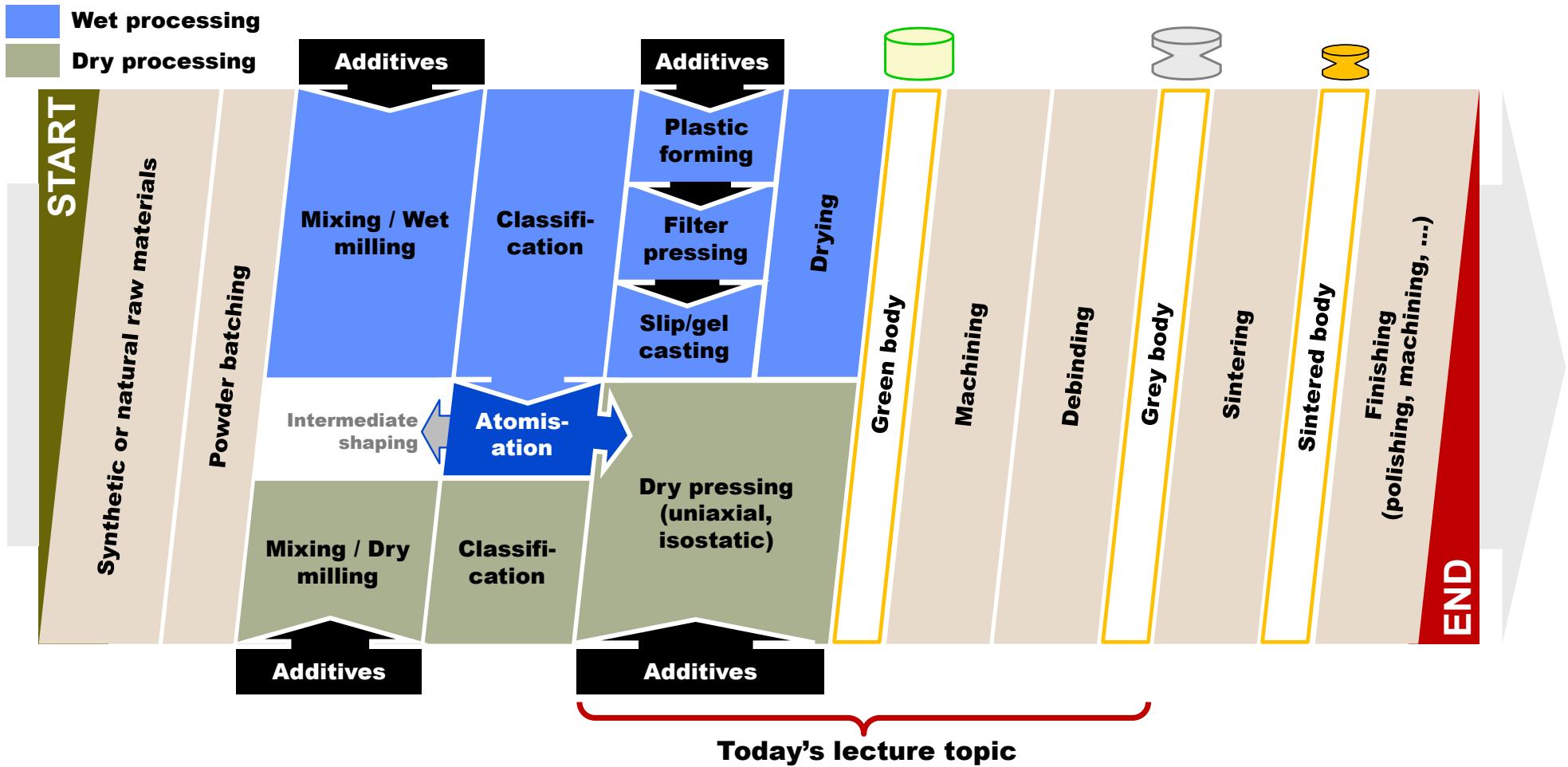
## Outline

- ◆ Shaping in the global ceramic process map
- ◆ Dry pressing
- ◆ Wet shaping
  - From slurries: Slip-casting, filtre pressing and tape casting
  - From pastes: extrusion, jiggering and injection
- ◆ Additive manufacturing
  - 3D shaping techniques
- ◆ Drying and debinding
- ◆ Case study: Shaping of micro-medical devices

### **Learning objectives:**

- ◆ Refresh typical shaping techniques
- ◆ Be able to chose the appropriate shaping technique
- ◆ Awareness of critical process steps
- ◆ Prepare for knowledge-based decision taking

# Global ceramic powder processing chain



## General shaping aspects

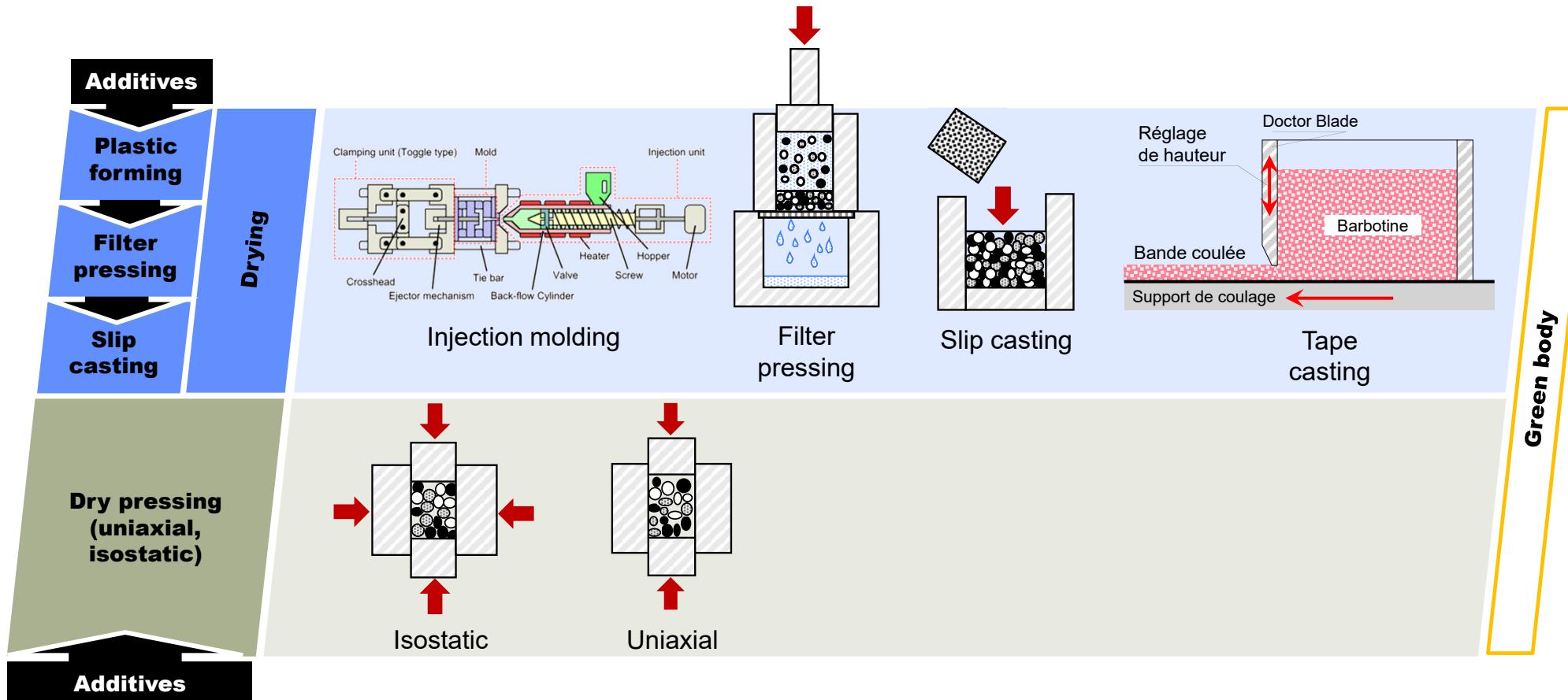
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- ◆ **Shaping:**

**Transformation** of a ceramic **powder** into a **green body** which shall be ideally **near-net-shape** due to the high machining costs of sintered ceramics!

- ◆ **Numerous shaping methods** exist or are under development. There is **no “single best method”**.
- ◆ The shaping method shall be **selected based on part**:
  - **Size**
  - **Shape complexity**
  - **Dimensional tolerances**
  - **Production volume**
  - **Production rate**

# Typical established wet and dry shaping techniques



If machining required, **preferential to machine in the green/pre-sintered state!** (e.g. machining costs)

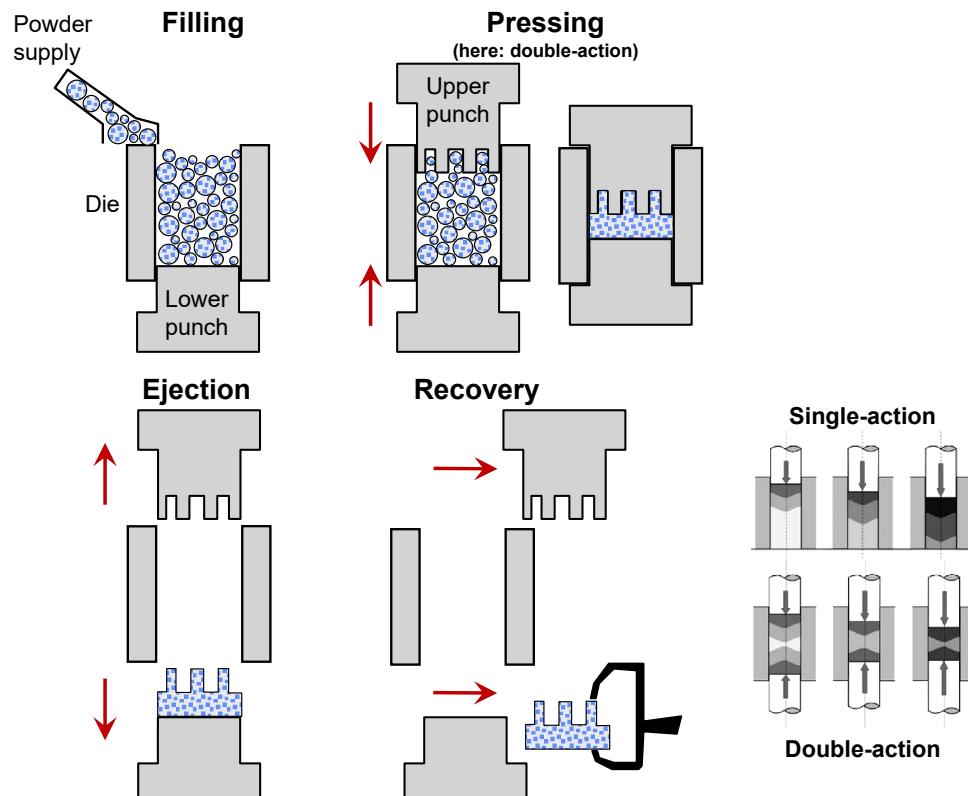
## - DRY SHAPING -

**Dry powder pressing**

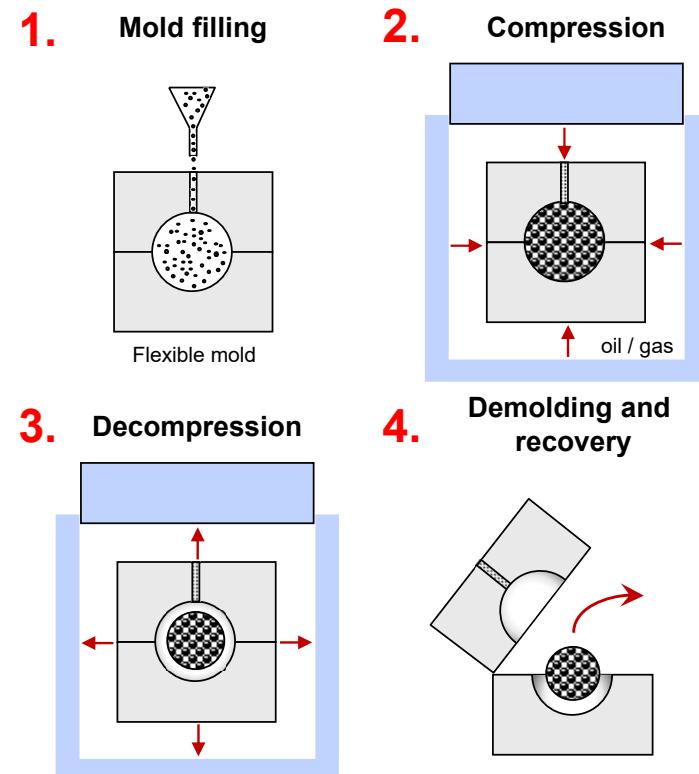


# Dry pressing: Schematic process representations

## #1\_uniaxial pressing



## #2\_cold isostatic pressing (CIP)



Watch here: <https://www.azom.com/materials-video-details.aspx?VidID=386>

Watch here: [https://youtu.be/77fu\\_aiEJkM](https://youtu.be/77fu_aiEJkM)

# Dry pressing: Key process facts

## #1 \_ uniaxial pressing

### Typical use cases:

Rather parts with simple shapes (2D+) and typically millimeter scale thickness

### Advantages:

- High **production rate** (up to 1000 pcs/hr)
- Can be widely **automatized**
- **Dimensional tolerances**  
(± related to mold except for the thickness)

### Limitations:

- **Tooling costs**
- **Part complexity** is limited
- **Part size** is limited
- Pressure/ejection cycle induced **part flaws**
- **Part thickness tolerances** depend on mold filling variability

## #2 \_ cold isostatic pressing (cIP)

### Typical use cases:

Parts with simple or complex shapes and dimensions unsuitable for uniaxial pressing

### Advantages:

- **Homogeneity** of green body
- **Density** of green body
- Compatibility with **large and complex part shapes**

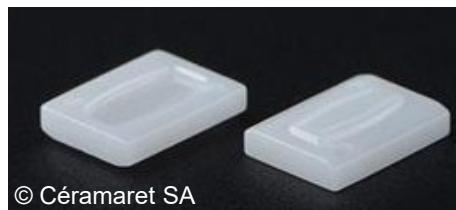
### Limitations:

- Can only be **semi-automated**  
→ low production rates (and volumes)
- **Dimensional tolerances** (deformable molds)  
→ resurfacing typically required



## Dry pressing: Process use case examples

### #1\_uniaxial pressing



### #2\_cold isostatic pressing

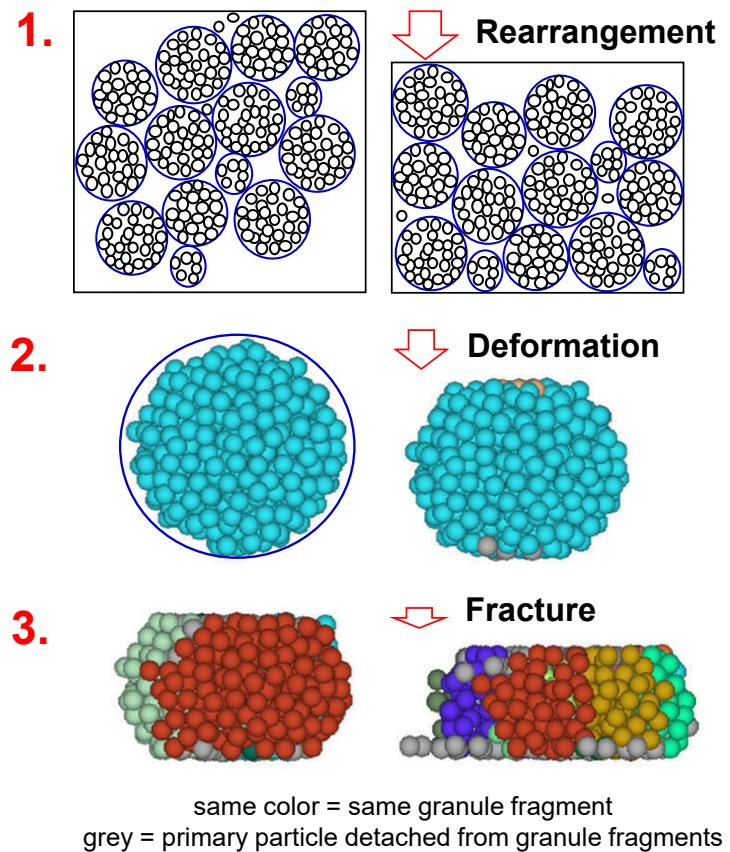


# Dry pressing: Objectives and process stages

- ◆ **Dry pressing** (ideally) requires **granulated powder feedstocks** (e.g. flow to fill molds)
- ◆ Dry pressing aims at achieving green bodies that are:
  - **dense**
    - Promote sintering and minimize shrinkage
  - **homogeneous** and **defect free**
    - Prevents **distortions** during sintering
    - Prevents **microstructural flaws** and inhomogeneities
- ◆ Dry pressing is **influenced by** properties of:
  - **Primary particles** (morphology, shape, size distribution)
  - **Granules** (flowability, additives, density, size distribution, ...)
- ◆ Dry pressing follows **3 stages:**
  1. Granule **rearrangement**
  2. Granule **deformation**
  3. Granule **fracture** and **densification**

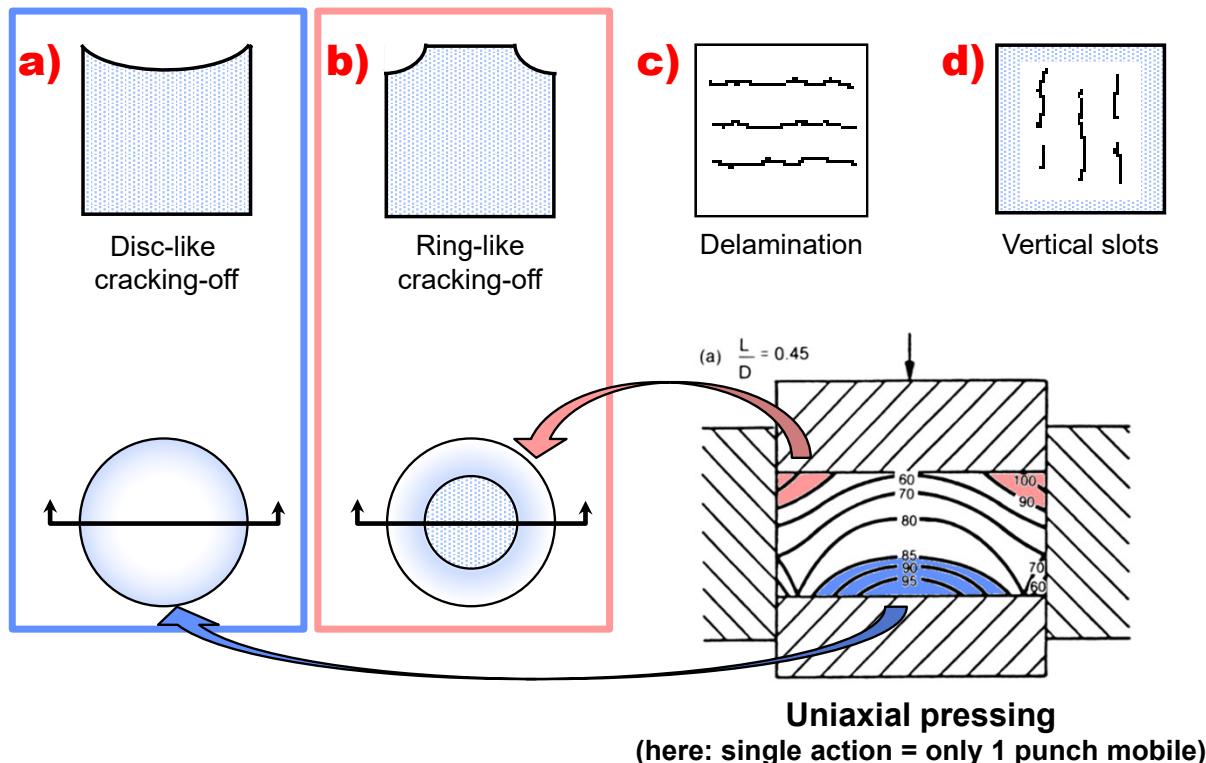
◦ Primary particle (p.ex. 0.2-1  $\mu\text{m}$ )

◦ Powder granule (50-300  $\mu\text{m}$ )



## Dry pressing: Defect occurrence during uniaxial pressing

**Inappropriate pressing conditions** (pressure, lubrication, tool adjustment) and **granule properties** may generate various types of part defects:



### Potential root causes by defect type:

- a) b):** Excessive pressure, friction, bad tool adjustment
- c):** Excessive pressure
- d):** Granules' solid loading too small  
→ compaction rate too high



# Dry pressing: Density gradients

**Density and internal energy gradients** occur due **friction** between

- particles themselves (particle-particle interaction)
- particles and the mold (particle-mould interaction)

leading to **pressure drops**

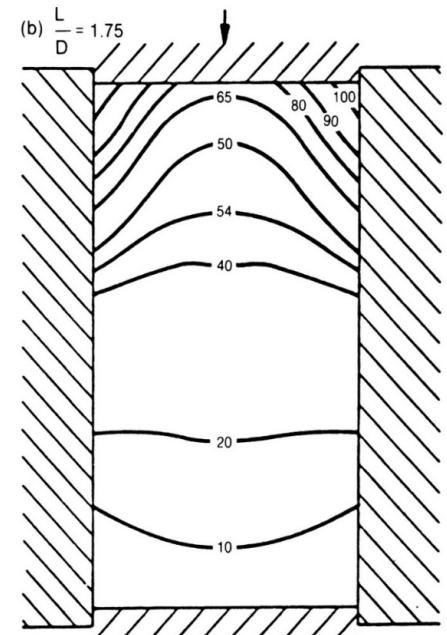
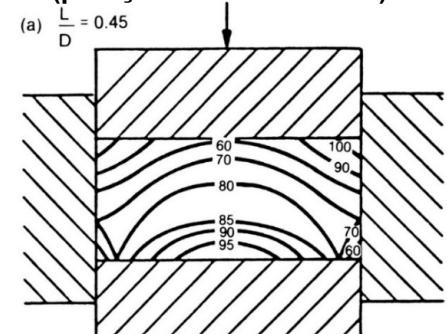
Resulting gradients cause

- Part **deformations** or **failures** during part ejection or sintering
- Microstructural **inhomogeneities** due to non-uniform sintering

## Corrective measures:

- Use of double-action mold → transmission de pression
- Addition of lubricants → oleic acid, stearic acid, ...
- Isostatic pressing → reduces gradient substantially but slow and expensive

Pressage uniaxial simple  
(poinçon du haut mobile)



## - WET SHAPING -

### Shaping of suspensions

*M. Stuer*



# Slip casting

## Use:

- ◆ Large parts and/or parts with complex shapes (ex. sinks, cups, ...)

## Advantages:

- ◆ Green body density close to theoretical maximum (= stable suspensions, dense particles arrangement)
- ◆ Compatible with large and complex shapes

## Limitations:

- ◆ **Wall thickness** limited to **~1 cm maximum**
- ◆ **Slow** process:
  - Slip casting time: Open: 80 minutes  
Pressure: 2 minutes
  - But **drying step:** **6 days**
- ◆ **Partially automatable**

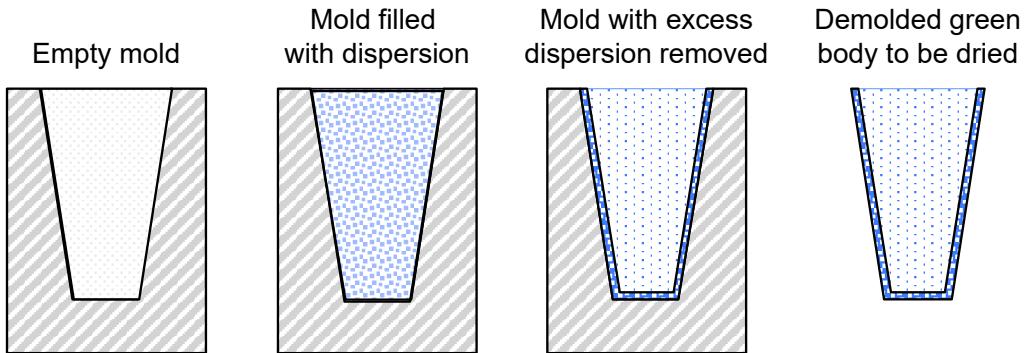
**Note:** It takes 1 week of preparation to mix, homogenize and mill the dispersion!



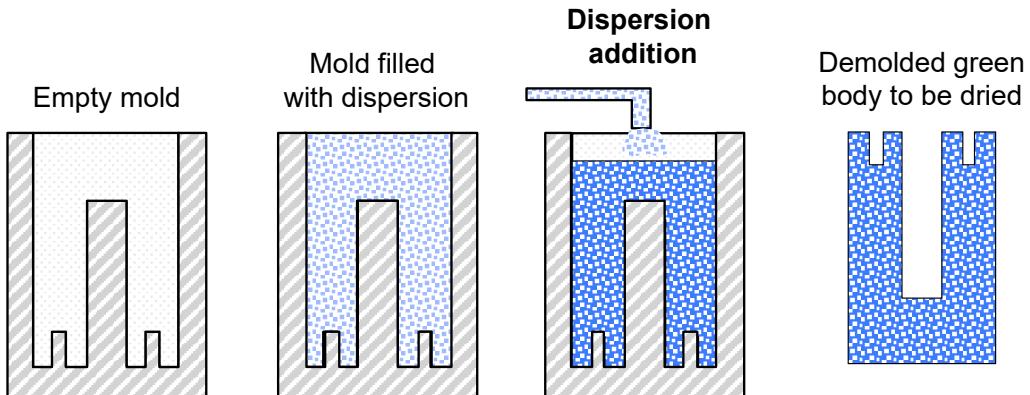
Suspensions

# Illustration of the slip casting process

## “open” or “recessed” slip casting



## “between plasters” or “plain” slip casting



## ◆ Principle:

- **Porous mold** absorbs water by capillary action
- Ceramic particles carried by the water form a **dense layer on the mold**.
- Layer follow Darcy's law (next slide)

## ◆ Variants:

- Open
- Between plasters

Slip casting can be **accelerated** by adding **pressure** (e.g. pressure casting)

A specific variant of slip casting is filter pressing, replacing the plaster with a filter (next slide)



# Filter pressing

## Advantages:

- ◆ Maximal green body density
- ◆ Possibility to seed texture at high flow rates (e.g. anisometric particles)

## Limitations:

- ◆ **Part thickness** limited at **~1 cm maximum**
- ◆ **Simple part shapes** (2D+ like for uniaxial pressing)

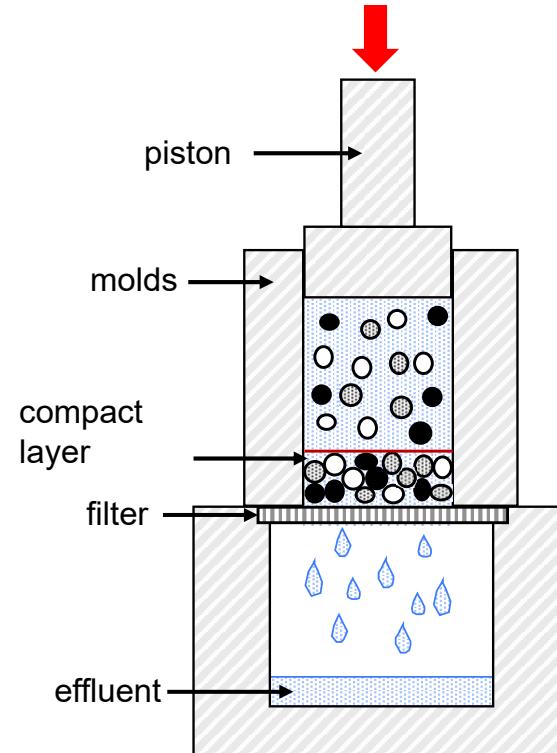
## Filtration kinetics:

- ◆ Follow a parabolic law derived from **Darcy's law** (also applicable to slip casting: pressure = capillary pressure and pressure casting: pressure = applied pressure)
- ◆ Compact layer thickness  $d$  can be expressed by:

$$d = \left[ \frac{2K_p P t}{\eta} \left( \frac{v_1}{v_0} - 1 \right) \right]^{\frac{1}{2}} \Rightarrow d \propto \sqrt{t}$$

- $K_p$  : permeability of compact layer
- $P$  : applied pressure
- $t$  : time (since process start)

- $\eta$  : viscosity of liquid (effluent)
- $v_1$  : volume fraction of particles in compact layer
- $v_0$  : volume fraction of particles in suspension/disperion



Suspensions

# Tape casting

## Use:

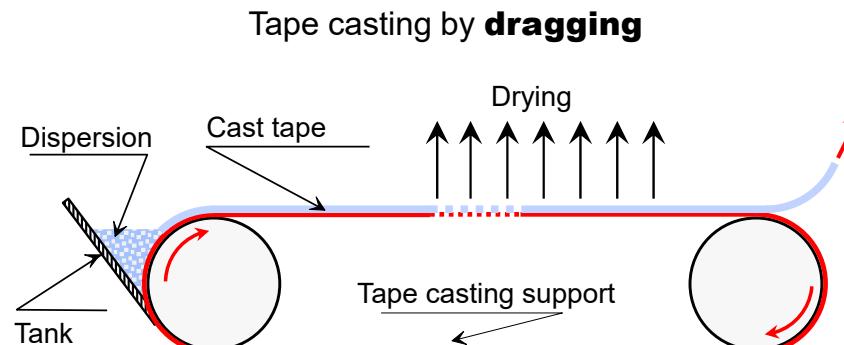
- ◆ Production of thick (10-1250  $\mu\text{m}$ ) ceramic tapes (typically large and flat)

## Advantages:

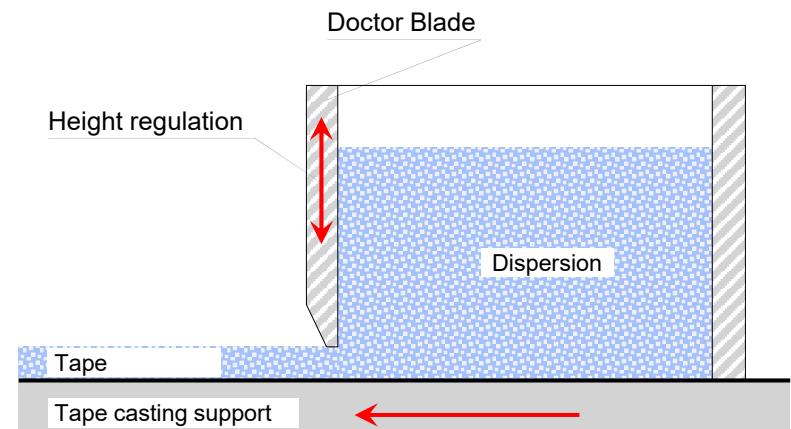
- ◆ High production **volume and speed** (automatable)
- ◆ Tape thickness control
- ◆ Possibility to generate **texture** with anisometric particles (e.g. shear stresses under doctor blade)
- ◆ Possibility to **cosinter tapes** of different materials (ex. electrodes and dielectric ceramics → capacitors)

## Limitations:

- ◆ Maximum film thickness is limited
- ◆ Typically uses (expensive) organic binders and solvents
- ◆ Careful debinding needed



Tape casting by **dragging**



Suspensions

## - WET SHAPING -

### Shaping of pastes

*M. Stuer*



# Common plastic forming techniques

## Extrusion

### Use:

Shapes with 2D structures extruded from **aqueous or organic** pastes

### Advantages:

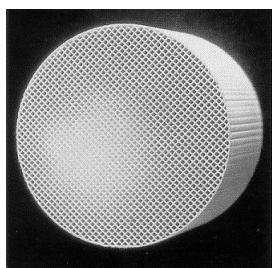
- ◆ **Continuous production**
- ◆ Production volume and speed
- ◆ Automatable

### Limitations:

- ◆ Stretched 2D shapes only
- ◆ Equipment and tooling **costs**

### Examples:

- ◆ Bricks
- ◆ Catalyst supports



Catalyst support for car

## Jiggering

### Use:

Relatively simple shapes from **aqueous** pastes

### Advantages:

- ◆ Production volume **flexibility** (single parts to large volumes)

### Limitations:

- ◆ Ideally shapes with circular symmetry
- ◆ Dimensional precision and tolerances

### Examples:

- ◆ Tableware
- ◆ Artisanal potteries



Tableware

## Injection

### Use:

Rather small precision parts with complex shapes from **organic** pastes

### Advantages:

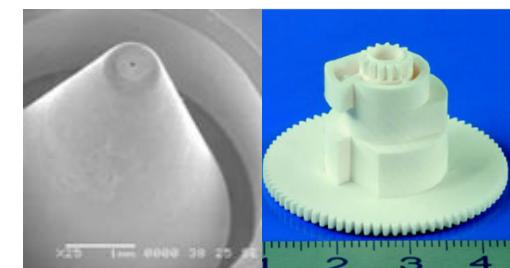
- ◆ Production volume and speed
- ◆ Automatable
- ◆ Precision and tolerances

### Limitations:

- ◆ Equipment and tooling **costs**
- ◆ Careful debinding

### Examples:

- ◆ Nozzles
- ◆ Gears



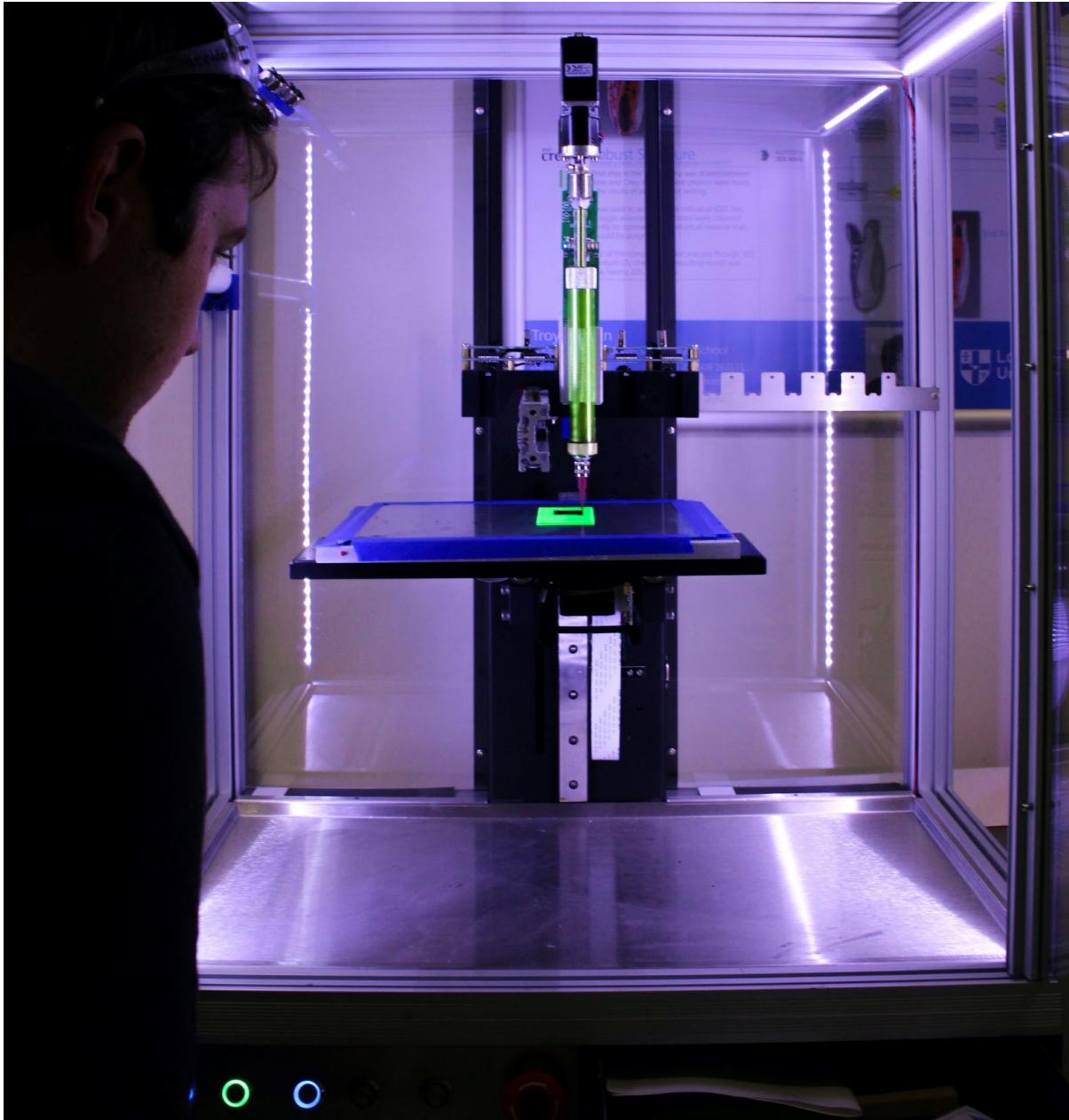
Nozzle and gear  
Source: SPT-Group



# EPFL

## Additive manufacturing

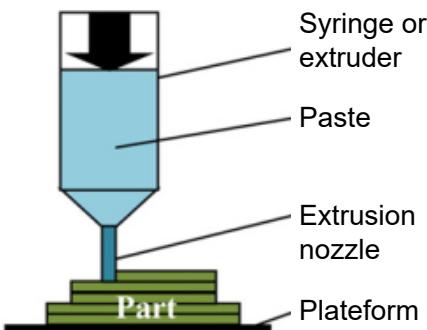
M. Stuer



# Selection of additive manufacturing techniques

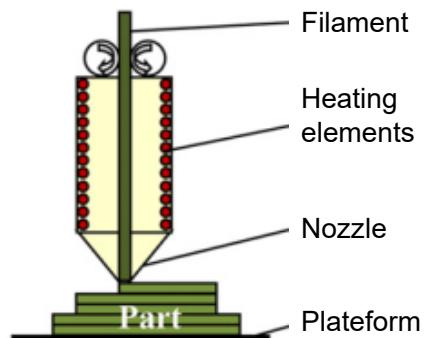
## Direct ink writing (DIW)

Paste extruded continuously through a cylindrical nozzle



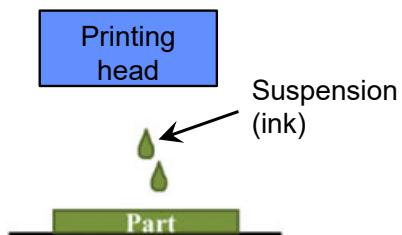
## Fused deposition modelling (FDM)

Deposition of a thermoplastic wire filled with ceramic powder



## Direct inkjet printing (DIP)

Inkjet printing of a particle loaded ink (i.e. suspension)



### Characteristics:

- ♦ Resolution and speed depend on the nozzle and rheological properties of the paste
- ♦ Delicate drying required
- ♦ Fast debinding due to low organic additives content

Watch: <https://youtu.be/N1LF14QhNyY>

### Characteristics:

- ♦ Resolution depends on nozzle and rheological properties of thermoplastic composite filament
- ♦ Delicate debinding required
- ♦ Fast building speed compared to other 3D techniques

Watch: <https://youtu.be/GxLjDNrQBgs>

### Characteristics:

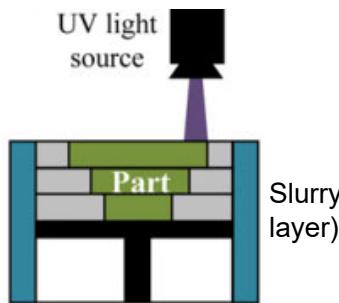
- ♦ Colloidal suspensions only
- ♦ Layer thicknesses and therefore manufacturing speeds are very low
- ♦ Mainly used in microelectronics and micromechanics

Watch: <https://youtu.be/0ba7KkUV-GA>

# Selection of additive manufacturing techniques

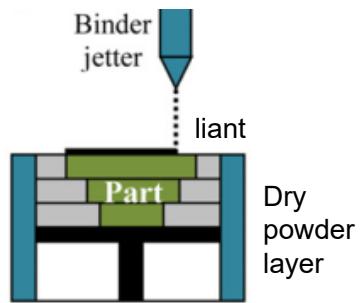
## Stereolithography (SLA/DLP)

Photo-polymerization of a photosensitive powder slurry



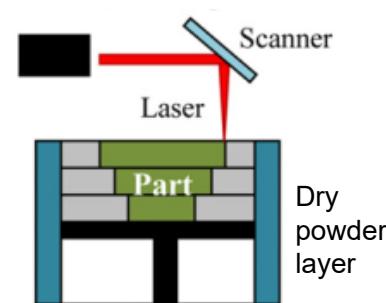
## Powder-based 3D Printing (3DP)

Deposition/printing of an organic binder on a powder bed



## Selective laser sintering (SLS)

Densification of a powder bed by laser sintering or laser fusion



## Characteristics:

- ◆ Homogeneous layers require highly stable dispersions
- ◆ Dispersions with compatible viscosity and high solid loadings preferred
- ◆ Difficult with powders that absorb light

Watch: <https://youtu.be/NM55ct5KwiI>

## Characteristics:

- ◆ Compatible with powders or granules that do not easily form aerosols
- ◆ Avoids typically need of support structures
- ◆ Resolution depends on particle size and binder diffusion

Watch: [https://youtu.be/4Bftt\\_4DQKE](https://youtu.be/4Bftt_4DQKE)

## Characteristics:

- ◆ Difficult due to thermal shock sensitivity of ceramics
- ◆ Controlled laser-material coupling required
- ◆ Easier with glass / glass-ceramics with low glass transition temperature

source: <https://doi.org/10.1111/jace.14705>

# Stereolithography based techniques

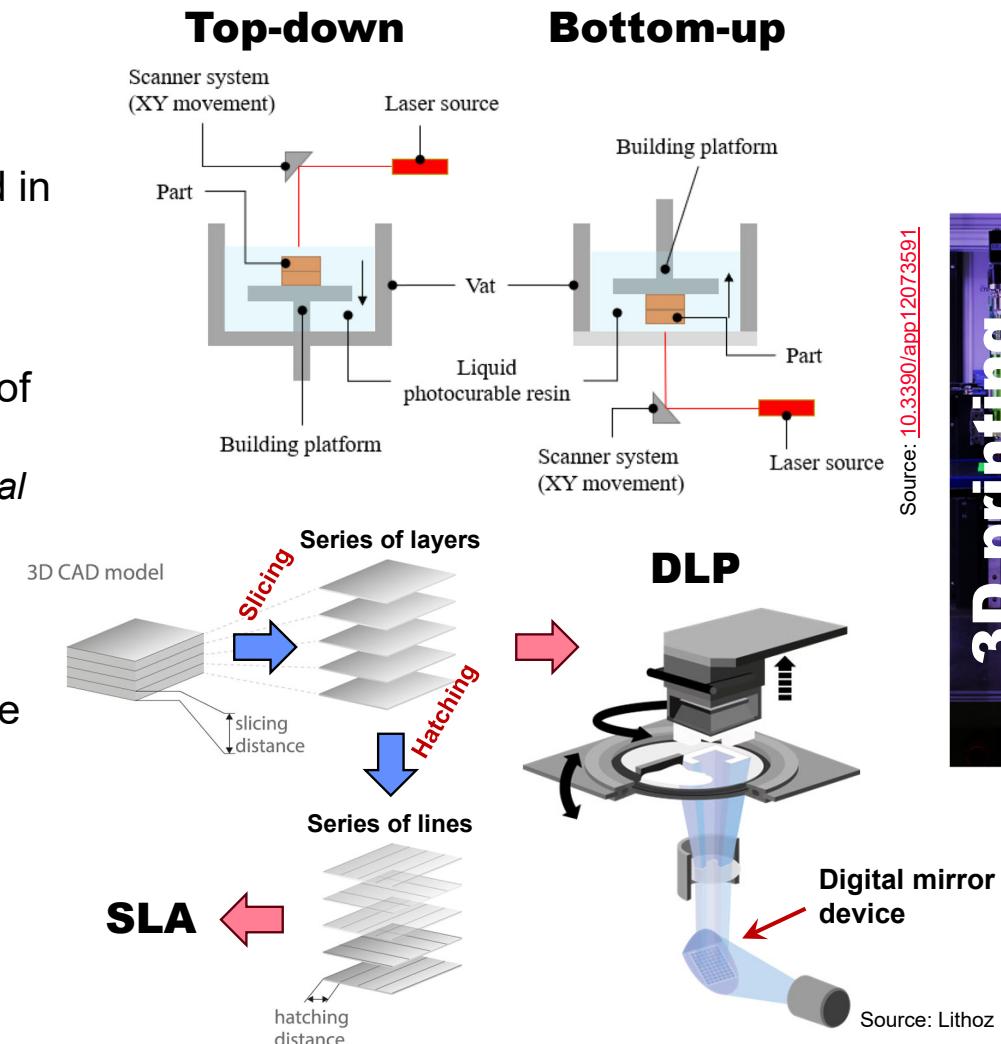
Techniques can be distinguished based on:

## ◆ Built-direction

- **Top-down:** The part is descending and immersed in the photocurable slurry
  - Pros:** No peeling forces
  - Cons:** Slurry volume
- **Bottom-up:** The part is ascending and pulled-out of the slurry
  - Pros:** Uses less slurry, multi-material
  - Cons:** Peeling forces/deformation

## ◆ Light exposure strategy

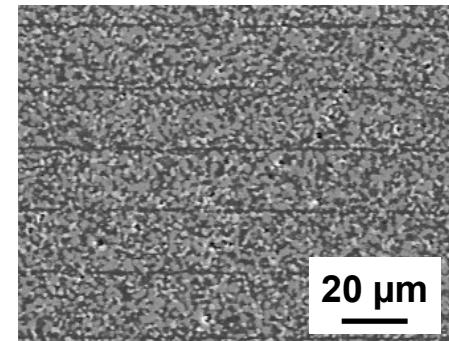
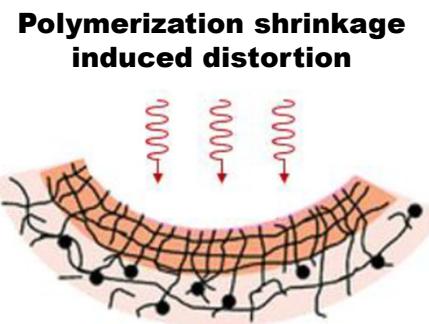
- **Vector scanning:** Vector-by-vector scanning of the resin surface by a single-point laser with a galvanoscanner.  
→ slicing + hatching
- **Projection:**  
→ slicing  
Use of a digital mirror device to project the layer onto the resin



# General aspects related to slurry formulation for SLA/DLP

## Slurry formulation for stereolithographic printing is delicate

- ◆ Formulations depend on the ceramic material powder properties as well as the AM system requirements
- ◆ They typically contain:
  - Monomers
    - Low viscosity
    - High polymerization shrinkage (c.f. internal stresses → deformation/delamination risk)
  - Oligomers
    - High viscosity
    - Low polymerization shrinkage
    - Strengthening of green body
  - Dispersants
    - Powder dispersion stabilization
  - Photoinitiators(/Sensitizers)
    - Initiate local photopolymerization
  - Photoinhibitors/dyes
    - Confine photopolymerization reaction
    - Increase printing resolution
  - Non-reactive diluents
    - Reduce slurry viscosity
    - Facilitate debinding
- ◆ Weakly **flocculated suspensions** are preferred to **prevent powder setting** between layers (see right image).



## SLA/DLP related metrics

- ◆ The cure depth ( $C_d$ ) is given by:

$$C_d = D_p \ln \left( \frac{E}{E_c} \right)$$

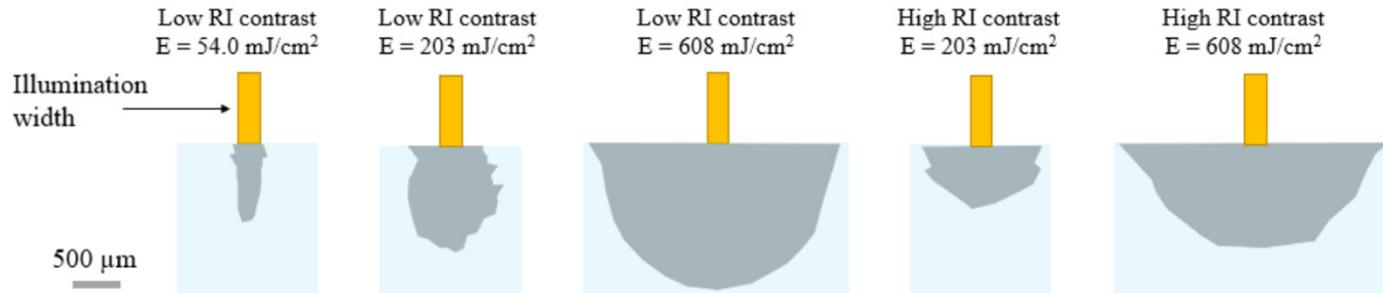
- $E$ : applied energy dose
- $E_c$ : critical energy dose for gelation
- $D_p$ : attenuation length

- ◆ For a given particle size and distribution

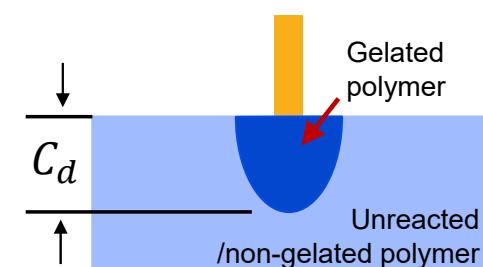
$$C_d \propto \frac{1}{(\Delta n)^2}$$

- $\Delta n$ : refractive index (RI) contrast between powder and resin

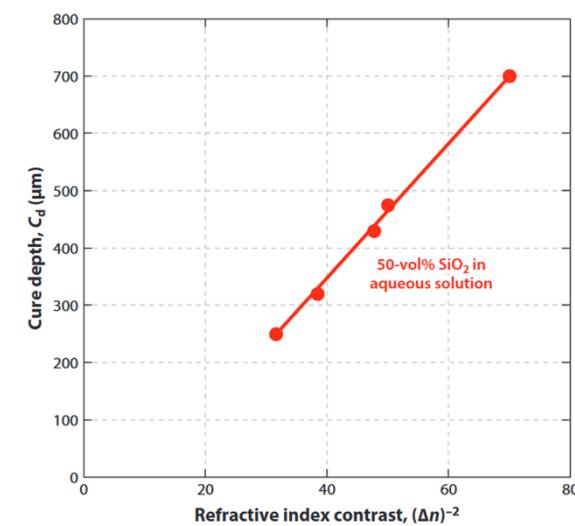
- ◆ Refractive index contrast and applied energy dose affect gelation volume



**Cure depth illustration**



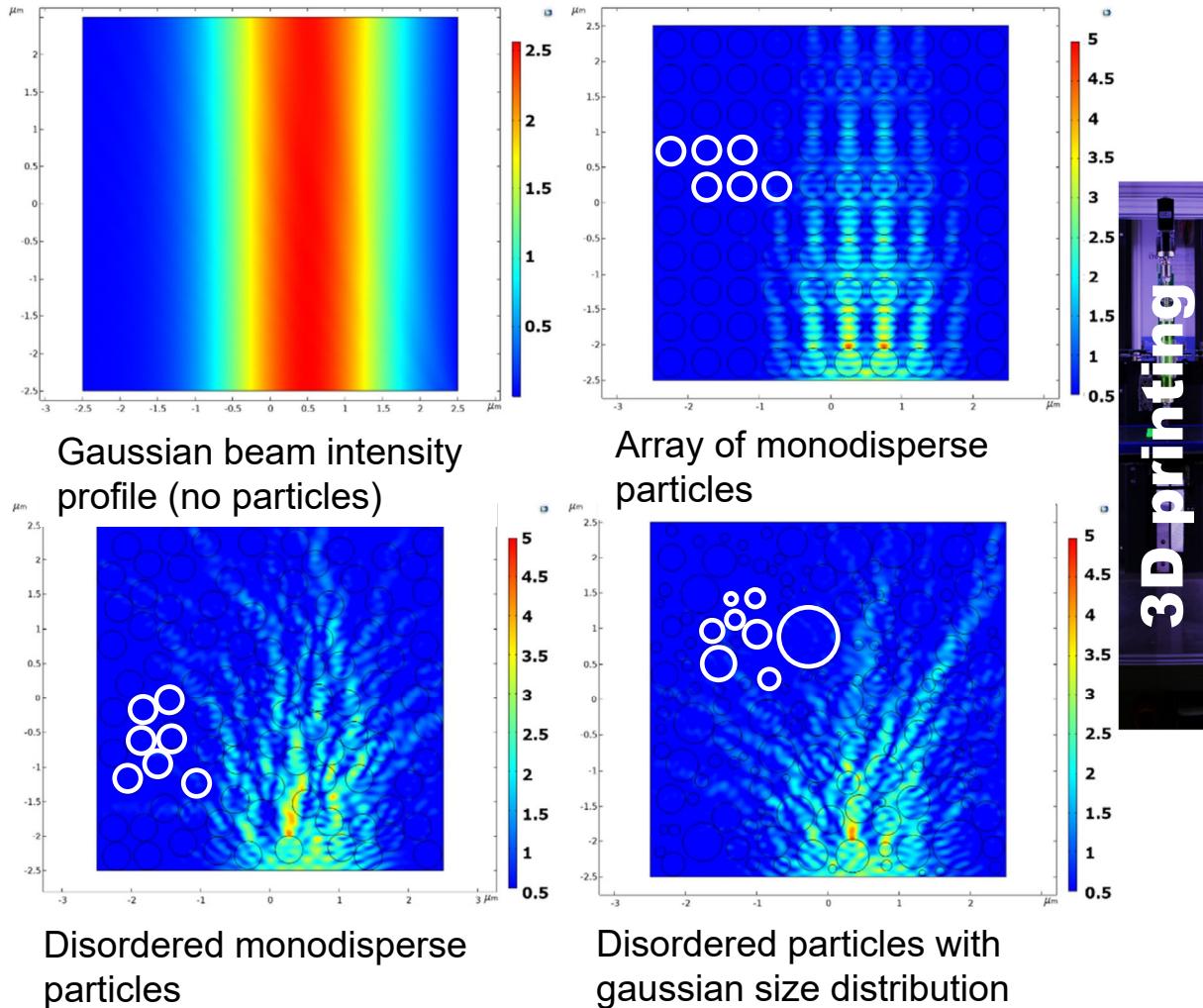
**Cure depth vs RI contrast**



## Light scattering due to particle size/distribution

- ◆ In concentrated powder suspensions, Rayleigh-Gans or Mie scattering approaches are not applicable
- ◆ Light scattering depends on particle arrangement, size and distribution

→ **Fine tuning** of the energy dose, photoinitiator and photoinhibitor concentration needed **for each powder**

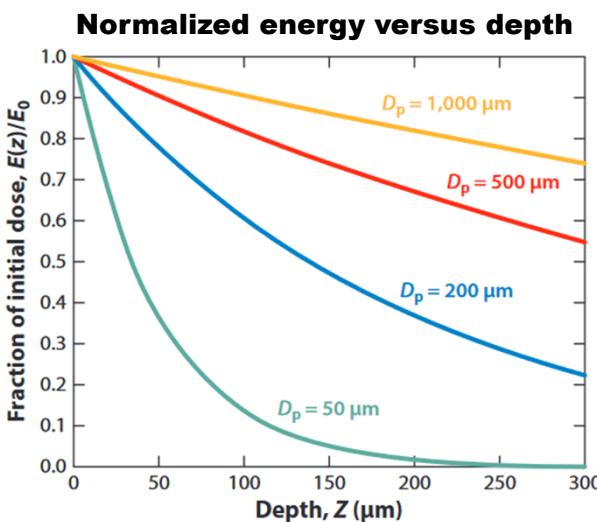


# Subsurface energy dose at z-depth

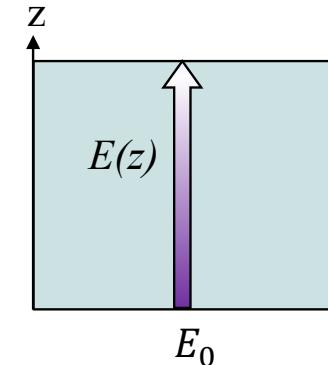
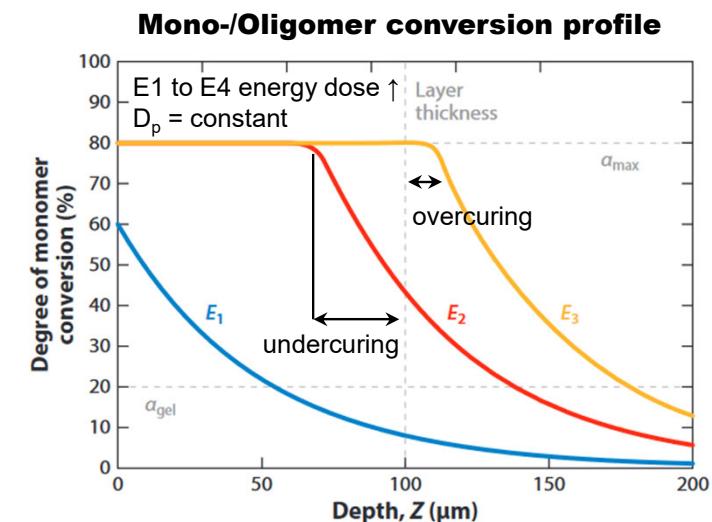
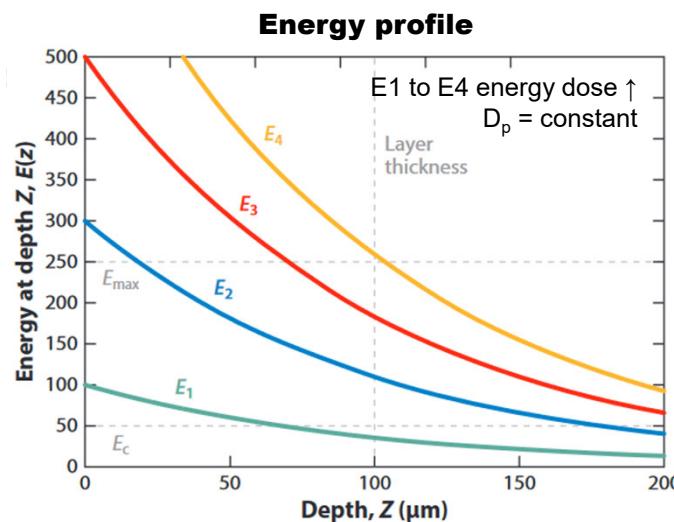
- ◆ The energy dose  $E(z)$  at position  $z$  from the surface can be approximated following a Lambert-Beer law:

$$E(z) = E_0 \exp\left(\frac{-z}{D_p}\right)$$

- ◆ To ensure optimal layer curing the energy dose and or layer thickness have to be adjusted depending on  $D_p$



$E_{\max}$ : Energy dose to reach max conversion rate  $\alpha_{\max}$   
 $E_c$ : Energy dose for gelation



# **Pre-sintering parts conditioning:**

## **Drying**

*M. Stuer*



## Green body drying

- ◆ **Drying** is a delicate and **slow** process after wet part shaping, since associated **shrinkage** may lead to part **cracking**!
- ◆ **Level of shrinkage** depends on:
  - **Shaping method**
  - **Solid loading** of the suspensions (=volume of powder in a suspension)
  - **Examples:**
    - **Slip casting:**  
→ linear shrinkage **between 1.5 and 7%**
    - **Tape casting or sol-gel:**  
→ linear shrinkage **between 50 and 90%**
- ◆ Note that invisible defects in a wet tablet may lead to part failure during drying.



### Example slip casting

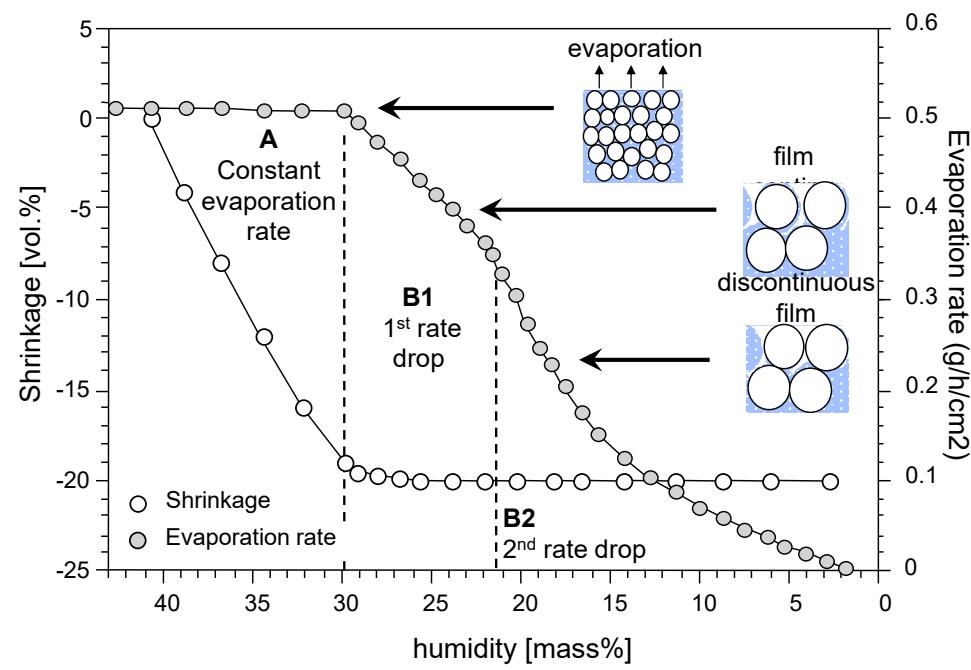
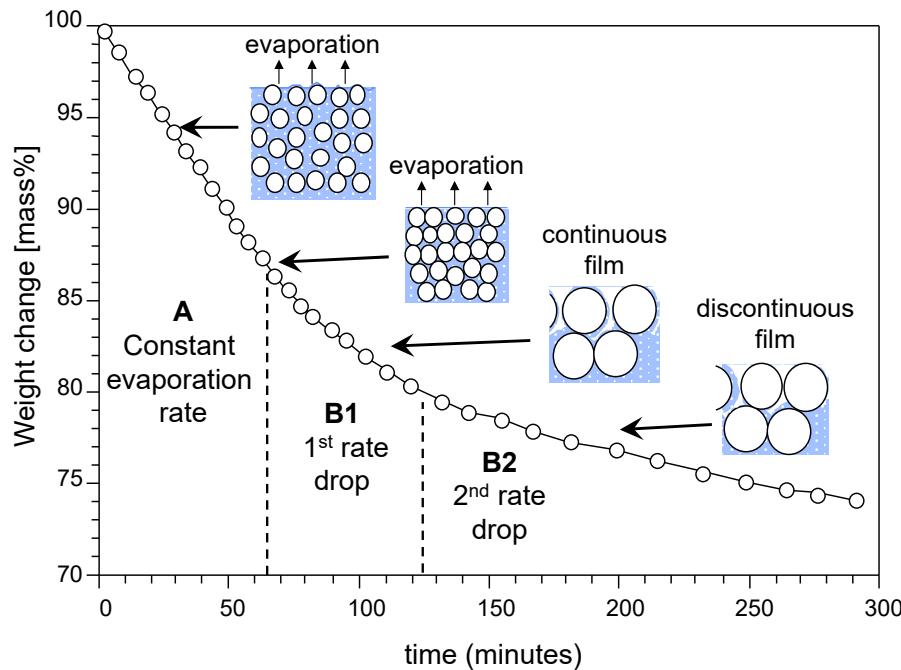
- ◆ **Shaping:**
  - without pressure → **80 mins**
  - with pressure → **2 mins!!**
- ◆ **Drying:**
  - Maximum wall thickness 1 cm  
→ **6 days!!!!**



# Drying stages and graphical representations

## A: Constant evaporation rate phase

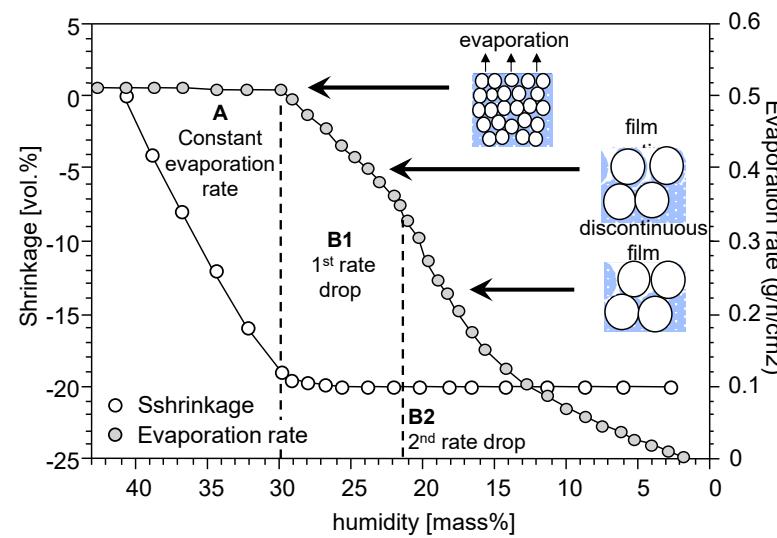
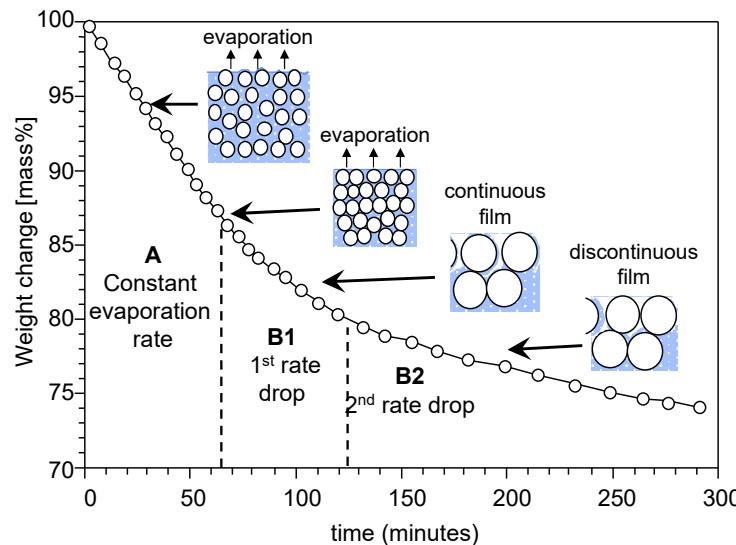
- ◆ Pores saturated with liquid and **evaporation** takes place directly at the **surface**
- ◆ **Surface tension** of liquid **compresses** part
  - particles rearrange and move closer
  - **drying shrinkage** nearly complete at end of this stage



# Drying stages and graphical representations

## B: Decreasing evaporation rate phase

- ◆ **Transition** from stage A to B when **liquid-gas interface moves into the pore network**.  
The dry area is no longer under compression → **cracking may occur !**
- ◆ Distinction made between decreasing evap. rate phase with:
  - **Continuous film**  
→ liquid is transported by **capillarity** to the surface of the part where it evaporates
  - **Discontinuous film**  
→ liquid evaporates within the part and vapor diffuses through pore network → reduced drying rate



## Mechanisms leading to part failure during drying

- ◆ **Cracking** occurs due to **capillary pressures** in the pore network within the green body
- ◆ Local capillary pressures  $P_c$  vary following:

$$P_c = \frac{-2\gamma_{lv} \cos \theta}{r}$$

- $\gamma_{lv}$ : surface tension (liq/vap)
- $\theta$  : wetting angle (l/v)
- $r$  : (cylindric) pore radius

If  $r$  **changes** locally, the **pressure changes** locally as well

It follows that:

- **Inhomogeneous particle packing** leads to a inhomogeneous porosity
  - Resulting **pressure gradients** lead to differential shrinkage and thus **internal stresses**
  - Associated **inhomogeneous permeability** (e.g. fluid dynamics) further amplify the internal stresses (see next slide)



Drying

## Mécanisme provoquant la fissuration pendant le séchage

- ◆ **Internal stresses** also depend on the **liquid transport dynamics** in the pore network, and thus on the **evaporation rate**.
- ◆ Empirically, the drying stresses  $\sigma$  follow a type law:

$$\sigma \propto \frac{L \eta_l V_e \gamma_{lv}}{3 K_p}$$

- $L$ : characteristic part dimension
- $\eta_l$ : viscosity of liquid
- $V_e$ : evaporation rate
- $K_p$ : permeability of pore network
- $\gamma_{lv}$ : surface tension(liq/vap)

It is necessary to ensure **even drying** and reduce de **surface tension** and/or dry **slowly**



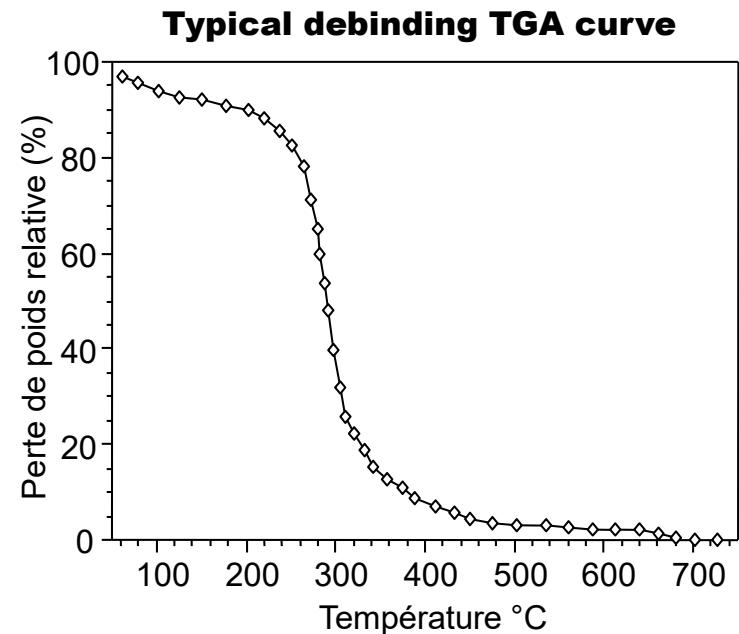
## Pre-sintering parts conditioning: Debinding



## General aspects related to debinding

**Organic additives** have to be **removed before sintering** to prevent part cracking and/or carbonaceous residues during sintering

- ◆ Most widespread method is thermal **decomposition and pyrolysis in air** with dwells at different temperature ranging from 300-600°C.  
(slow heating rates (0.25-1°/min), temperature profiles established from thermogravimetric analysis (TGA))
- ◆ **Mechanisms:**
  - Evaporation
  - Oxydation
  - Decomposition
- ◆ **Cross-linking of organics** (e.g. oxygen deficiency, heating rate too fast, ...) must be avoided as it can result in **carbon residues** (e.g. graphite-like) that are difficult to remove.
- ◆ Upon **incomplete organics removal**, rapid gas formation during sintering may lead to part **swelling and even cracking**.



# Estimation of thermal debinding temperatures

**Average decomposition temperature ( $T_d$ )** of polymers based on their **chemical structure**

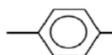
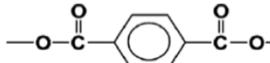
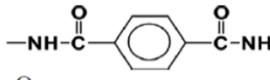
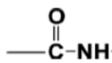
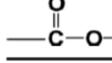
## ◆ Assumptions:

- Numerical value ( $Y_{d,i}$ ) assigned to each chemical group in binder molecule
- For repeating units (e.g. polymers) account only once for repeating chemical structure

## ◆ Decomposition temp. estimation:

$$T_d = (\sum_i N_i Y_{d,i}) / \sum_i M_i$$

- $Y_{d,i}$ : Contribution of  $i^{\text{th}}$  group to the decomposition temperature of the binder
- $N_i$ : Amount of  $i^{\text{th}}$  group in the binder
- $M_i$ : Molecular weight of  $i^{\text{th}}$  group in the binder

Group	Source (e.g.)	$Y_d$ (K · kg/mol)
-CH <sub>2</sub> -	Ethylene	9.5
-CH(CH <sub>3</sub> )-	Propylene	18.5
-CH(C <sub>6</sub> H <sub>5</sub> )-	Styrene	60
-CH(COOCH <sub>3</sub> )-	methyl acrylate	56.5
-CH(OCOCH <sub>3</sub> )	vinyl acetate	42.5
-C(CH <sub>3</sub> )(COOCH <sub>3</sub> )	methyl methacrylate	37.5
-CHF-	vinyl fluoride	18
-CHCl-	vinyl chloride	23.5
-CH(CN)-	Acrylonitrile	28
-CH(OH)-	vinyl alcohol	14
-CF <sub>2</sub> -	tetrafluoroethylene	38.5
-CH=CH-	Neoprene	18
	Phenyl	64
	aromatic polyester	119
	aromatic polyamide	135
-O-	Oxide	8
-S-	Sulfide	33
-NH-	Amine	16
>C=O	Ketone	20
	Amide	22.5
	Ester	33.5

## CASE STUDY: Shaping process development



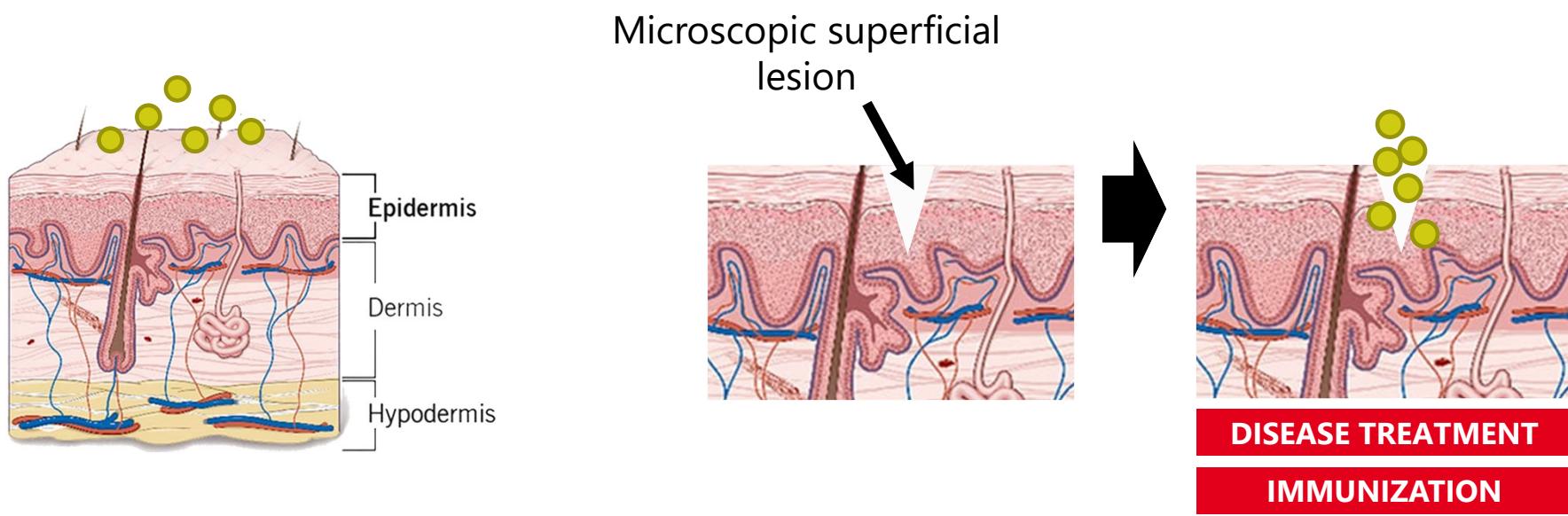
## Background and motivation

- **Background and motivation**
- Functional analysis and fabrication strategy
- Mold fabrication
- Slurry formulation and casting
- Demolding, debinding and sintering



## Background and motivation: Skin barrier

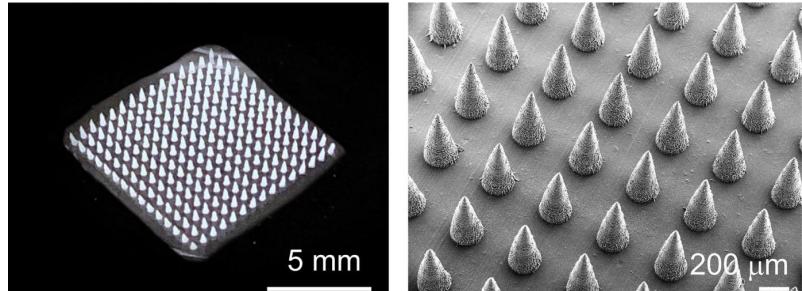
- ◆ **Stratum corneum barrier** layer prevents permeation of large drugs



- ◆ **Microscopic lesions required** for local skin delivery of drugs and vaccines

## Background and motivation: Microlesion creation

- ◆ **Microneedle patches** exist but limited to treatment of
  - **Small** areas
  - Areas with **good accessibility**



Credit: NC State University

→ Ceramic **STAR particles** developed to overcome those issues

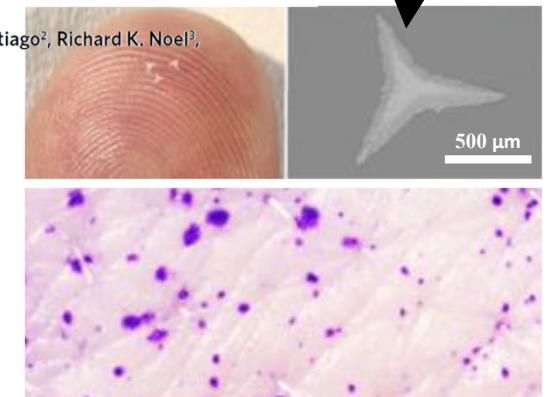
- ◆ For large scale application **reliable** and **economical** fabrication approach needed



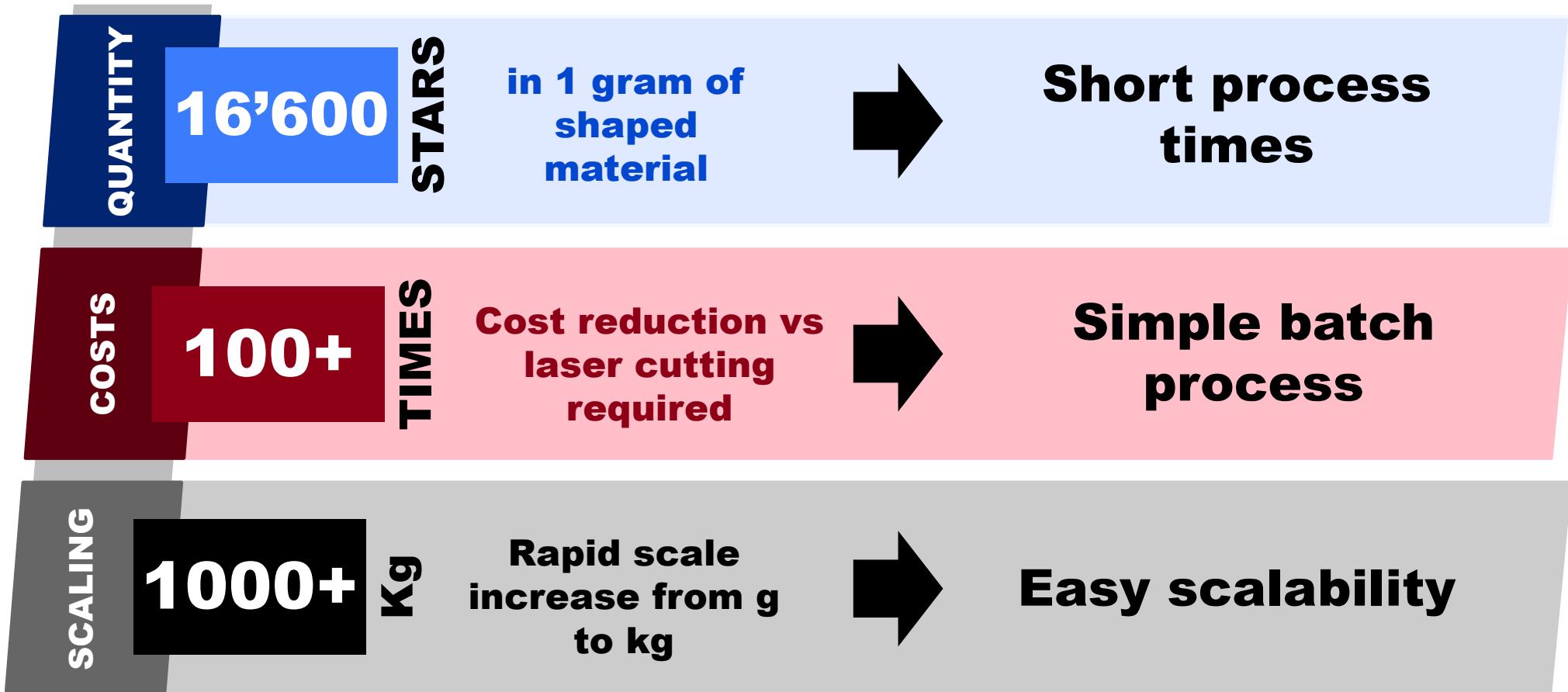
**STAR particles for enhanced topical drug and vaccine delivery**

Andrew R. Tadros<sup>1</sup>, Andrey Romanyuk<sup>1</sup>, Ian C. Miller<sup>2</sup>, Andrea Santiago<sup>2</sup>, Richard K. Noel<sup>3</sup>, Laura O'Farrell<sup>3</sup>, Gabriel A. Kwong<sup>2</sup> and Mark R. Prausnitz<sup>1,2</sup> 

**Laser machined!**  
\$\$\$



## Background and motivation: Some key figures



- Background and motivation
- **Functional analysis and fabrication strategy**
- Mold fabrication
- Slurry formulation and casting
- Demolding, debinding and sintering

## Functional analysis and fabrication strategy

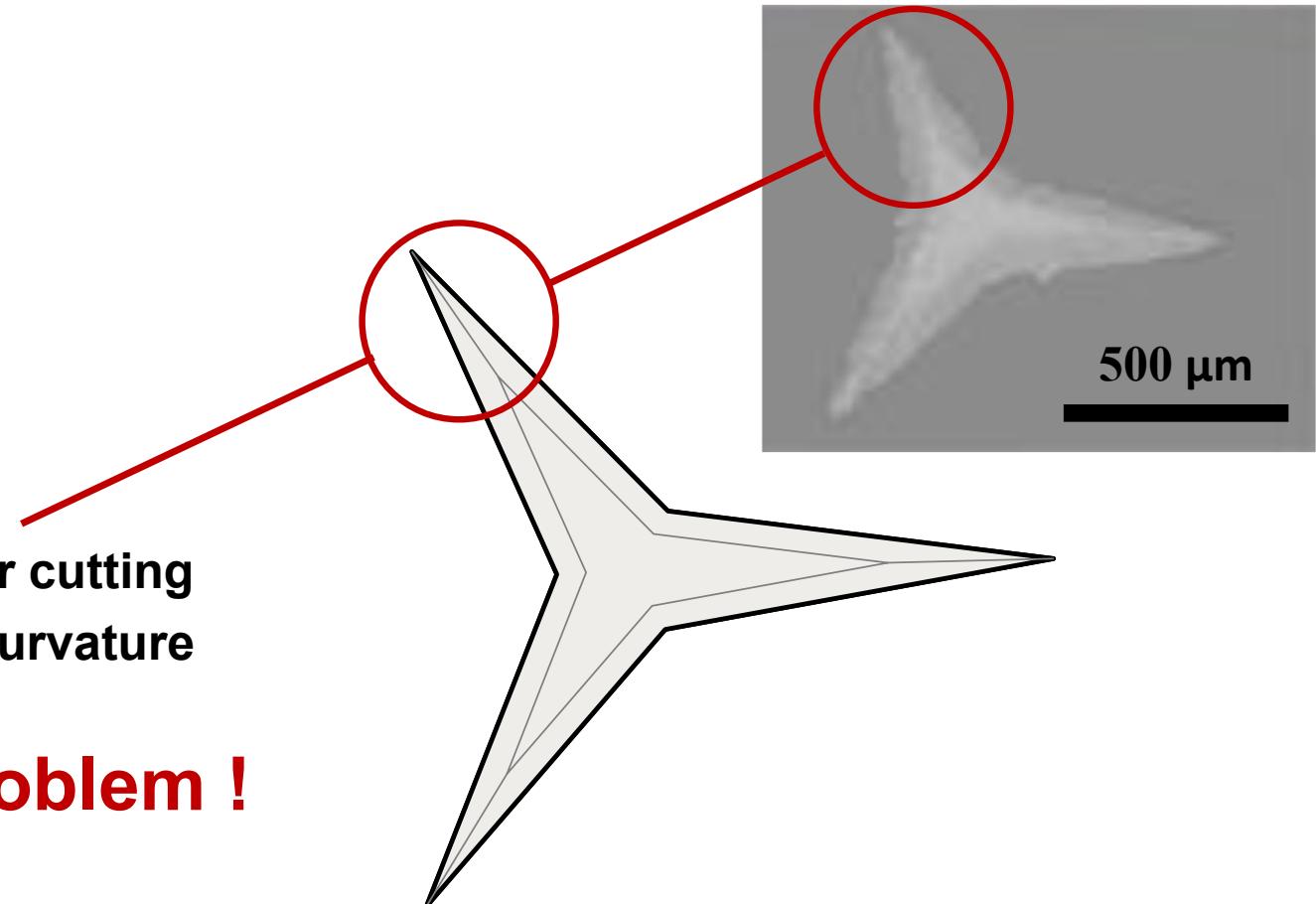


## Functional analysis

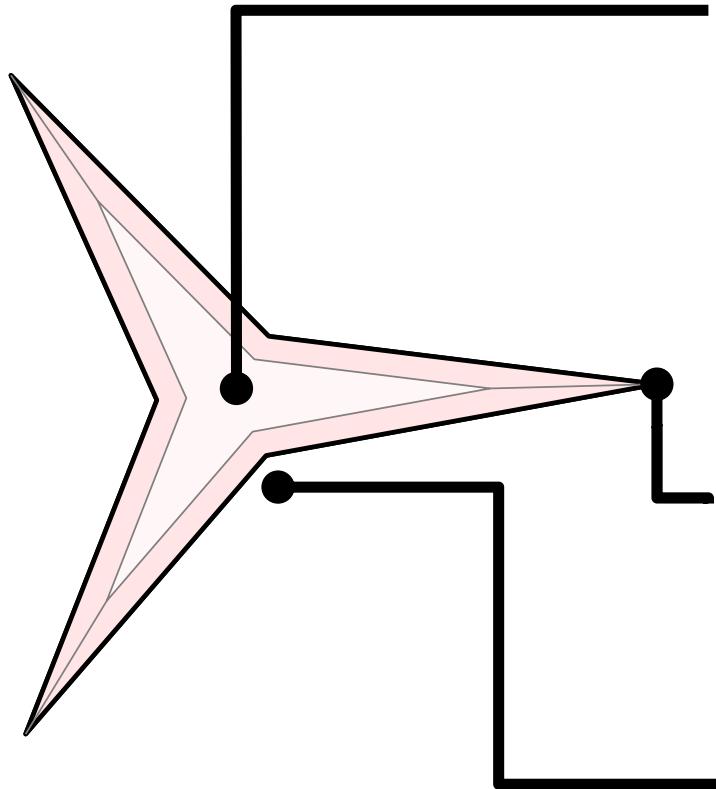
- ◆ Critical part features

- Material
- Arm length
- Tip radius
- Part thickness
- **Tapering of the tip**
  - Effect from CO<sub>2</sub> laser cutting
  - Mean surface curvature

→ 3D shaping problem !



# Functional analysis and fabrication strategy



- ◆ **Material**

- Alumina-based
- Biocompatible / non-allergenic
- No failure risk during use  
(e.g. no debris in skin)

→ **Aim for zirconia toughened alumina**

- ◆ **Tip diameter**

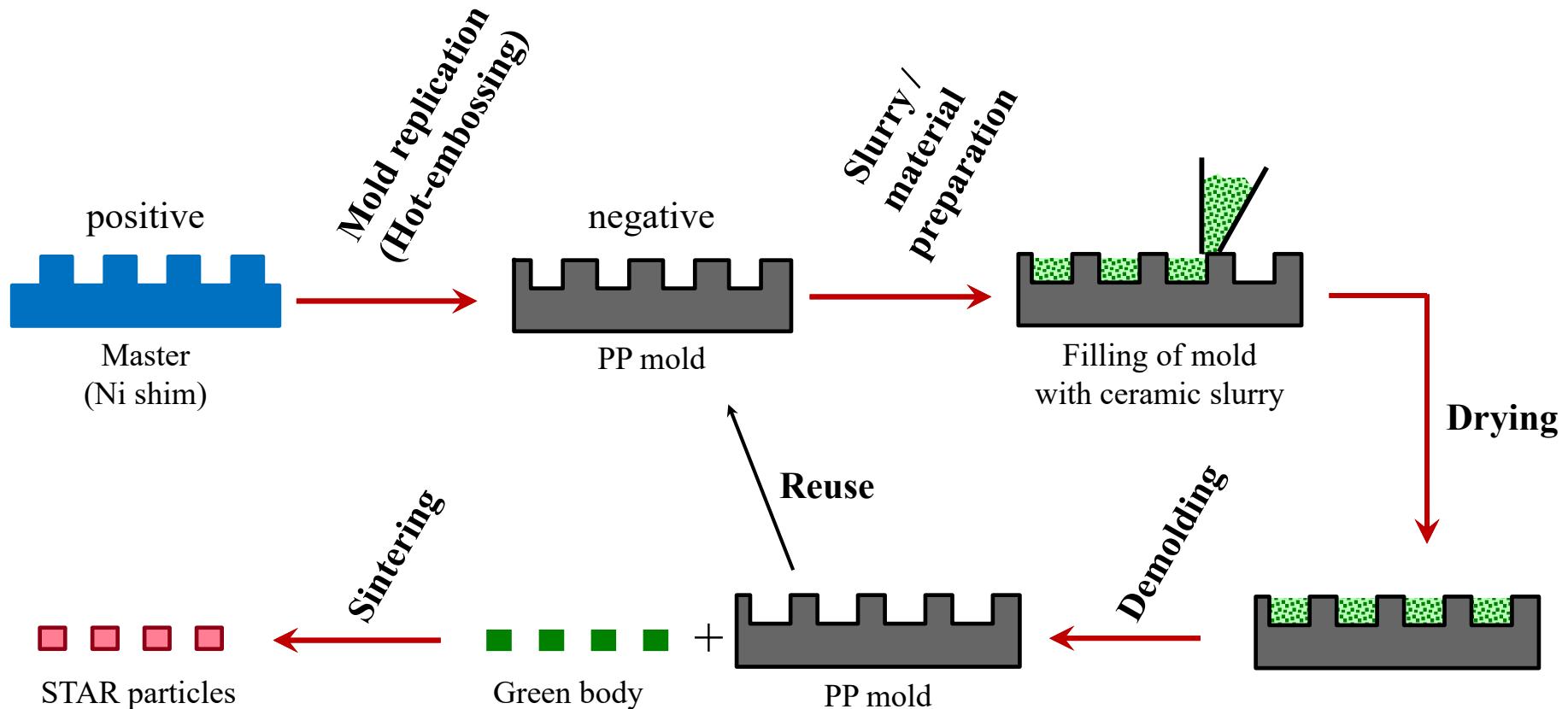
- The sharper (XY and XZ) the better

→ **Aim for diameter <10 µm**

- ◆ **No stress concentration zones**

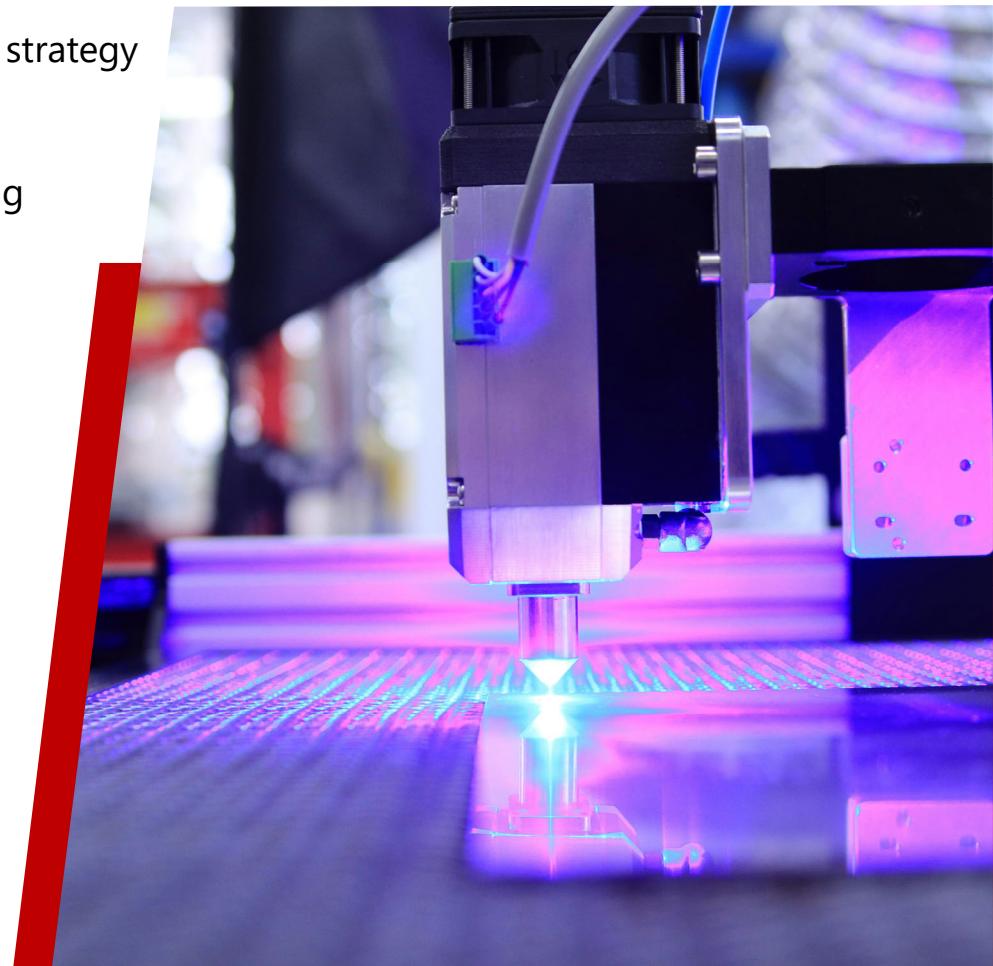
→ **No sharp corners**

## Selected fabrication strategy (simplified)



- Background and motivation
- Functional analysis and fabrication strategy
- **Mold fabrication**
- Slurry formulation and casting
- Demolding, debinding and sintering

## Mold manufacturing

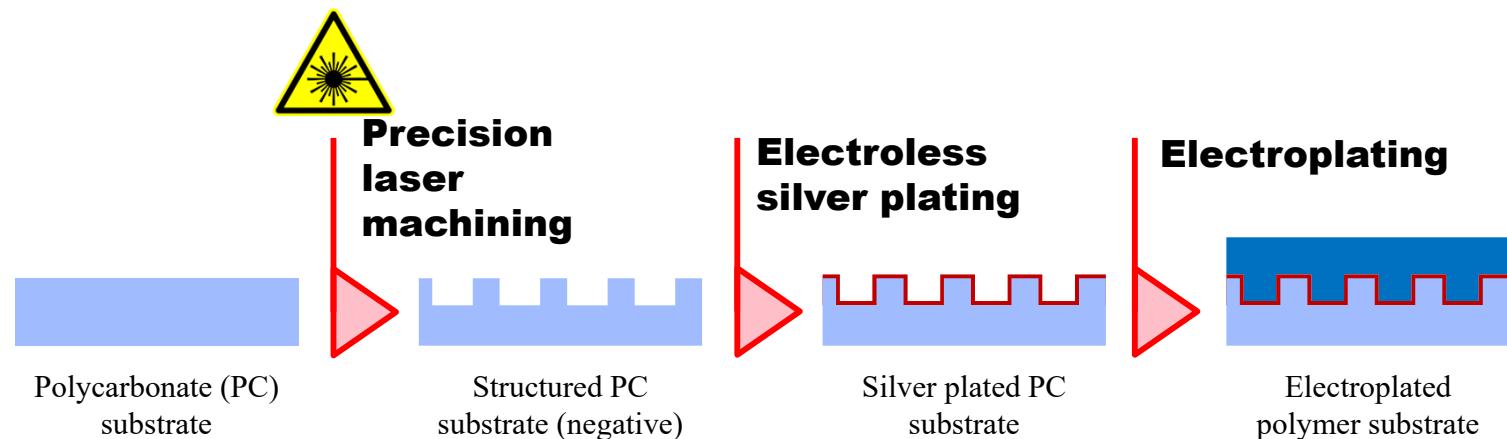
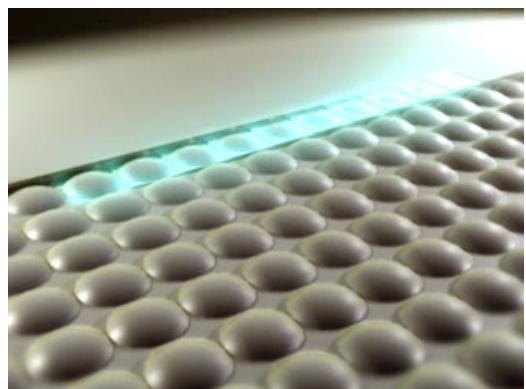


## A two-step approach

2-step mold fabrication process (mass replication compatible):

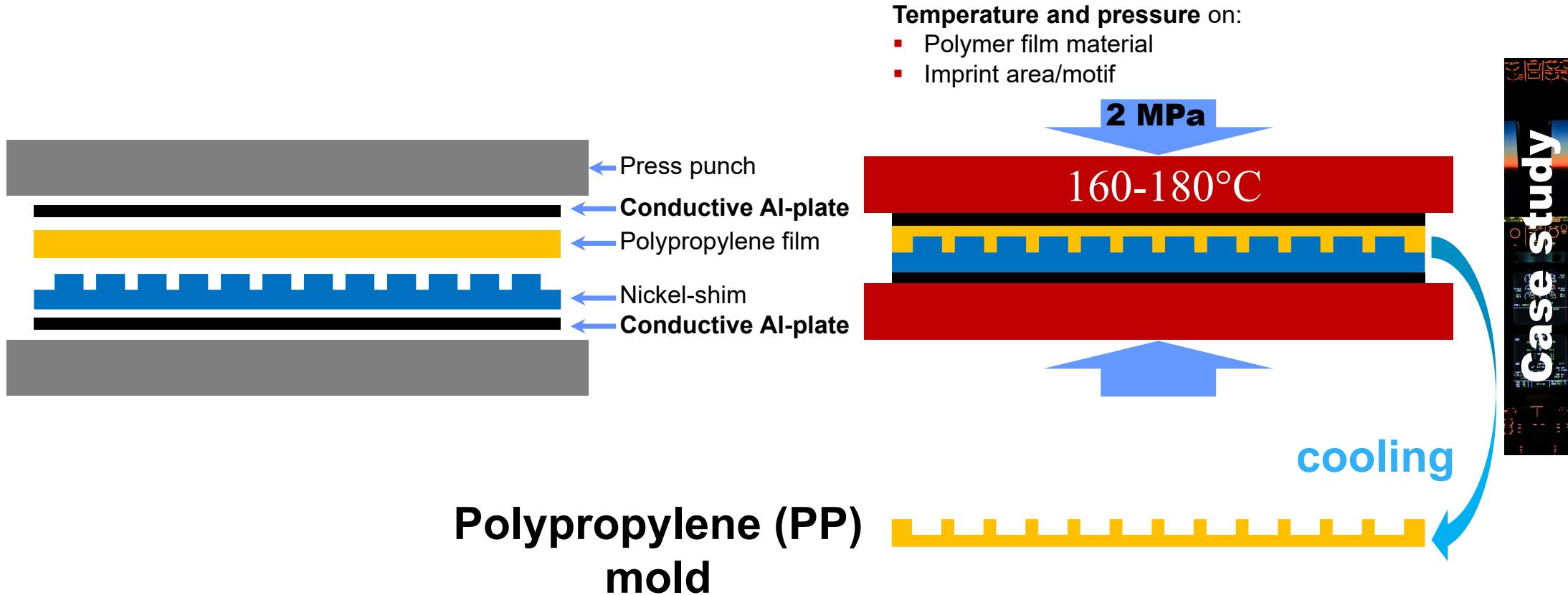
- 1. Nickel shim:** Laser machining + electroplating (**cost intensive**)
- 2. PP mold:** Hot embossing with Nickel shim (**low cost**)

### 1. Nickel shim fabrication



## Nickel shim replication process

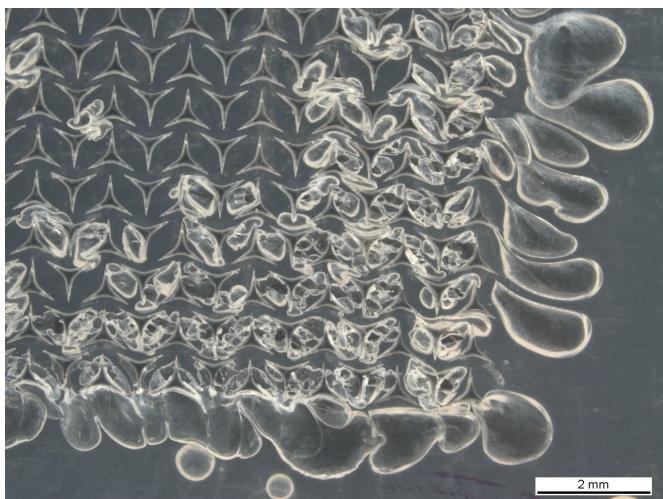
### 2. Hot embossing process



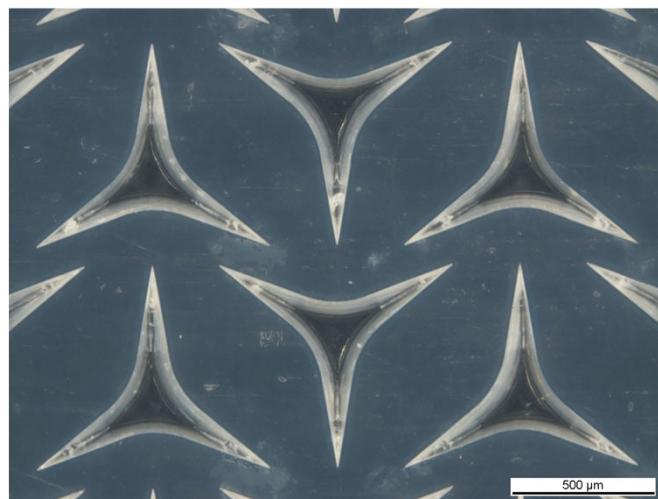
# Effect of temperature and pressure

## Mold artefacts

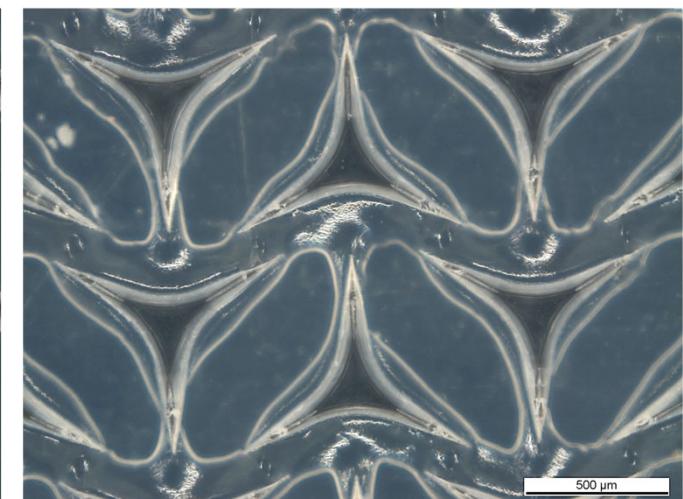
### Bubbles and flaws



Unstable temperature /  
Temperature too high



### Rounded edges



Pressure too low

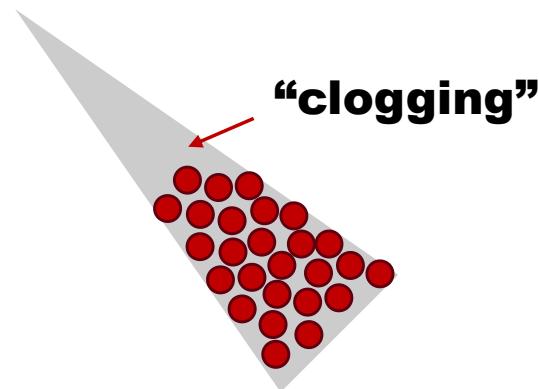
- Background and motivation
- Functional analysis and fabrication strategy
- Mold fabrication
- **Slurry formulation and casting**
- Demolding, debinding and sintering

## Slurry formulation and casting



# Target

## 1 Reliable tip filling



## 2 Low shrinkage and high GB strength

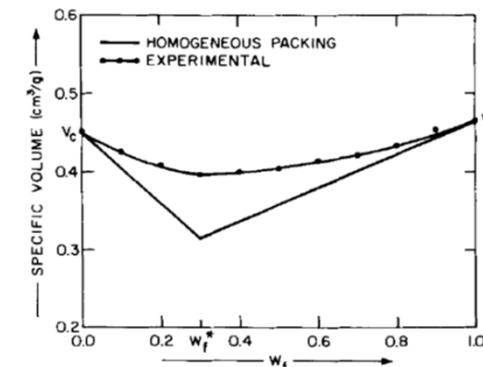
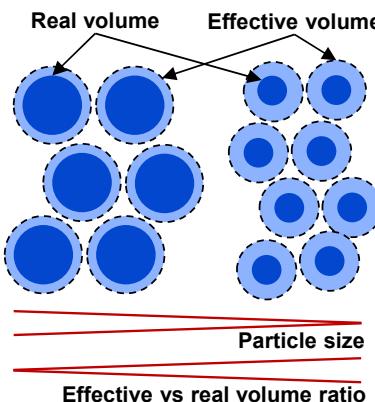


Fig. 1. Theoretical and experimental packing results for binary mixtures compacted at 207 MPa.

- ◆ Clogging prevention
  - Particle size ( $D_{v50}$ ) ~6-8x lower than target tip radius
  - $D_{v50} < 0.38 \mu\text{m}$  (for 3  $\mu\text{m}$ )

- ◆ High solid loading
  - Bimodal particle size distribution to preserve flowability

Smith, J. P. and G. L. Messing (1984).  
doi: [10.1111/j.1151-2916.1984.tb18938.x](https://doi.org/10.1111/j.1151-2916.1984.tb18938.x)



# Casting requirements

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**Reliable casting requires:**

- ◆ Full mold **filling**
  - a. **Contact angle**
- ◆ **Defect-free** drying
  - b. **Solid loading & drying stress**
- ◆ Good green body **strength**
  - c. **Interparticle forces and/or binders**

# Materials & dispersant concentration

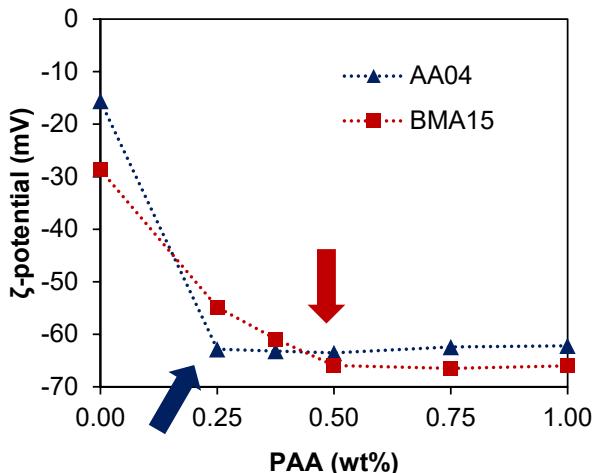
## Raw materials

- Inorganic binder
- Microstructure refiner (Zener pinning)
- Material toughener

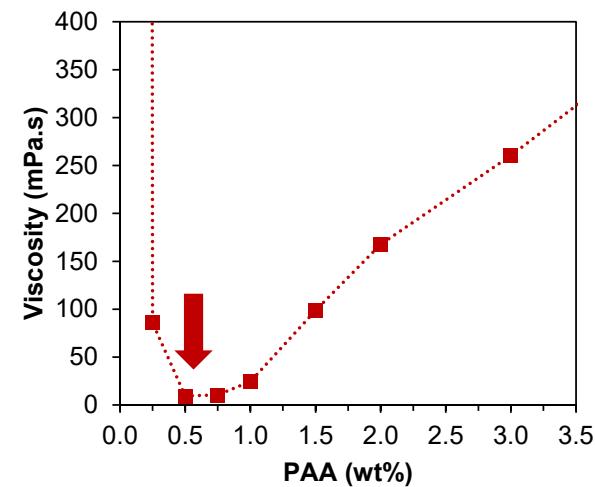
	AA04	BMA15	ZrO <sub>2</sub> sol
D <sub>v50</sub> [nm]	450	120	10
Span	1.77	1.33	-
S <sub>BET</sub> [m <sup>2</sup> /g]	4.4	14.6	180

- Dispersant: PAA M<sub>w</sub>=2000

Titration with PAA (R1.5) @ pH 10

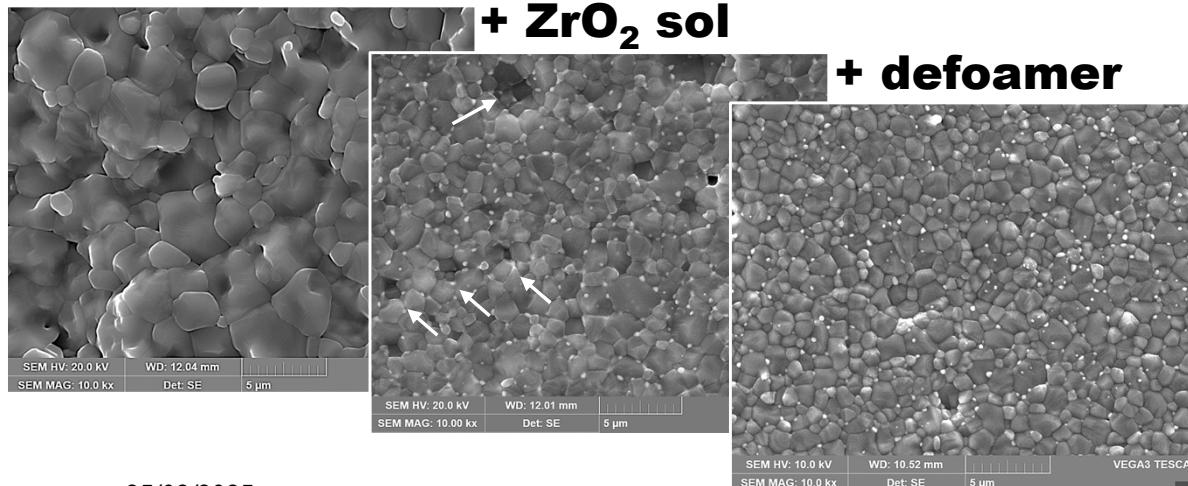


Viscosity @ 100 s<sup>-1</sup> for 30 vol.% BMA15



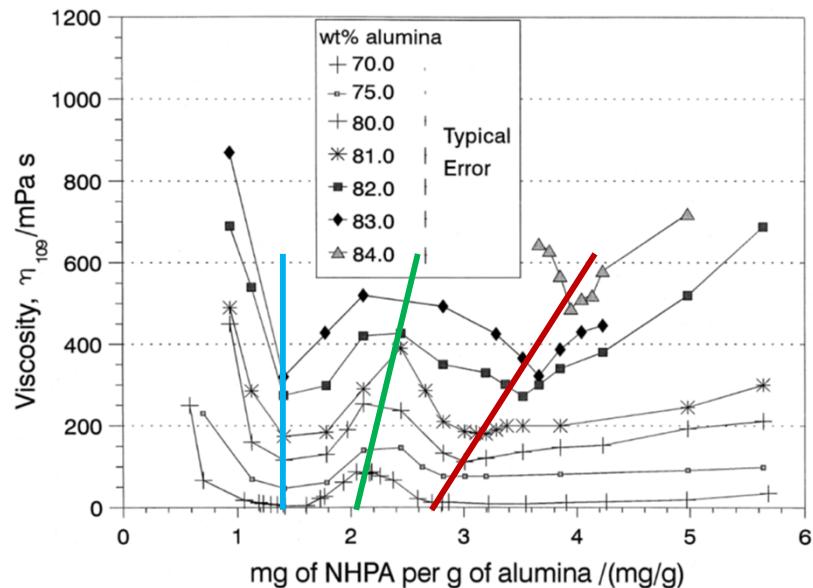
## Slurry preparation

- ◆ Ball milling
- ◆ Homogenization (>30 hrs) on rolling bench (see next slide)
- ◆ Defoamer addition
- ◆ Degazing

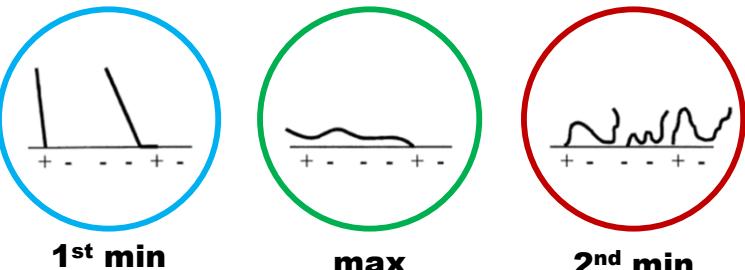


# Concentration/time effect on PAA configuration

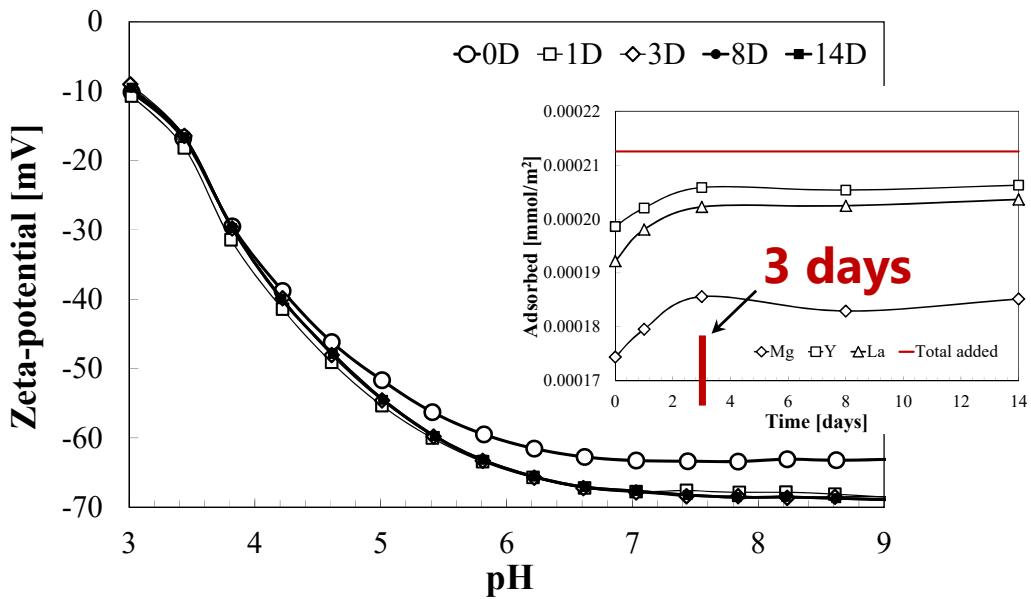
## Concentration dependent configuration



Davies, J. and J. G. P. Binner (2000).  
doi: [10.1016/S0955-2219\(00\)00012-1](https://doi.org/10.1016/S0955-2219(00)00012-1)

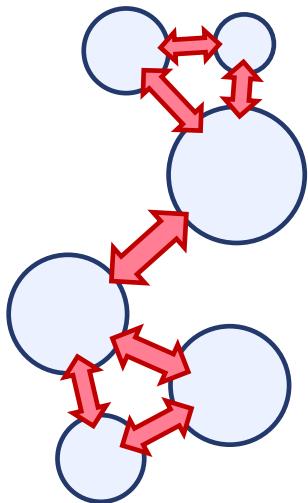


## Time dependent configuration



Speciation reactions may take **up to 3 days to reach steady state**

# Interparticle potential and solid loading effect on yield stress

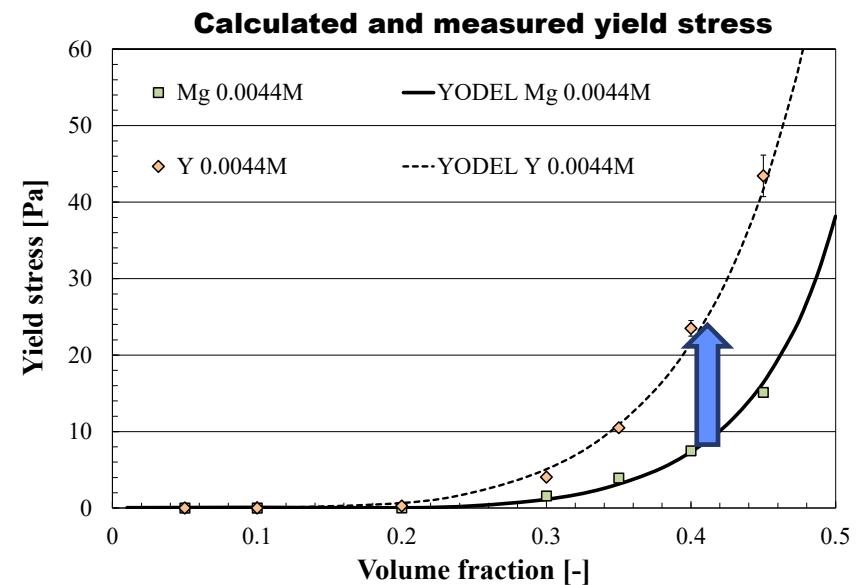
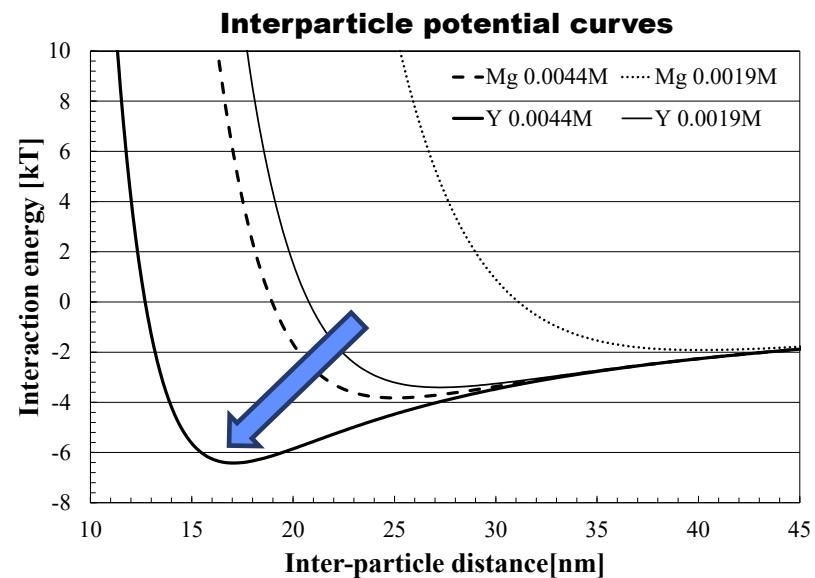


Yield stress related to **attractive interaction** between particles

- ◆ YODEL model:

$$\tau = m_1 \frac{\Phi(\Phi - \Phi_0)^2}{\Phi_{max}(\Phi_{max} - \Phi)}$$

- $\Phi$ : vol. solid loading
- $\Phi_{max}$ : vol. solid loading at maximum packing
- $\Phi_0$ : vol. solid loading at percolation threshold
- $m_1$ : function of the interparticle potential and particle morphology



# Solid loading effect on viscosity

## Viscosity increases with solid loading

- ◆ For dilute suspensions ( $\Phi < 0.01$ ) following the Einstein equation:

$$\frac{\mu_{eff}}{\mu_{bf}} = \frac{1}{(1-\Phi)[\eta]}$$

Very limited  
particle interactions

- ◆ For medium concentrated suspensions ( $0.01 < \Phi < 0.1$ ) a Taylor series development is necessary. For spherical particles:

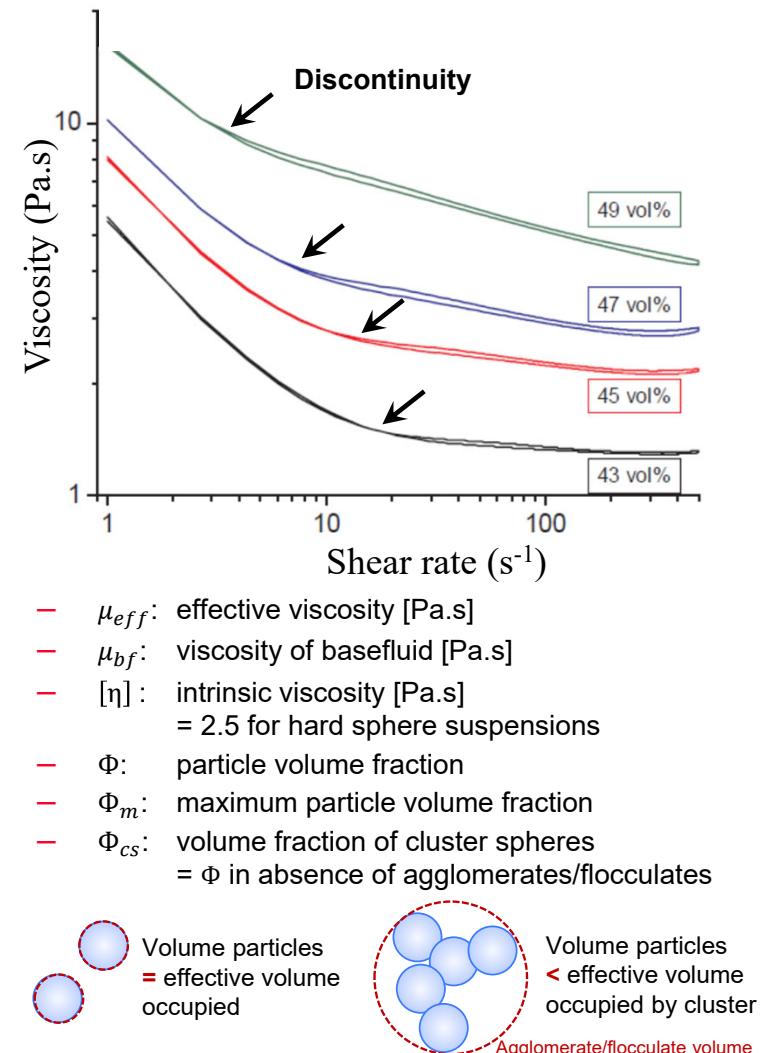
$$\frac{\mu_{eff}}{\mu_{bf}} = 1 + 2.5\Phi + 6.2\Phi^2 + O(\Phi^3)$$

Single particle and  
pair-particle interactions

- ◆ For concentration suspensions ( $\Phi > 0.1$ ) following the Krieger-Dougherty equation:

$$\frac{\mu_{eff}}{\mu_{bf}} = \left(1 - \frac{\Phi_{cs}}{\Phi_m}\right)^{[\eta]\Phi_m}$$

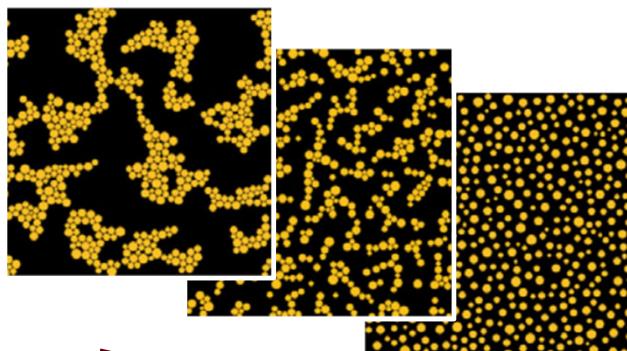
Multi-particle interactions



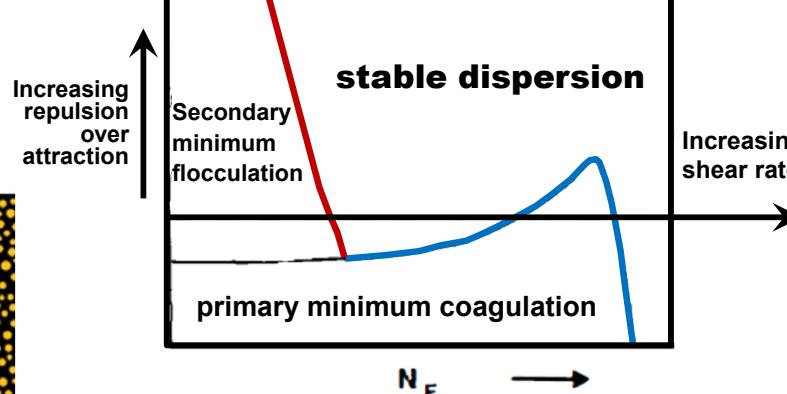
# Shear stress effect on deflocculation and coagulation

**Discontinuities** in the viscosity-shear rate curves may be related to:

**Deflocculation** induced morphology changes



Deflocculation



Importance of hydrodynamic vs attractive forces:

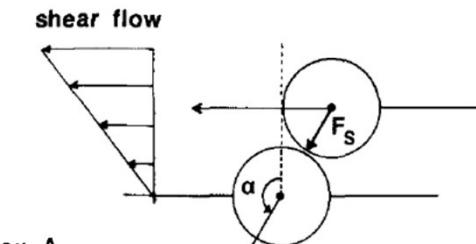
$$N_F = 6\pi\eta a^3 \dot{\gamma} / A [-]$$

—  $\eta$ : viscosity      —  $\dot{\gamma}$ : shear rate  
—  $a$ : particle radius      —  $A$ : Hamaker constant

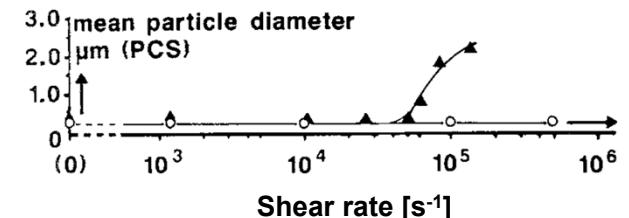
Morphology changes due to **shear induced coagulation**

For equal spheres:

$$F_S^{(max)} \propto 6\pi\eta a^2 \dot{\gamma}$$

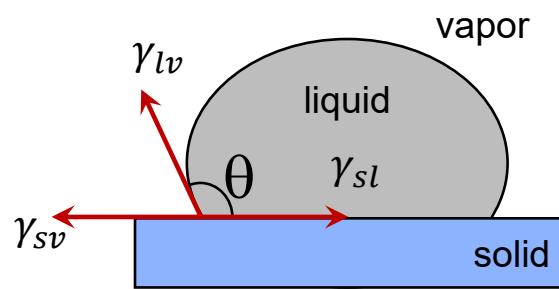


(a) Latex A



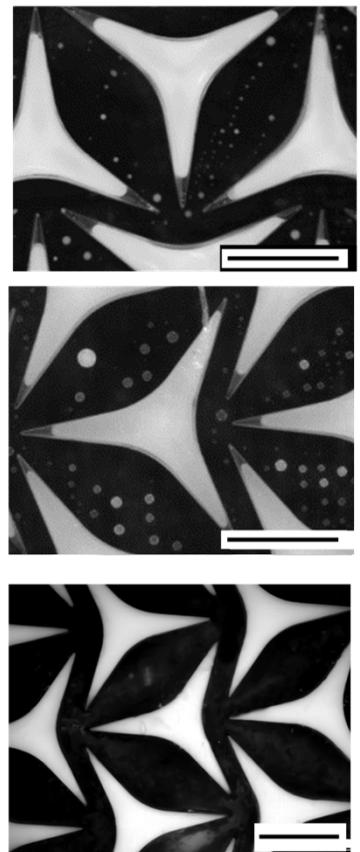
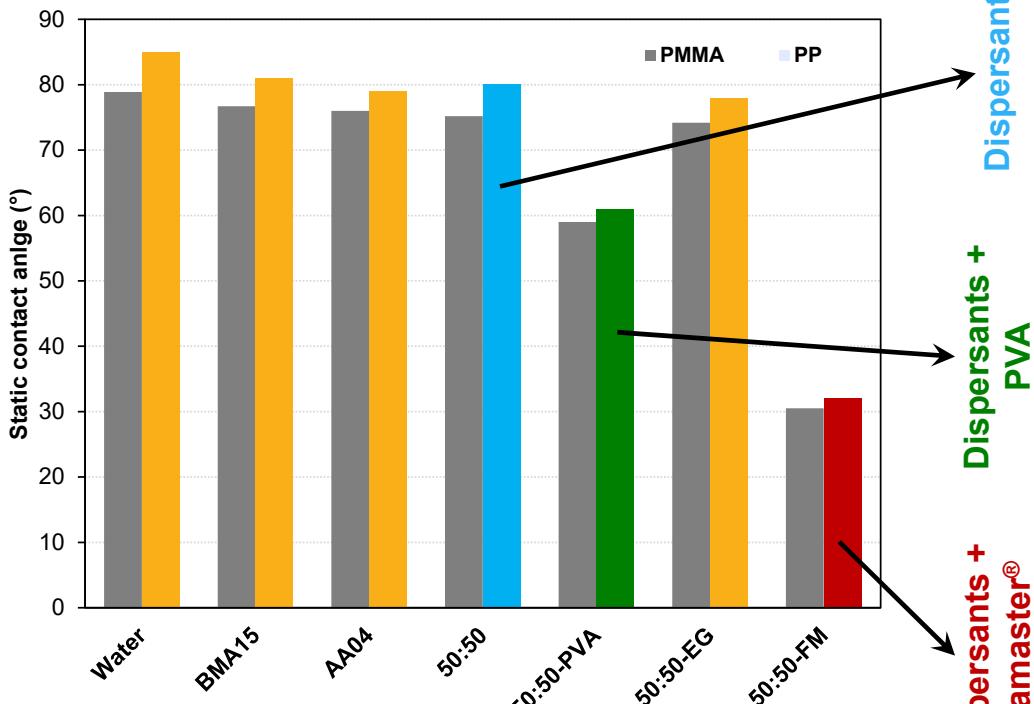
Husband, J. C. and J. M. Adams (1992). doi: [10.1007/Bf01095060](https://doi.org/10.1007/Bf01095060)

## a. Effect of contact angle on mold filling

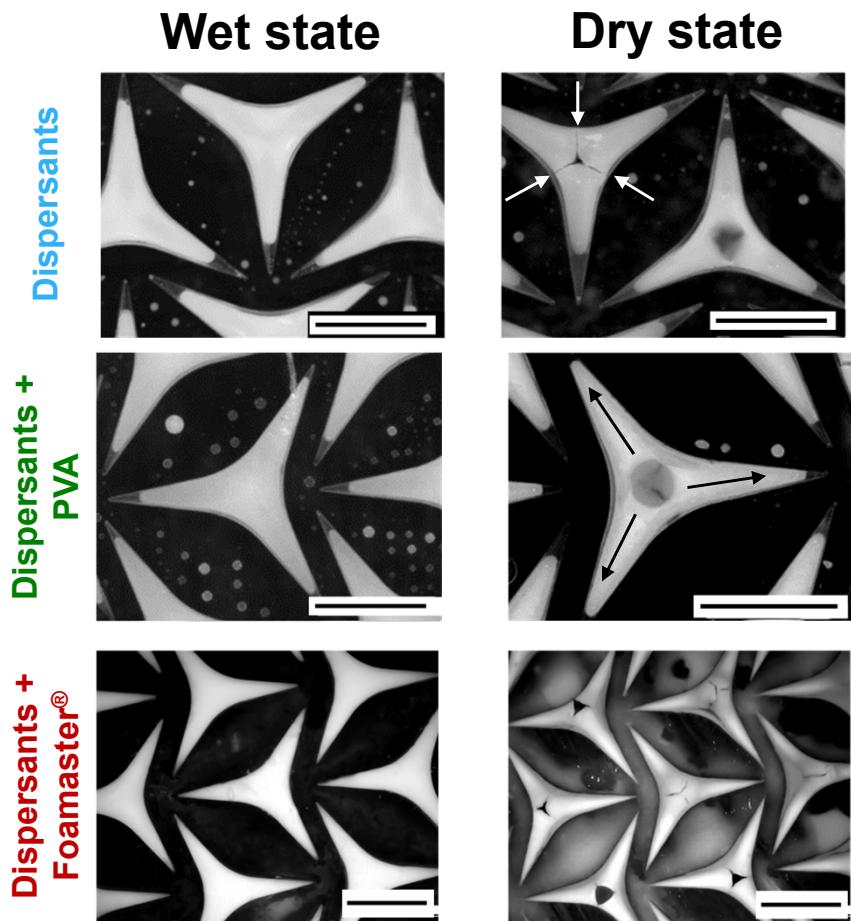


$$\cos \theta = \frac{\gamma_{sv} - \gamma_{sl}}{\gamma_{lv}}$$

Young equation



## b. Defect formation during drying



### Green body defects may occur due to:

1. **Particle migration** towards star tips due to capillary pressure “pulling” slurry during drying (e.g. drying flow + capillary pressure gradient)

$$P_c = \frac{-2\gamma_{lv} \cos \theta}{r}$$

- $\gamma_{lv}$ : surface tension (liq/vap)
- $\theta$  : contact angle (l/v)
- $r$  : characteristic pore radius

2. **Crack formation** due to drying stresses

$$\sigma \propto \frac{L \eta_l V_e \gamma_{lv}}{3 K_p}$$

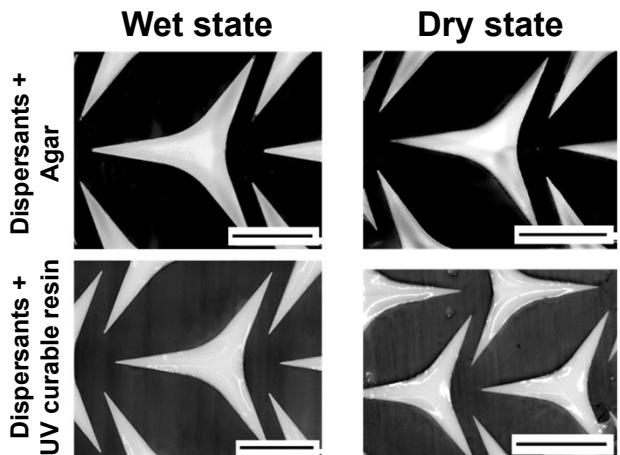
- $L$ : characteristic part dimension
- $\eta_l$ : viscosity of liquid
- $V_e$ : evaporation rate of liquid
- $K_p$ : permeability of pore network
- $\gamma_{lv}$ : surface tension (liq/vap)

## b.1. Mitigation of particle migration

### Considered approaches

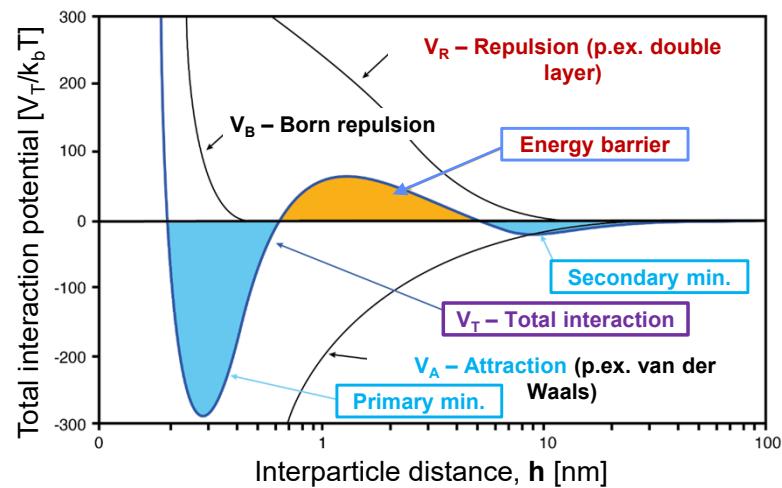
1

#### Gel-casting / Curing



2

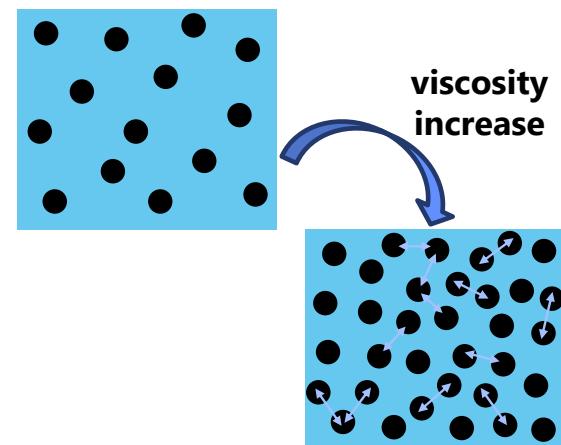
#### Flocculate dispersion (yield stress)



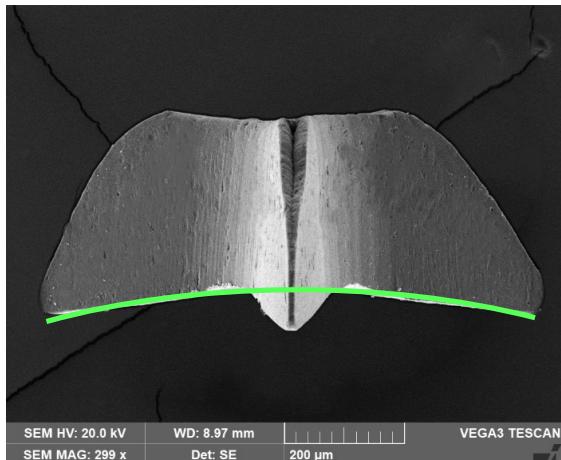
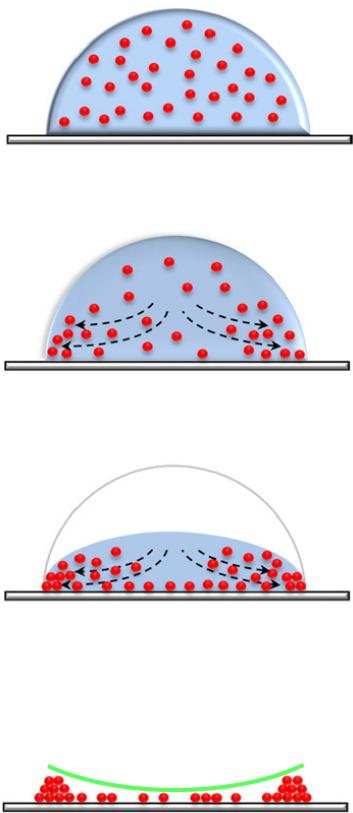
source: Wikipedia

3

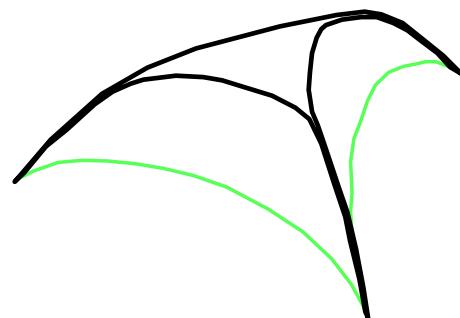
#### Increase particle loading



## b.1. Mitigation of particle migration: Design benefits

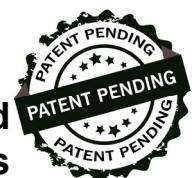


**Curved star particles with appropriate contact angle and solid loading**  
→ Improved rolling and punctuation performance



Source: Al-Milaji, K. N. and H. Zhao (2019).  
doi: [10.1021/acs.langmuir.8b03406](https://doi.org/10.1021/acs.langmuir.8b03406)

Shape and process patents



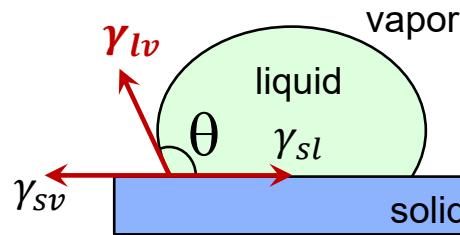
## b.2. Mitigation of drying cracks

### Decrease drying stress

- ◆ Addition non-volatile co-solvent preserving appropriate contact angle

$$\sigma \propto \frac{L \eta_l V_e \gamma_{lv}}{3 K_p}$$

- $L$ : characteristic part dimension
- $\eta_l$ : viscosity of liquid
- $V_e$ : evaporation rate of liquid
- $K_p$ : permeability of pore network
- $\gamma_{lv}$ : surface tension (liq/vap)



$$\cos \theta = \frac{\gamma_{sv} - \gamma_{sl}}{\gamma_{lv}}$$

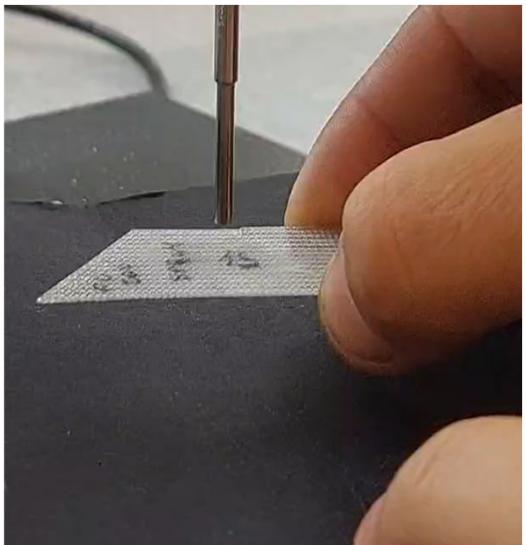
Young equation

- Background and motivation
- Functional analysis and fabrication strategy
- Mold fabrication
- Slurry formulation and casting
- **Demolding, debinding and sintering**

## Demolding, debinding and sintering



# Ultrasound demolding

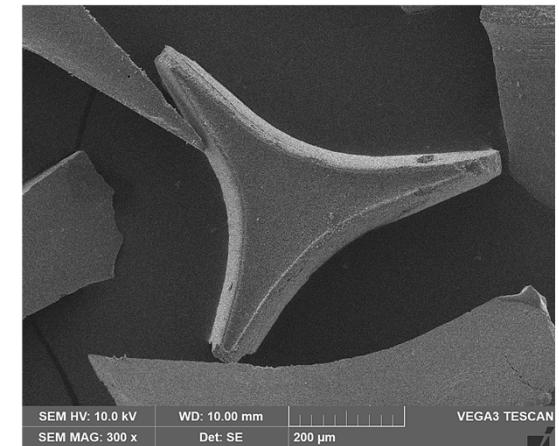
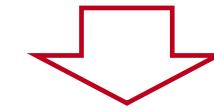


- ◆ **Part removal by mold ashing**
  - Part **cracking** due to mold deformation (i.e. thermoplastic)
  - **Carbon footprint** of process
- ◆ **Demolding by ultrasound**
  - Part cracking due to adhesion forces

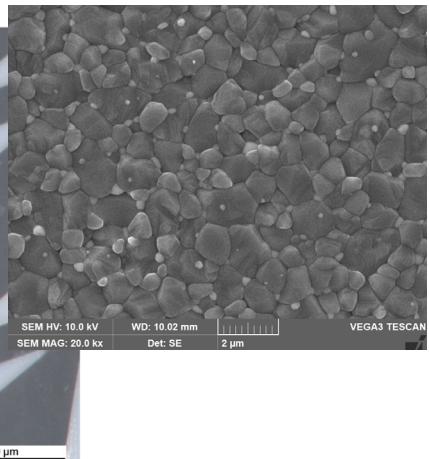
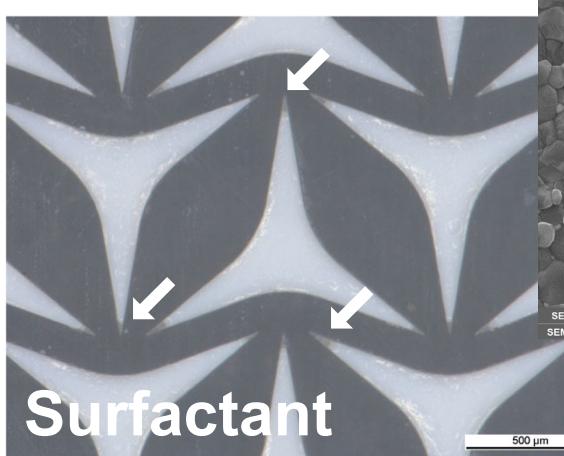
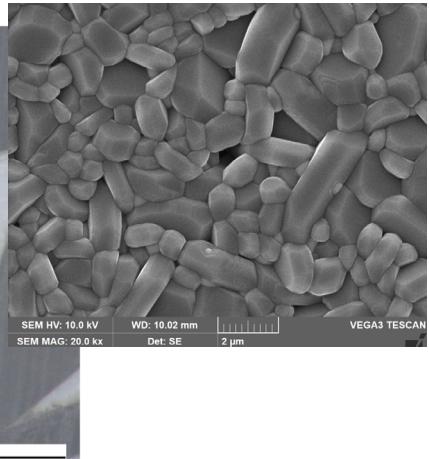
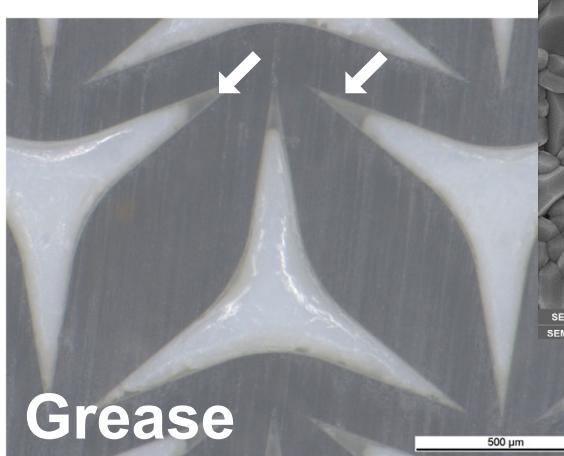
**Demolding strategy needed**



- Mold surface modification**
- Slurry modification**



## Mold modification



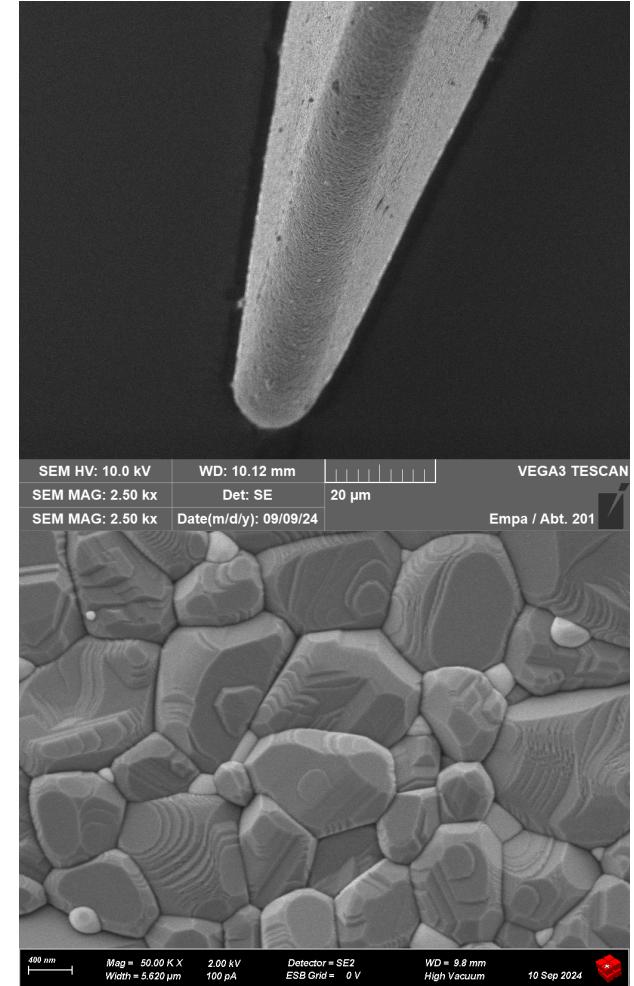
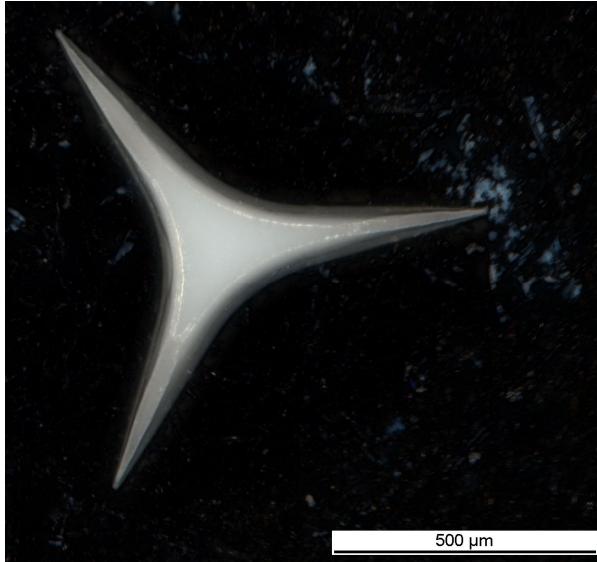
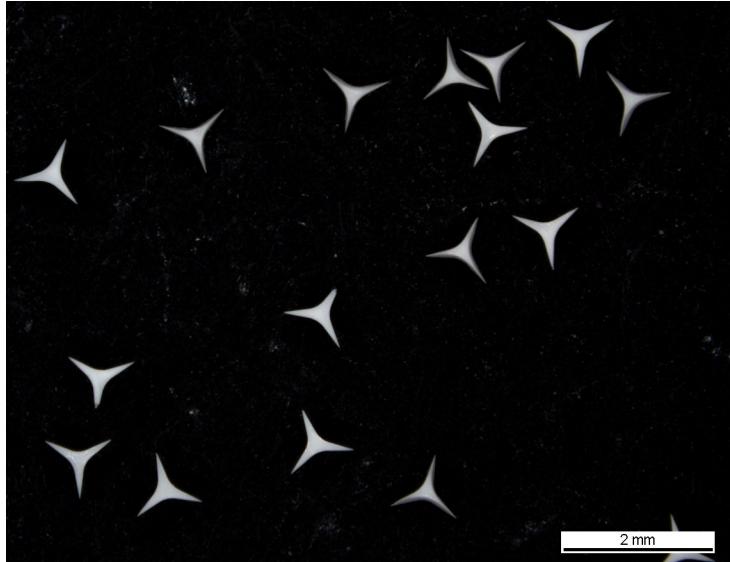
### Application of grease film

- Filling issues
- Removal of zirconia sol at surface
- But significantly decreases part loss

### Application of surfactant film

- Filling issues with excess surfactant
- **Combination of mold treatment with surfactant and surfactant addition to slurry**

## After slurry modification with surfactant



Case study

### Some process metrics

- ◆ Process yield >90%
- ◆ Tip diameter <10 μm

# The end

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*What was on the  
- MENU -  
today?*



## ◆ Summary

- Review of typical dry and wet shaping techniques
- Additive manufacturing techniques with focus on stereolithography-based ones
- Critical aspects related to drying and debinding
- Use case: Development of a scalable process for large scale shaping of ceramic microdevices

## ◆ Questions?