

## ENV-413: Thermodynamics of the Earth systems

### Exercise session for Lecture 11

20. Aerosol particles that attract water are called

- a) hydrophobic
- b) hydrophilic

21. Cloud condensation nuclei (CCN) are aerosol particles that nucleate water drops at supersaturations less than

- a) 100%
- b) 10%
- c) 1%

22. Which of these aerosol particles are likely to act as cloud condensation nuclei (circle all that apply)

- a) clay
- b) NaCl
- c)  $(\text{NH}_4)_2\text{SO}_4$
- d) AgI
- e) pollen
- f) sand

Combination of Kelvin's Law (#16, pure droplets with surface tension effects) with Raoult's Law (4.48) for solutions yields Kohler's equation:

$$\frac{e_s(r, m_{\text{salt}})}{e_s} = \left[ 1 - \frac{b}{r^3} \right] \exp(a/r) \quad (5.17)$$

where  $a = 2\sigma_{lv}/(\rho_l R_v T)$ . Use the Kohler curve below to estimate:

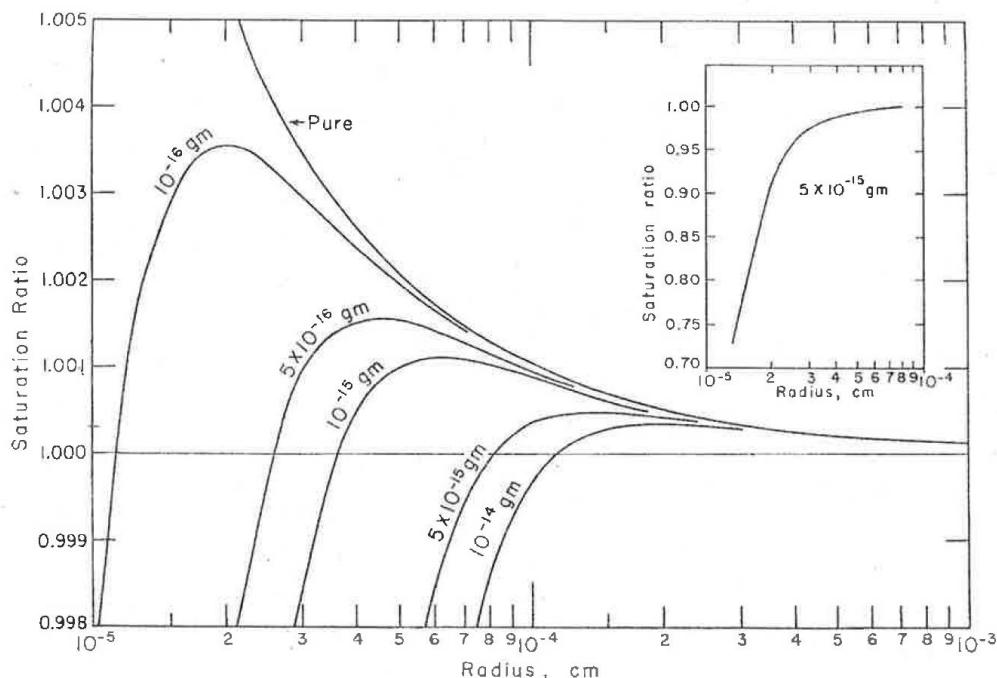


FIG. 2.4.---Curves of equilibrium saturation ratio of water droplets containing the stated mass of sodium chloride compared with Kelvin curve for pure water droplets. *Inset:* curve for  $5 \times 10^{-15}$  g NaCl on a compressed scale extended to the droplet size at which the given amount of NaCl would form a saturated salt solution in the droplet. All computations are made for a temperature of  $25^\circ \text{C}$ , but the values are very nearly the same at other atmospheric temperatures.

25. The radius of the droplet that will be in equilibrium on a NaCl particle of mass  $10^{-15}$  g in air which is 0.1% supersaturated.

26. The relative humidity of the air adjacent to a droplet of 0.3 microns with  $10^{-15}$  g NaCl.

27. The critical supersaturation required for a NaCl particle of mass  $10^{-16}$  g to grow beyond the haze state.

28. Consider the droplet with radius  $r$  with a CCN consisting of  $10^{-15}$  g NaCl in an environment with saturation ratio  $S$ , as indicated by point A. Is the droplet growing, evaporating, or in equilibrium?
29. Consider the droplet with radius  $r$  with a CCN consisting of  $10^{-15}$  g NaCl in an environment with saturation ratio  $S$ , as indicated by point B. Is the droplet growing, evaporating, or in equilibrium?
30. Use the Excel CCN calculator sheet and solve the problems therein.