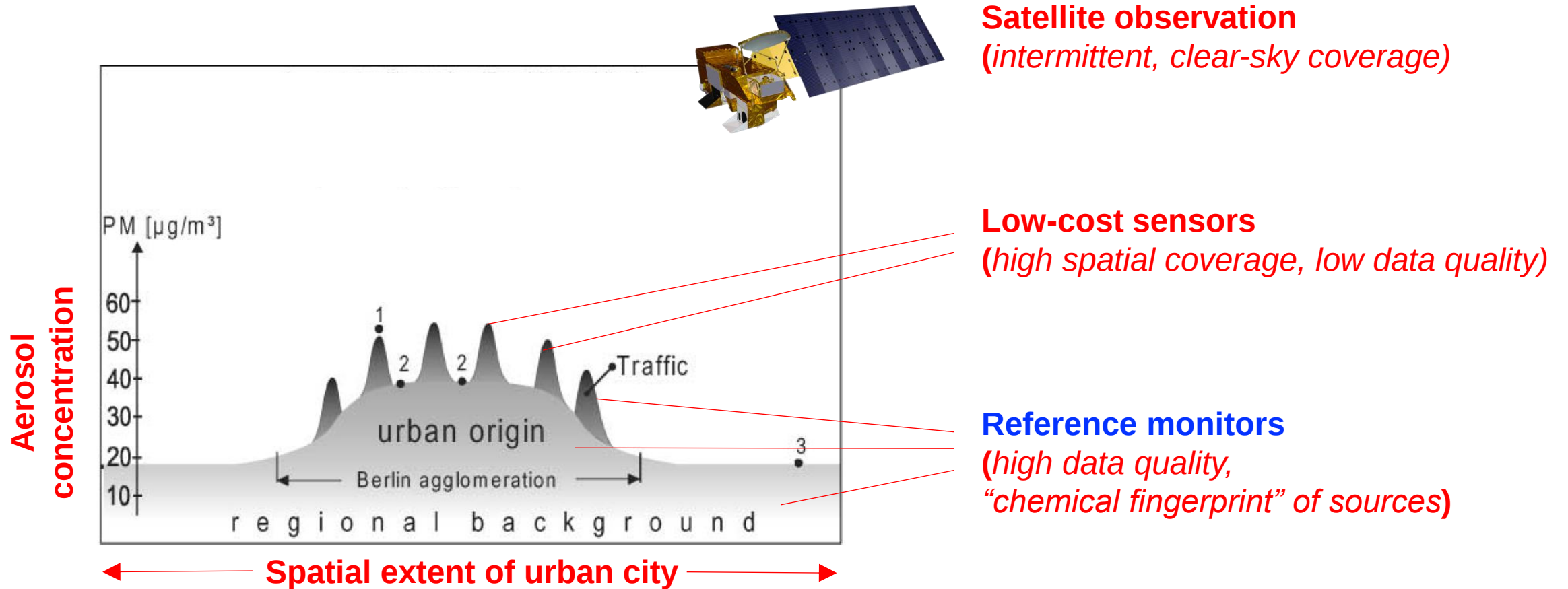


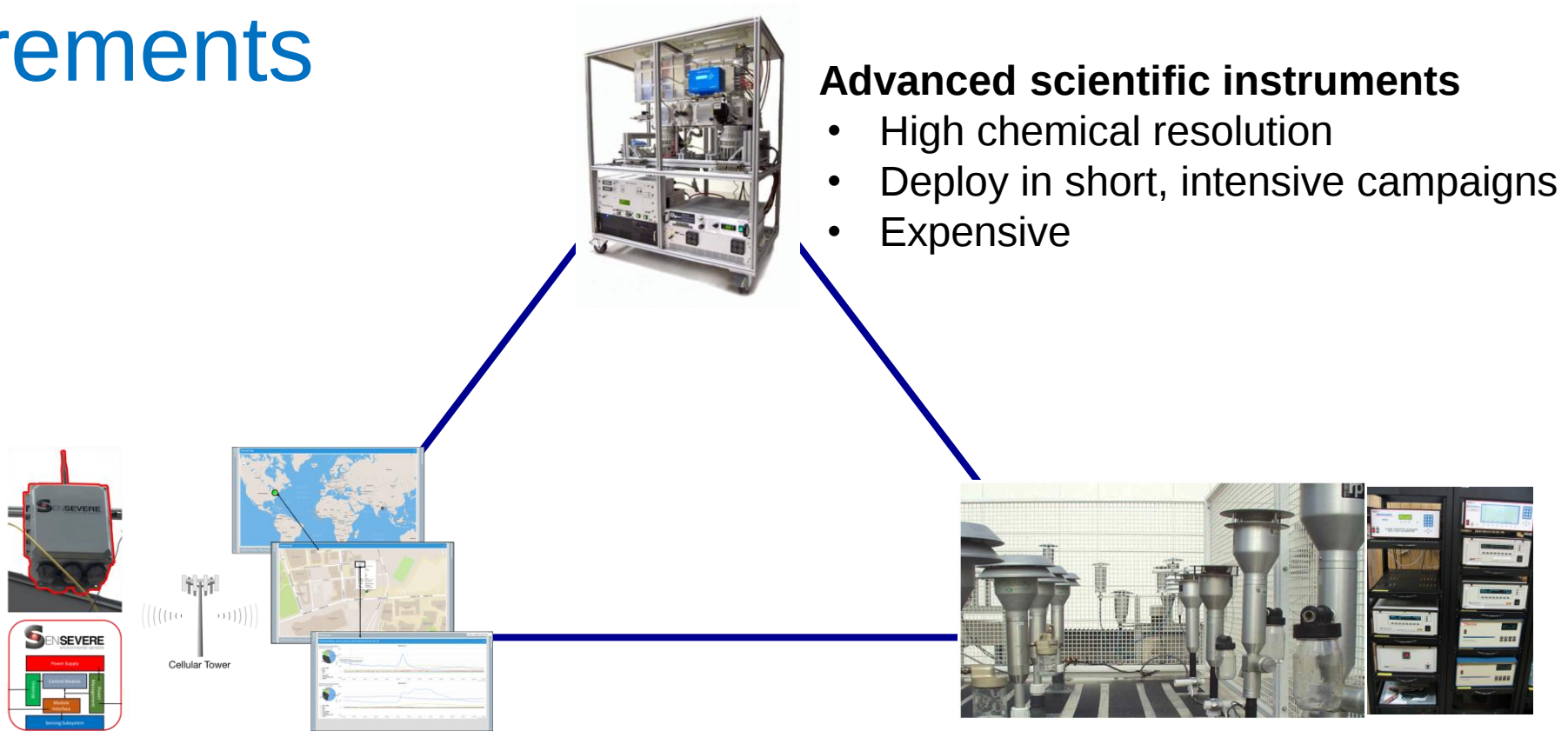
Aerosol chemical analysis

ENV-409

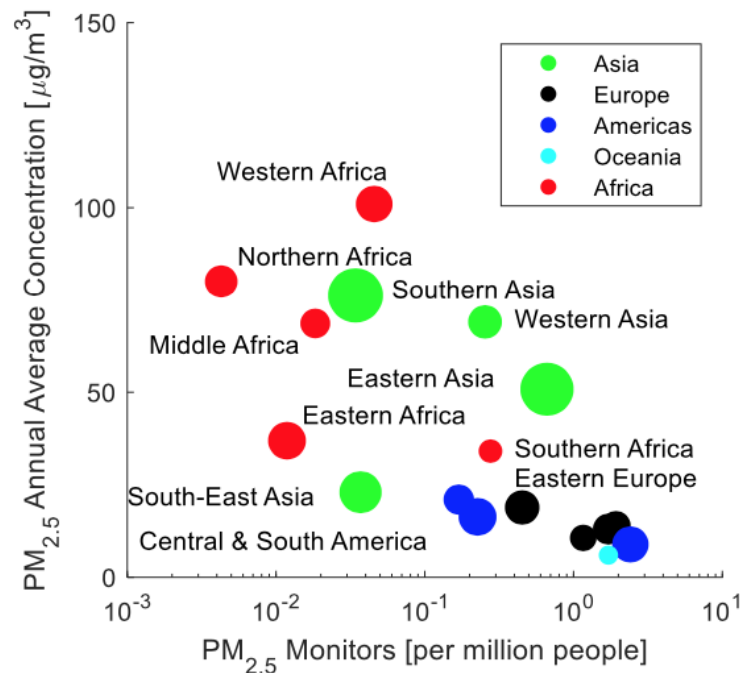
Complementary monitoring strategies



Ground-based aerosol measurements



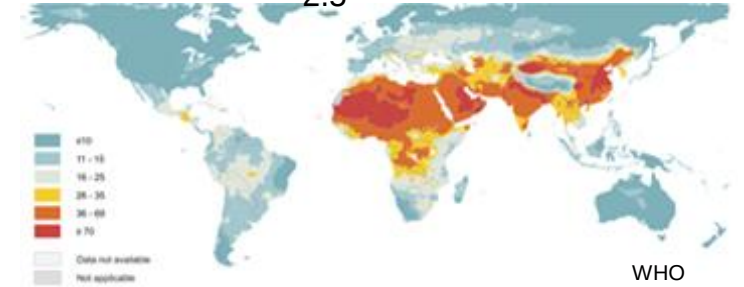
Need for cost-effective monitoring of air pollution



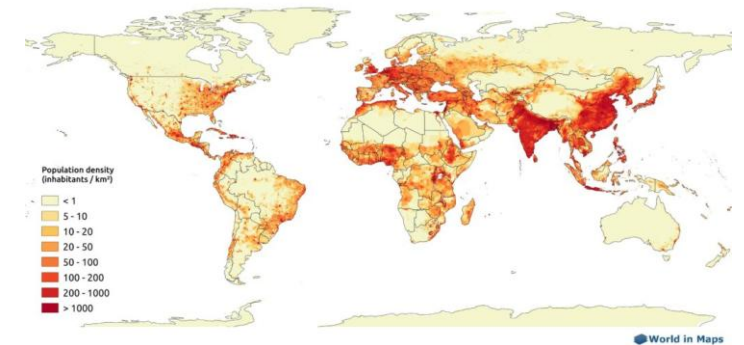
*Lack of monitors in
high population areas*

Malings et al., 2020

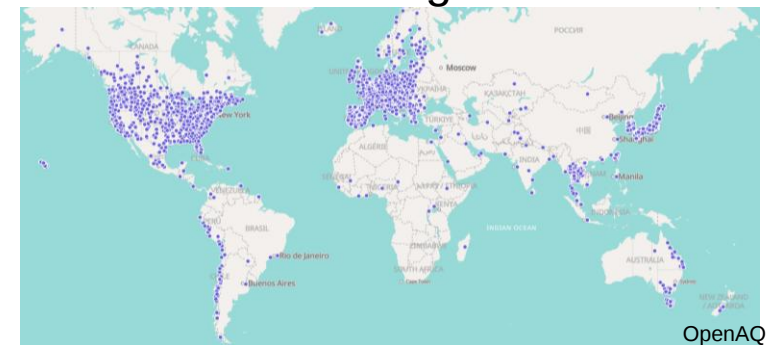
Estimated PM_{2.5} concentrations



Population density



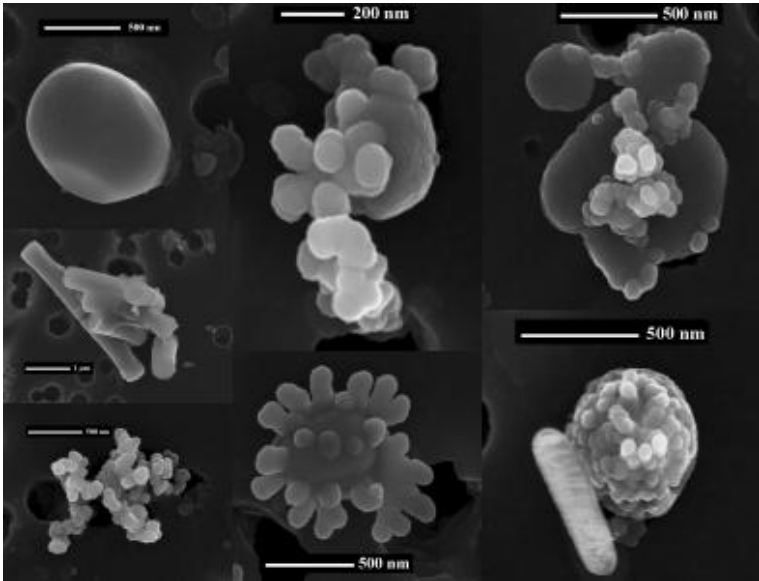
Available monitoring data



Atmospheric aerosols

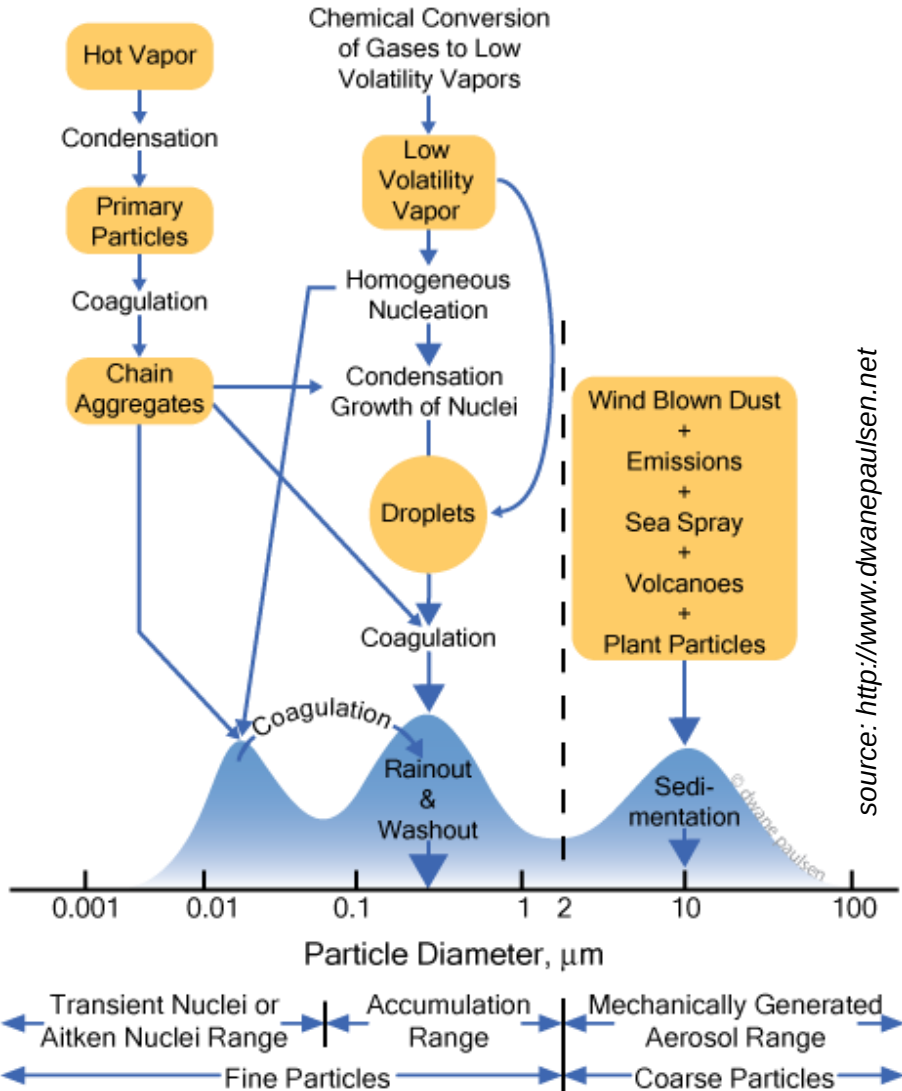
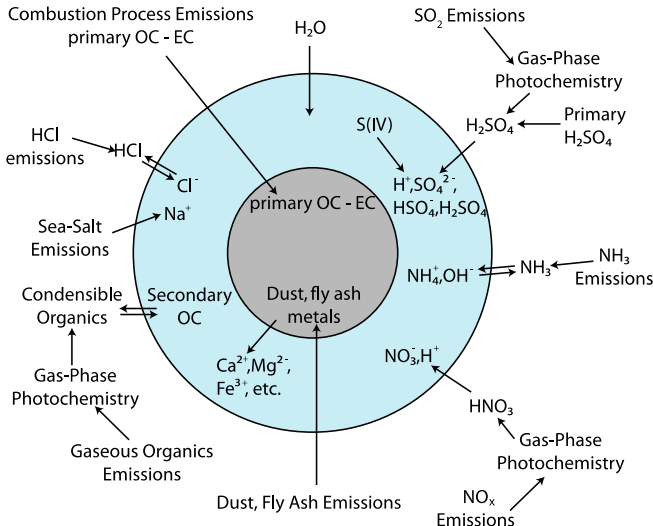
Scanning electron microscopy images

Coz et al., Aerosol Sci. Tech., 2008



- Varies in
- size
 - shape
 - phase
 - composition

adapted from Pandis et al.,
J. Phys. Chem., 1995



Aerosol sources

Mixed



Forest fires



Sea spray



Dust

Natural



Volcanic eruptions



Traffic / Transport



Domestic activities



Industry



Agriculture

Anthropogenic

High variation in aerosol composition

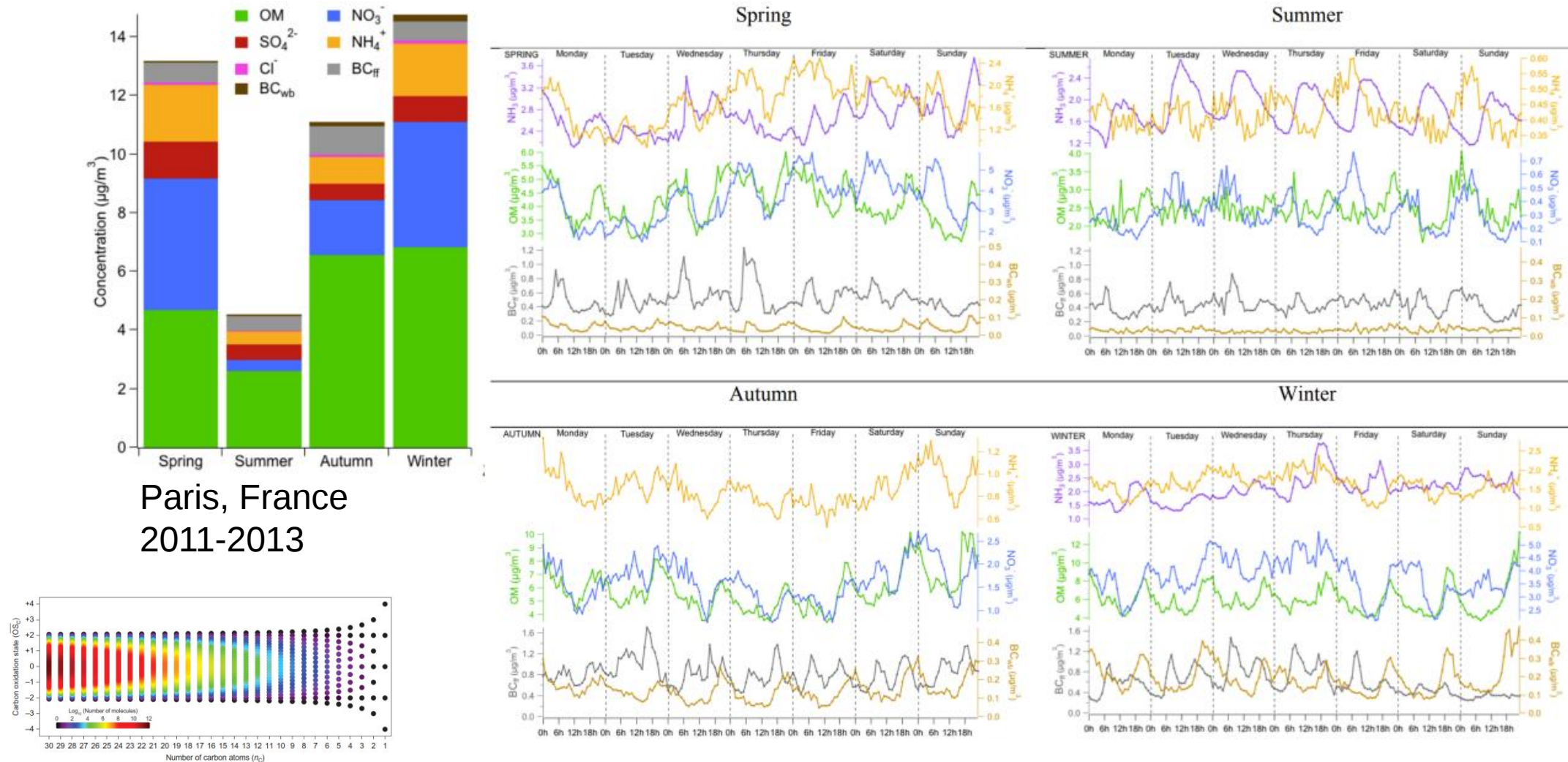


Figure 3 | Chemical complexity of organics as a function of oxidation state and carbon number. Points are coloured by the logarithm (base 10) of the number of possible compounds at a given OS_C and n_C , assuming an unbranched, acyclic carbon skeleton, and the addition of carbonyl, alcohol and acid groups only. Including a wider range of carbon skeletons or functional groups will lead to a dramatically steeper increase in chemical complexity with OS_C and n_C (ref. 31).

Kroll et al., *Nat. Chem.*, 2011

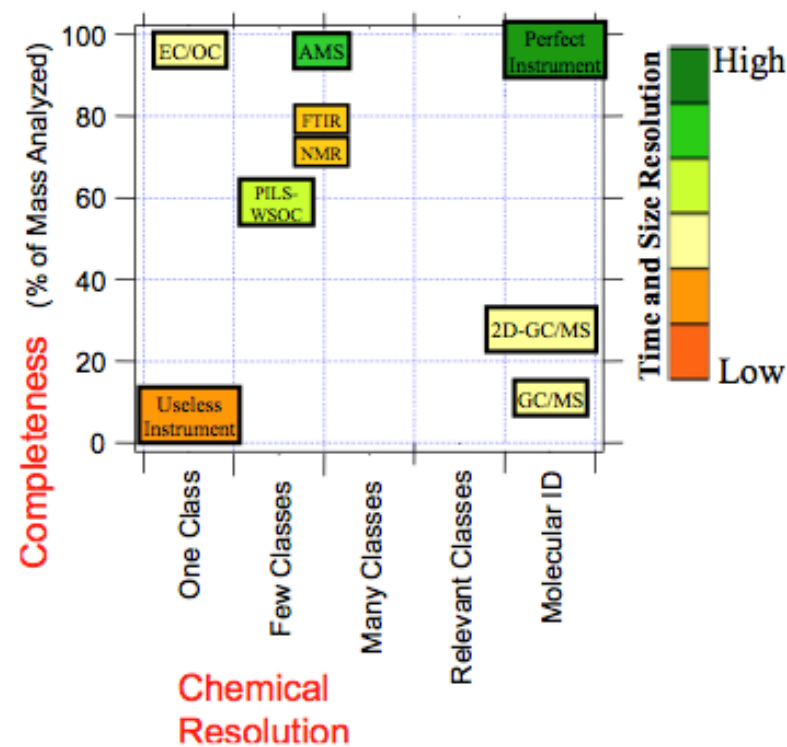
Petit et al., *Atmos. Chem. Phys.*, 2015

Chemical measurement technology tradeoffs

Diverse range of molecular classes, size, polarity in atmospheric particles

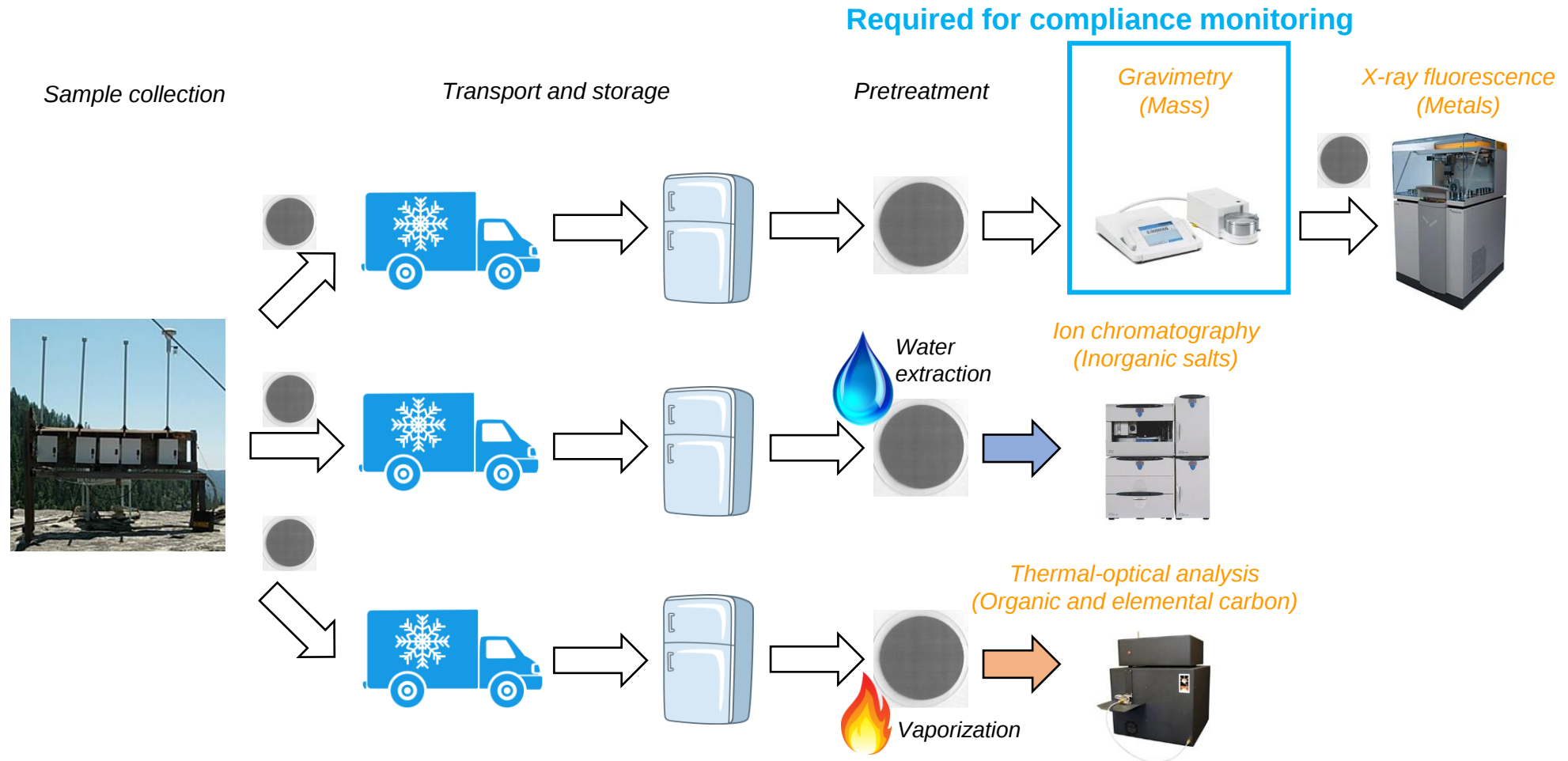
No single measurement technique for measurement of all molecules. Tradeoffs:

- offline / online
- time resolution
- chemical resolution (number of species or classes)
- completeness
- quantitative / qualitative
- Separation
 - physical / chemical
 - algorithmic
- scalability – labor and cost

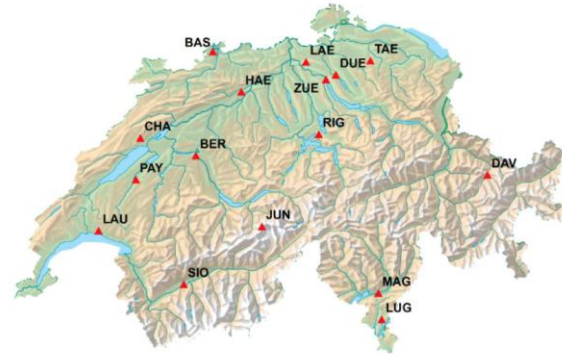


Hallquist et al., *Atmos. Chem. Phys.*, 2009

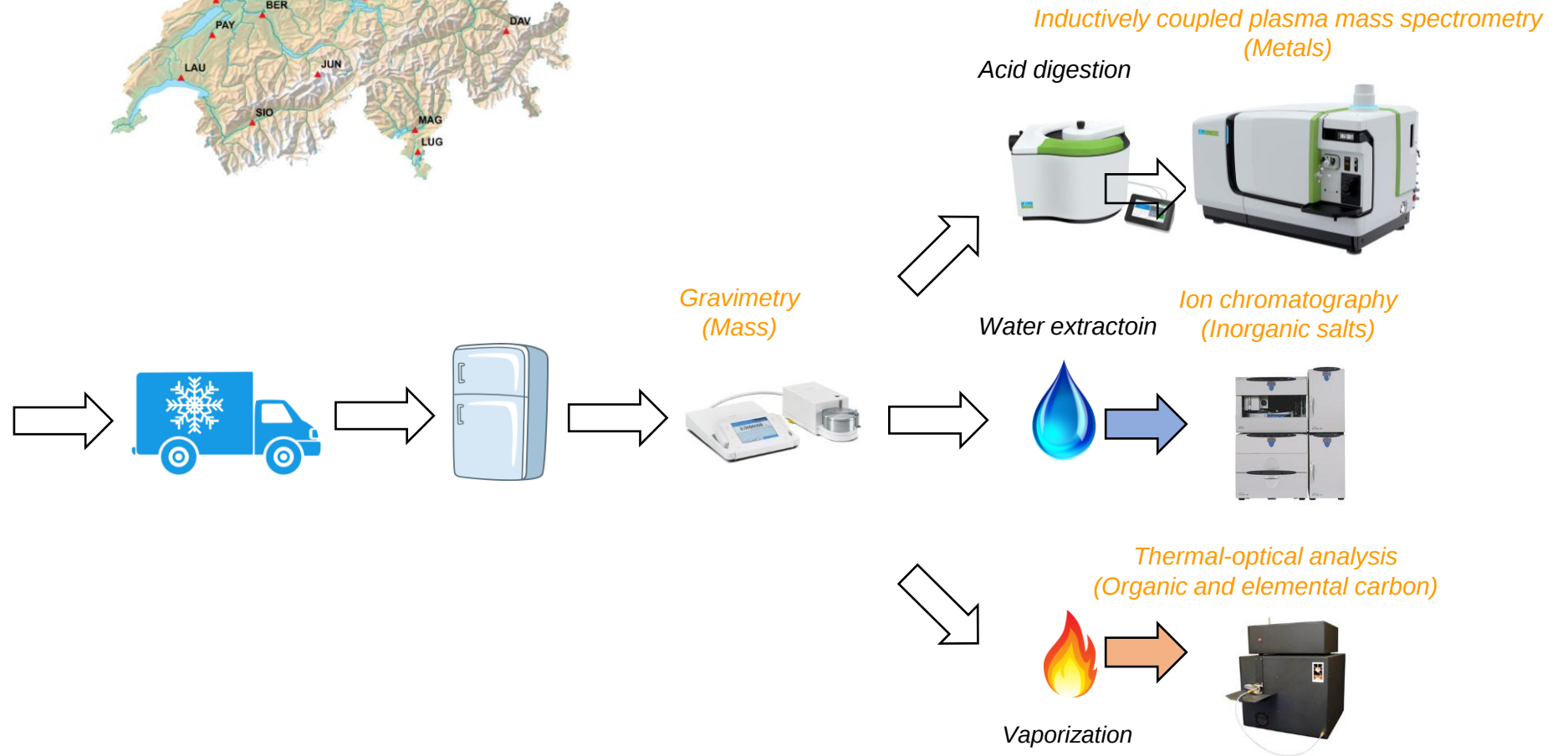
Conventional chemical speciation methods



NABEL Network

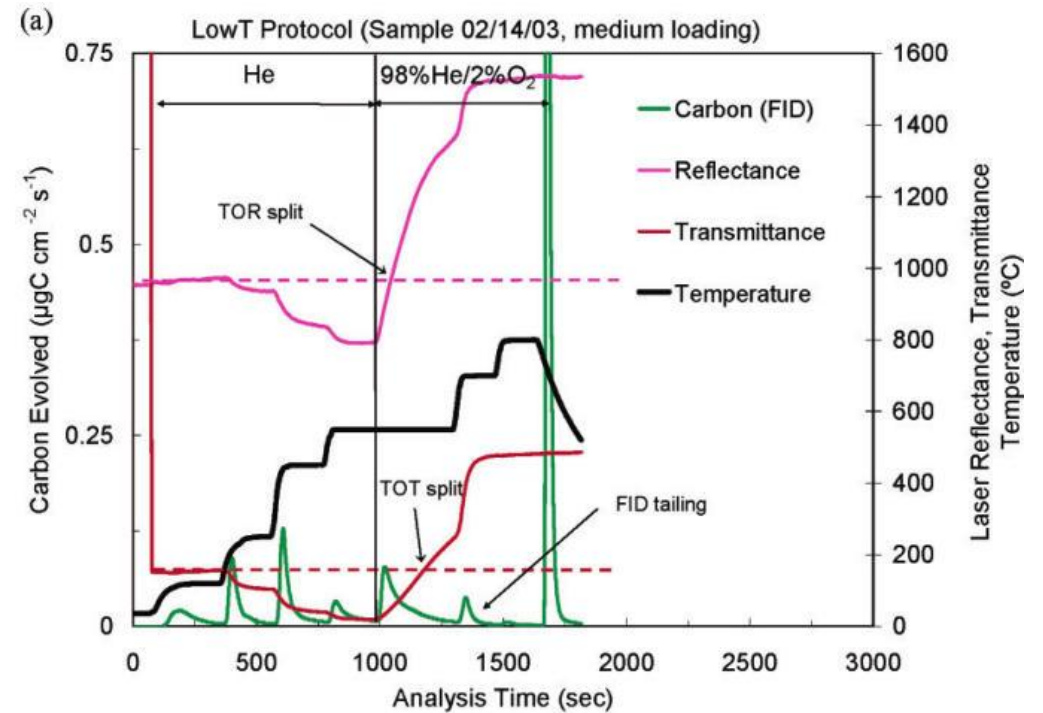


Single quartz filter



Thermal optical methods

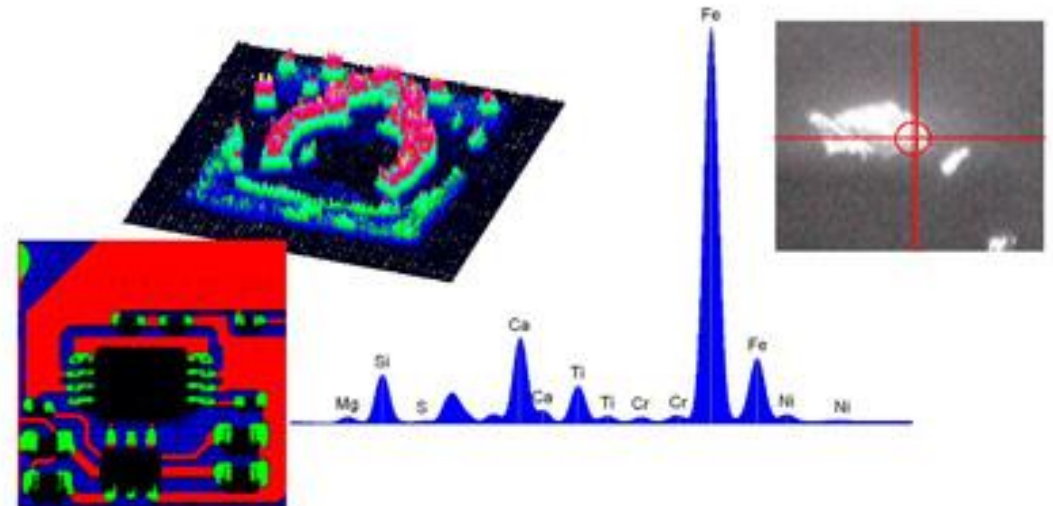
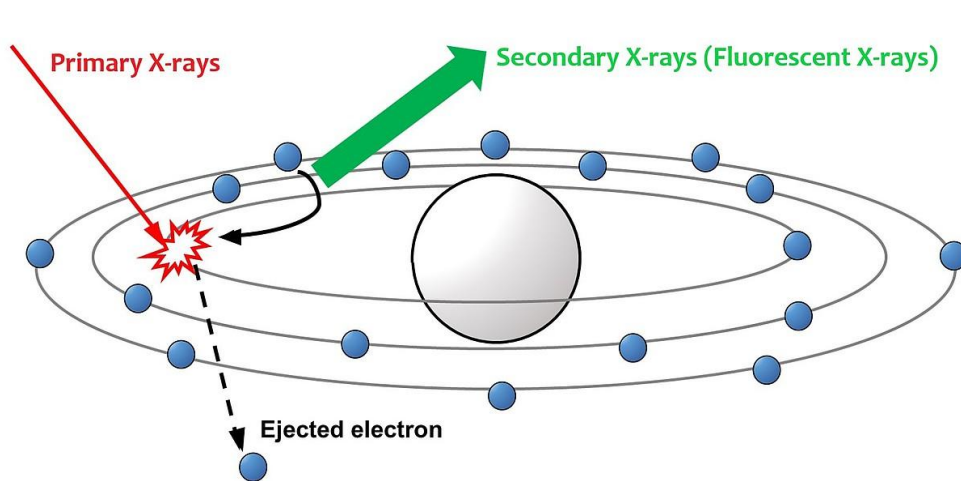
- Thermal Optical Reflectance (TOR) and Thermal Optical Transmission (TOT)
- Detector (flame ionization detector or non-dispersive IR) measures total carbon (TC)
- TC is split into organic carbon (OC) and elemental carbon (EC) via *evolved gas analysis*
- Organic matter mass can be ~1.4-2.5 times OC



Chow et al., *Environ. Sci. Tech.*, 2004

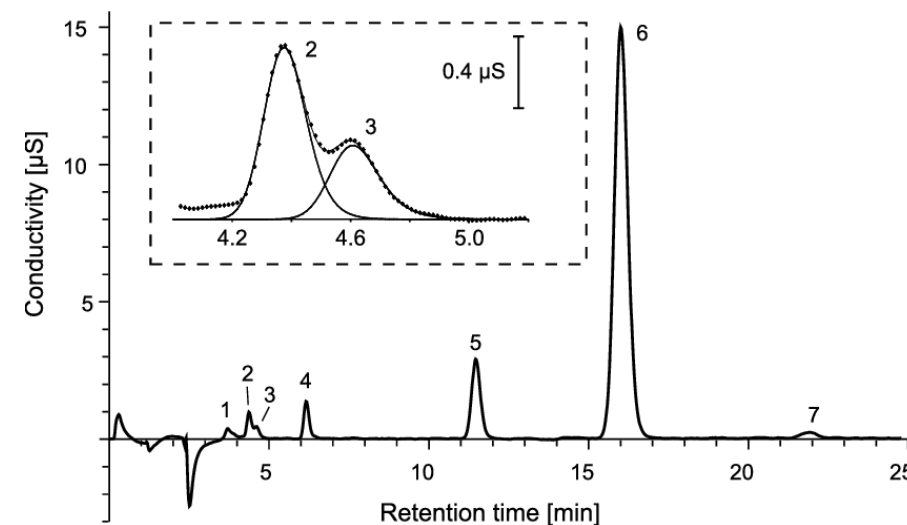
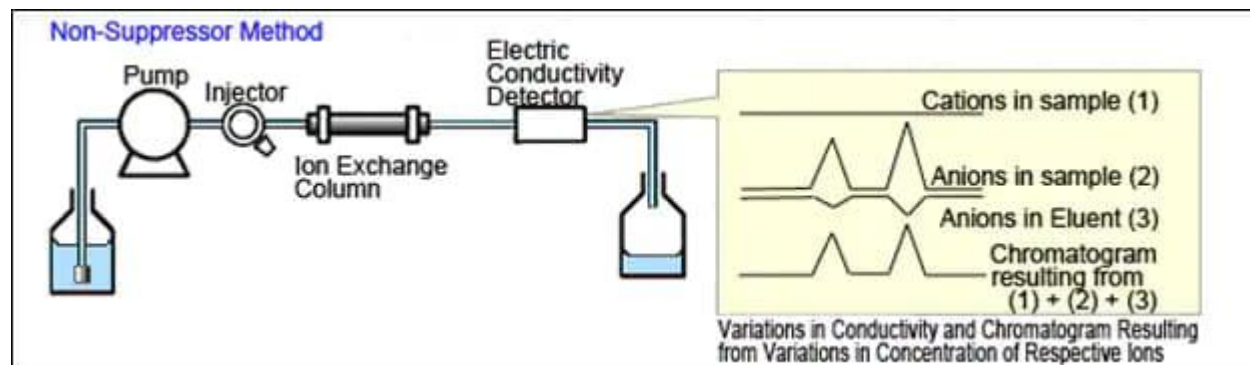
X-Ray fluorescence

- Analysis of ~40 elements
- Most comprise a few percent of the total mass



Ion chromatograph

- Major water-soluble ions (inorganic salts, dust)
- Cations and anions require separate columns and mobile phases – both are not always analyzed



Automation

Carbon analyzer



Magee Scientific

*Ion monitors
(ICs not shown)*



Metrohm
URG Corporation

Metals analyzer



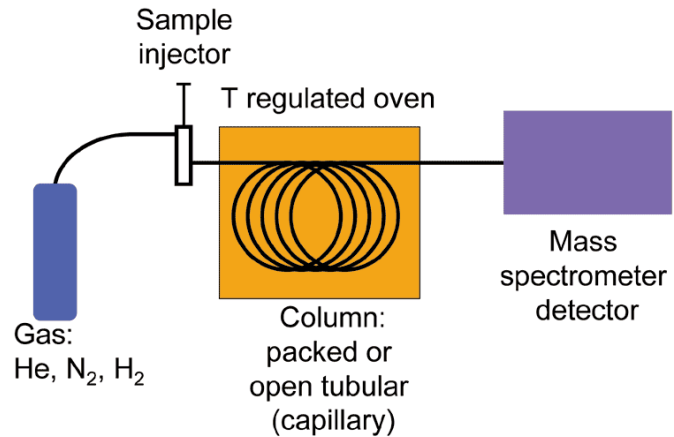
Cooper Scientific

Scientific workhorses

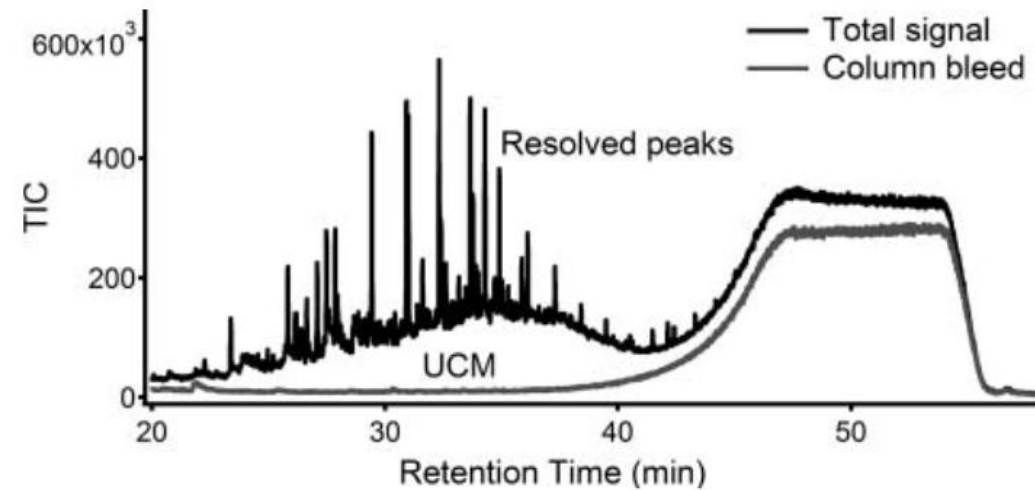
Gas chromatography

- Preparation: extraction / derivatization / vaporization
- Molecular identification possible with mass spectrometer
- 10-20% mass recovery

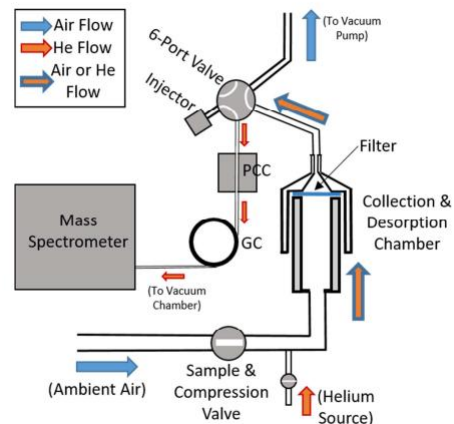
Zhang et al., *Aerosol Sci. Tech.*, 2014



Wikipedia.com



Online version



Cropper et al., *Atmos. Environ.*, 2017



Agilent Technologies

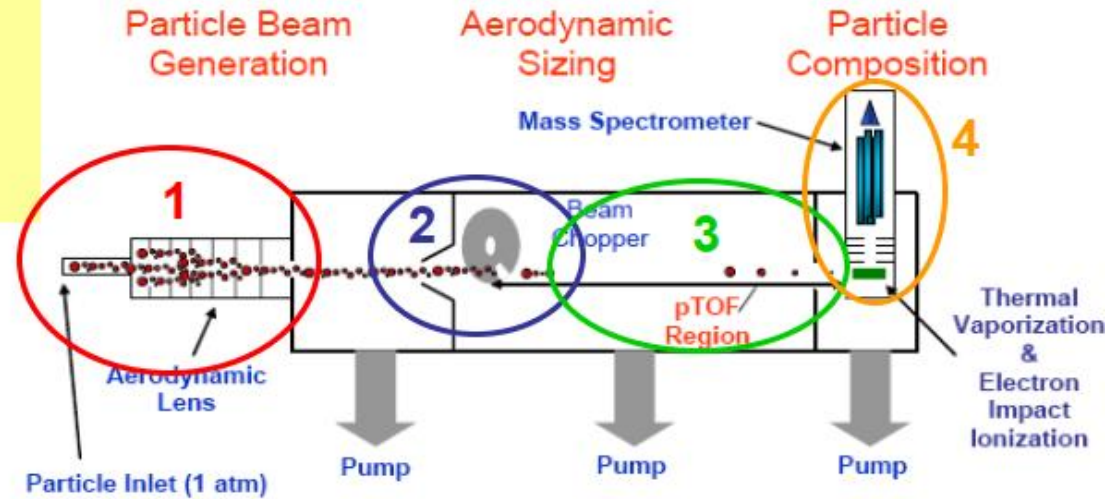
measurement of particle ingredients

Aerosol Mass Spectrometer (AMS)

on-line analysis of particulate composition
elements: C, H, O, N, Fe, Zn, Pb, Hg
and PAHs

- High time resolution
- Inorganic + organics (typically not dust)
- Collection efficiency (50-100%)
- Molecular mass fragments

1. Aerosol particles (0.04-1 μm) are sampled into a vacuum system and focused
2. A rotating beam chopper provides packages of particles
3. Size separation with the time-of-flight (TOF) technique
4. Detection of elements by quadrupole mass spectrometry (or time of flight mass spectrometer)



Aerodyne and TSI

