

Measurements of atmospheric trace gases

Stefan Reimann
stefan.reimann@empa.ch

Empa

Laboratoire fédéral d'essai des matériaux et de recherche



Empa

Materials Science and Technology

do-it yourself

?

?

What is the heaviest thing in this room?

?

?

?

?

?

?

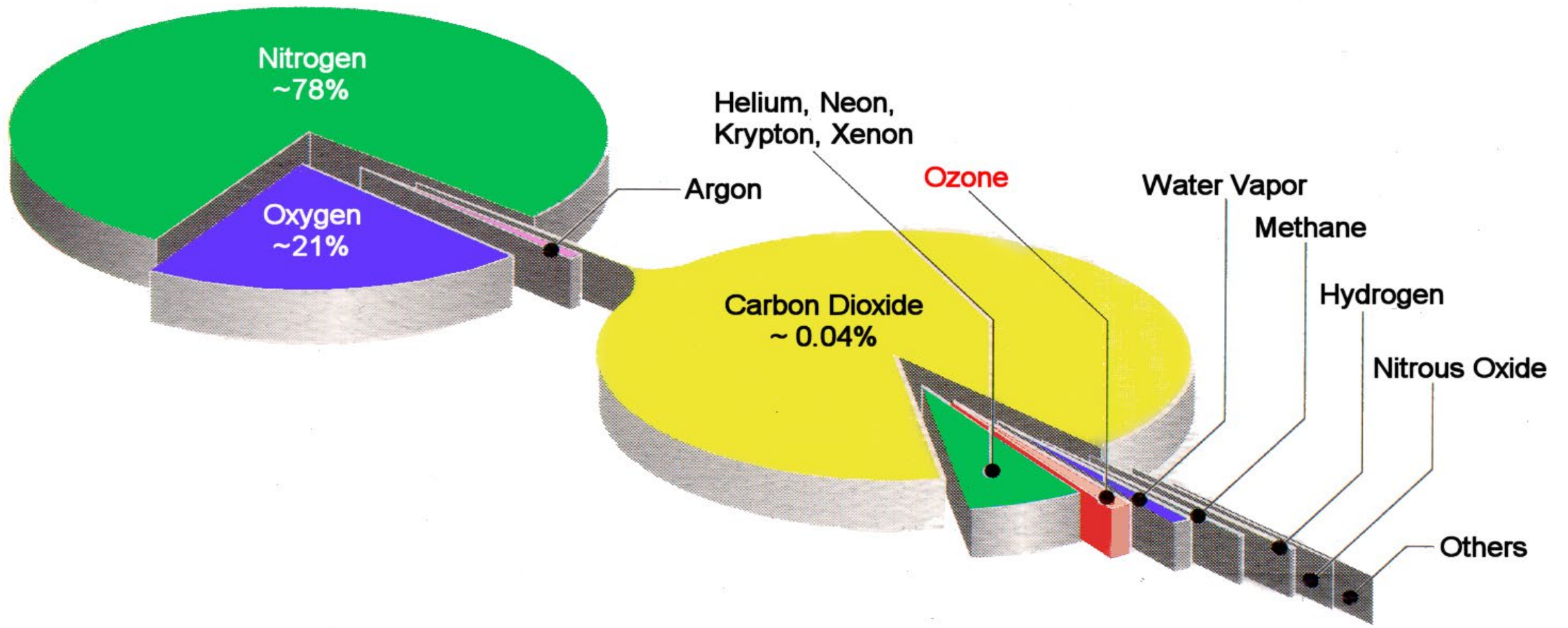
?

?

do-it yourself

What is the heaviest thing in this room?

trace components in the atmosphere



do-it yourself

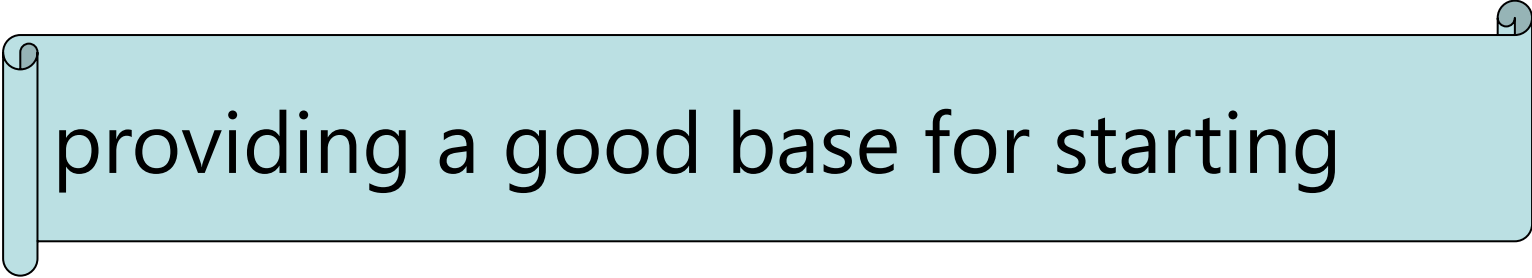
400 ppm:

What is the volume of CO₂ in this room?

10 x 10 x 3 m

Aims of the course

- overview on atmospheric trace compounds
- understanding of how trace compounds are monitored
- measurement practice – good and bad
- using atmospheric measurements for interpretation



providing a good base for starting

Schedule of the lessons

08.03.23:

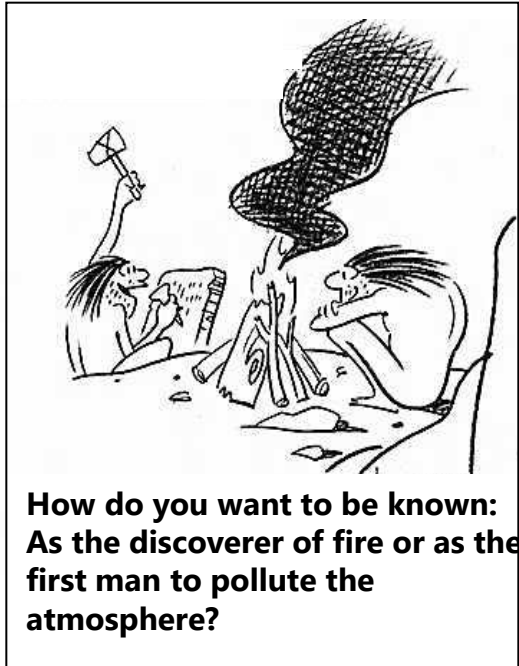
- introduction
- measurement strategiesdesign of a station
- measurements of trace compounds (1)
- exercise: measurement strategy

15.03.23:

- measurements of trace compounds (2)
- exercise: quality assurance
- instrument exhibition/exercise: contamination

historic development of air pollution

local



global



regional



Measurement strategies and quality assurance

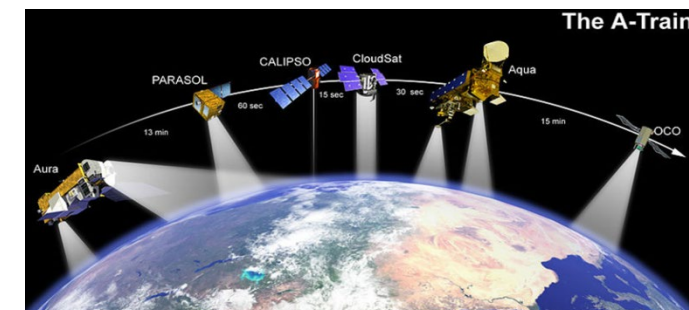
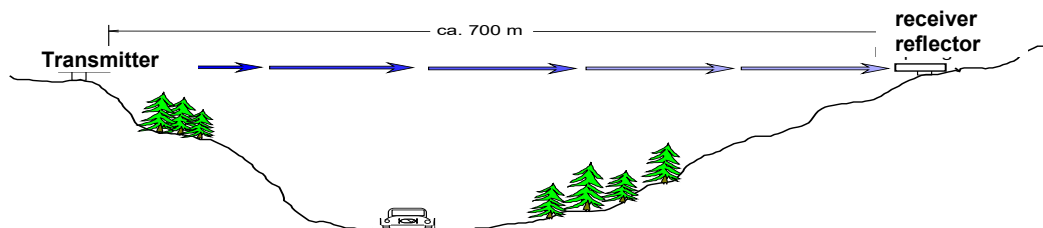


Measurement strategies

- *why are ambient air measurements performed?*
legislation (limit values), research
- *which compounds should be measured?*
gases, particles
- *time resolution of the measurements?*
continuous (<second, minute, hour, day)? non-continuous?
limit values: hourly/annual means?
- *location of the measurements?*
stack emissions (chimney), urban air, background air, personal samplers
(exposition) , fluxes (ecosystems)
is the sampled air representative?
are there undesirable sources nearby?)

two principal measurement strategies

- **with sampling (in-situ)**
 - online, automatic analysis at site (monitor)
 - quasi online (automated GC)
 - offline, sampling at site, analysis in laboratory (passive samplers)
- **without sampling (remote sensing, ground-based/satellites)**
 - DOAS (Differential Optical Absorption Spectroscopy)
 - FTIR (Fourier Transformation Infrared Spectroscopy)
 - LIDAR (Light Detection and Ranging)
 - TDLS (Tuneable Diode Laser Spectroscopy)



Introduction to measurements of trace gases in the atmosphere

1. Regional/National networks (mostly for legislation)

- e.g. NABEL network, Switzerland

2. Global networks (for detection of trends/source regions)

- in-situ (NOAA/AGAGE)
- airplanes (e.g. CARIBIC)
- Remote sensing (ground-based, e.g. FTIR, NDAAC)
- Remote sensing (satellites)



3. Emission measurements

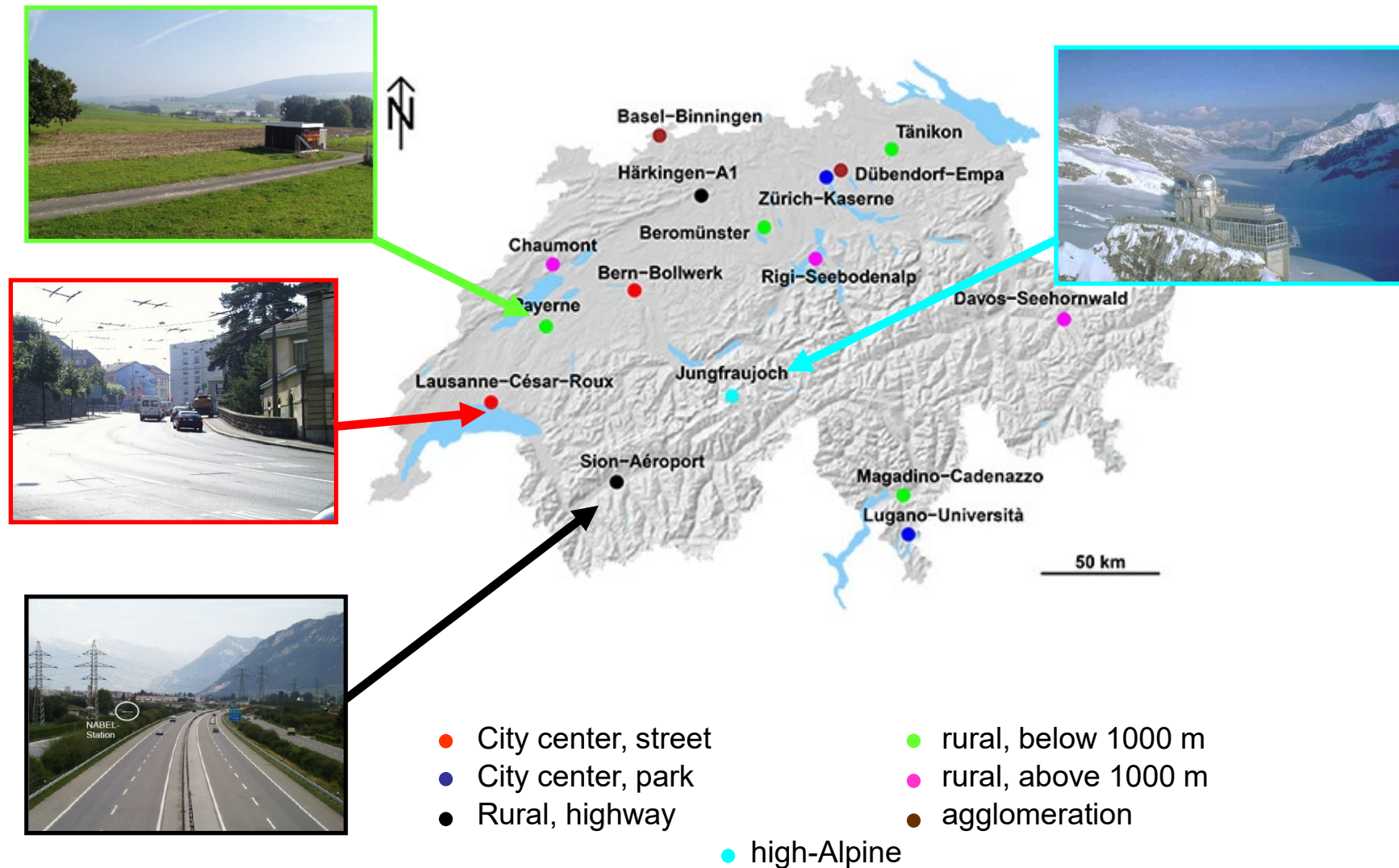
- stack emissions/tail-pipe emissions (in-situ/remote sensing)
- tunnel measurements/street canyon/highway
- flux measurements (ecosystems)

4. Exposition measurements

- Personal samplers, models including human behaviour (indoor vs. outdoor)

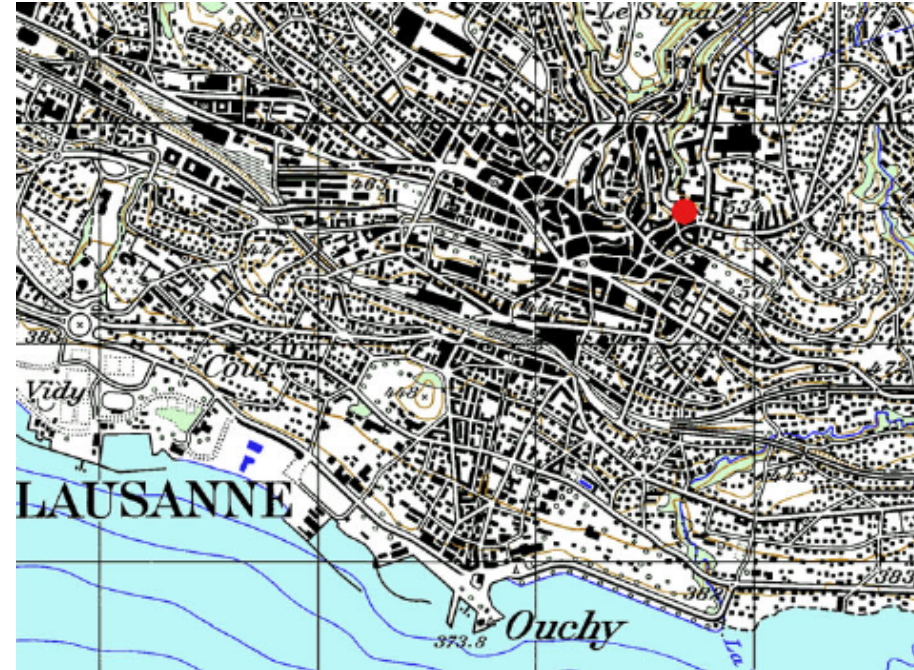
5. Scientific campaigns (various aims)

1. Swiss Network for Air Pollution Monitoring NABEL



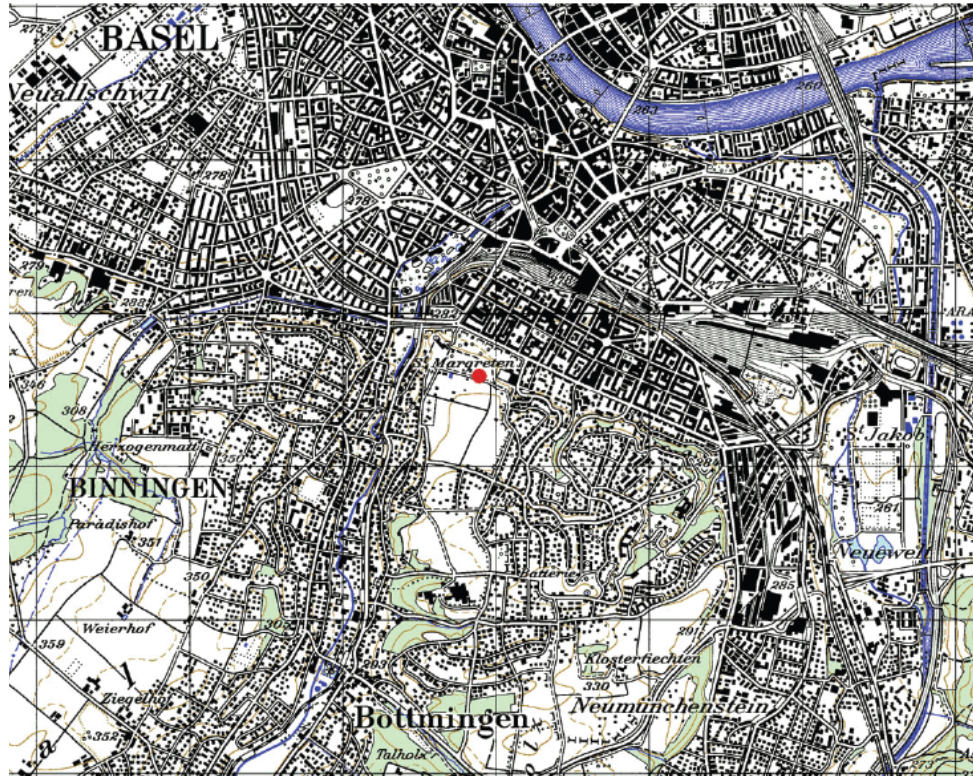
NABEL

Station Lausanne



NABEL

Station Basel-Binningen



Canton Vaud

<https://www.vd.ch/themes/environnement/air/qualite-de-lair/>
App: airCheck

Sunday, 06.03.22

Situation actuelle de la qualité de l'air

Pollution de l'air

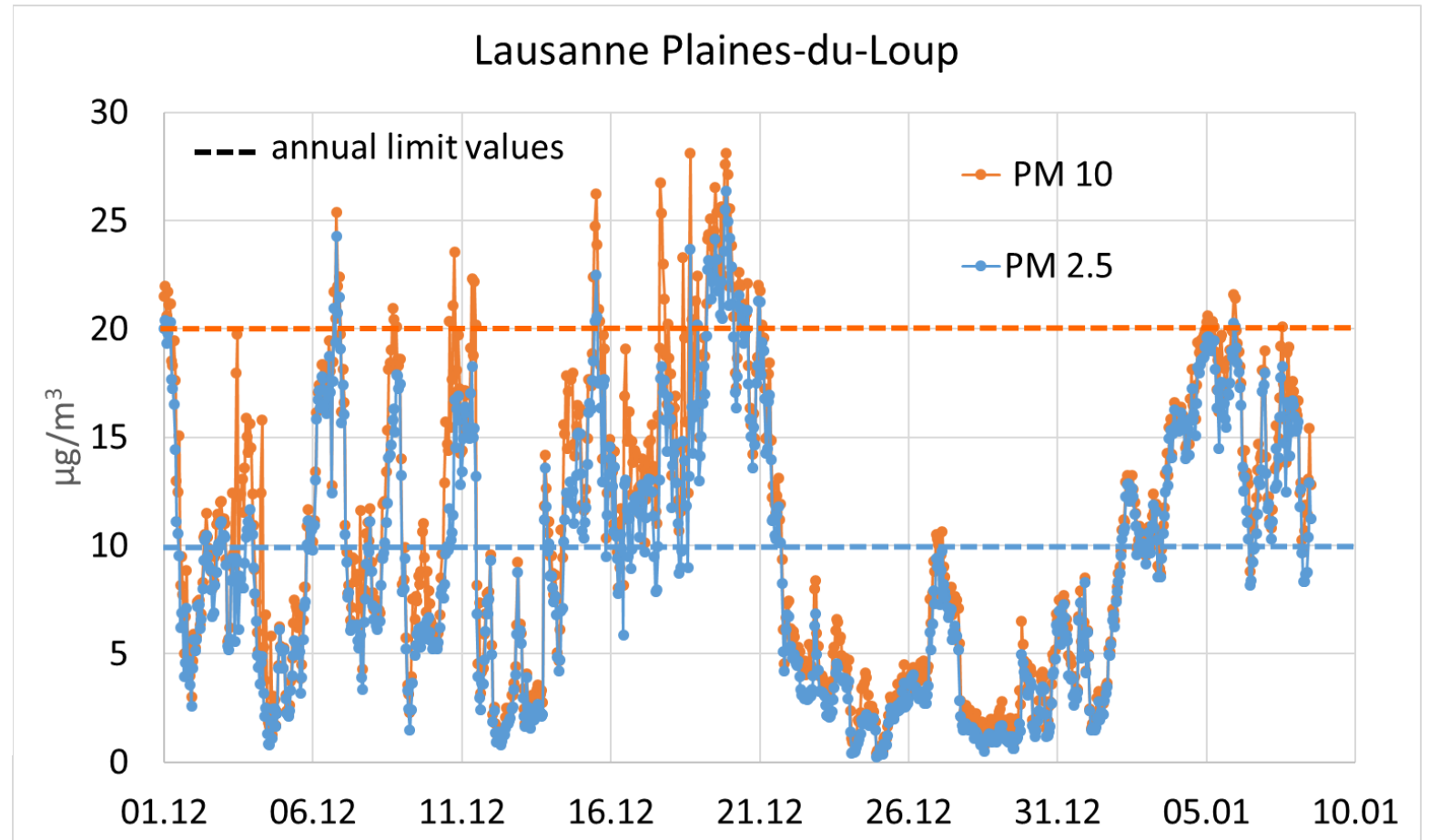
- Très élevée
- Elevée
- Marquée
- Significative
- Valeurs limites OPAir
- Modérée
- Faible

Donnée manquante



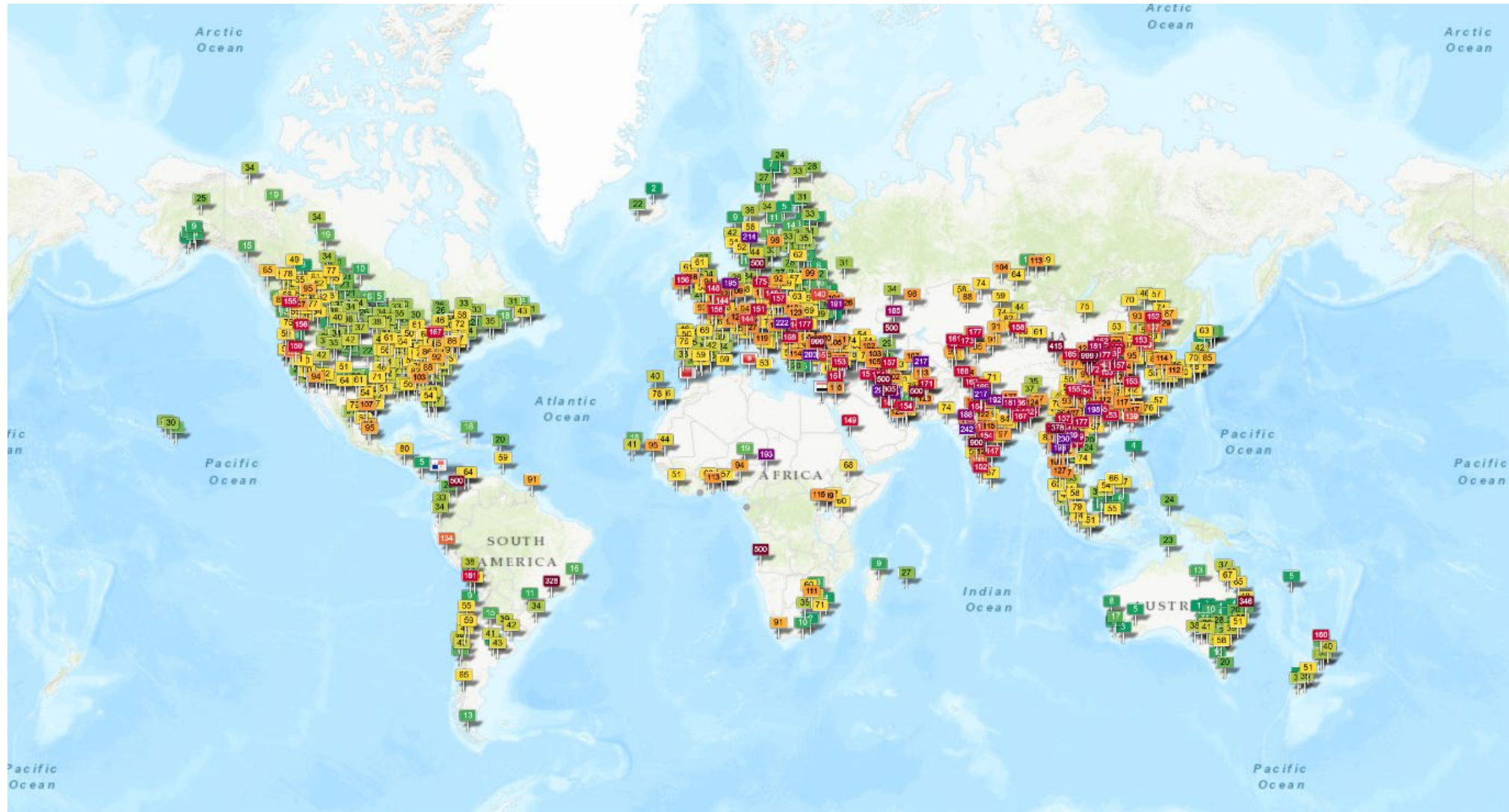
Indice de la pollution de l'air le 6 mars 2022 à 11h00

December 2021



Global air pollution resource
<http://aqicn.org/map/world>

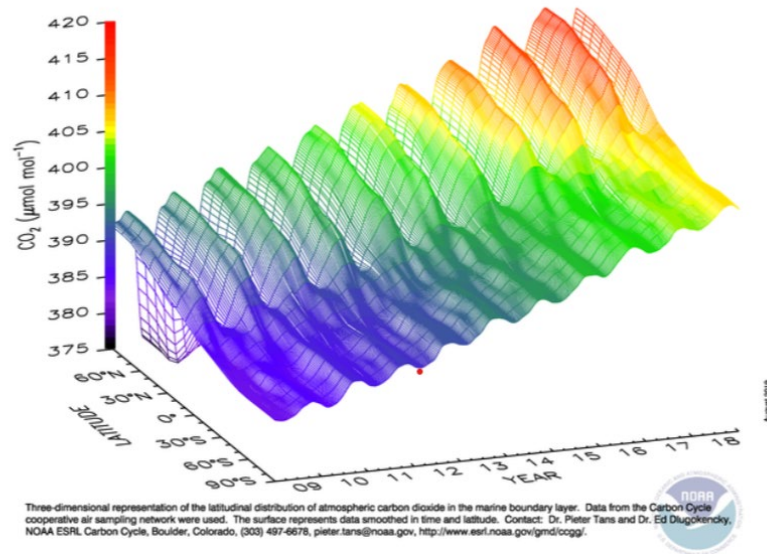
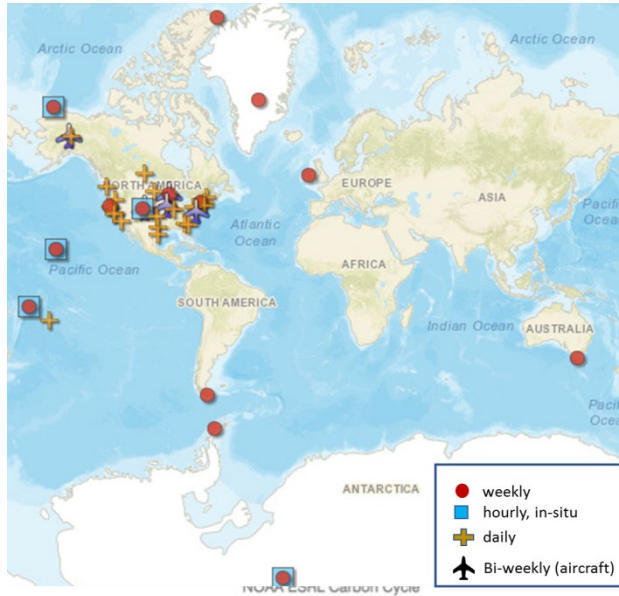
Air Quality Indices



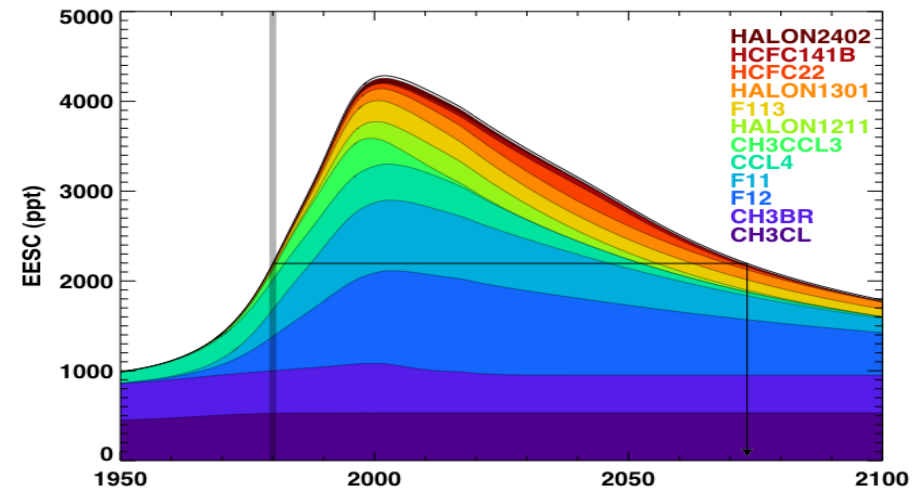
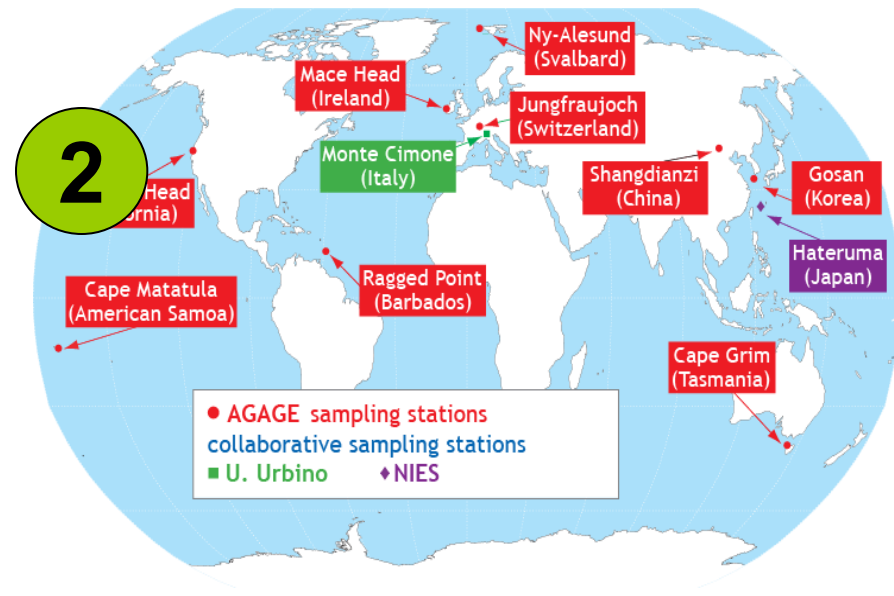
Global networks (ground-based, continuous)

1. NOAA flask measurement programme for CO₂/CH₄/N₂O/halocarbons
2. AGAGE network for halocarbons/CH₄/N₂O

1

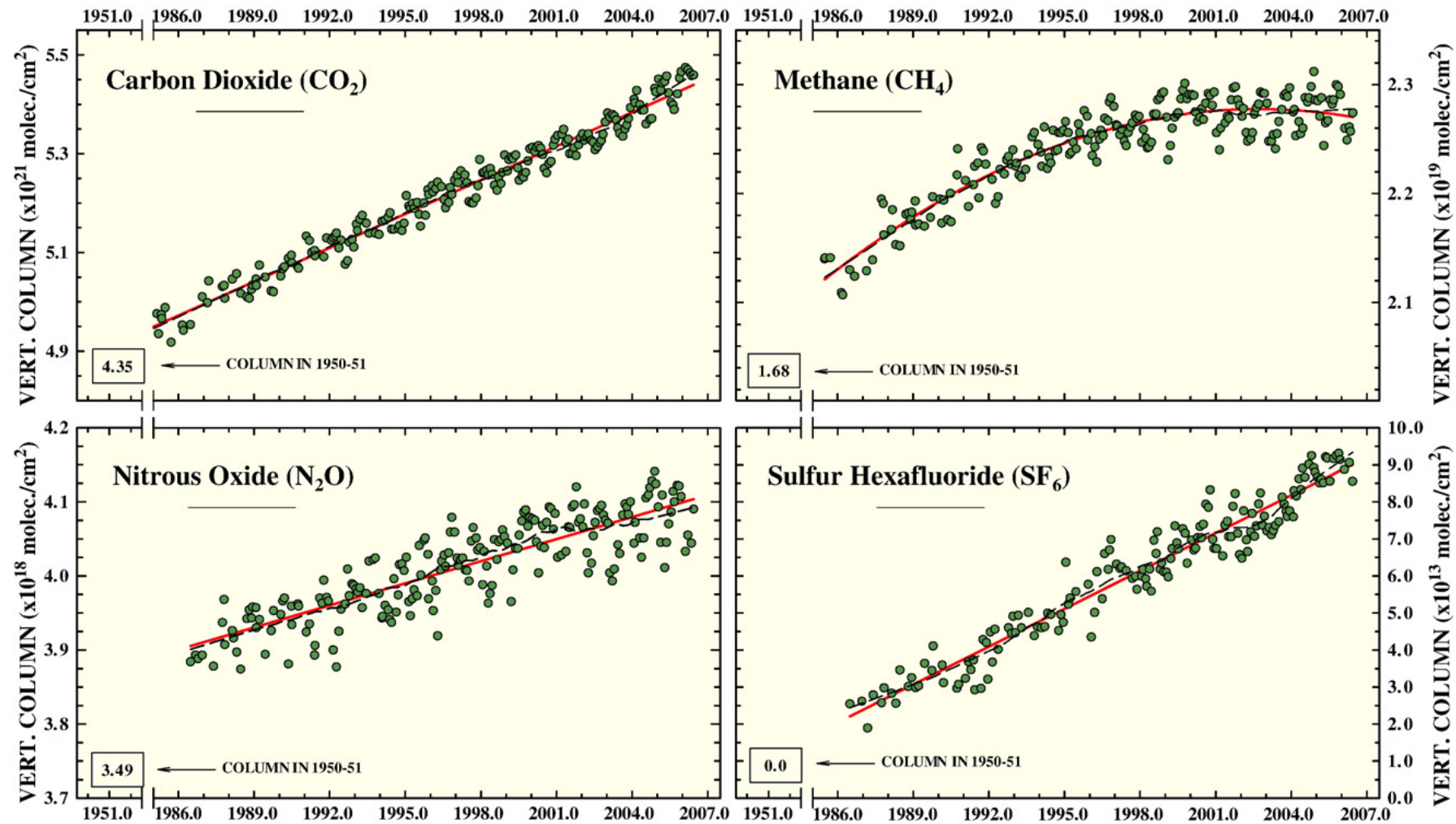


2

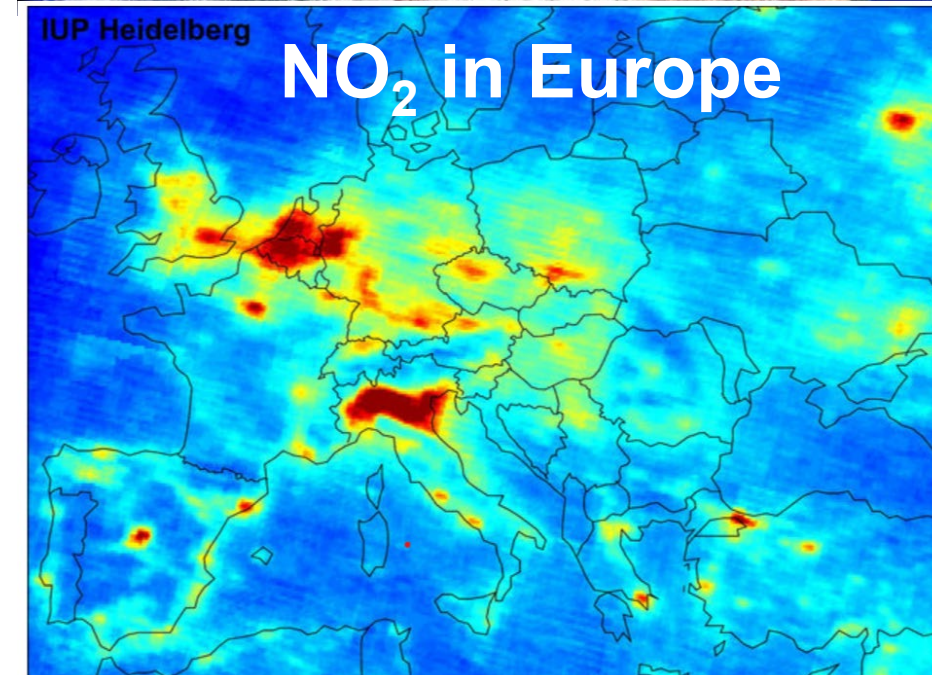
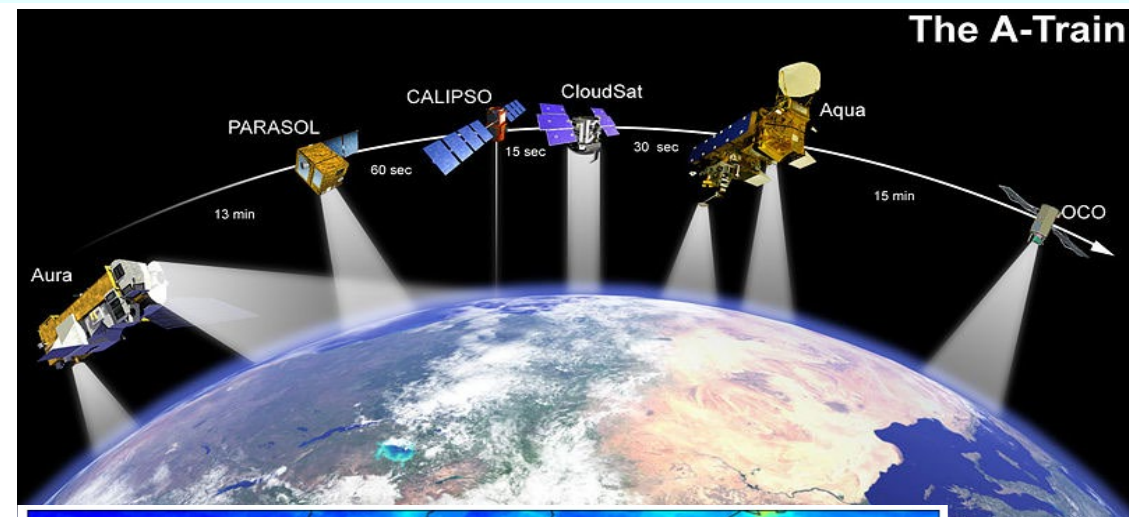


2. Global networks

NDACC
FTIR measurements at Jungfraujoch



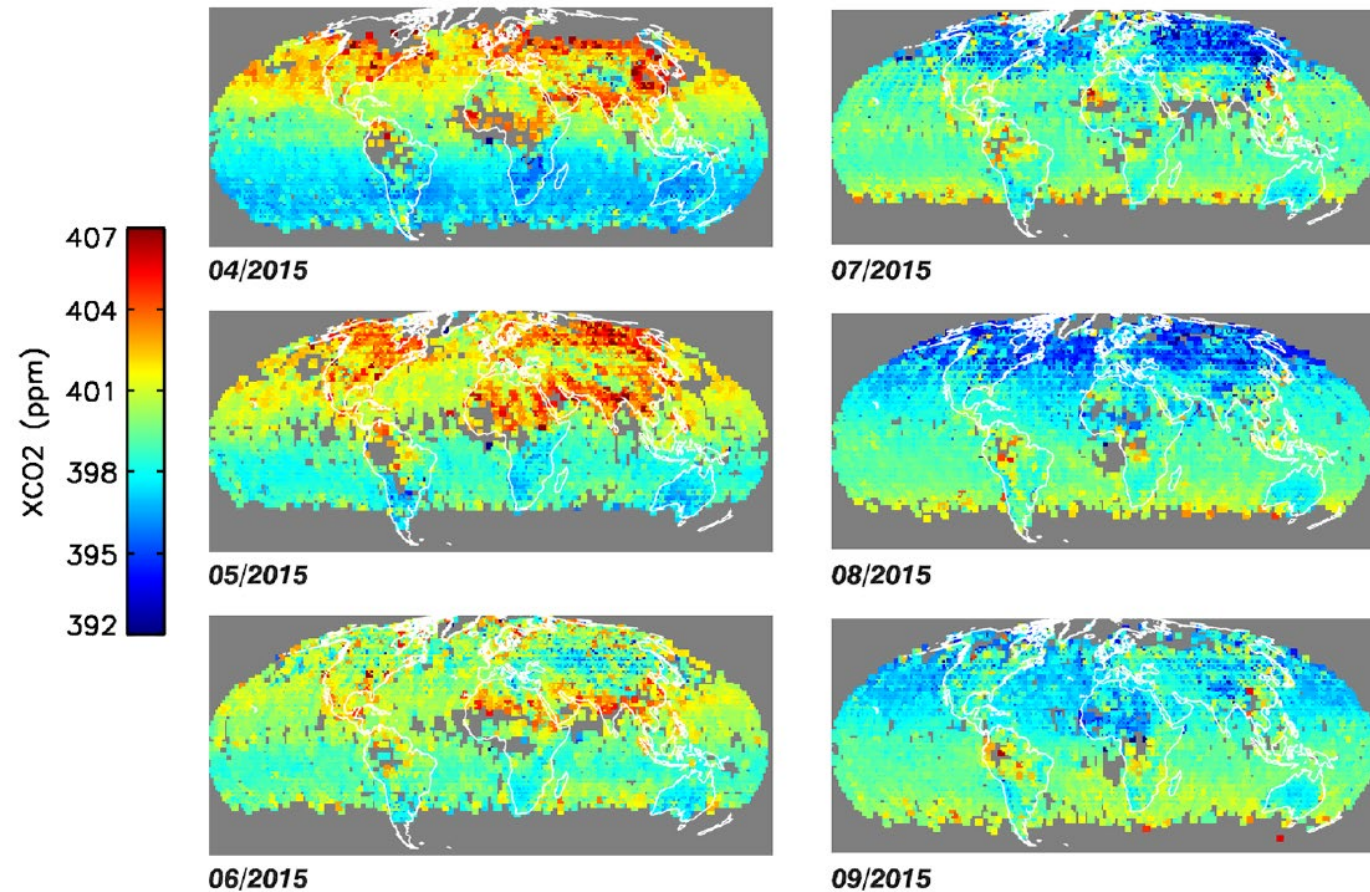
Remote sensing: Global networks satellites



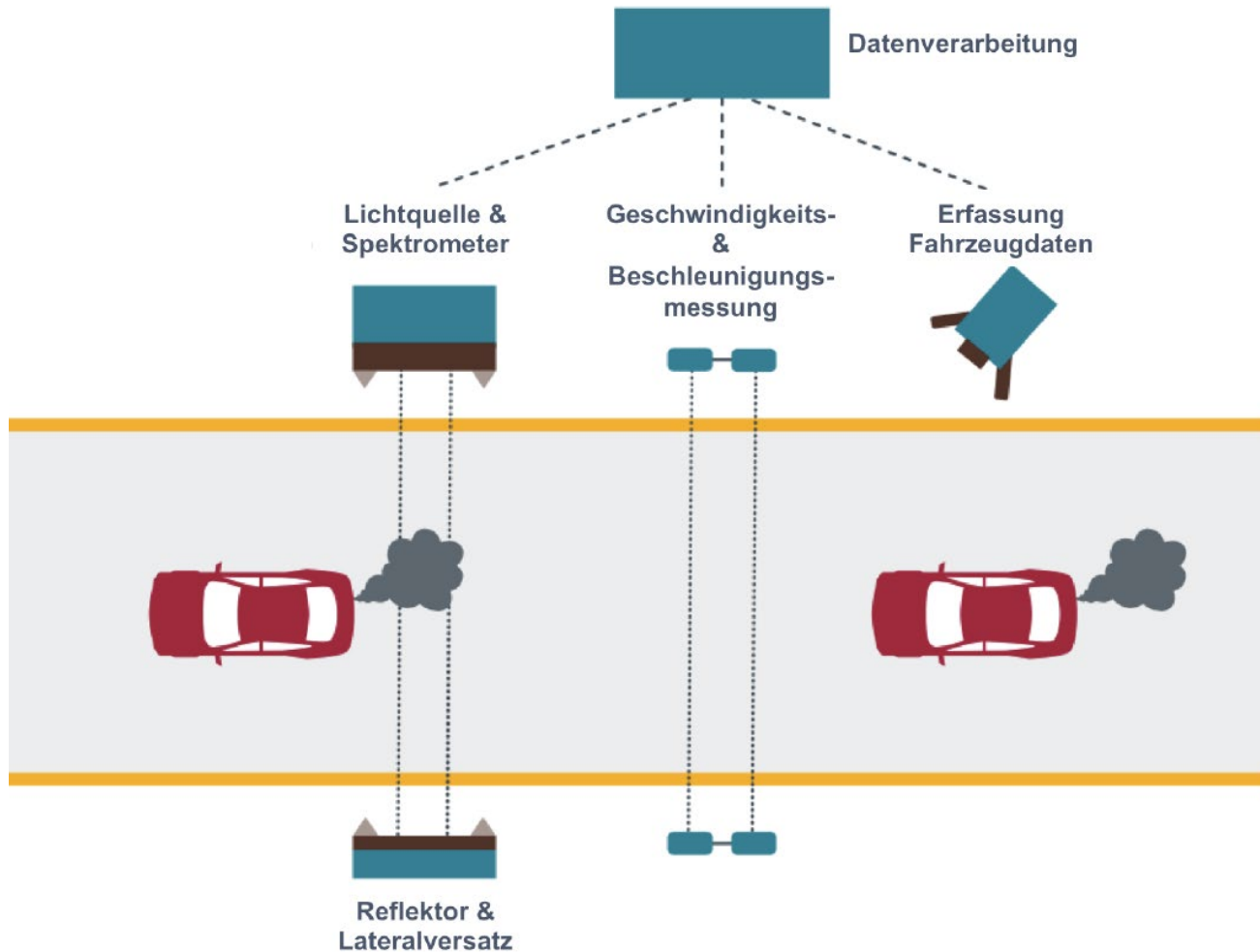
http://www.esa.int/Our_Activities/Observing_the_Earth/Envisat/

CO₂ from OCO-2

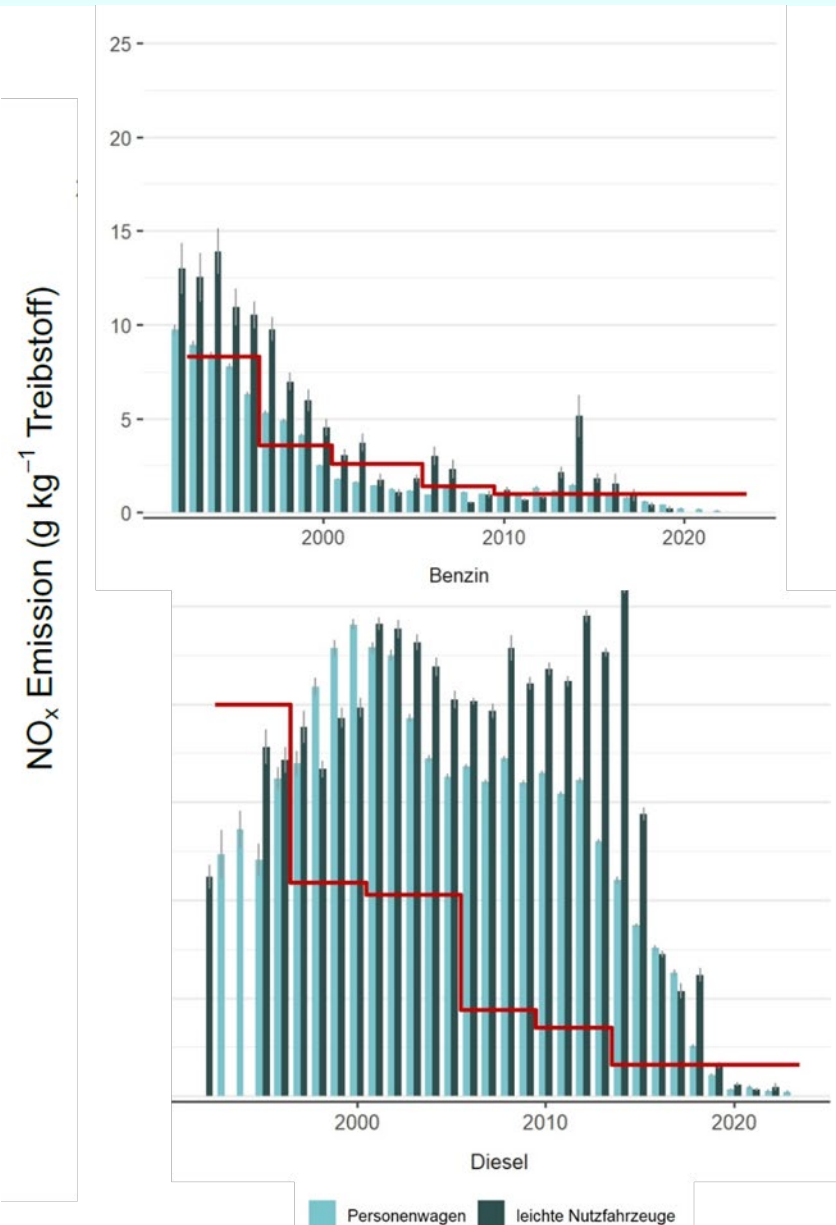
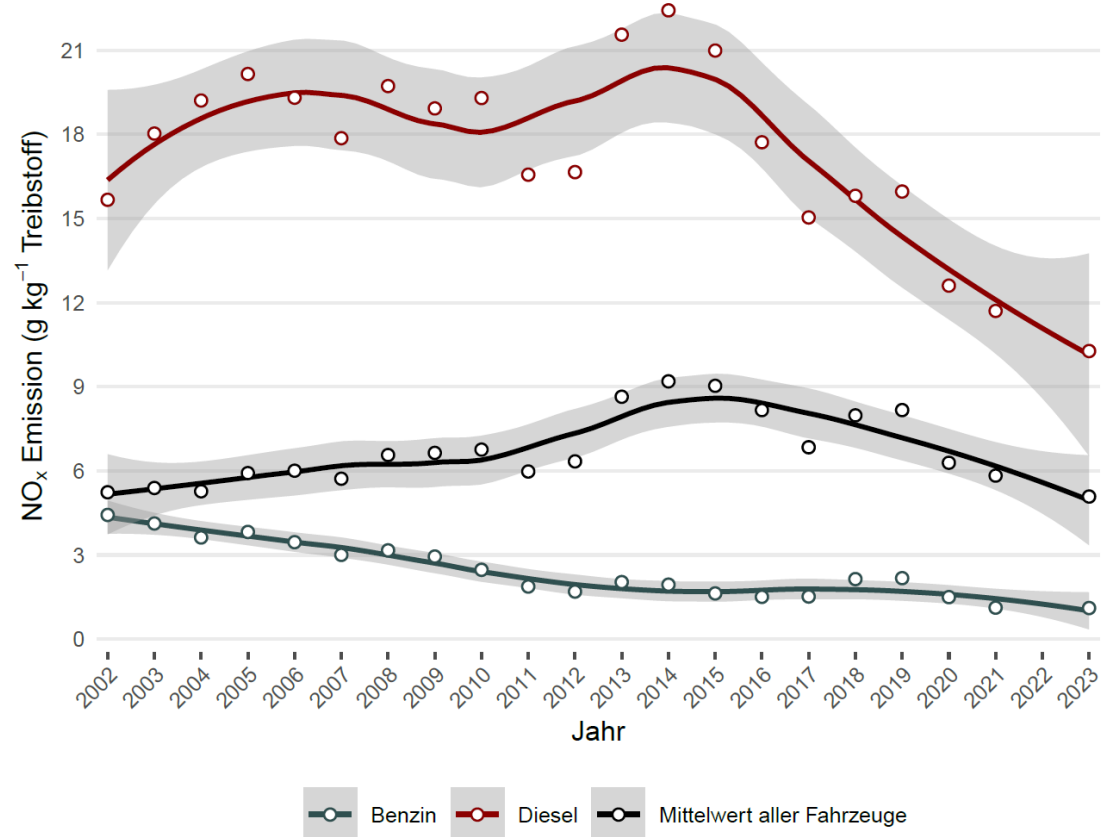
Eldering et al., 2016



3. Emission in-situ: Remote Sensing: NO_x Messungen Kanton Zürich



NO_x measurements Canton Zürich



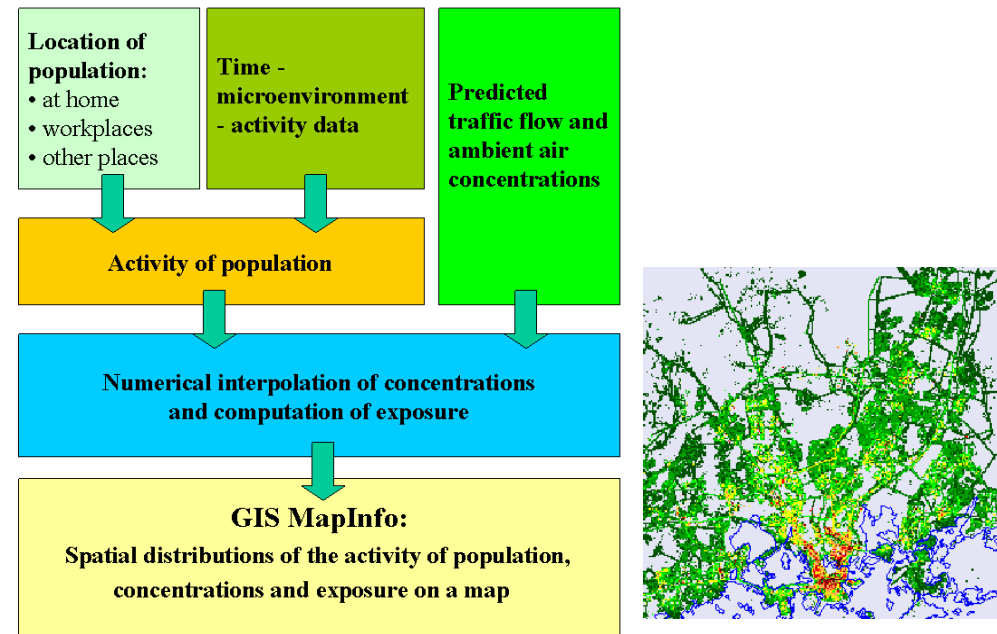
Mobile in-situ measurements for NO₂, particles, CO and CO₂ PEMS (Portable Emission Monitoring System)



4. Exposure measurements personal samplers



Personal passive samplers



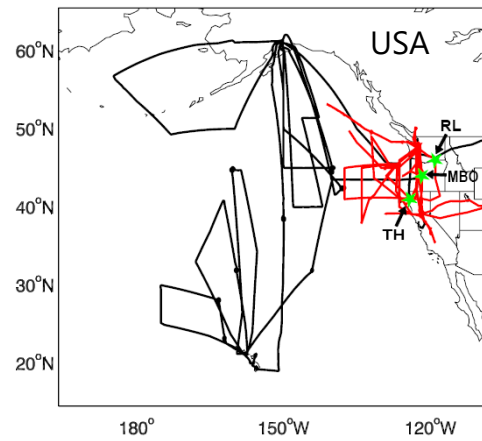
The predicted exposure of population to NO_2 in the afternoon (3 - 6 p.m.), in March 1996, ($\mu\text{g}/\text{m}^3 \cdot \text{persons}$).

Ref. Kousa, A., et al., 2002.

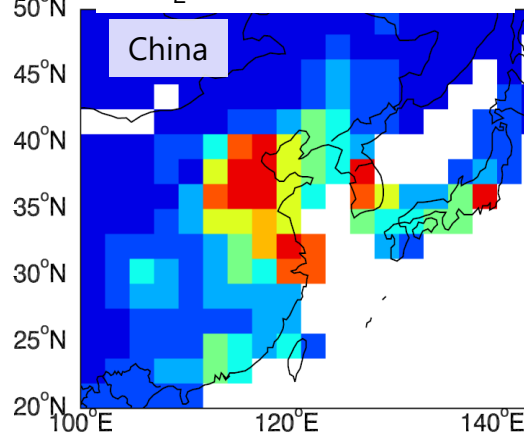
5. Scientific campaigns (in-situ, air planes, satellites, modelling)

INTEX-B campaign (2006)

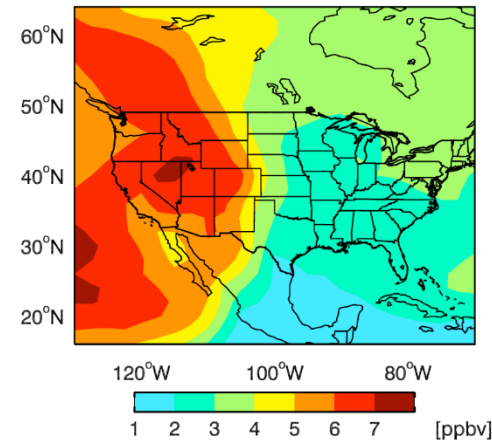
in-situ/air planes



NO₂ from OMI satellite



Chemical-Transport Model



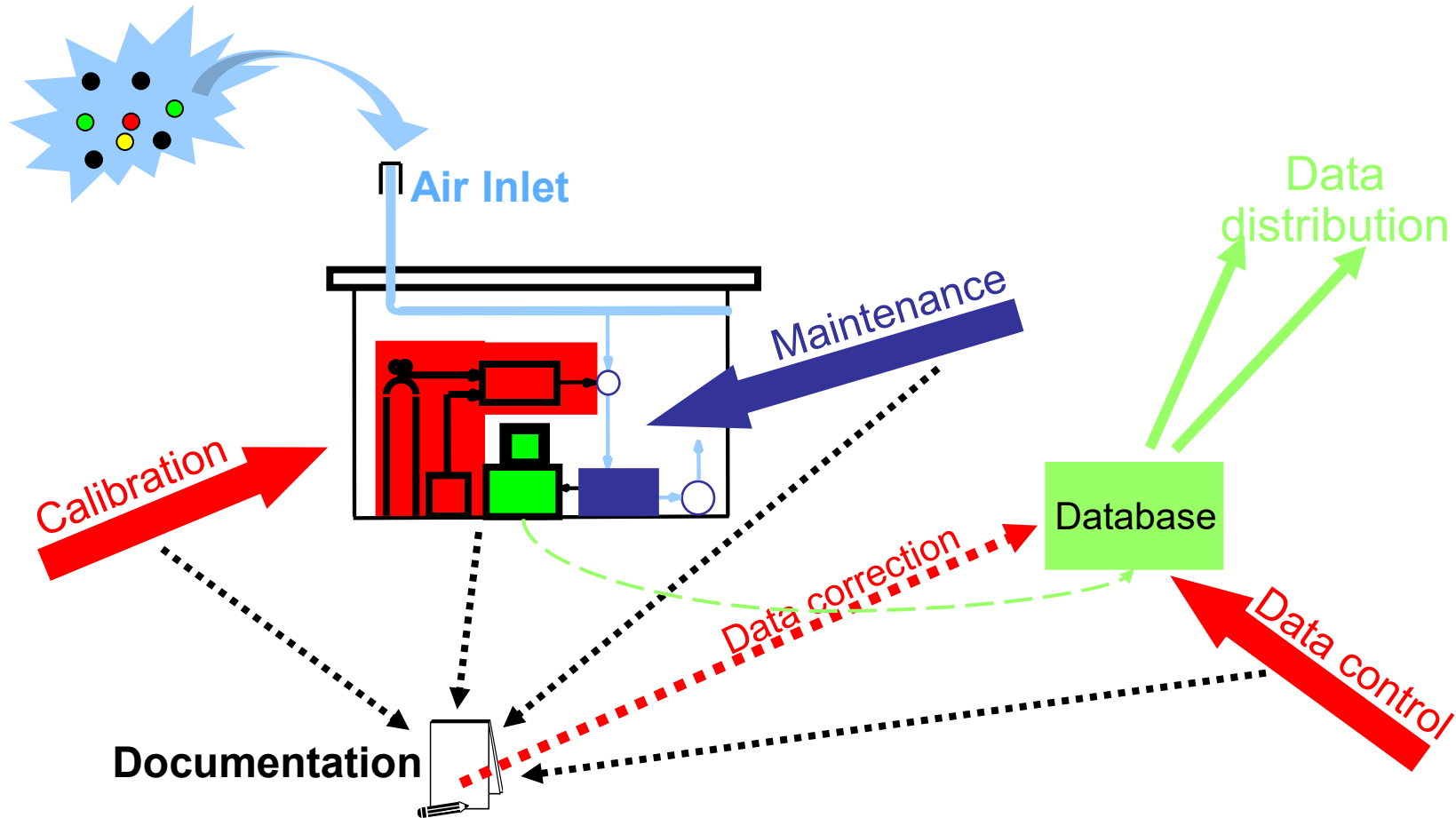
Asian O₃ enhancement

Zhang et al., ACP, 2008

quality assurance

example NABEL

Swiss Network for Air Pollutant Monitoring



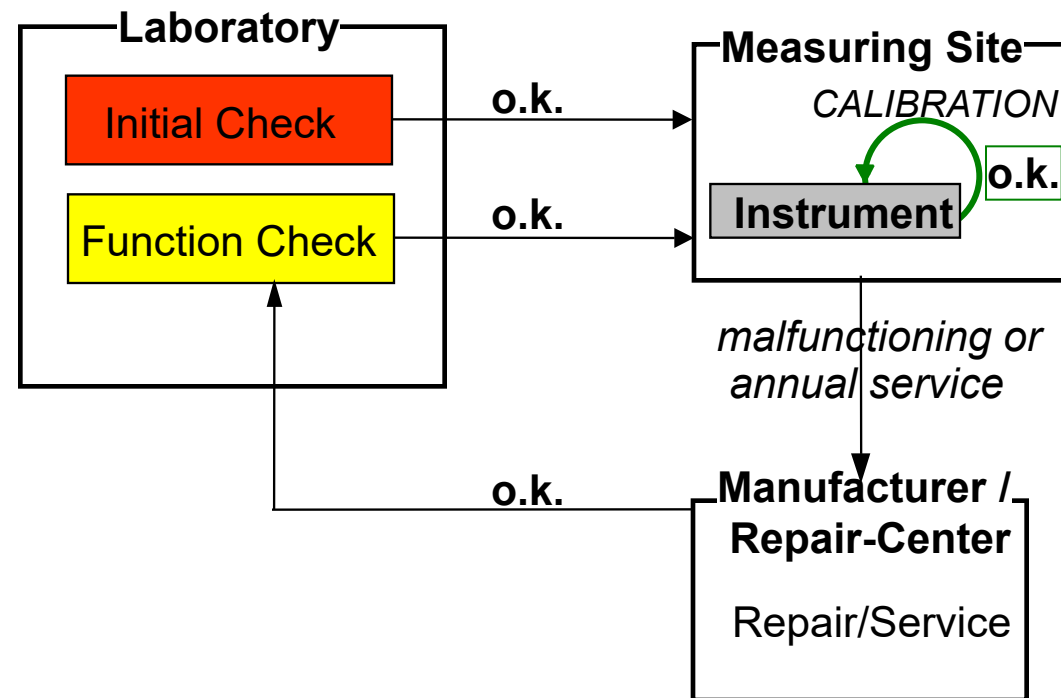
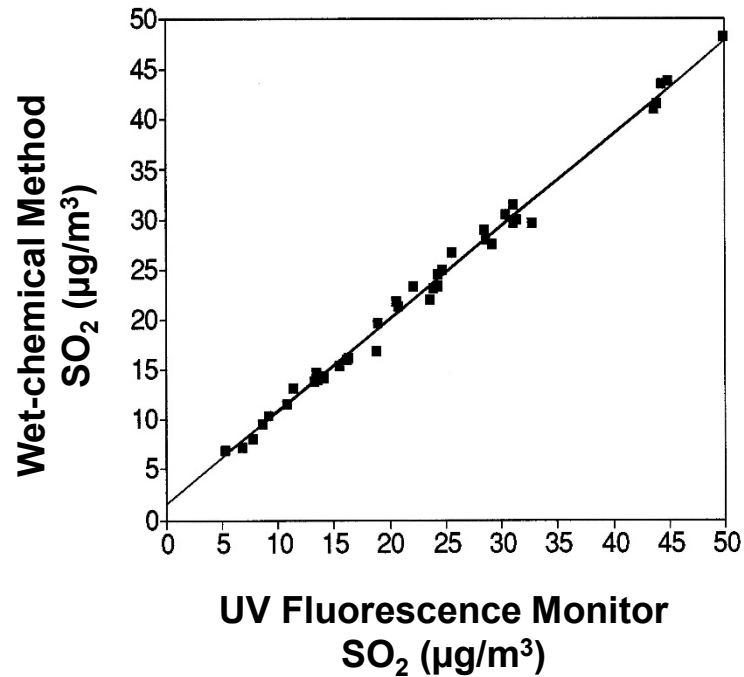
NABEL

Quality Assurance

- adequate measuring instrument
- air inlet/distribution device/pump
- data acquisition system
- traceability of calibration gases
(to international references)
- calibration device/dilution unit
- calibration gases/zero air
- periodic maintenance
- manual data control/documentation
- meteorological data

NABEL

Equipment - Initial checks



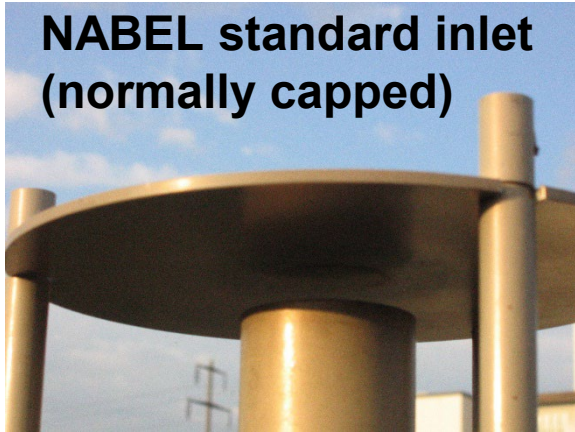
NABEL

Quality Assurance

- adequate measuring instrument
- air inlet/distribution device/pump
- data acquisition system
- traceability of calibration gases (to international references)
- calibration device/dilution unit
- calibration gases/zero air
- periodic maintenance
- manual data control/documentation
- meteorological data

NABEL air inlet

**NABEL standard inlet
(normally capped)**



other solutions



Alert, Canada



Mt. Kenia, Kenia



Ny-Alesund, Spitsbergen



Mace Head, Ireland

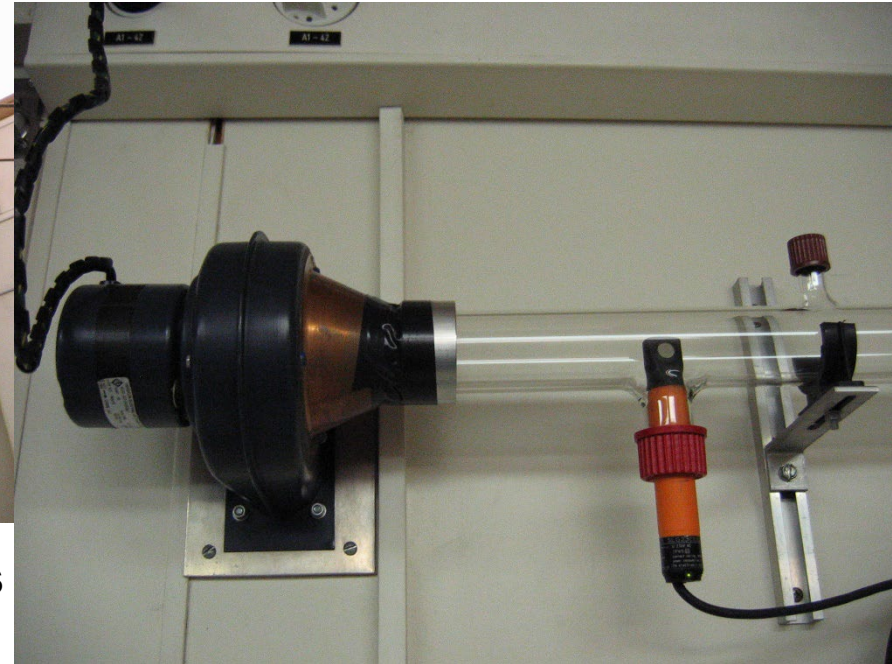
NABEL

distribution device/pump



Glass manifold/tubing to instruments

Flow: 100 liter/min



Pump/air flow meter

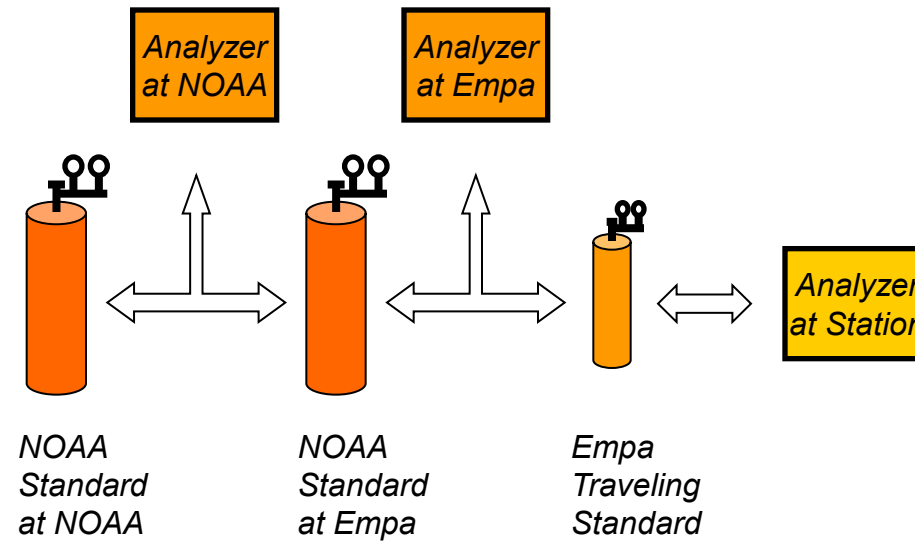
NABEL

Equipment - Quality Assurance

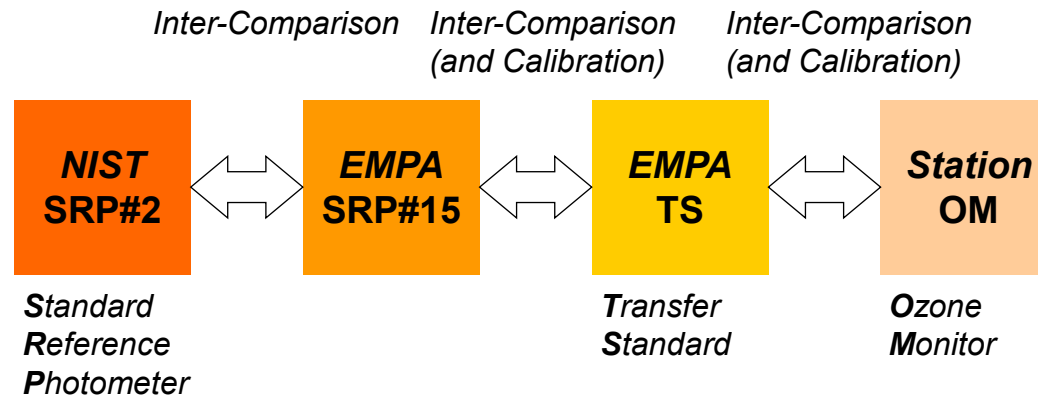
- adequate measuring instrument
- air inlet/distribution device/pump
- data acquisition system
- traceability of calibration gases
(to international references)
- periodic maintenance
- calibration device/dilution unit
- calibration gases/zero air
- manual data control/documentation
- meteorological data

NABEL international traceability

Stable gases
(CO, NO₂, CO, SO₂)

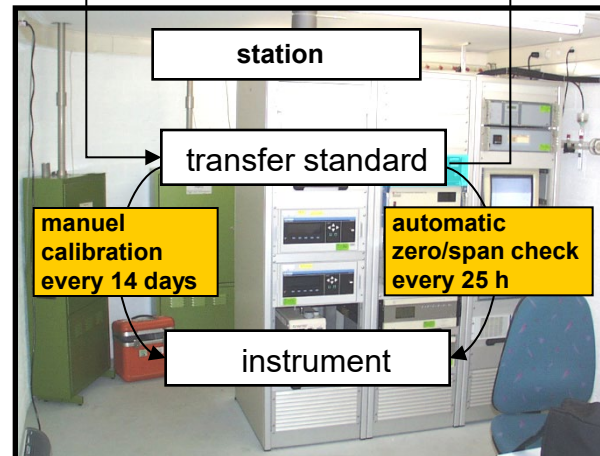
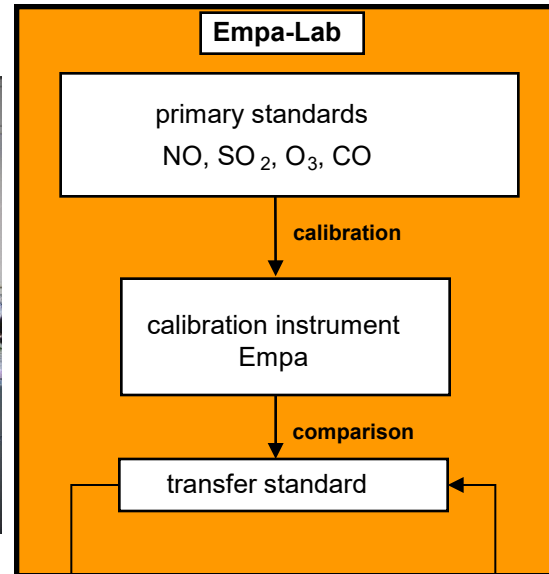
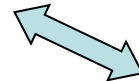


Non-stable gases
O₃

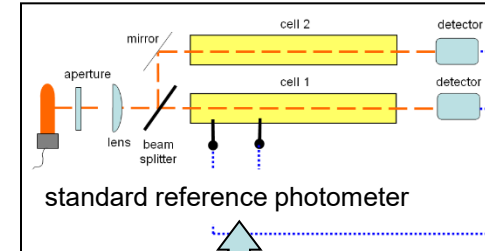


NABEL traceability

stable gases



ozone



NABEL

Quality Assurance - periodic maintenance

- daily: check data availability/instrument status
calibration zero / span checks
- 2-weekly: general preventive maintenance
manual calibration
check impact on data series
- annualy: final data check, submission to BAFU
instrument check/replacement
update standard operation procedures

NABEL

Quality Assurance - Documentation

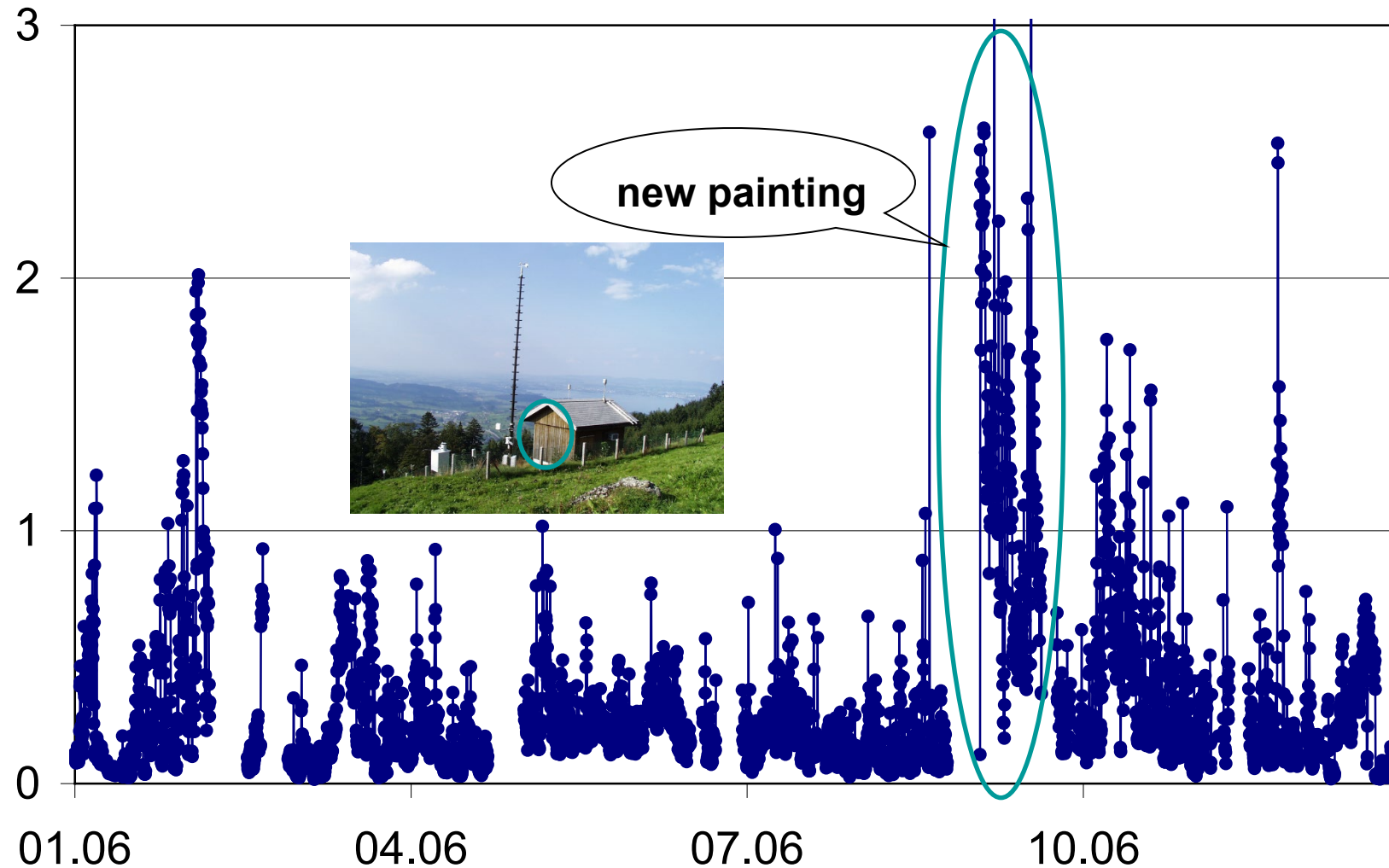
- instrument conditions (log book)
- maintenance procedures (SOP)
- calibration procedures (SOP)
- calibration data (zero/span)
- data control procedures
- data correction functions

NABEL

Quality Assurance - Data control

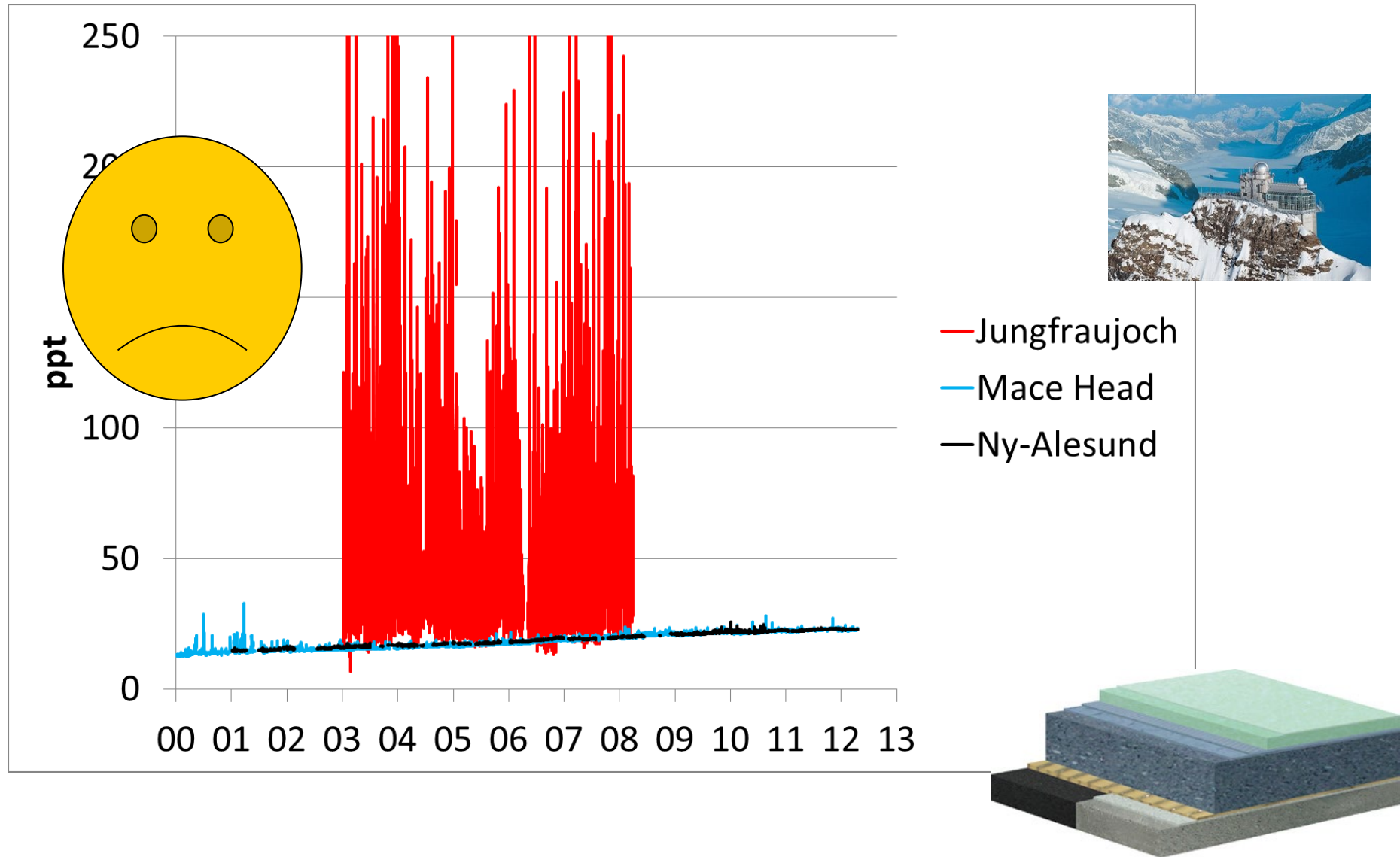
- apply calibration function
- include consequences of log book remarks
- internal data control procedure:
 - *check with plausibility limits/meteorology*
 - *look at daily and annual behaviour*
 - *check variability*
 - *compare with other compound*
 - *compare similar stations*
 - *check data before and after gap*
- possible external data control
- final data release

Data quality checks contaminations Example 1: toluene at Rigi (NABEL)



Quality Assurance - Data control

Example 2:
HCFC-142b at Jungfraujoch, Switzerland



Quality Assurance - Data control

Example 2:
HCFC-142b at Jungfraujoeh, Switzerland

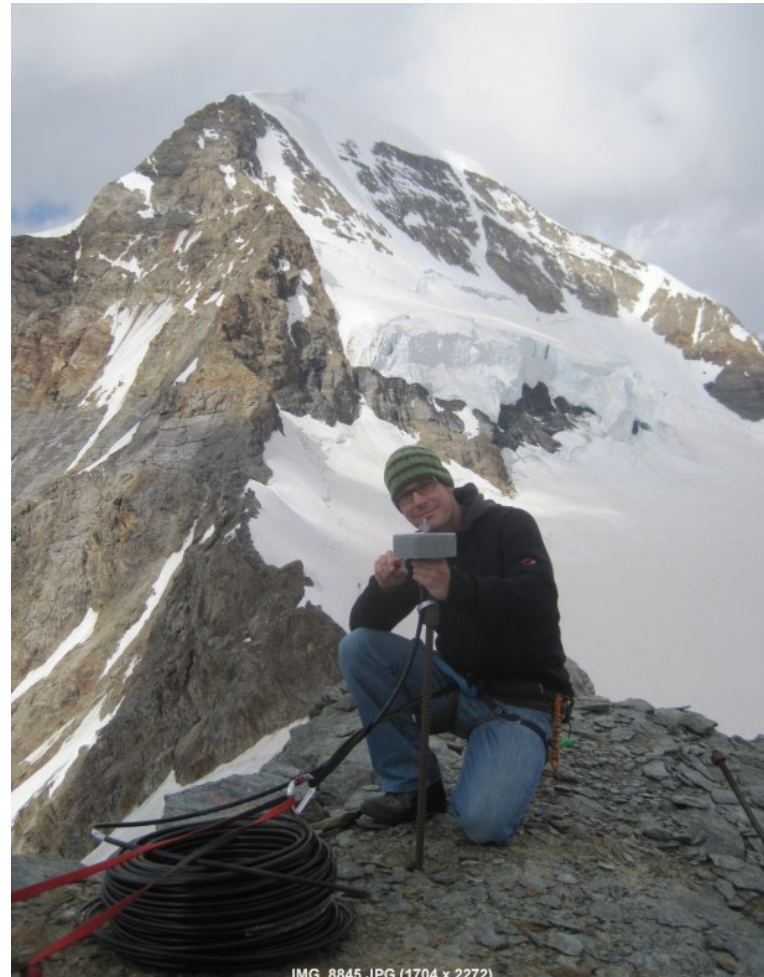
relocation of the inlet: August 2012



Quality Assurance - Data control

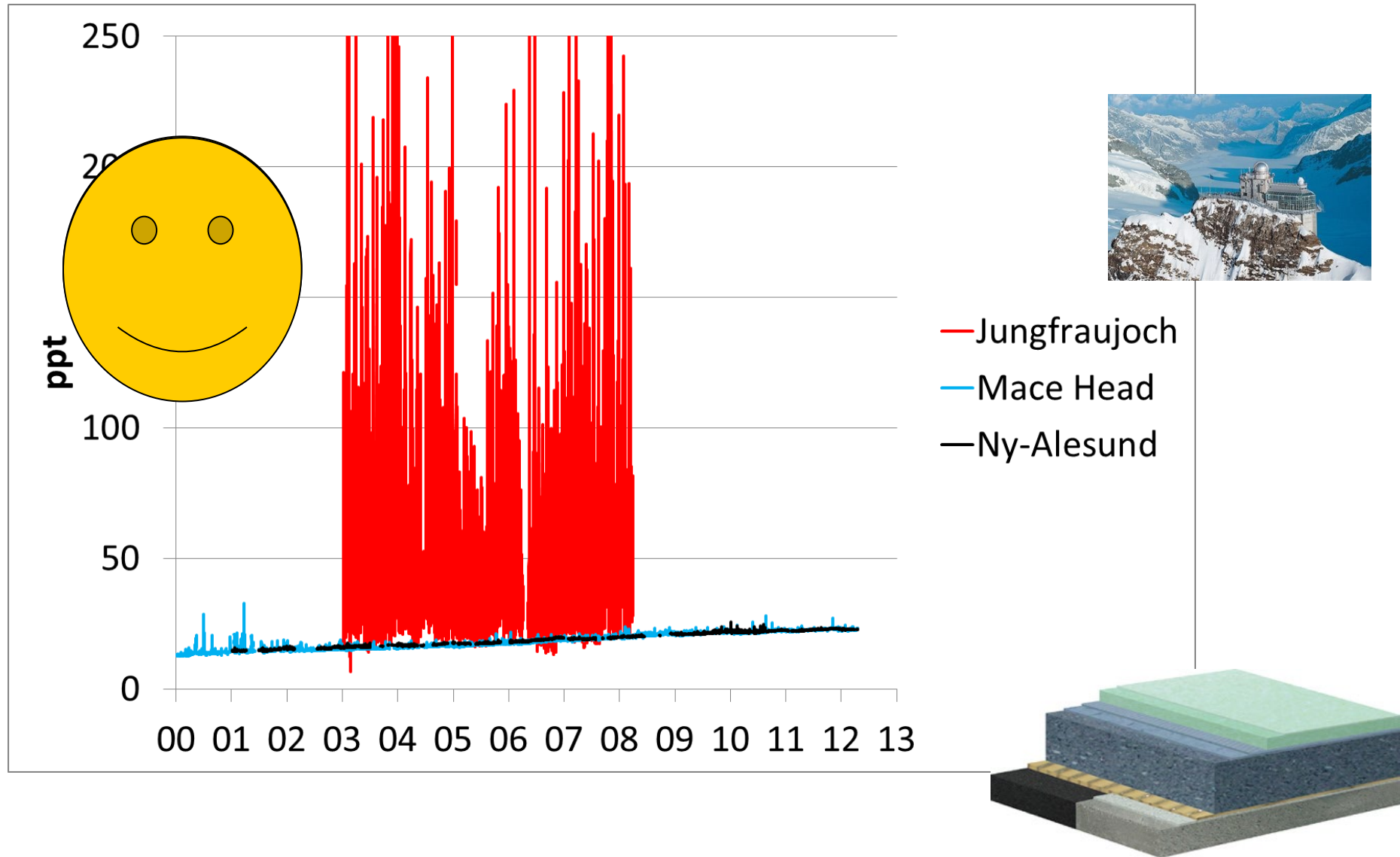
Example 2:
HCFC-142b at Jungfrauoch, Switzerland

relocation of the inlet: August 2012

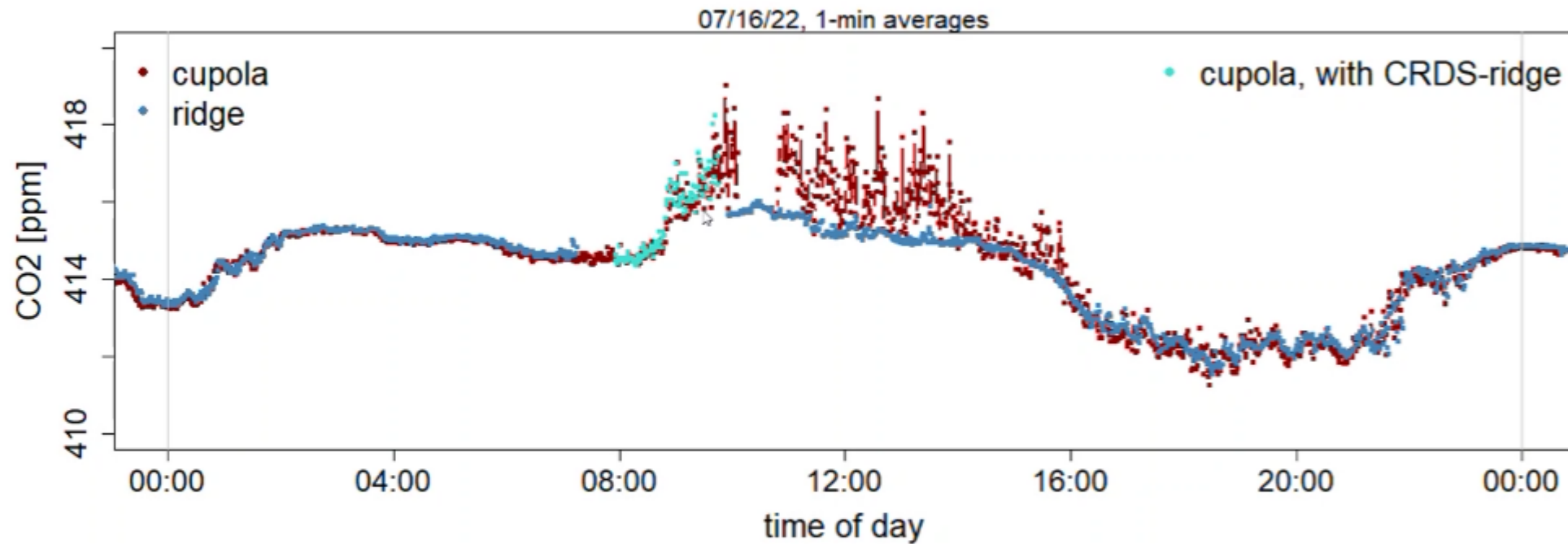


Quality Assurance - Data control

Example 2:
HCFC-142b at Jungfrauoch, Switzerland



CO₂ from the ridge and from the roof

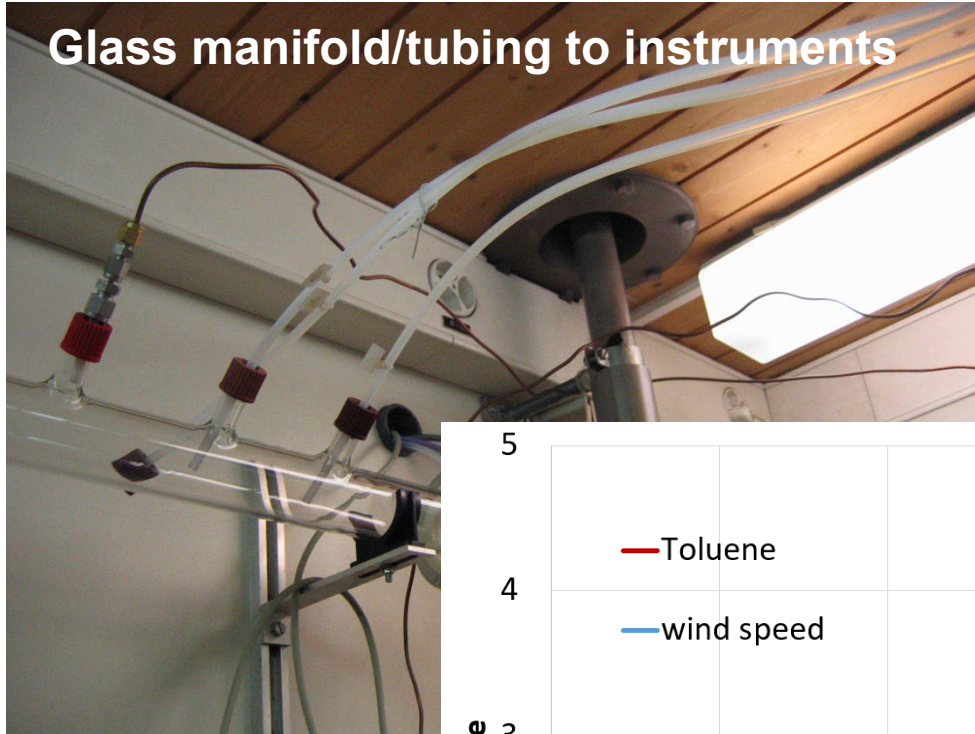


Time series of 1-min averages of CO₂ mole fractions from the cupola (red) and the ridge inlet (blue) for 16 July 2022. Light blue data (between 8 AM and 9:45 AM) show data from the cupola inlet sampled with CRDS-ridge.

NABEL

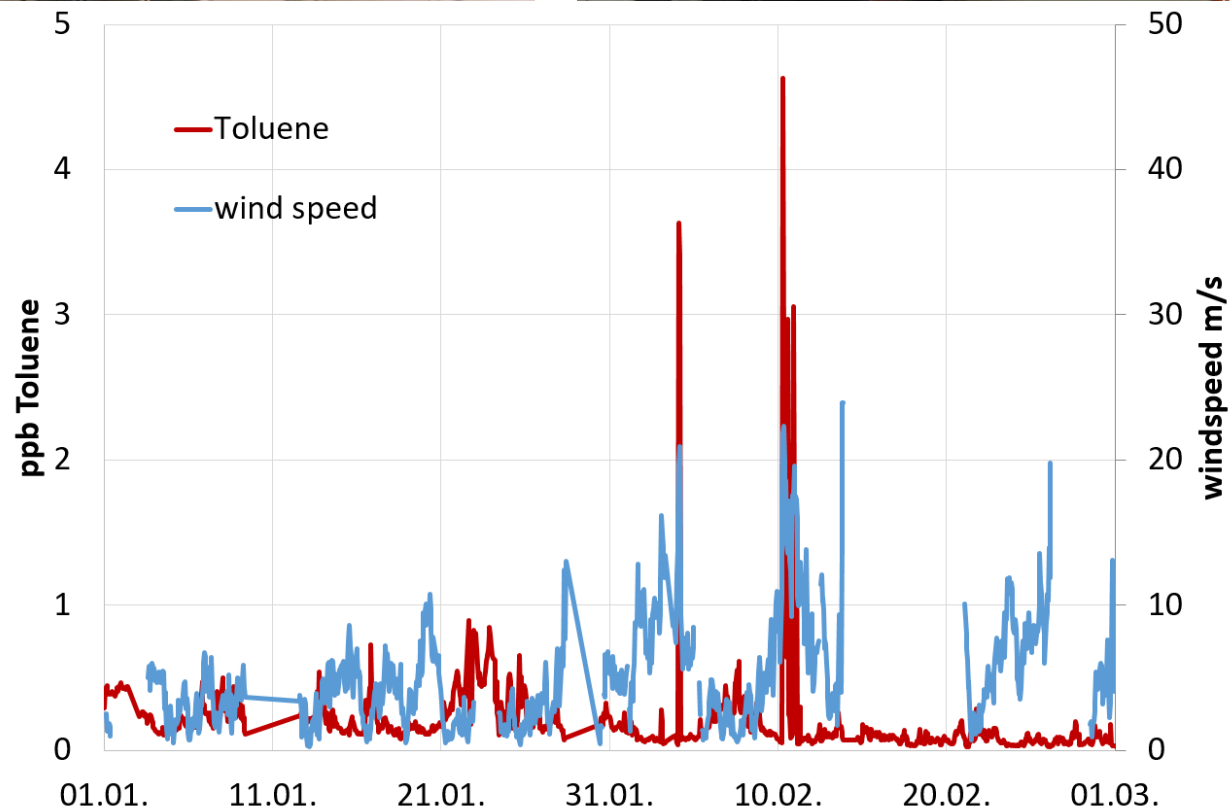
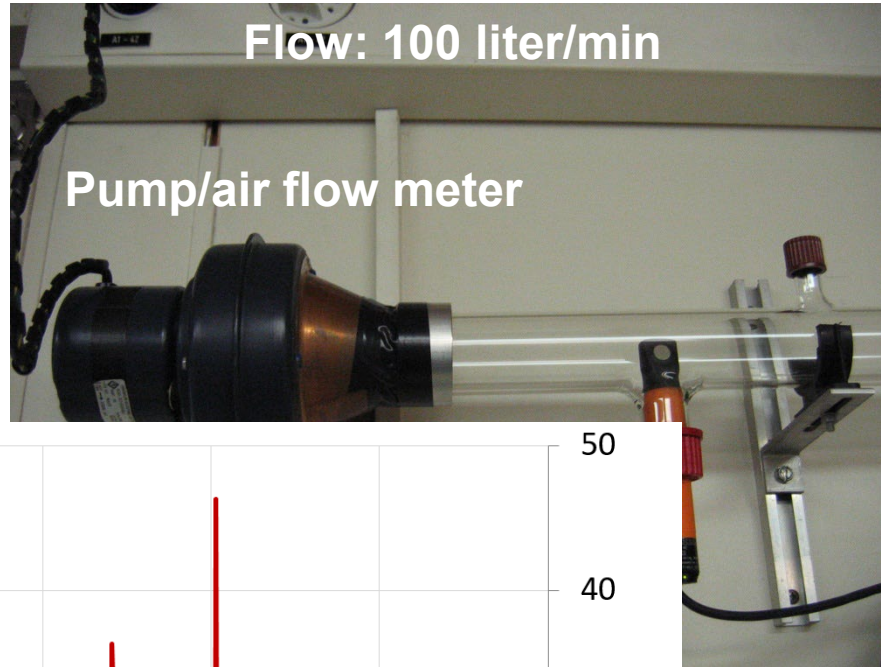
distribution device/pump

Glass manifold/tubing to instruments



Flow: 100 liter/min

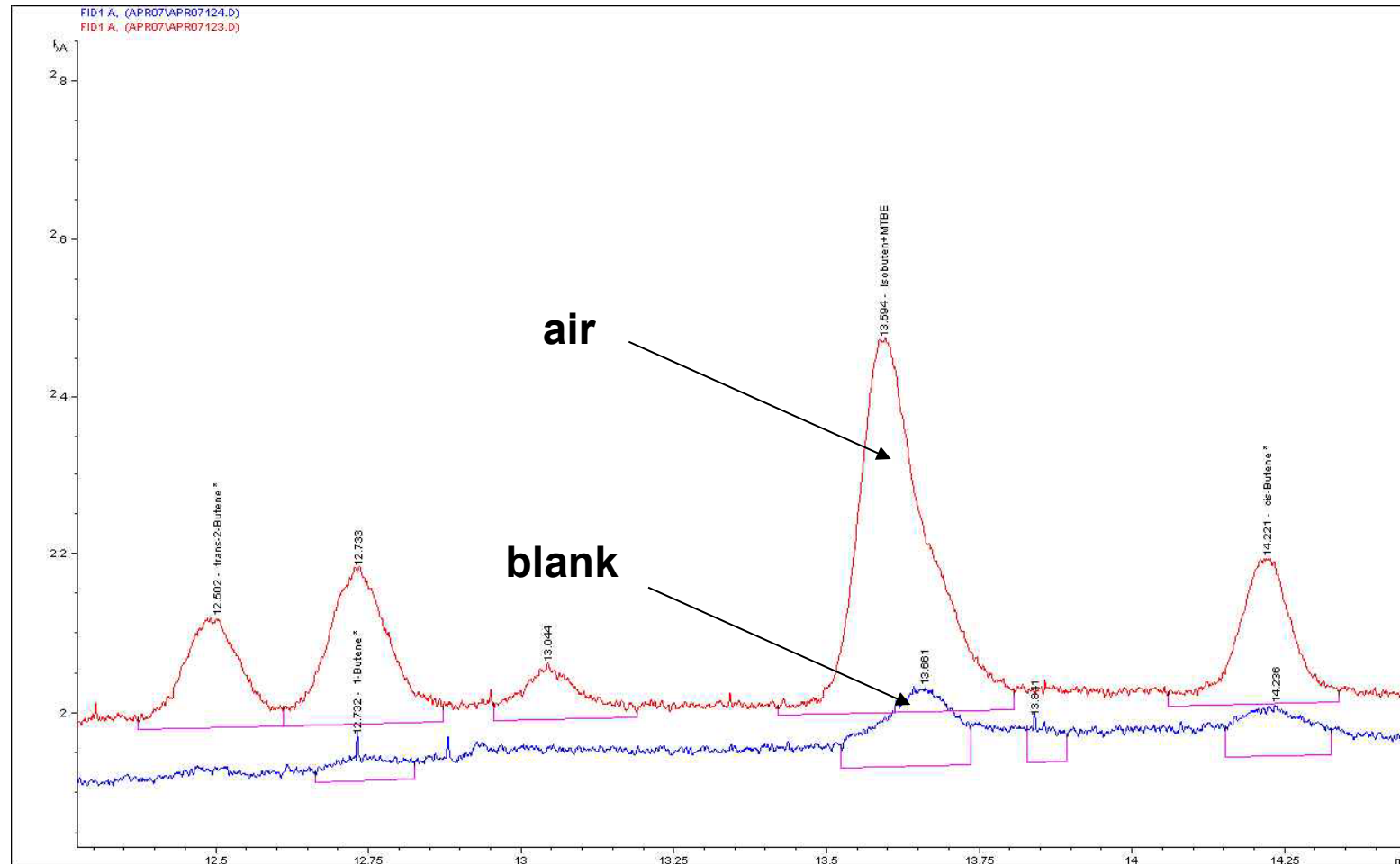
Pump/air flow meter



Quality Assurance - Data control

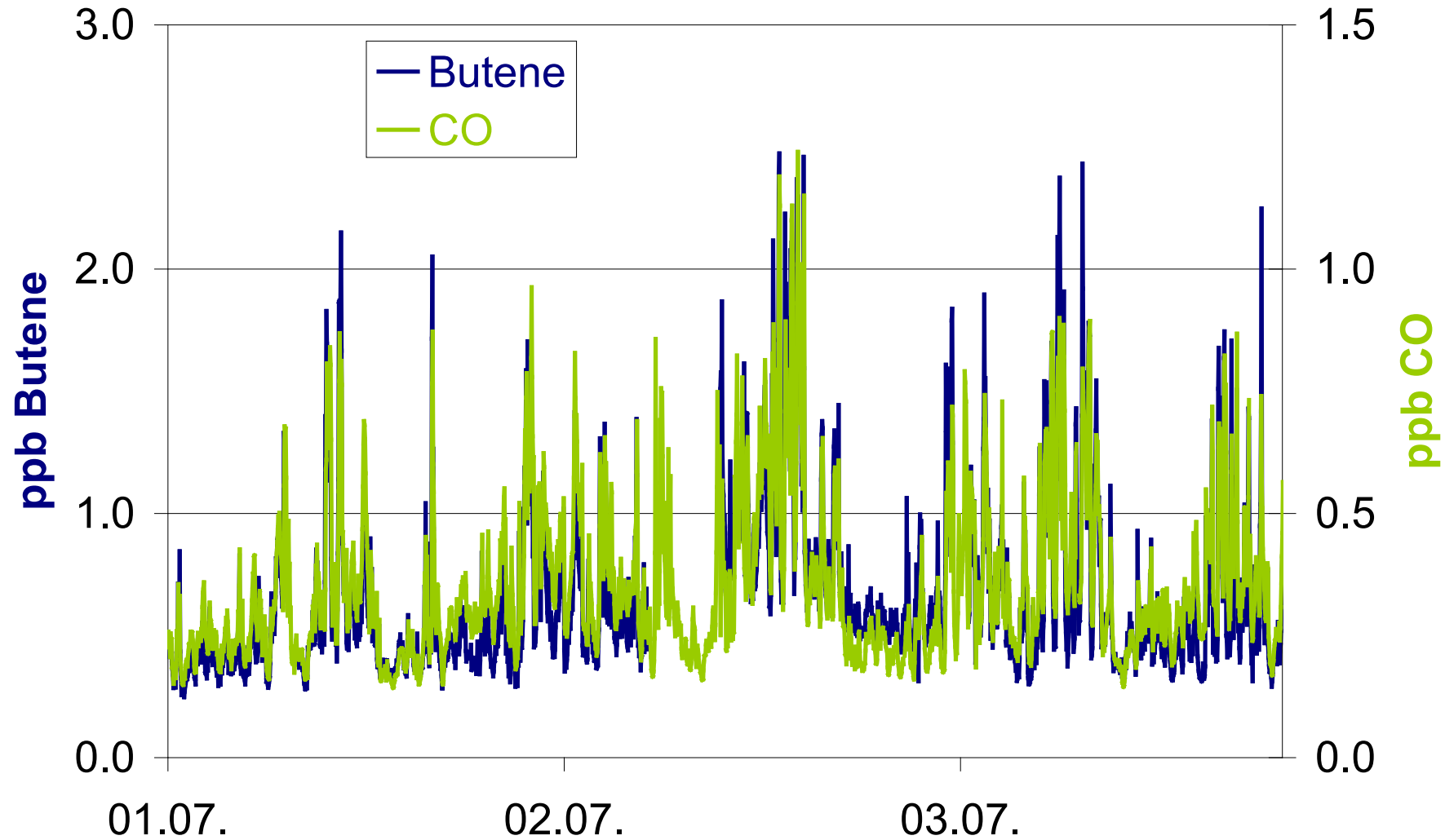
Blanks

Blank of butene (C_4) from the Nafion Dryer: 0.2 ppb



Quality Assurance - Data control

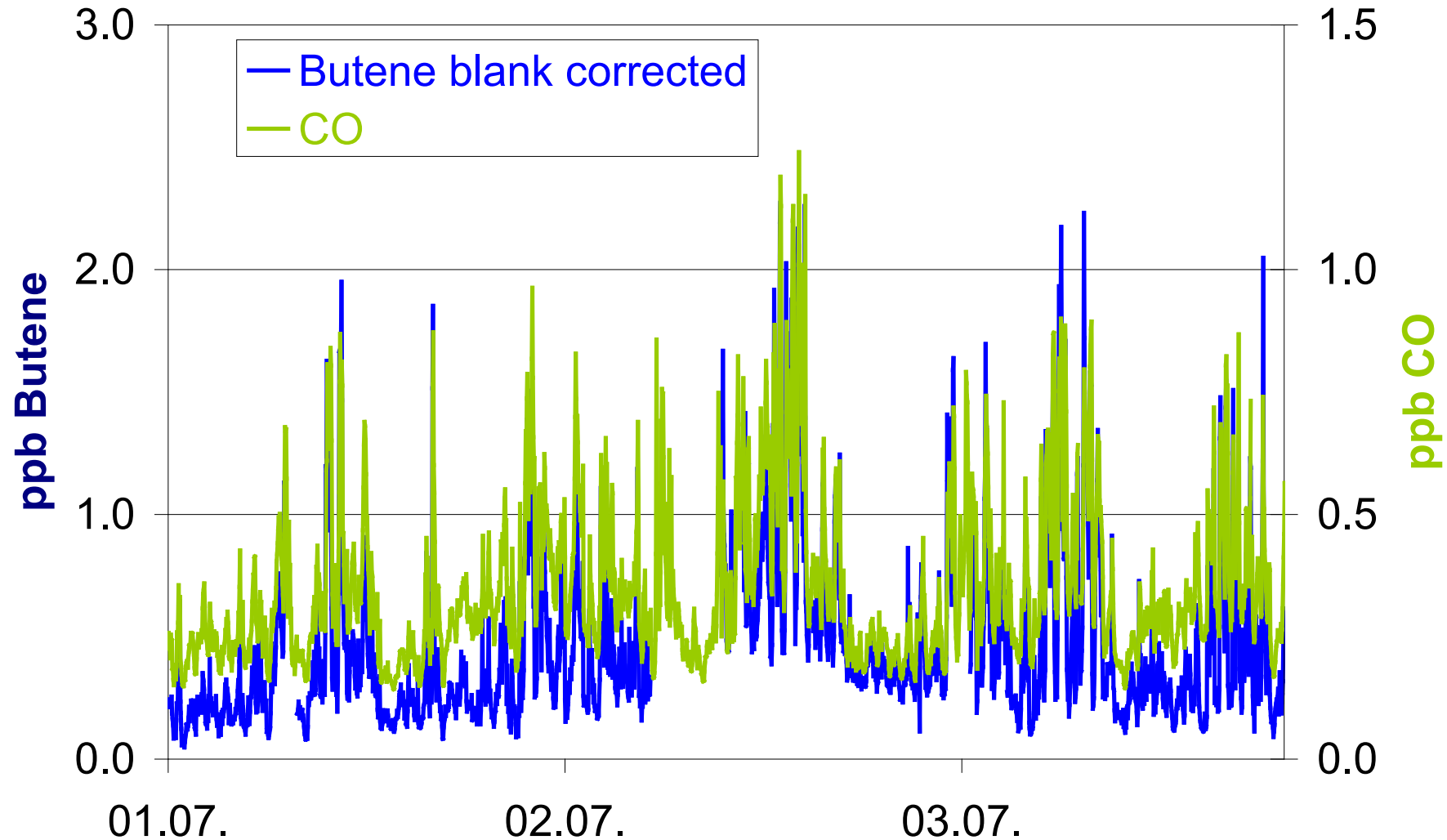
Blanks



Quality Assurance - Data control

Blanks

Blank of butene (C_4) from the Nafion Dryer: 0.2 ppb



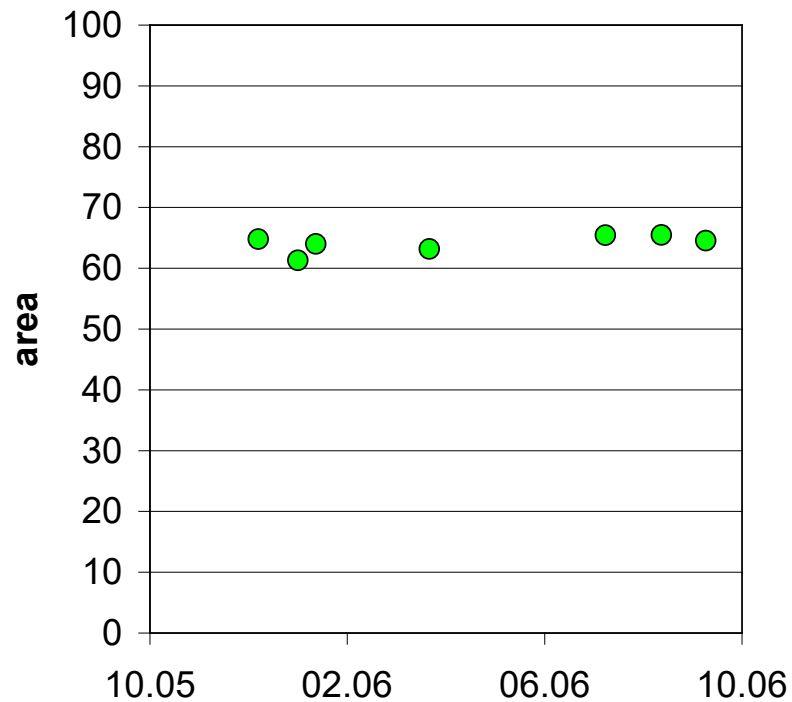
Quality Assurance - Data control

drift in detector stability

FID (stable)

peak areas of calibration:

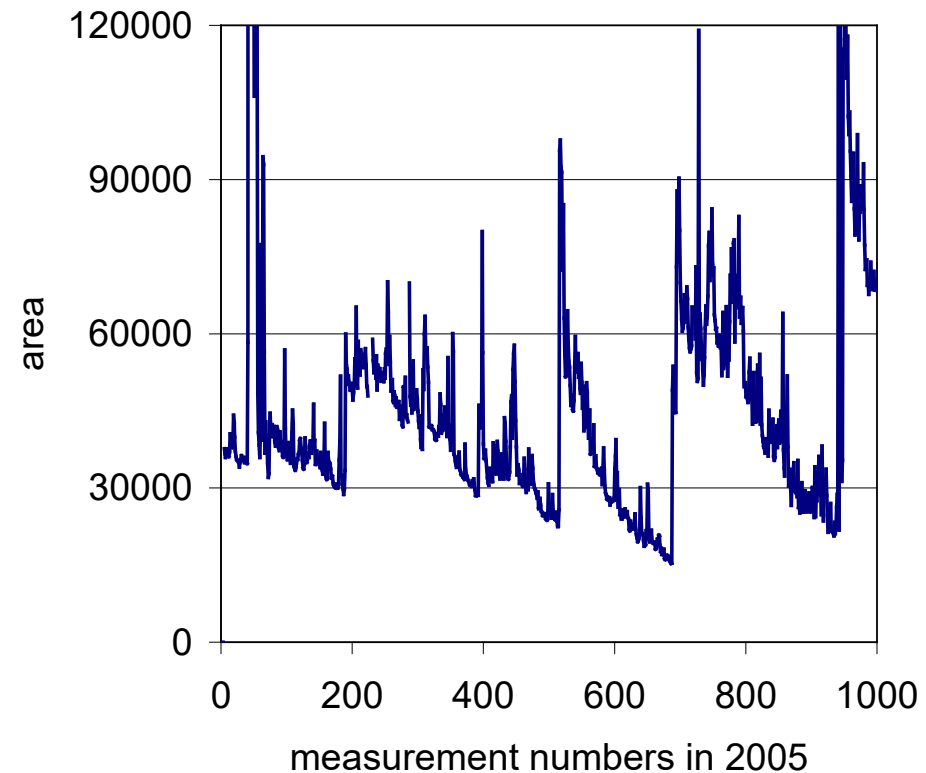
- stable over 1 year
- $\pm 5\%$ std. dev.



MS (unstable)

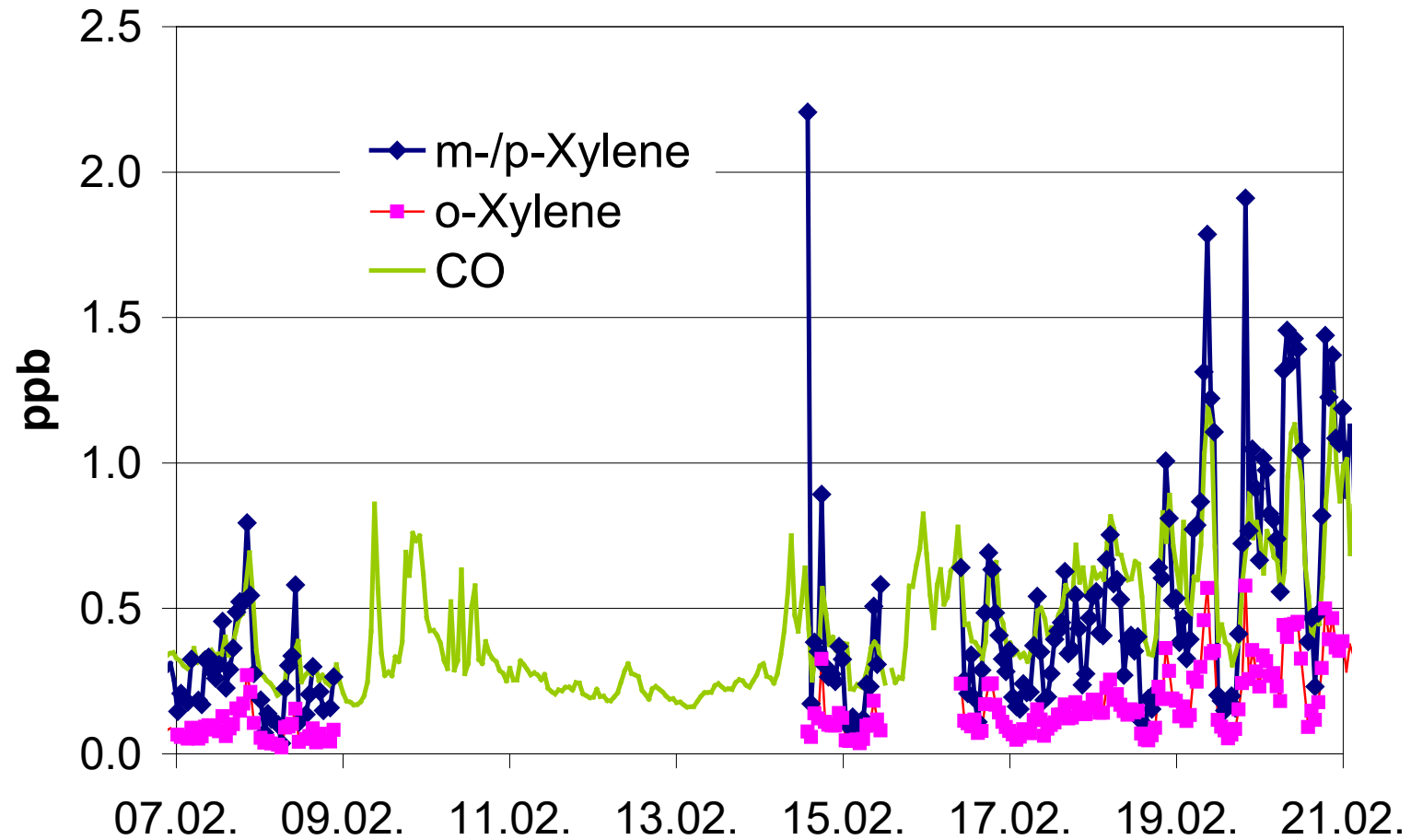
raw peak areas show:

- fast declining sensitivity
- jumps after tuning of source



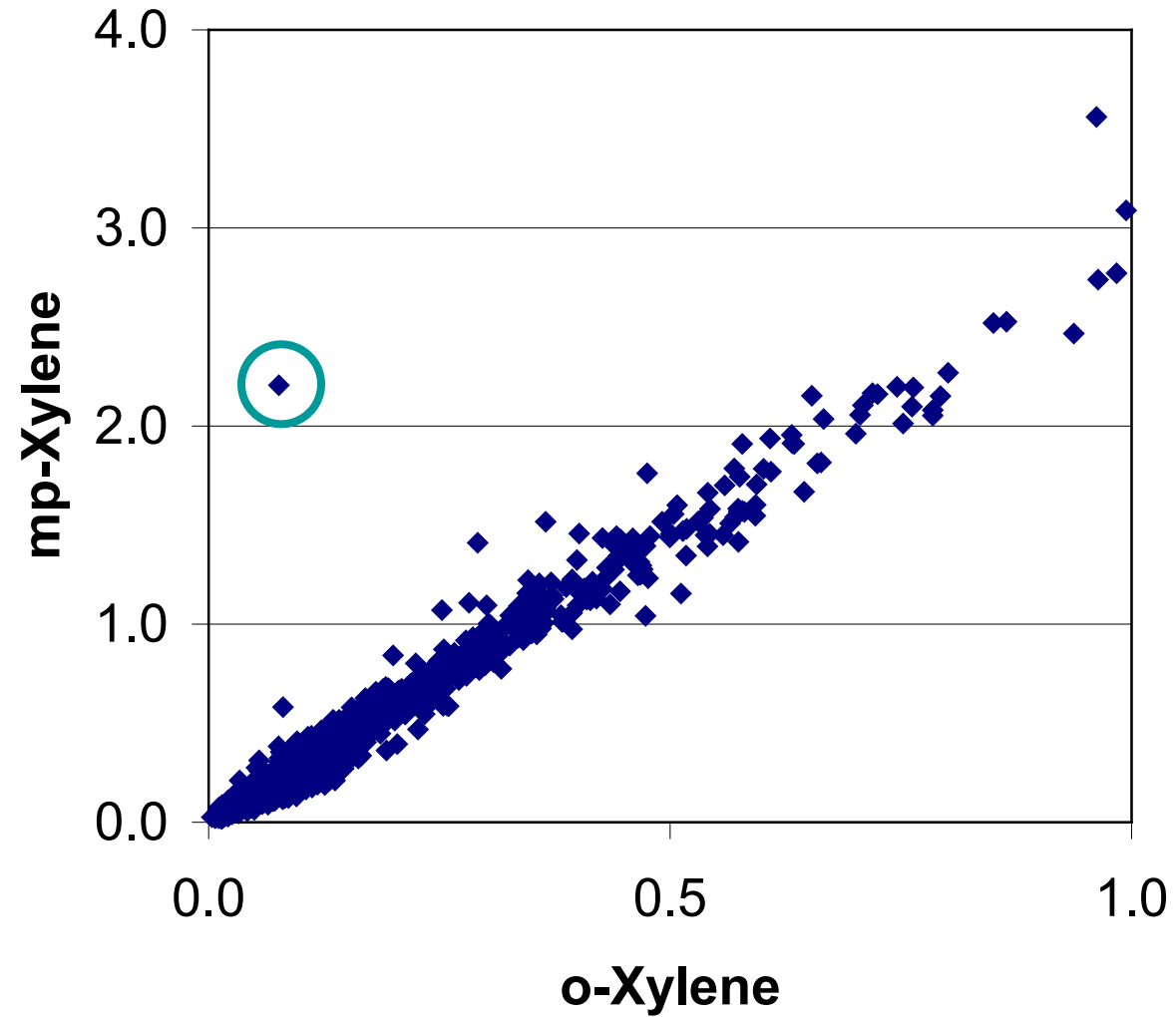
Data quality checks

Interdata comparison (xy-plots)



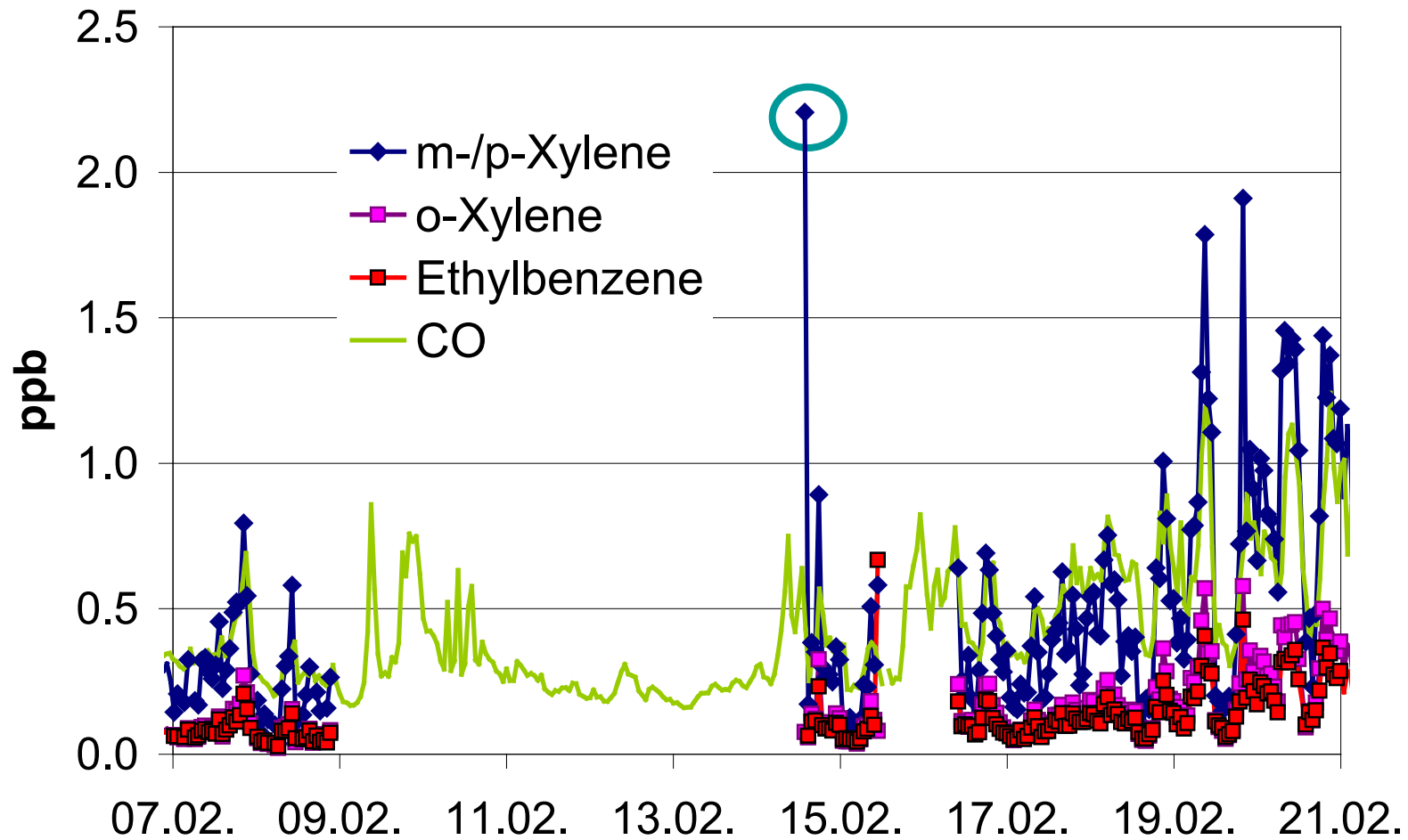
Data quality checks

Interdata comparison (xy-plots)



Data quality checks

Interdata comparison (xy-plots)



Quality Assurance - Data control

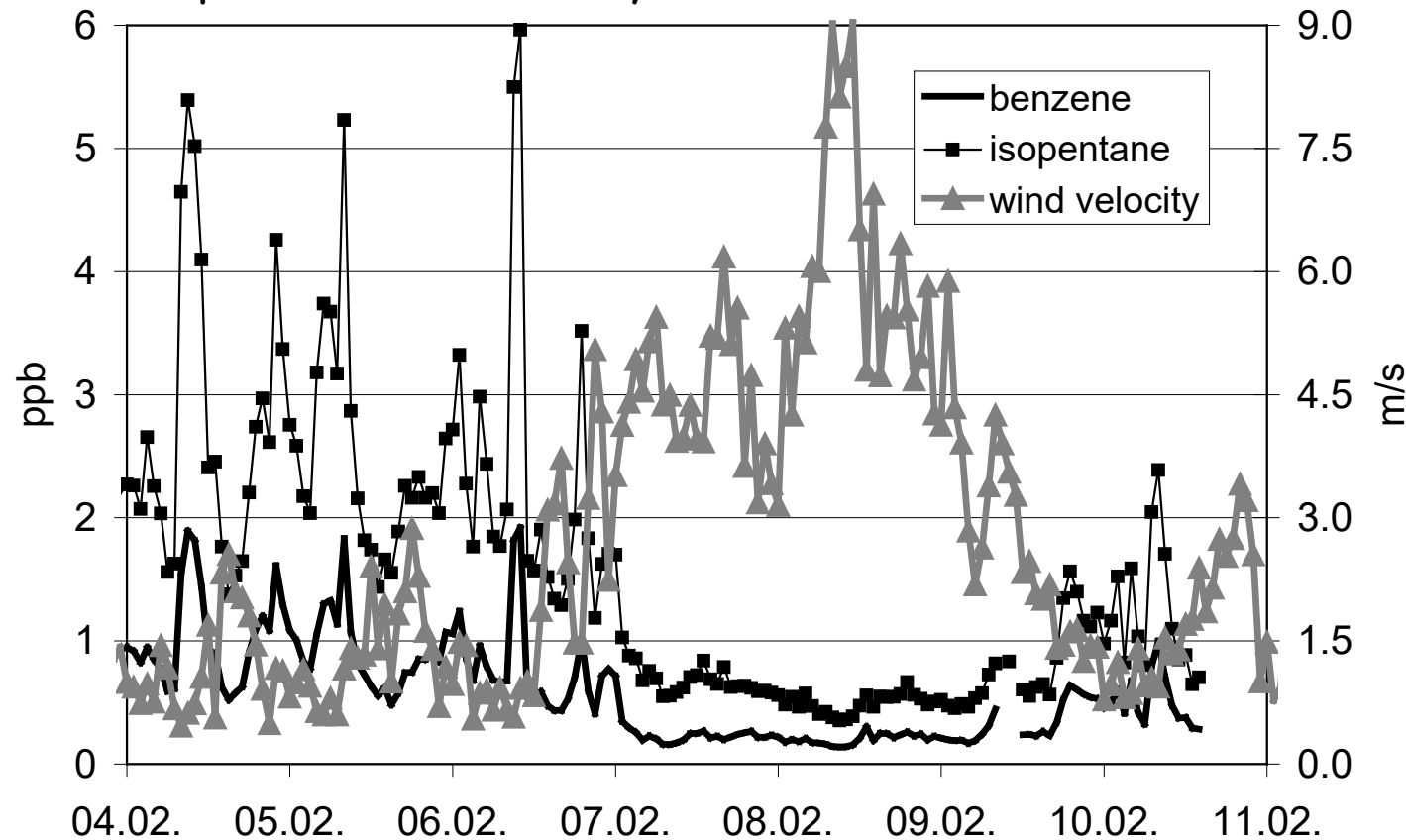
Tools for data verification

2. comparison with meteorological information

Wind speed and direction

important in cities and with known industrial sources

Example: VOCs in the city center of Zurich



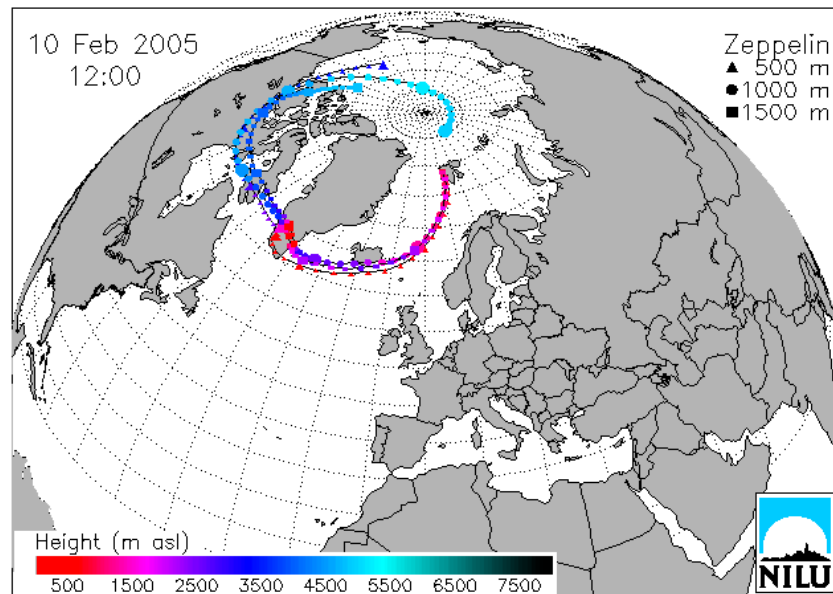
Quality Assurance - Data control

Tools for data verification

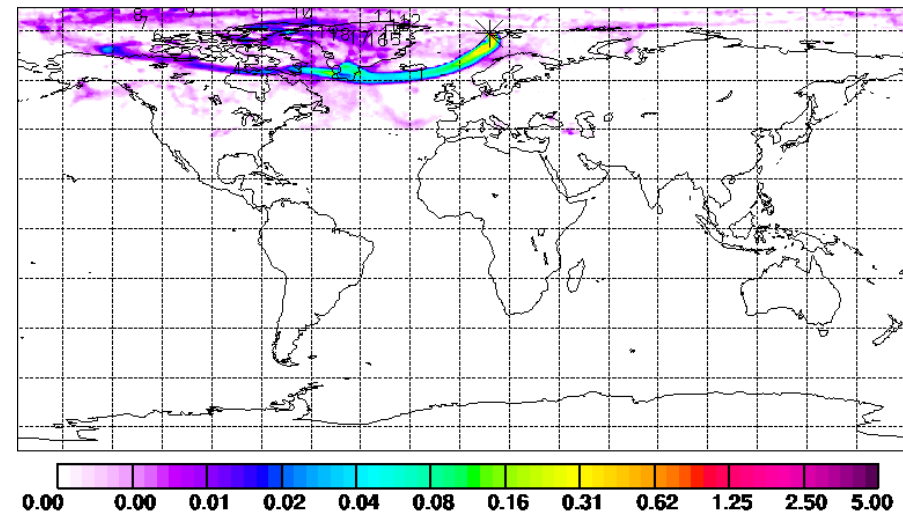
2. comparison with meteorological information

important at remote locations

b) Trajectories



c) Footprints of Lagrangian particle dispersion models (LPDM)

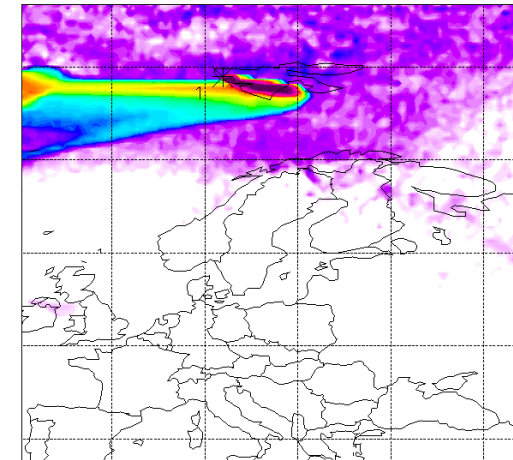
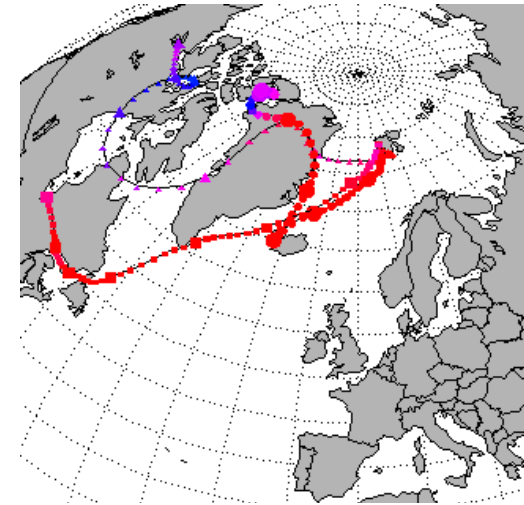
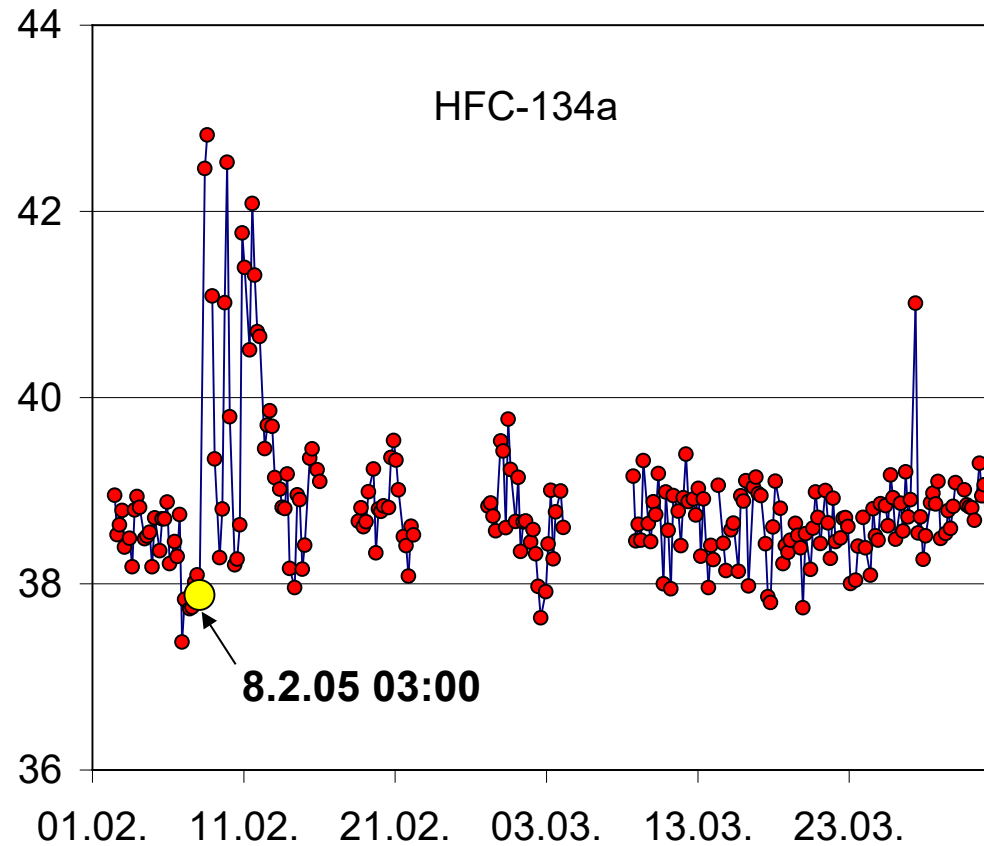


Quality Assurance - Data control

Tools for data verification

2. comparison with meteorological information

Example: Pollution event of HFC-134a in Spitsbergen

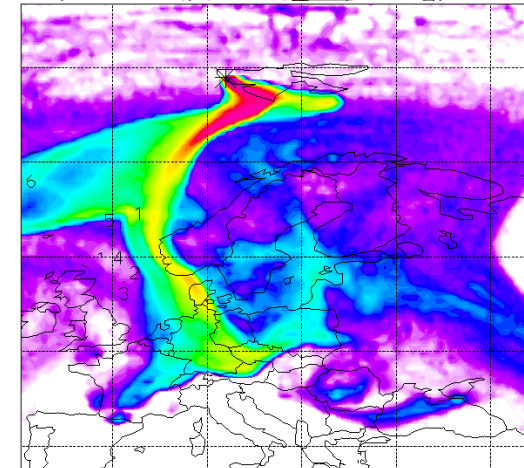
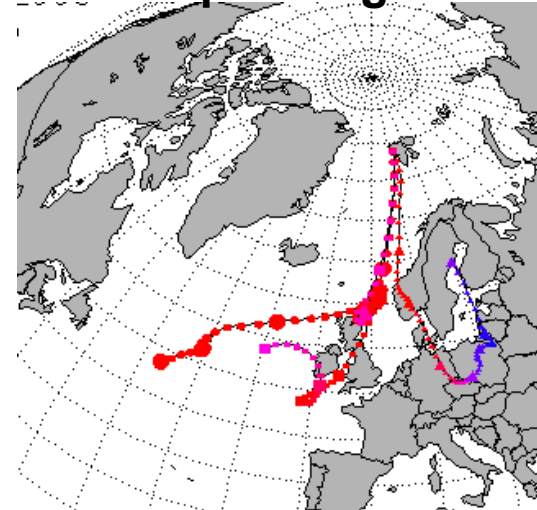
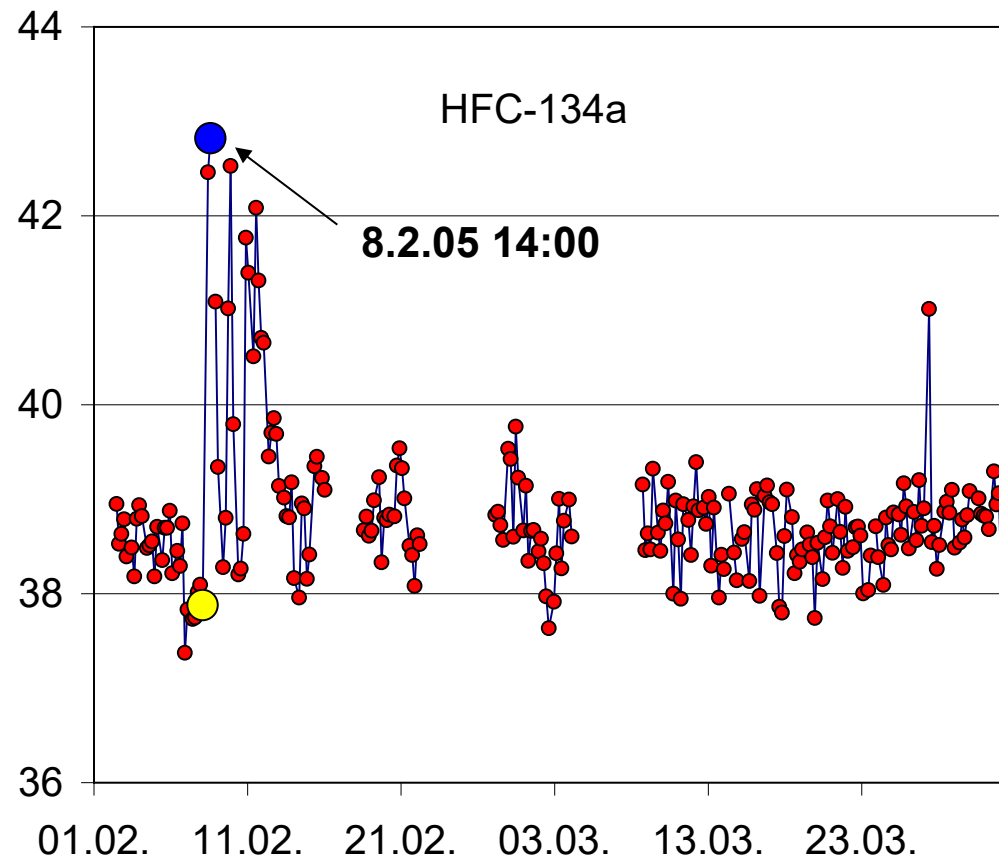


Quality Assurance - Data control

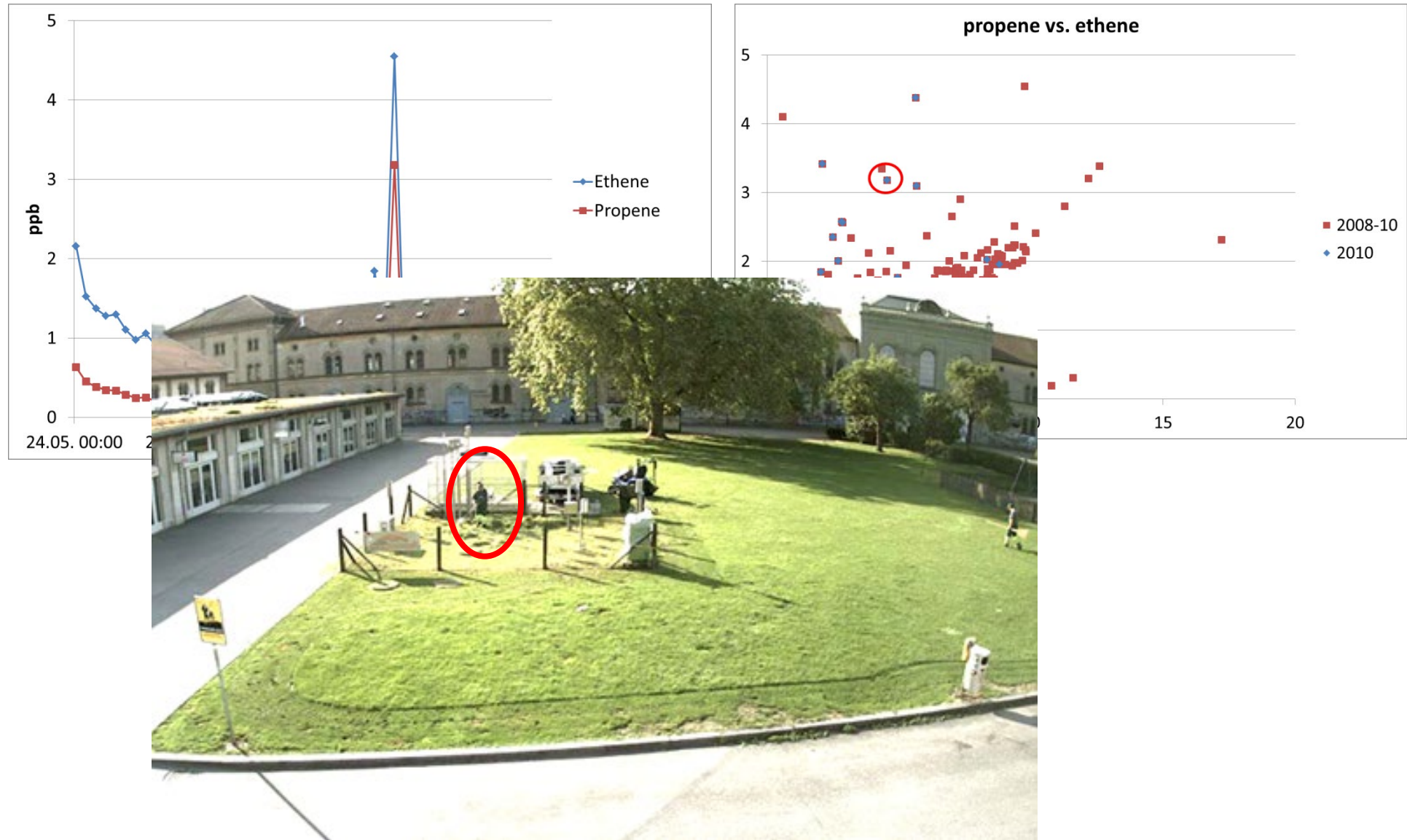
Tools for data verification

2. comparison with meteorological information

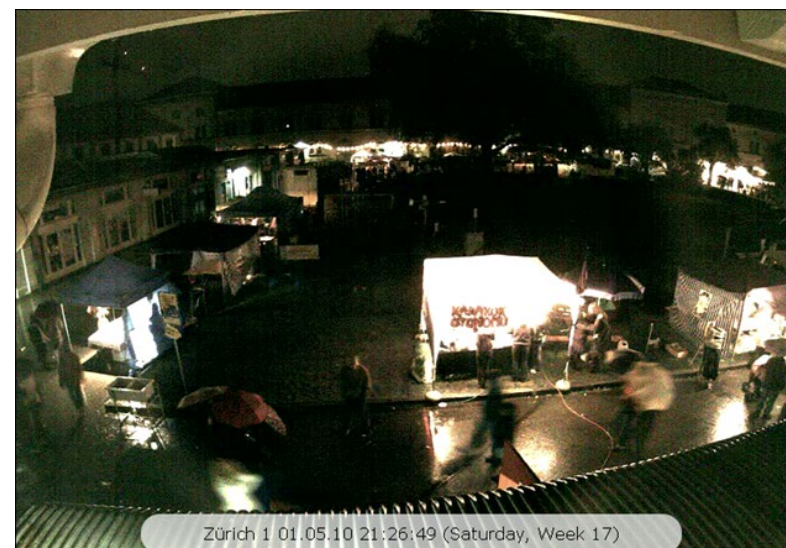
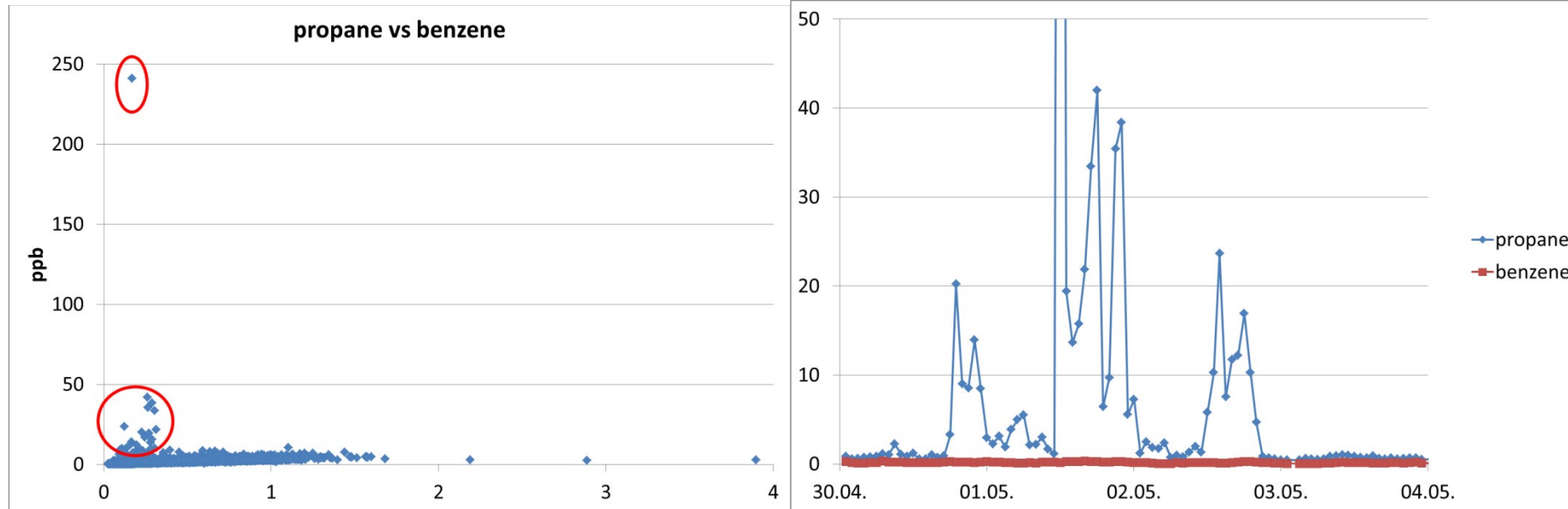
Example: Pollution event of HFC-134a in Spitsbergen



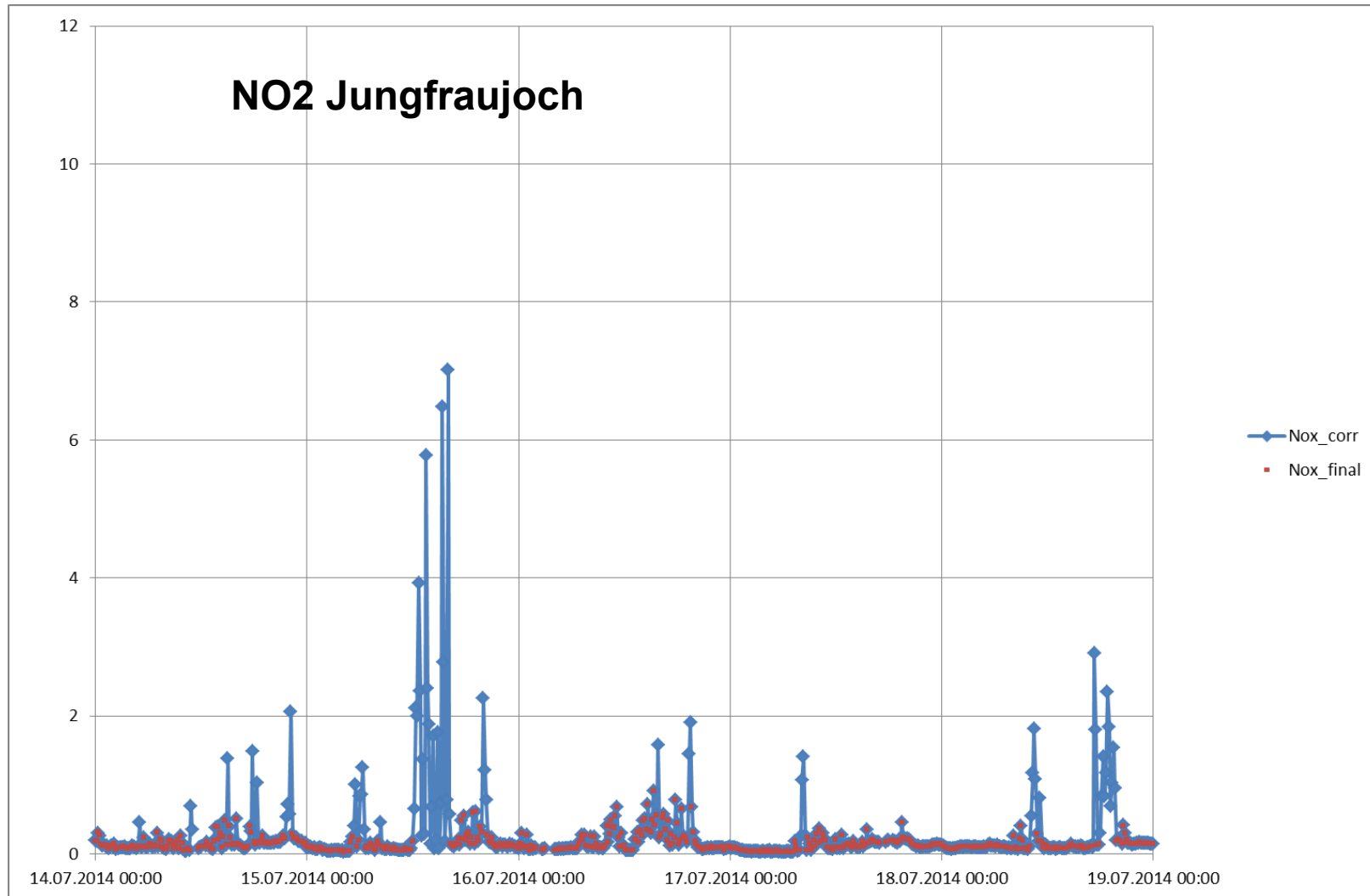
Is this a normal data point?



Is this a normal data point?



Is this data ok?



4 different ways to describe the "concentration" in the atmosphere

1. Mixing ratio C_X [mol mol⁻¹]
2. Mass concentration ρ_X [μg m⁻³]
3. Number density n_X [Moleküle cm⁻³]
4. Column abundance [Moleküle cm⁻²]

Mixing ratio C_X [mol mol⁻¹]

$$C_X = \frac{\text{\# moles of X}}{\text{mole of air}}$$

advantage:

Remains constant if pressure and/or temperature changes

GAS	MIXING RATIO (dry air) [mol mol ⁻¹]
Nitrogen (N ₂)	0.78
Oxygen (O ₂)	0.21
Argon (Ar)	0.0093
Carbon dioxide (CO ₂)	420x10 ⁻⁶
Neon (Ne)	18x10 ⁻⁶
Ozone (O ₃)	(0.01-10)x10 ⁻⁶
Helium (He)	5.2x10 ⁻⁶
Methane (CH ₄)	1.9x10 ⁻⁶
Krypton (Kr)	1.1x10 ⁻⁶

**Trace
gases**

• **units:**

1 ppm = 1x10⁻⁶ mol mol⁻¹

1 ppb = 1x10⁻⁹ mol mol⁻¹

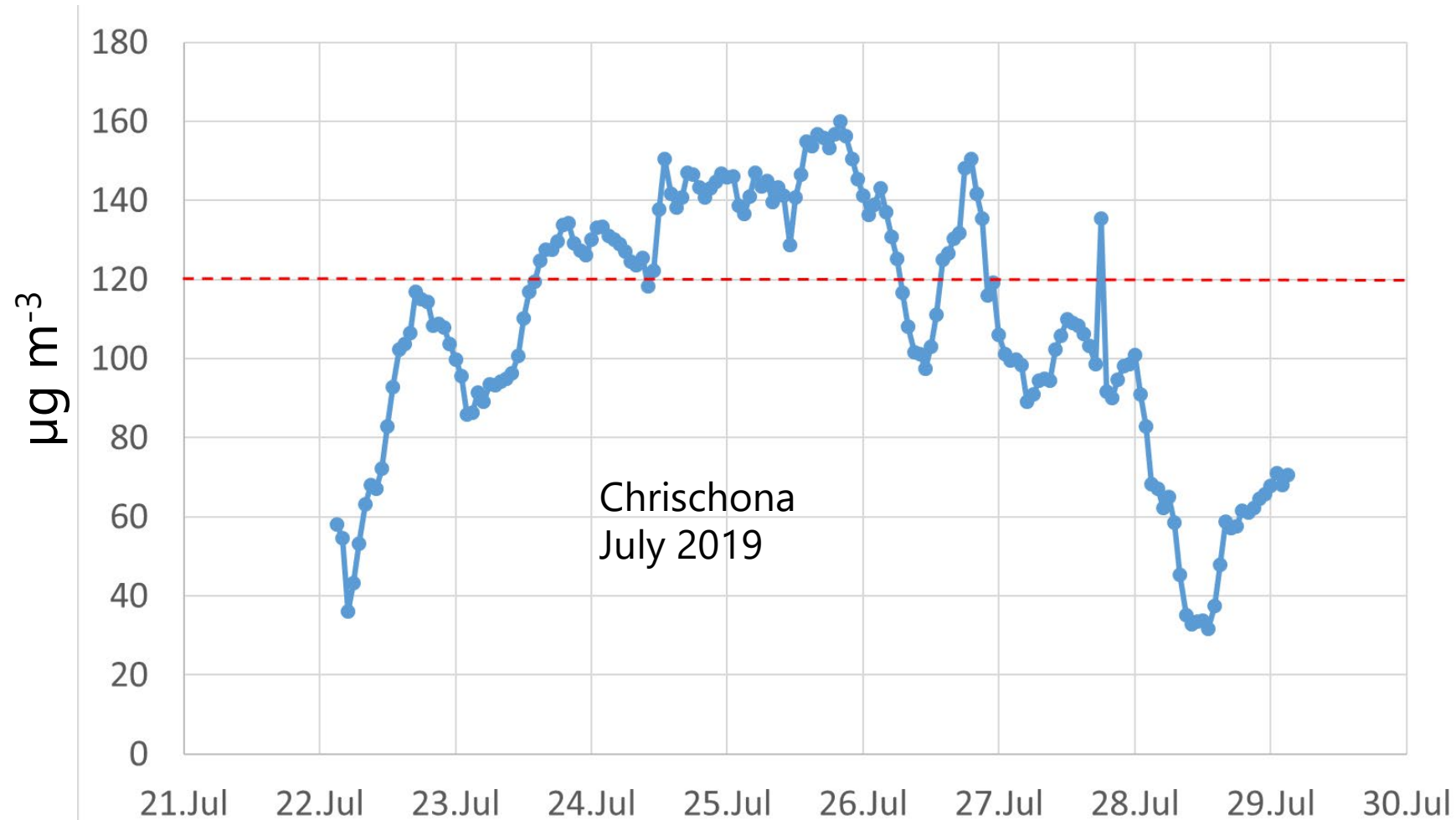
1 ppt = 1x10⁻¹² mol mol⁻¹

1 ppq = 1x10⁻¹⁵ mol mol⁻¹

Mass concentration ρ_x [$\mu\text{g m}^{-3}$]

Legal limit values

example: Swiss 1-hour limit value for ozone: $120 \mu\text{g m}^{-3}$



Conversion: Mass Concentration and Mixing Ratio

1/24.47
l/mol

25 °C

Mass concentration [$\mu\text{g}/\text{m}^3$] =
Mixing ratio [ppb] x molecular weight [g/mol] x 0.0409 [mol/l]

1 ppm => 1 mg/m³

1 ppb => 1 $\mu\text{g}/\text{m}^3$

1 ppt => 1 ng/m³

Mixing ratio [ppb] =
Mass concentration [$\mu\text{g}/\text{m}^3$] / molecular weight [g/mol] / 0.0409 [mol/l]

1 mg/m³ => 1 ppm

1 $\mu\text{g}/\text{m}^3$ => 1 ppb

1 ng/m³ => 1 ppt

Conversion: Mass Concentration and Mixing Ratio

Exercise:

How many $\mu\text{g}/\text{m}^3$ are 60 ppb of ozone (MW: 48 g/mol) at 1013 mbar/25 °C

$$60\text{ppb} \times 48 \text{ g/mol} \times 1/24.45\text{L} = \sim 120 \mu\text{g}/\text{m}^3$$

And at the Jungfraujoch (3580 m)? 1 mol air = 32.4 Liter (700 mbar/ 0° C)

$$60\text{ppb} \times 48 \text{ g/mol} \times 1/32.4\text{L} = 88 \mu\text{g}/\text{m}^3$$

$$\begin{aligned} \text{Mass concentration } [\mu\text{g}/\text{m}^3] = \\ \text{Mixing ratio [ppb]} \times \text{molecular weight [g/mol]} \times 1/24.45 [\text{mol/l}] \end{aligned}$$

Number density n_X [Molecules cm⁻³]

$$n_X = \frac{\text{\# molecules of X}}{\text{unit volume of air}}$$

Only used in reaction kinetics

n_X can be calculated from mixing ratio relative to the total amount of air molecules

$$n_X = n_a C_X$$

$$n_A (\text{total amount of air molecules}) = \frac{\text{\# air molecules}}{\text{ml}} = 2.46 \times 10^{19}$$

→ 1 ppm of a substance → 2.46×10^{13} molecules/ml

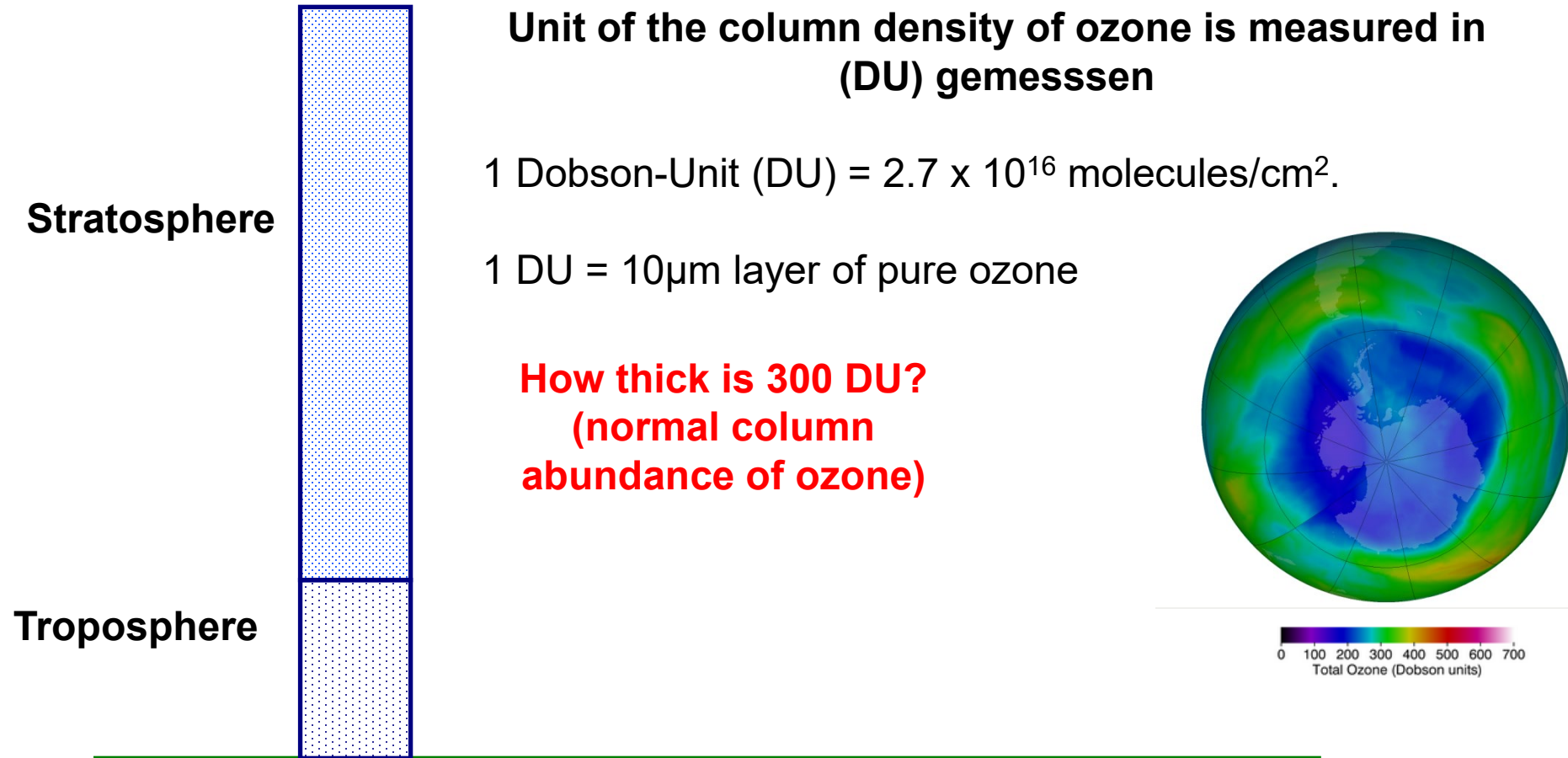
$$1 \text{ mol} = 24.45 \text{ L (25°C, 1013 mbar)}$$

$$\Rightarrow 1 \text{ ml} = 1/24.45/1000 = 4.09 \times 10^{-5} \text{ mol}$$

$$\Rightarrow 4.09 \times 10^{-5} * 6.022 \times 10^{23} = 2.46 \times 10^{19} \text{ air molecules/ml}$$

Column abundance: molecules/cm²

Example: Measurement of total ozone from space



SO₂ in the atmosphere

First local anthropogenic air pollution due to SO₂:

1306: **King Edward** I tried to ban coal from being used in London

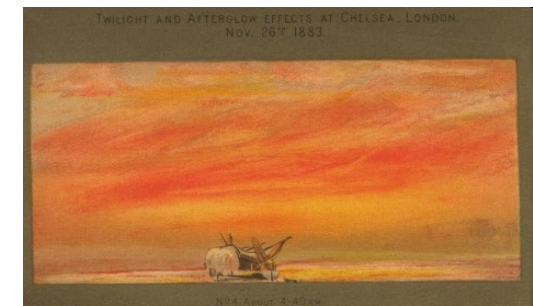
16th century: Replacing of smelters outside of towns

Effects on human health

- labored breathing
- permanent pulmonary damage
- Production of aerosols ("London smog")

Effects on the environment (Production of H₂SO₄, sulfuric acid)

- Forest decline
- Acidification of lakes and soils (Scandinavian lakes, 1970s)
- Global cooling (negative greenhouse effect)



Fluxes and effects of SO₂ in the atmosphere

Sources (Tg/y): **tons?**

Anthropogenic (~100 Tg/y)

Power plants:	35 Tg/y
Oil and gas production	10 Tg/y
Metal smelting:	10 Tg/y
Small scale use:	45 Tg/y

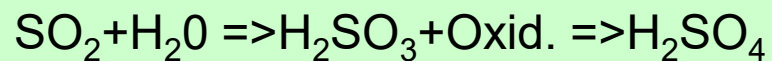
Natural

Volcanoes:	18 Tg/y
Degradation of oceanic DMS, H ₂ S, OCS :	50 Tg/y

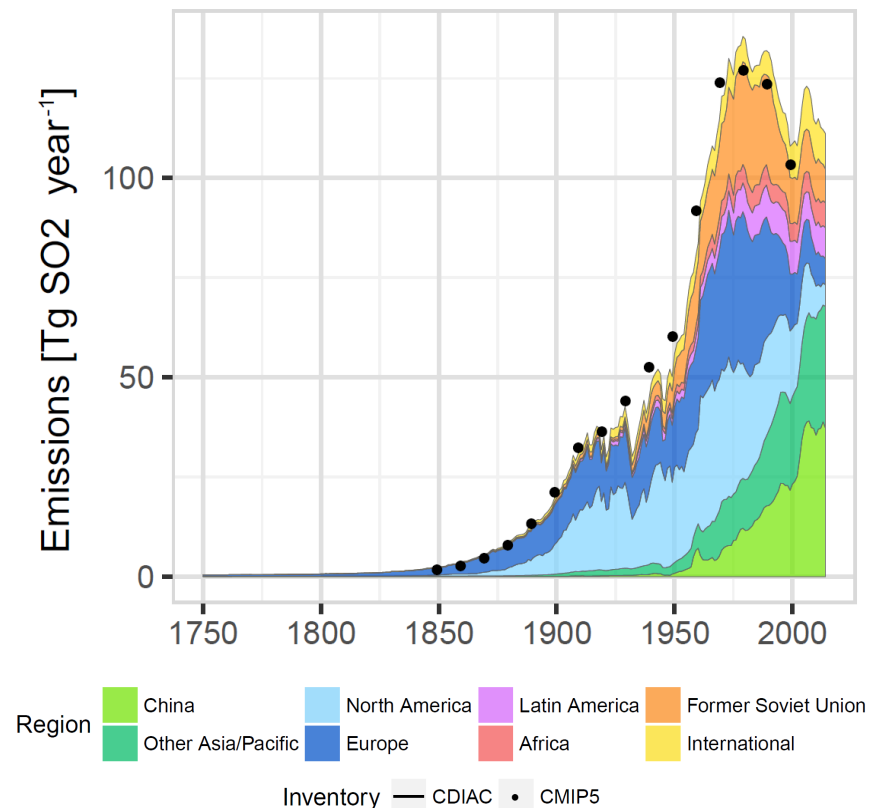
Sinks:

1. Dry deposition (40%)

2. Wet deposition (60%)



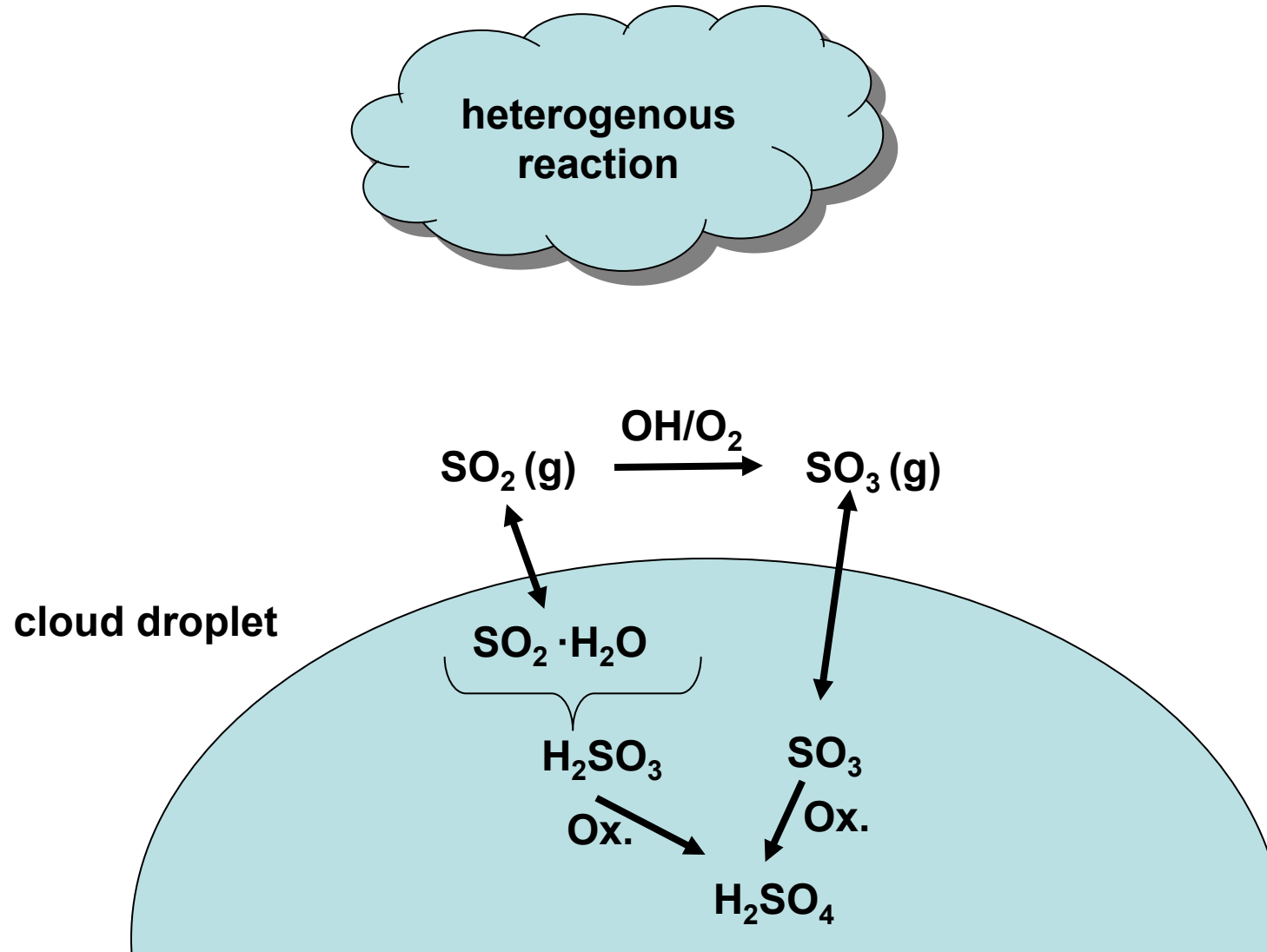
H₂SO₄ => deposition



Hoesly et al., Geosci. Model Dev., 11, 369–408, 2018

Lifetime:
<3 days

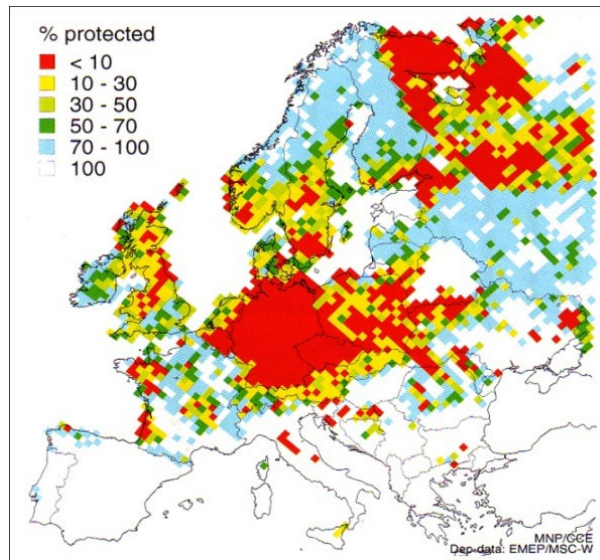
Formation of H_2SO_4 from SO_2 in the atmosphere



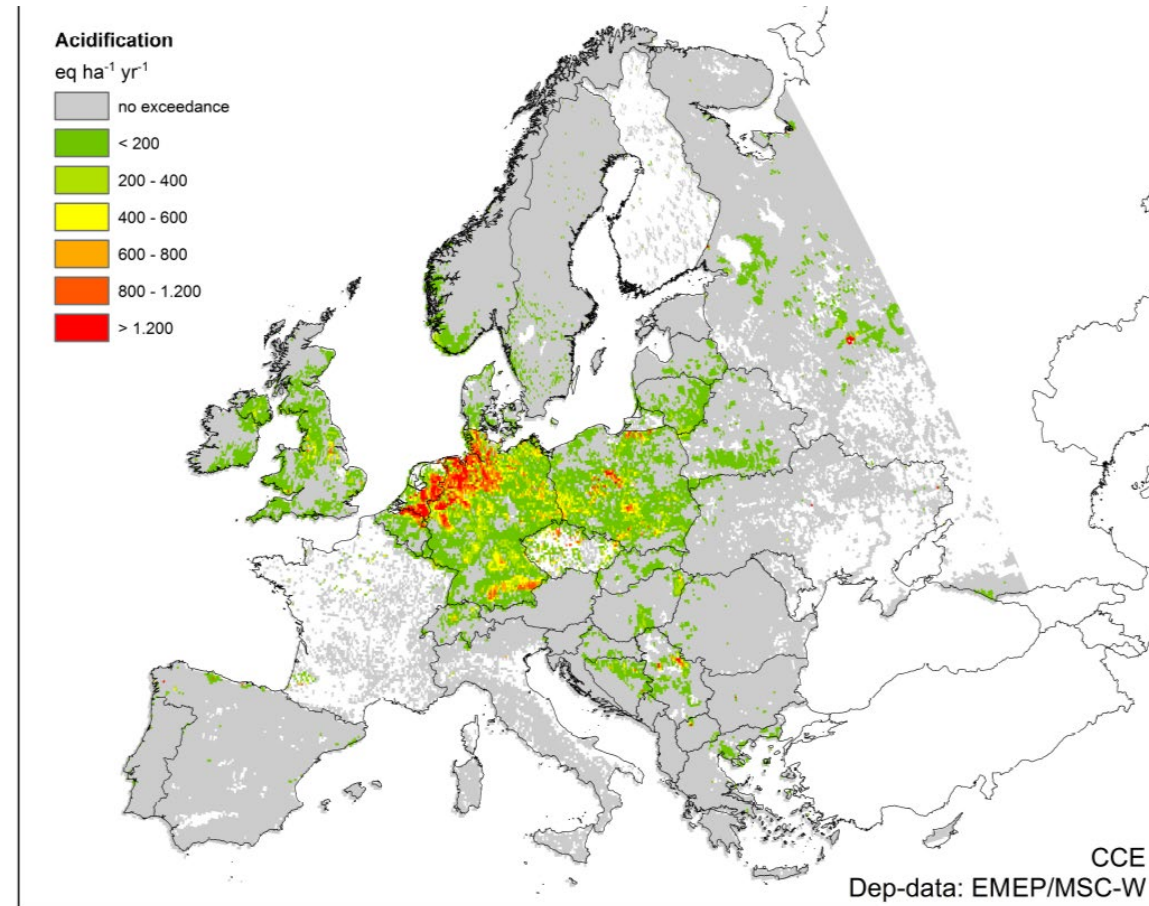
regional effects: Influence of acid rain

Exceedance of critical loads for acidification

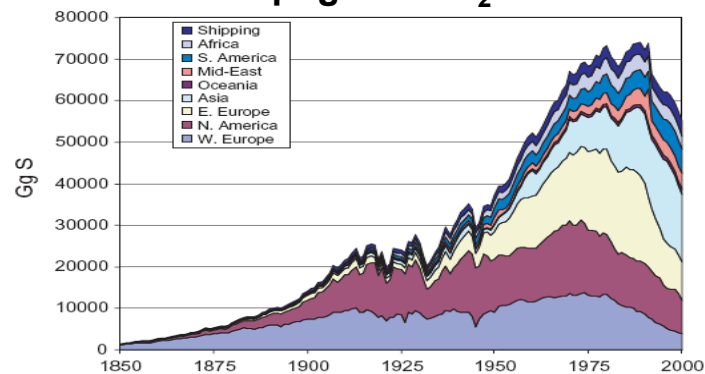
1980



2017



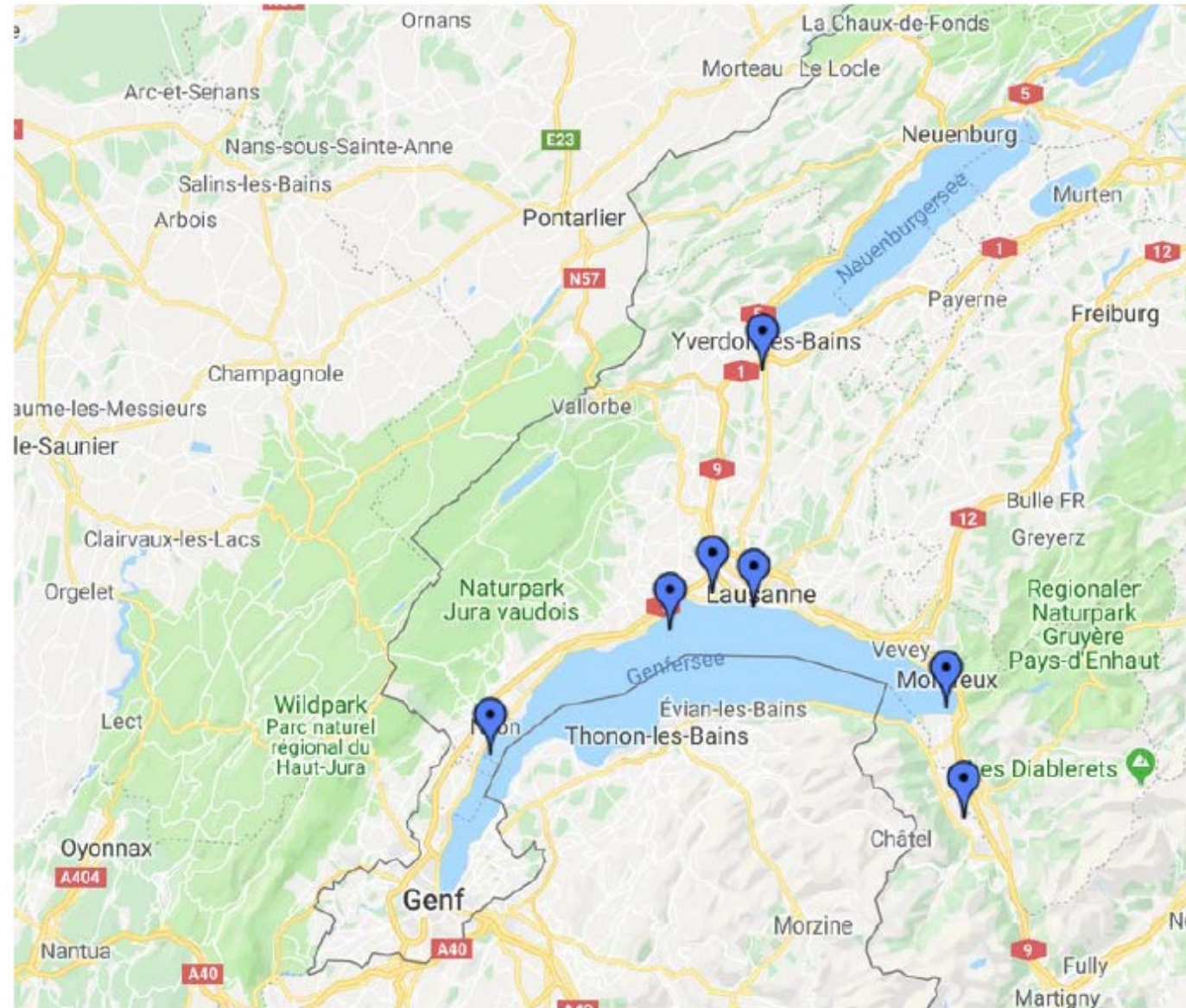
Global anthropogenic SO₂ emissions



EMEP status report 2018

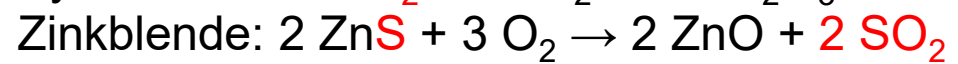
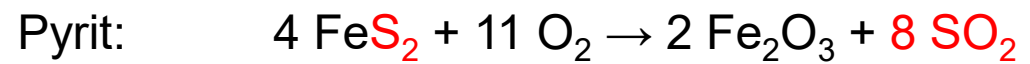
Exercise 1

The Canton Vaud is looking for a new background site
where would you place it and why?

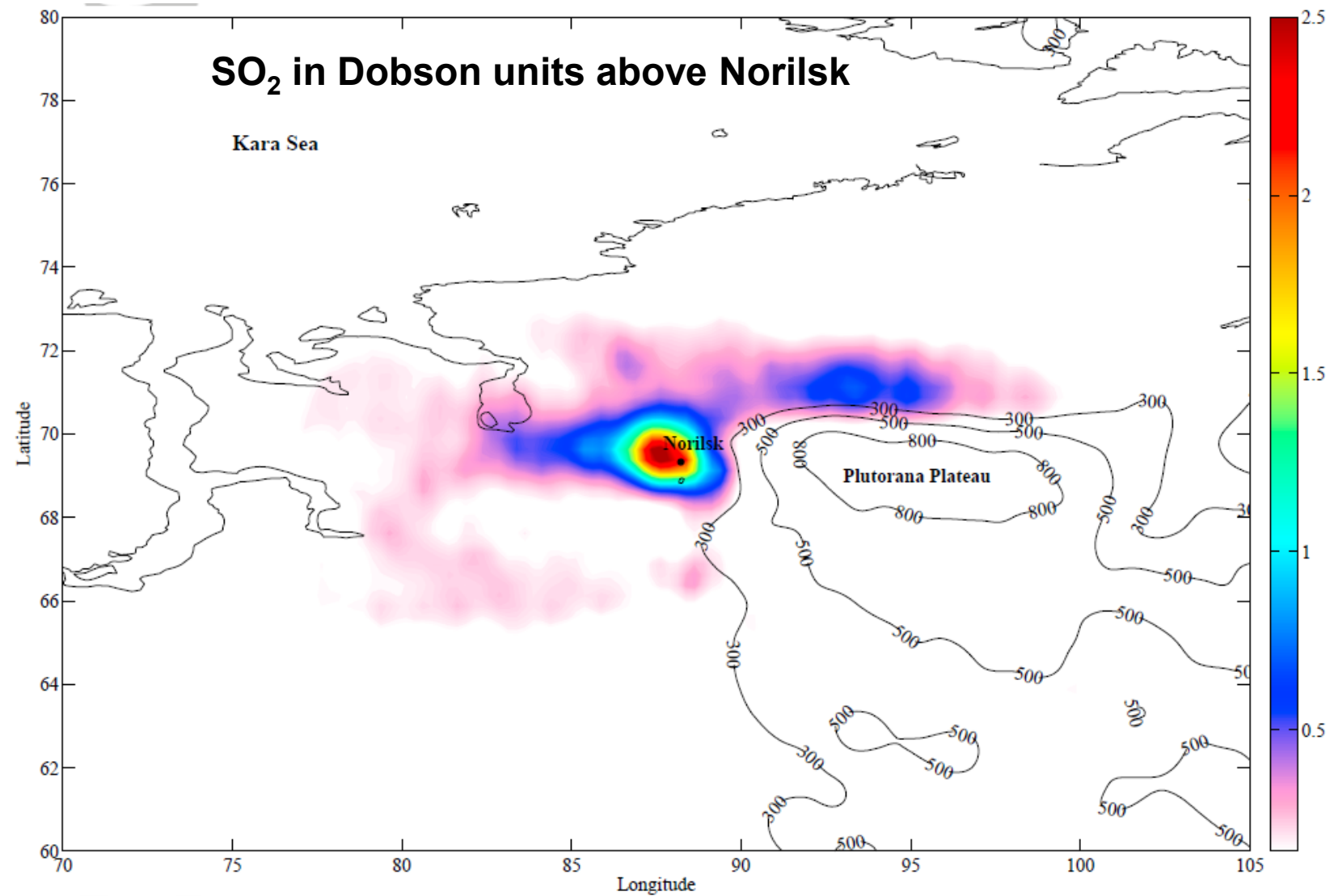


SO₂ Emissions from metal smelters

Norilsk (Russia)
2% of global emissions (2 Mio t/yr)

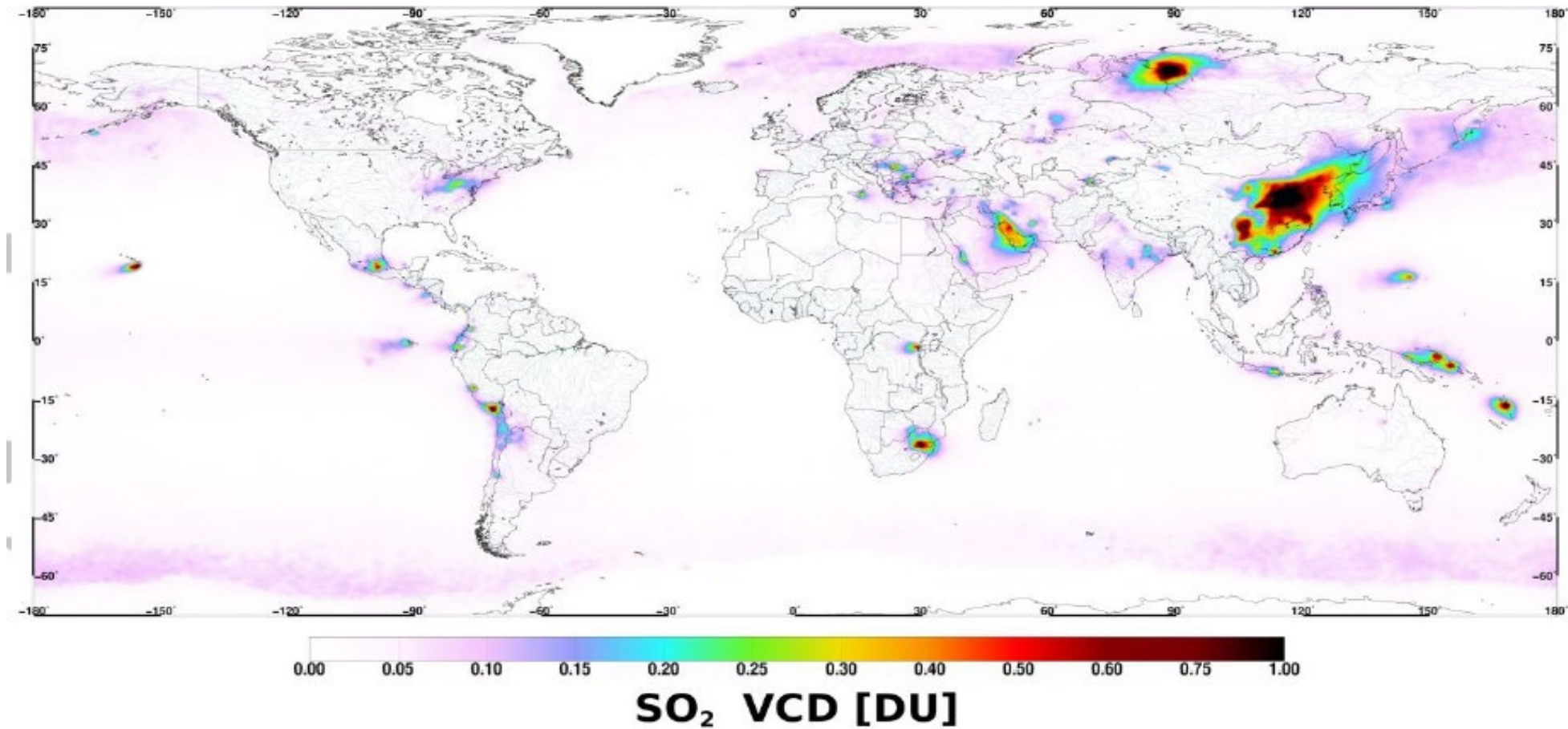


SO₂ Emissions from metal smelters



SO₂ emissions anthropogenic sources and volcanos

(Theys et al, JGR, 2015)



Coal-Fuelled Power Plants Oil and Gas Refineries

Coal-Fuelled Power Plant, South Africa



SO₂ from Coal Power Plants

Coal contains sulfur from the sulfur-containing proteins in the old biomass. Of biomass is oxidised to CO₂, sulfur is oxidised to SO₂.

Good practice: Stripping of SO₂ from chimneys and usage of produced H₂SO₄ for chemical processes.

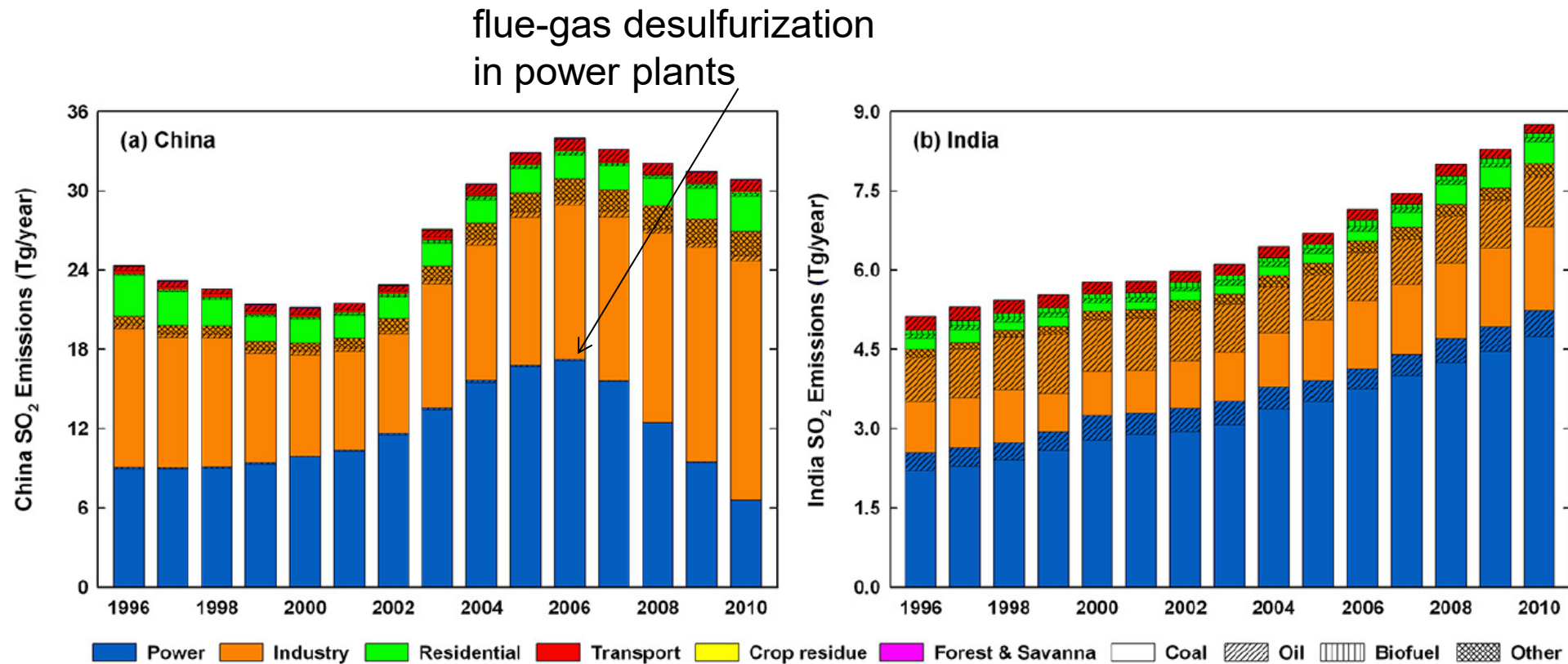
Aghar Oil and Gas Refinery, Iran



SO₂ from Refineries

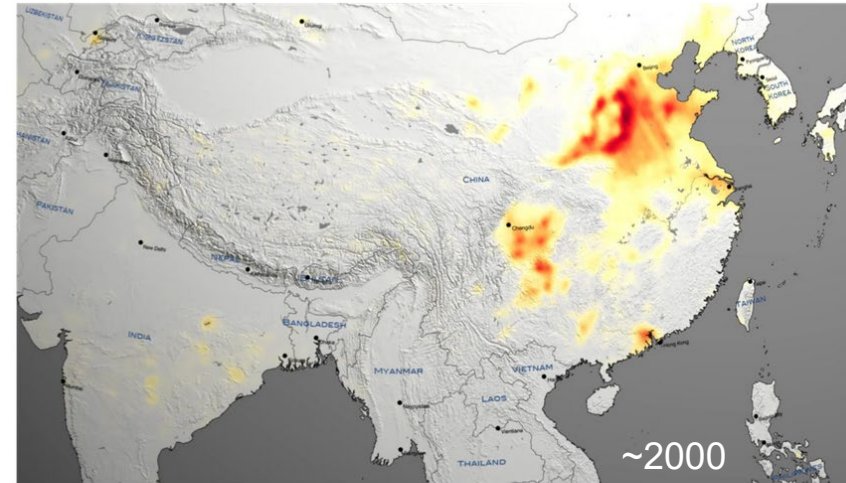
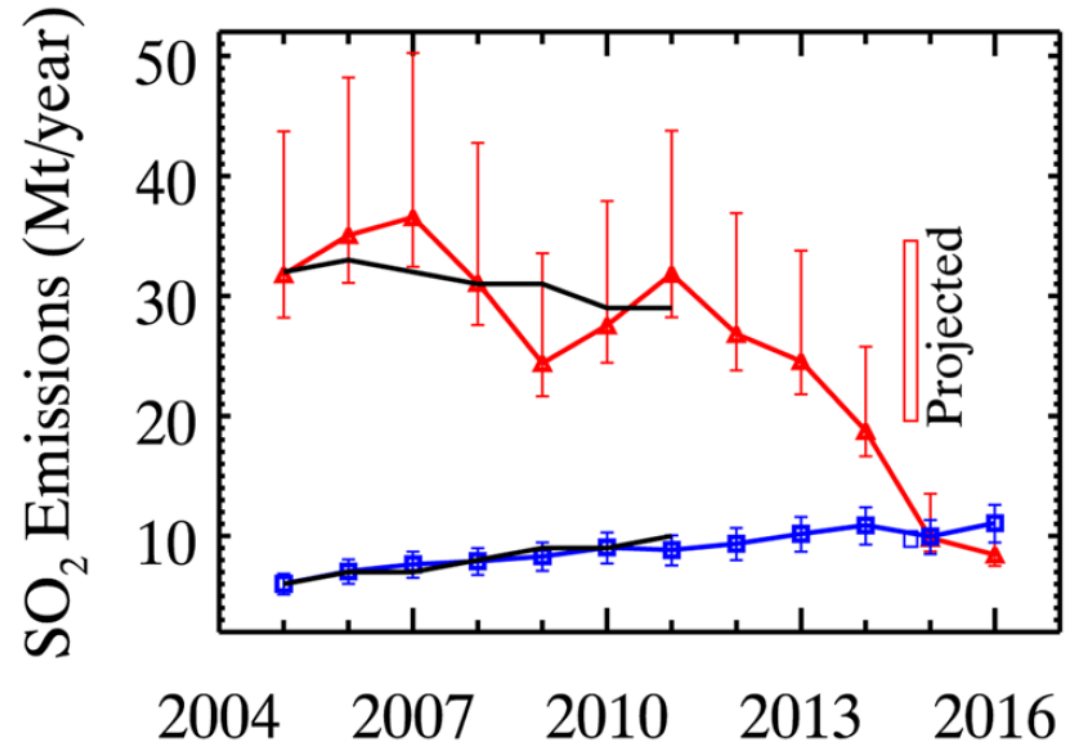
1. Regeneration of catalysts from cracking by oxidizing adsorbed coke to carbon monoxide. During regeneration, sulfur and sulfide deposits that also accumulated on the catalyst are oxidized to SO₂.
2. Hydrogen sulfide (H₂S) flaring. Good practice: > 95 % are stripped from the gas before it is are burned. If not, SO₂ is formed
3. Burning of sulfur-containing oil and gas on the spot for energy consumption.

SO₂ emissions from China and India

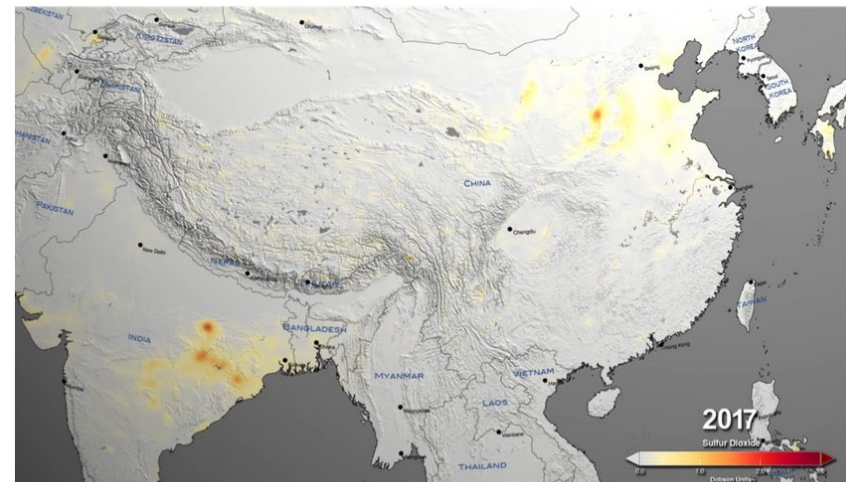


Sulfur dioxide and primary carbonaceous aerosol emissions in China and India, 1996–2010, Lu et al., ACP, 2011

In 2016 India Overtook China, based on Satellite Measurements)



NASA

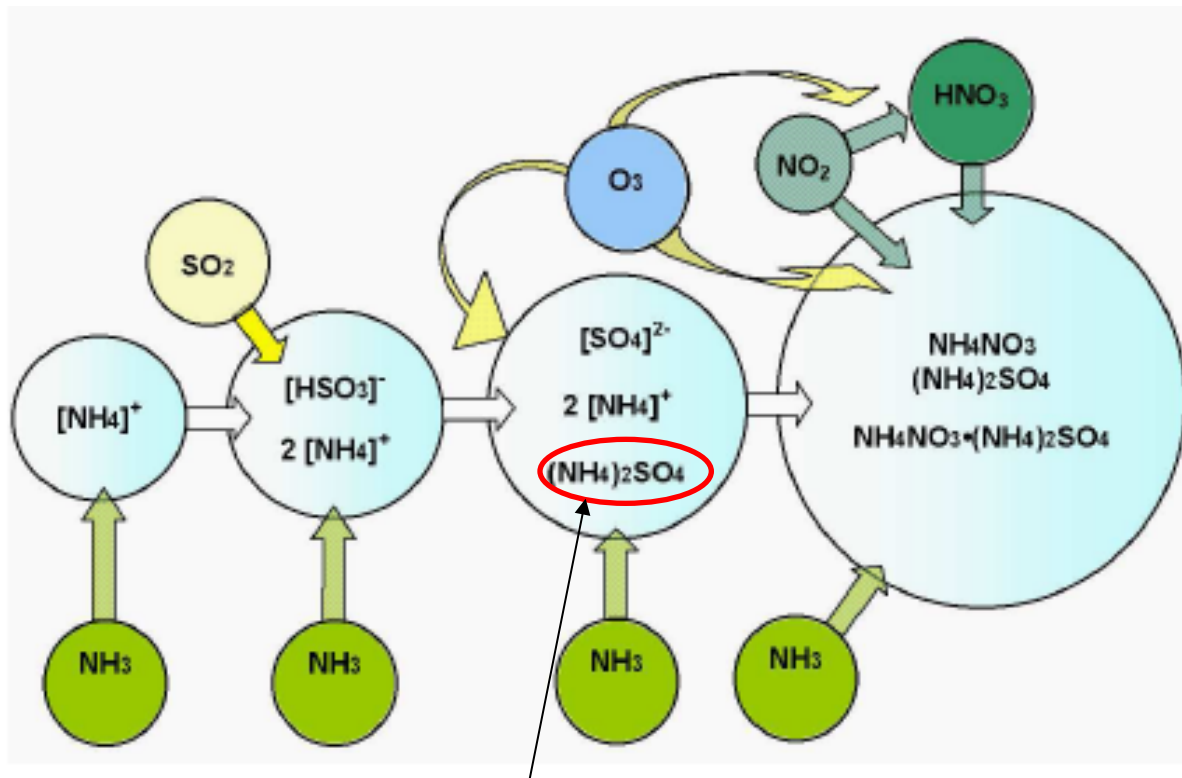


NASA

Li et al., Scientific Reports, 2017

Particle formation with the inclusion of SO_2

Key to secondary particle formation is NH_3 (ammonia)
mainly emitted from agriculture (manure/fertilizer)

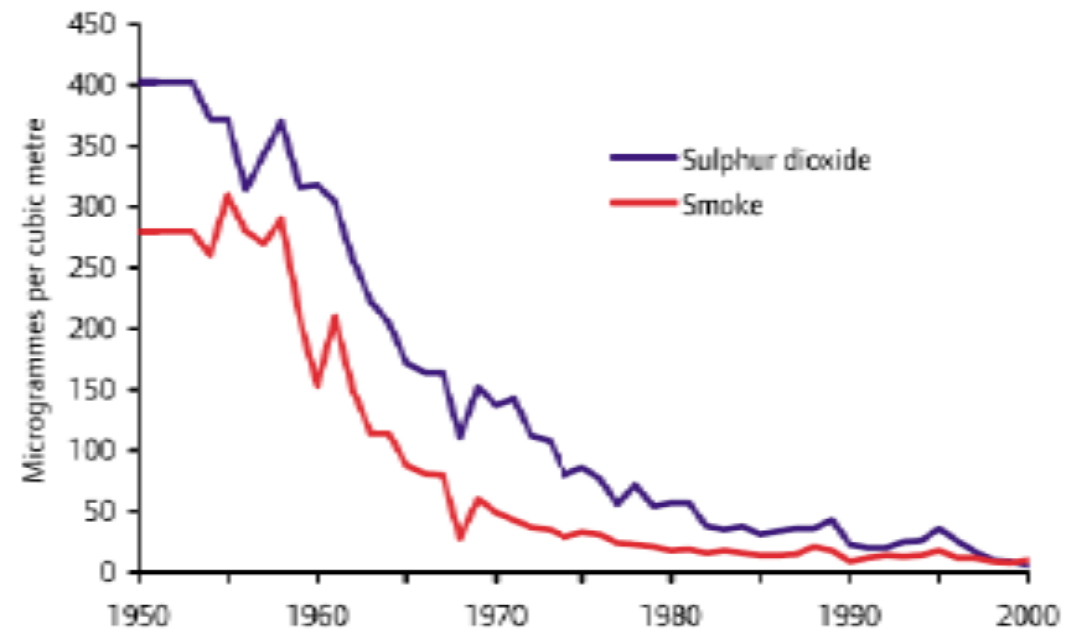


Ammoniumsulfate

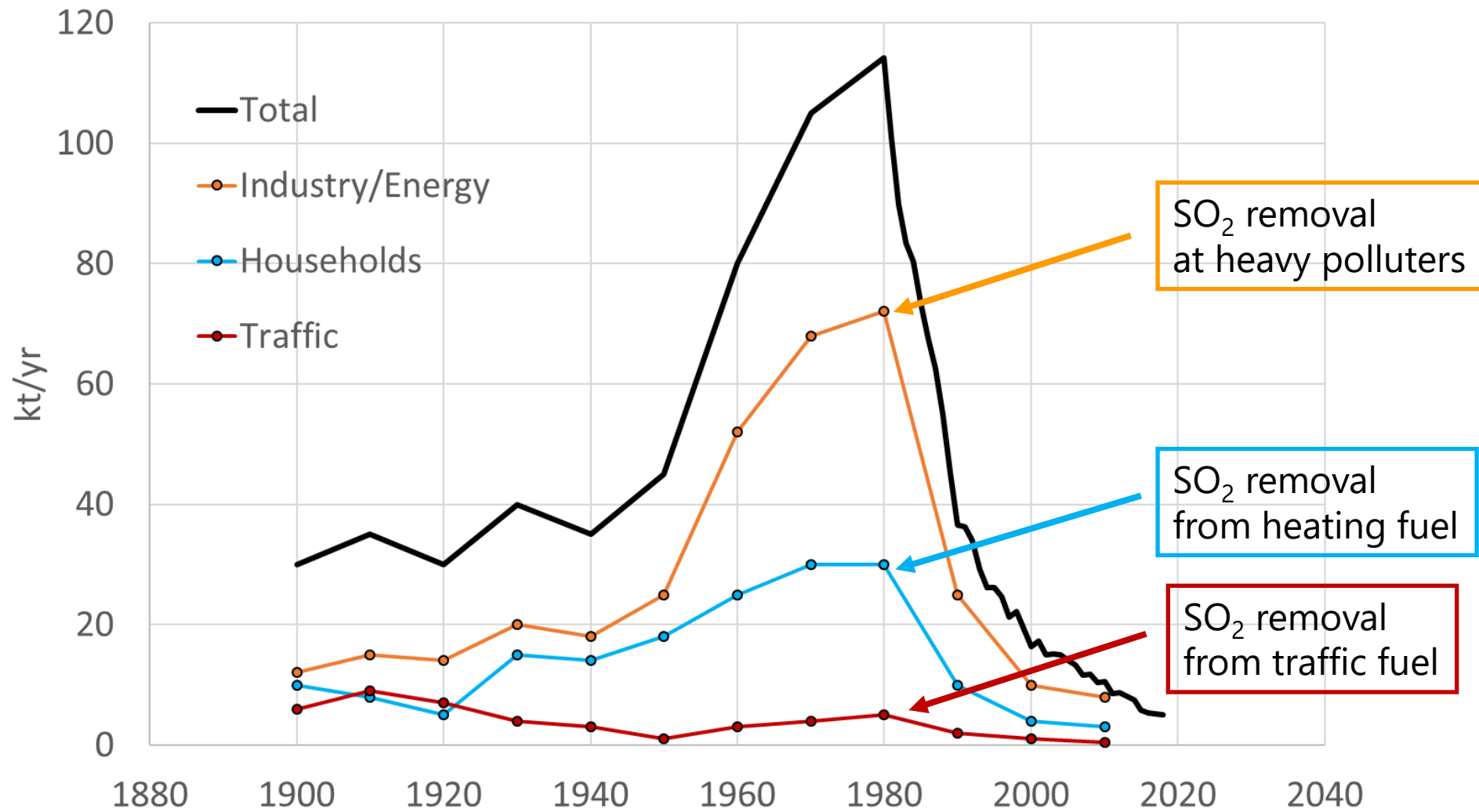
regional effects: The London smog (winter smog)



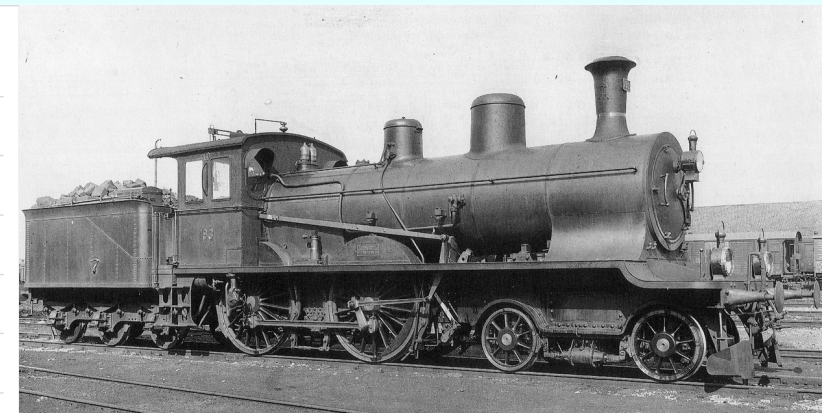
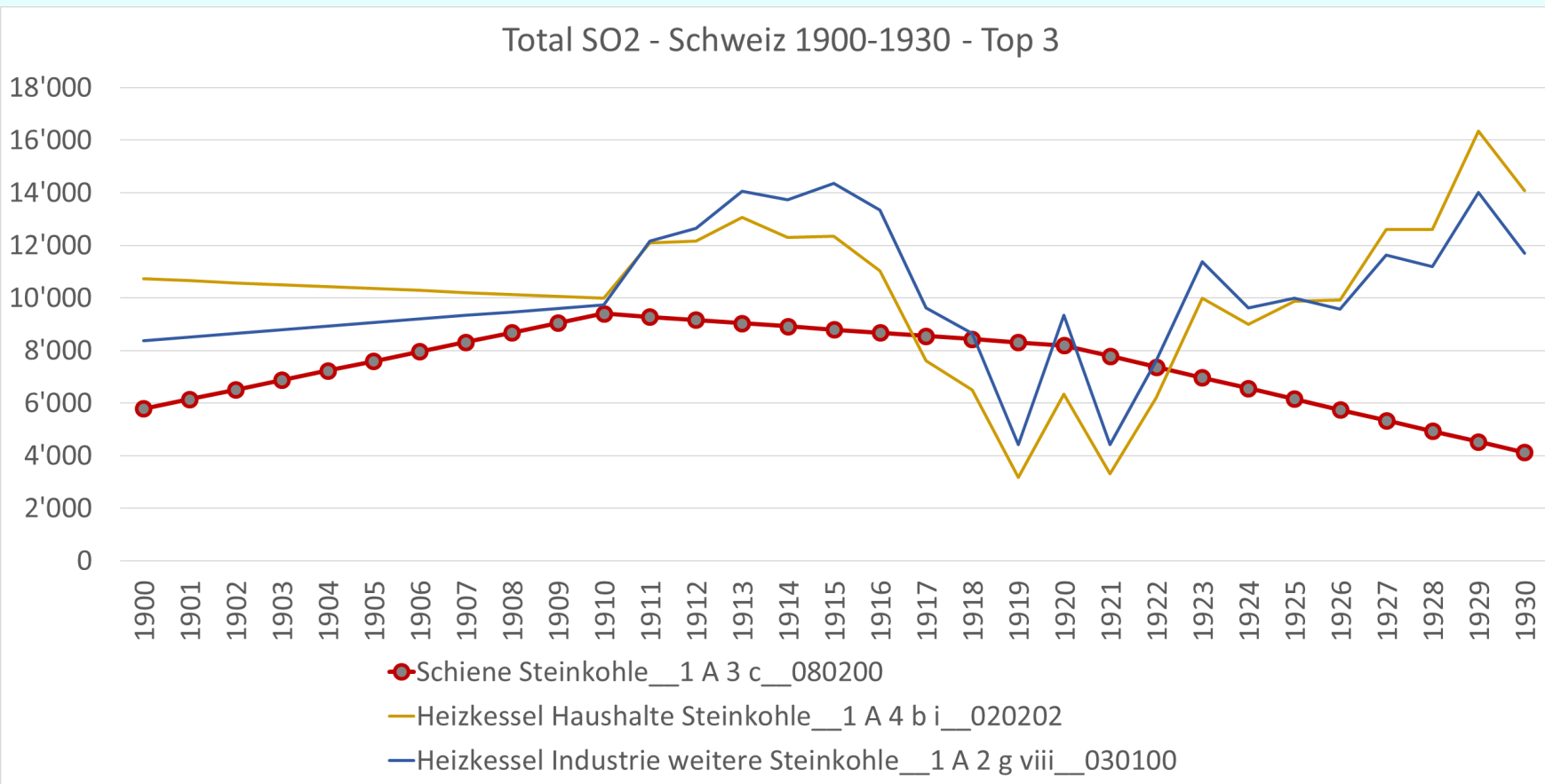
London, 1952



SO₂ emissions in Switzerland



SO₂ peak from traffic in 1910 in Switzerland



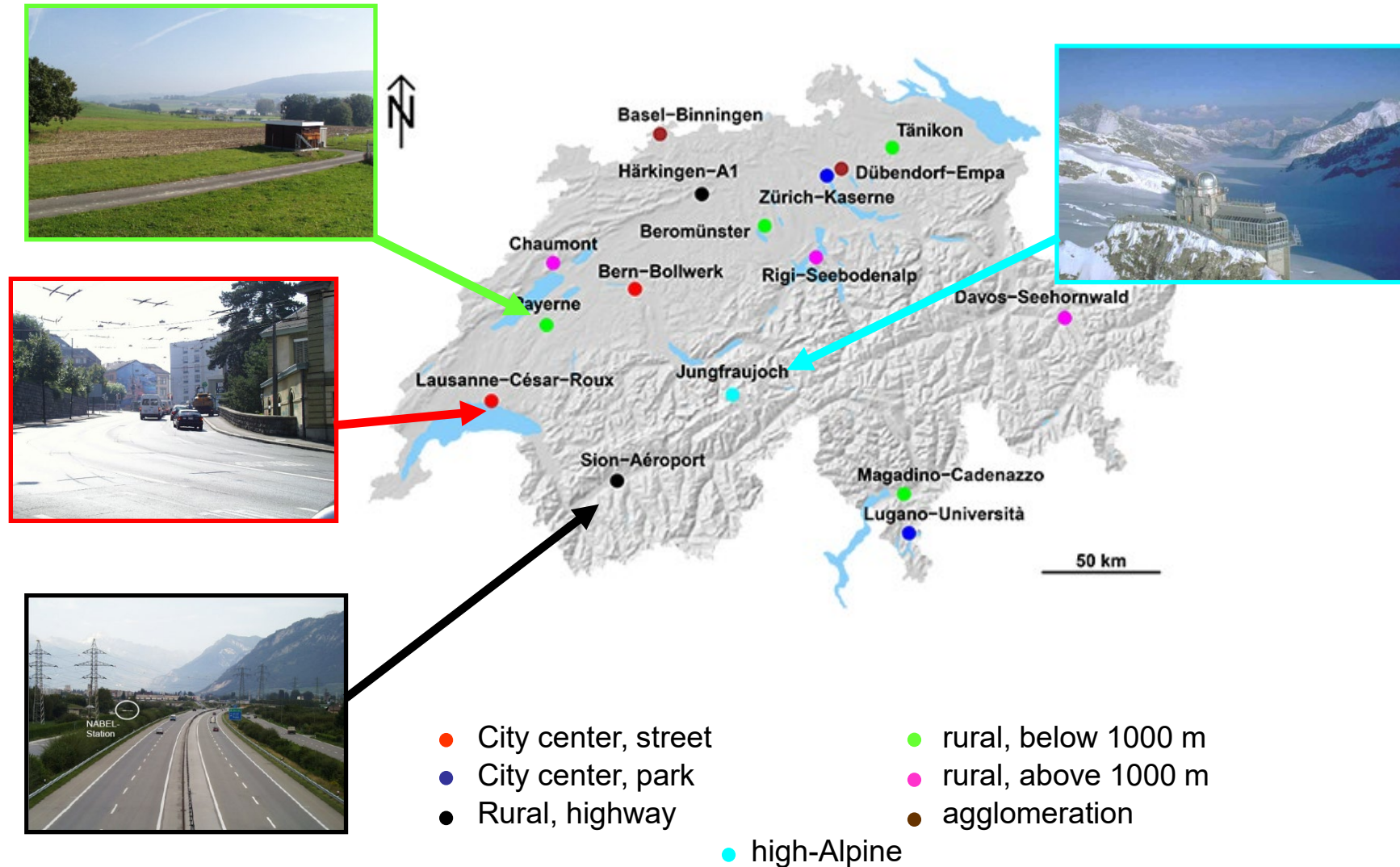
A2/4: Schnellzuglokomotive 1906
6t Kohle; bis 100 km/h



AE 3/6: ab 1920
bis 110 km/h

Antwort BAFU: Schaut man aber die Emissionen aus dem Schienenverkehr an, erkennt man tatsächlich einen schwachen Peak im Jahr 1910. Die Vermutung mit der Elektrifizierungswelle um und ab 1920 dürfte also korrekt sein.

Swiss Network for Air Pollution Monitoring NABEL



Swiss Legal Air Pollution Limits

https://www.ekl.admin.ch/inhalte/dateien/pdf/EKL-231120_de_orig.pdf

Pollutant	Averaging time	WHO AQG 2021	Current OAPC ambient limit value	FCAH recommendation 2023
Sulphur dioxide (SO ₂), µg/m ³ (see Chapter 8)	Annual average and new mean value over winter half-year	–	30 ^a	20 ^b
	95 % of ½ h mean value for a year	–	100	remove
	24h mean value	40 ^c	100 ^d	40 ^c
Nitrogen dioxide (NO ₂), µg/m ³ (see Chapter 7)	Annual average	10	30	10
	95 % of ½ h mean value for a year	–	100	remove
	24h mean value	25 ^c	80 ^d	25 ^c
Carbon monoxide (CO), mg/m ³ (see Chapter 9)	24h mean value	4 ^c	8 ^d	4 ^c
Ozone (O ₃), µg/m ³ (see Chapter 6)	Summer season ^e	60	–	60
	98 % of ½ h mean value for a month	–	100	100
	8h mean value	100 ^c	–	–
	1h mean value	–	120 ^d	120 ^d
Suspended particulates / particulate matter (PM ₁₀), µg/m ³ (see Chapter 4)	Annual average	15	20	15
	24h mean value	45 ^c	50 ^c	45 ^c
Suspended particulates / particulate matter (PM _{2.5}), µg/m ³ (see Chapter 5)	Annual average	5	10	5
	24h mean value	15 ^c	–	15 ^c

^a Ambient air quality standard, which also includes the protection of animals and plants, their biological communities and habitats according to Article 1 paragraph 1 EPA, and corresponds to the state of knowledge when the Air Pollution Control Ordinance was adopted in 1985.

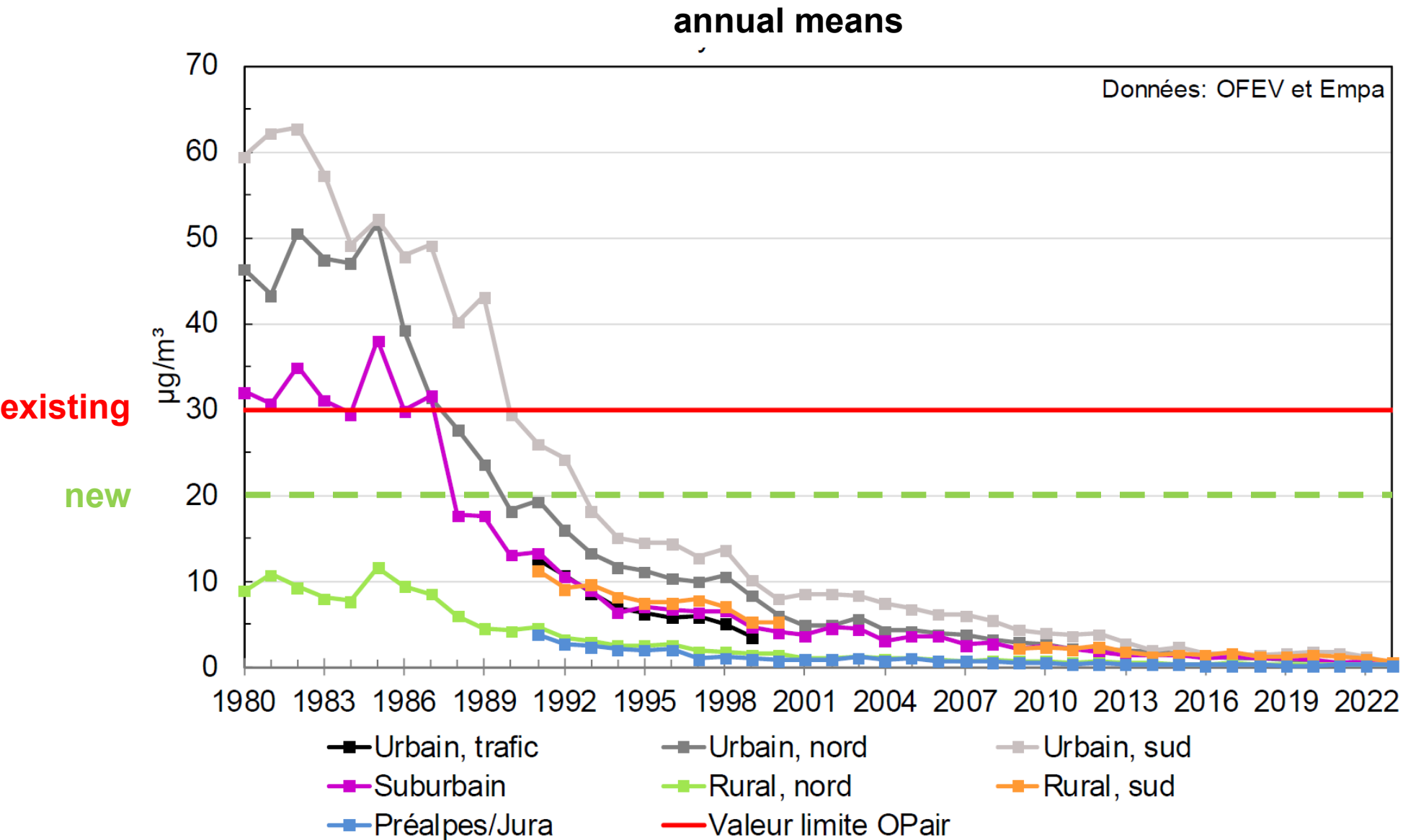
^b Value stipulated in the 2000 WHO AQGs (WHO, 2000) for the protection of forests and other seminatural ecosystems. Valid as an annual average as well as for the winter half-year. (October–March).

^c 99th percentile (i.e. limit value may be exceeded three times per year).

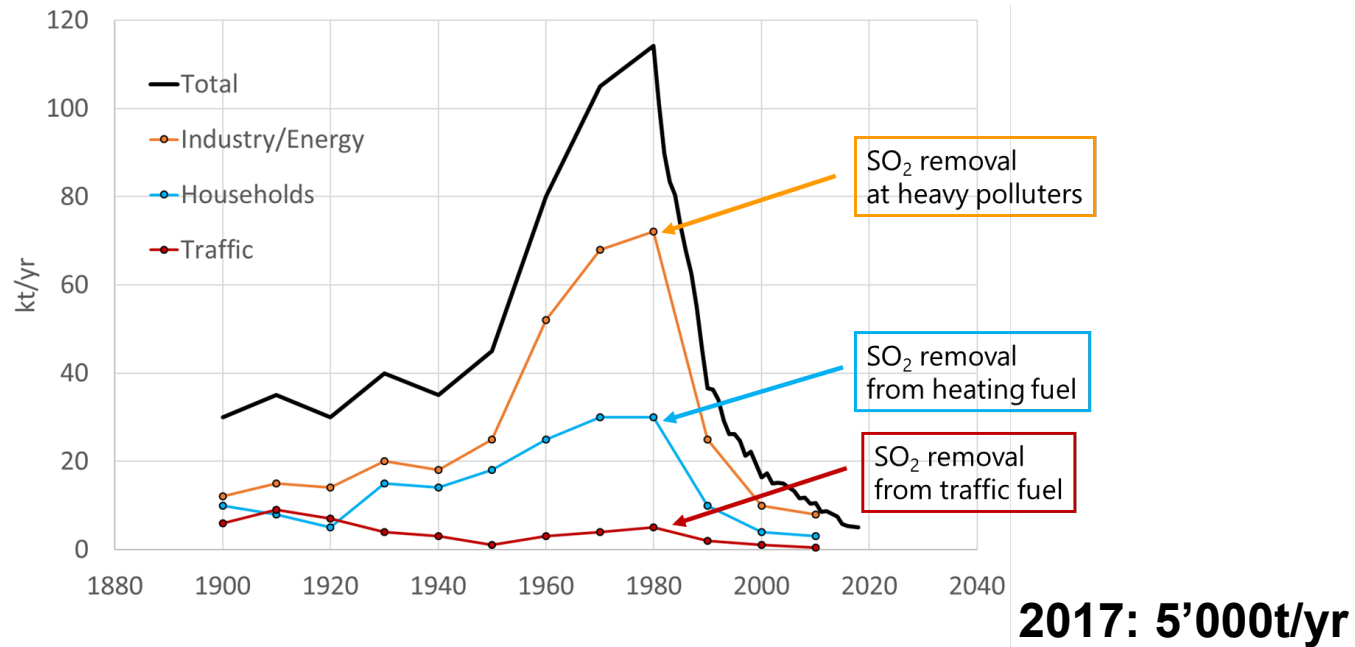
^d May only be exceeded once per year.

^e Average of the maximum daily 8h mean value ozone concentrations in the six consecutive months with the highest six-month average for ozone concentration. For Switzerland, this corresponds to April to September.

SO₂ concentrations in Switzerland (1980-2023)



Do-it yourself: SO₂ in Switzerland



Back-of-the envelope calculation

Assumption:
Lifetime: 3 days
(all emissions are replaced)

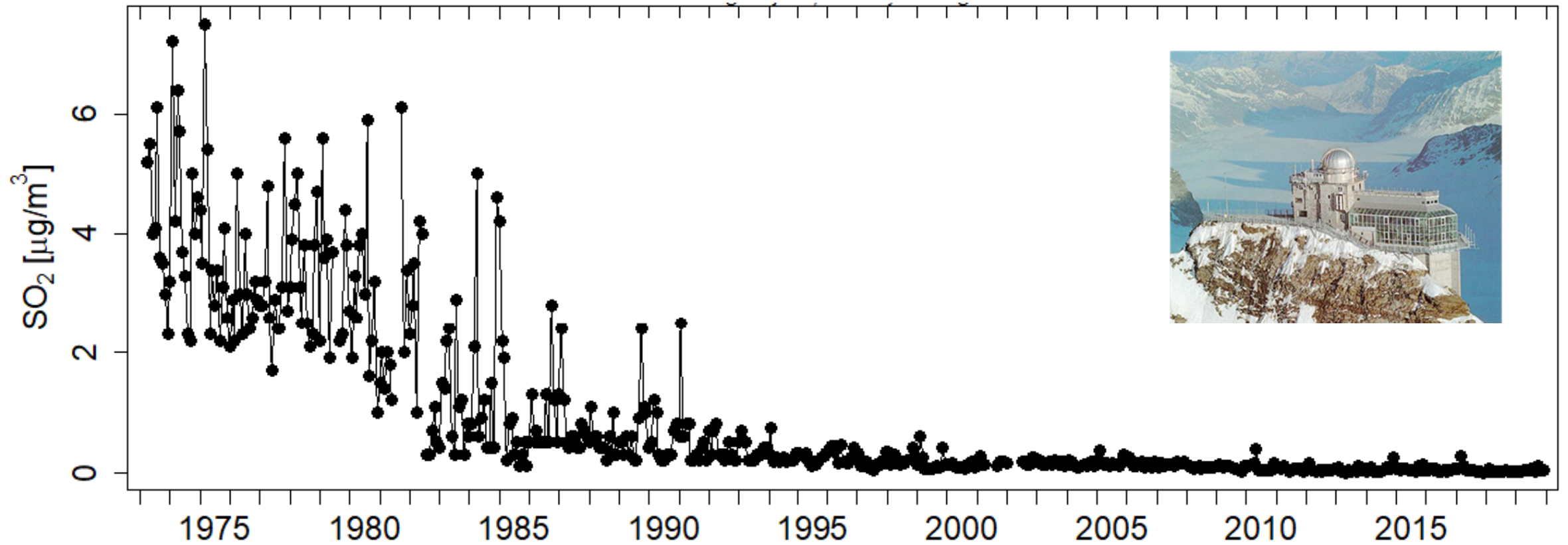
Mixing height: 1 km

Area CH: 42'000 km²

Concentration in $\mu\text{g}/\text{m}^3$?
1985 and 2017

Effect of restrictions on SO₂ in the atmosphere

Jungfraujoch 1973 – 2019



Overview of measurement-techniques for tropospheric trace compounds

	Compound	Monitor	GC – (MS)	Laser	Satellite
air pollutants in OPair	SO ₂	XX			X
	NO _x	XX			X
	CO	XX	X	X	X
	O ₃ (Troposph.)	XX			(X)
	Particles	X			(X)
	VOCs	(X)	X		
greenhouse gases in Kyoto Protocol	CO ₂	XX	X	XX	(X)
	CH ₄	X	XX	XX	X
	N ₂ O		XX	X	(X)
	Halocarbons*		X		(X)

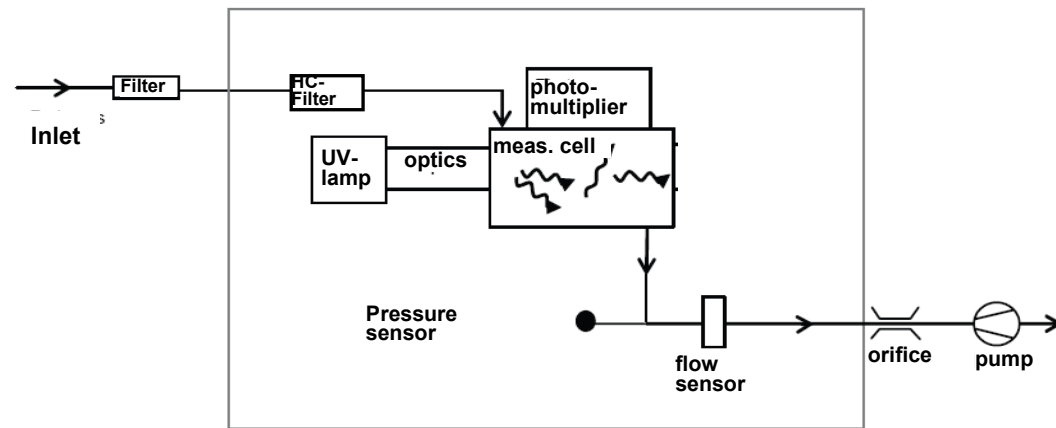
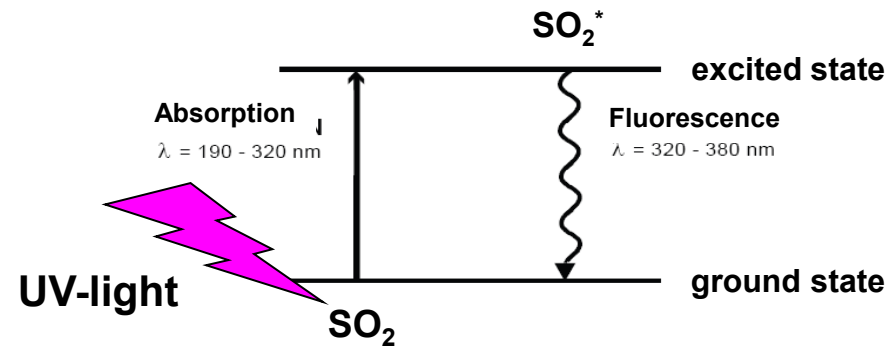
XX: best available method
X: good method
(X): possible

* also causing ozone depletion

Measurements of SO₂ in the atmosphere

Measurement principle: UV-fluorescence (monitor)

UV-fluorescence (monitor)
APSA-370



Swiss Legal Air Pollution Limits

Luftreinhalteverordnung bestehend und neu empfohlen

Schadstoff	Mittelungszeit	WHO AQG 2021	LRV IGW derzeit	Empfehlung EKL 2023
Schwefeldioxid (SO₂), µg/m³ (siehe Kapitel 8)	Jahresmittelwert und neu Mittelwert über das Winterhalbjahr	–	30 ^a	20 ^b
	95 % der ½-h-Mittelwerte eines Jahres	–	100	streichen
	24-h-Mittelwert	40 ^c	100 ^d	40 ^c
Stickstoffdioxid (NO₂), µg/m³ (siehe Kapitel 7)	Jahresmittelwert	10	30	10
	95 % der ½-h-Mittelwerte eines Jahres	–	100	streichen
	24-h-Mittelwert	25 ^c	80 ^d	25 ^c
Kohlenmonoxid (CO), mg/m³ (siehe Kapitel 9)	24-h-Mittelwert	4 ^c	8 ^d	4 ^c
Ozon (O₃), µg/m³ (siehe Kapitel 6)	Sommersaison ^e	60	–	60
	98 % der ½-h-Mittelwerte eines Monats	–	100	100
	8-h-Mittelwert	100 ^c	–	–
	1-h-Mittelwert	–	120 ^d	120 ^d
Schwebstaub / Feinstaub (PM₁₀), µg/m³ (siehe Kapitel 4)	Jahresmittelwert	15	20	15
	24-h-Mittelwert	45 ^c	50 ^c	45 ^c
Schwebstaub / Feinstaub (PM_{2.5}), µg/m³ (siehe Kapitel 5)	Jahresmittelwert	5	10	5
	24-h-Mittelwert	15 ^c	–	15 ^c

^a Immissionsgrenzwert, welcher auch den Schutz von Tieren und Pflanzen, ihren Lebensgemeinschaften und Lebensräumen nach USG Art. 1 Abs. 1 einschliesst und dem Stand des Wissens zum Zeitpunkt des Erlasses der Luftreinhalte-Verordnung im Jahre 1985 entspricht.

^b Wert der WHO-Luftqualitätsleitlinien 2000 (WHO, 2000), welcher zum Schutz von Wäldern und weiteren naturnahen Ökosystemen festgelegt wurde. Er gilt als Jahresmittelwert sowie auch für das Winterhalbjahr. (Oktober–März).

^c 99. Perzentil (d. h. 3 Überschreitungen pro Jahr sind zulässig).

^d Darf höchstens einmal pro Jahr überschritten werden.

^e Durchschnitt der maximalen täglichen 8-h-Mittelwerte der Ozon-Konzentration in den sechs aufeinanderfolgenden Monaten mit der höchsten Ozon-Konzentration im Sechsmonatsdurchschnitt. Für die Schweiz entspricht dies April bis September.

Rationale for CO measurements

- CO is toxic to humans
- CO is an intermediate product of the VOC degradation on the way to CO₂ (the final product)
- CO is a good tracer for anthropogenic pollution (e.g. in Europe) and for biomass burning (e.g. in Africa)
- CO sources are known well => relative emission of other anthropogenic pollutants can be estimated

Carbon monoxide (CO): global budget

Sources [Tg/y] (Duncan et al., 2007)

Fossil fuel combustion	476
Biofuel	189
Biomass burning	512
Oxidation of methane	812
Oxidation of hydrocarbons	177
Total	2166

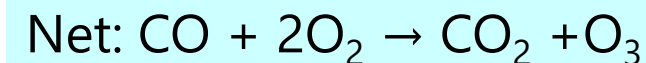
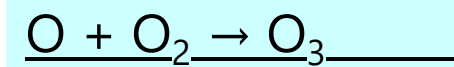
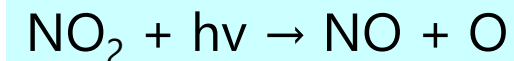
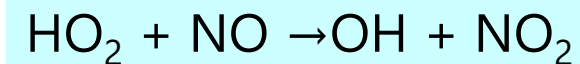
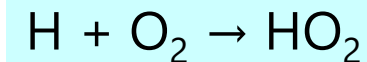
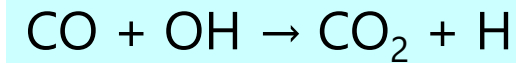
Sinks (approx.)

Oxidation by OH	78%
Soil uptake	17%
Stratosphere	4%

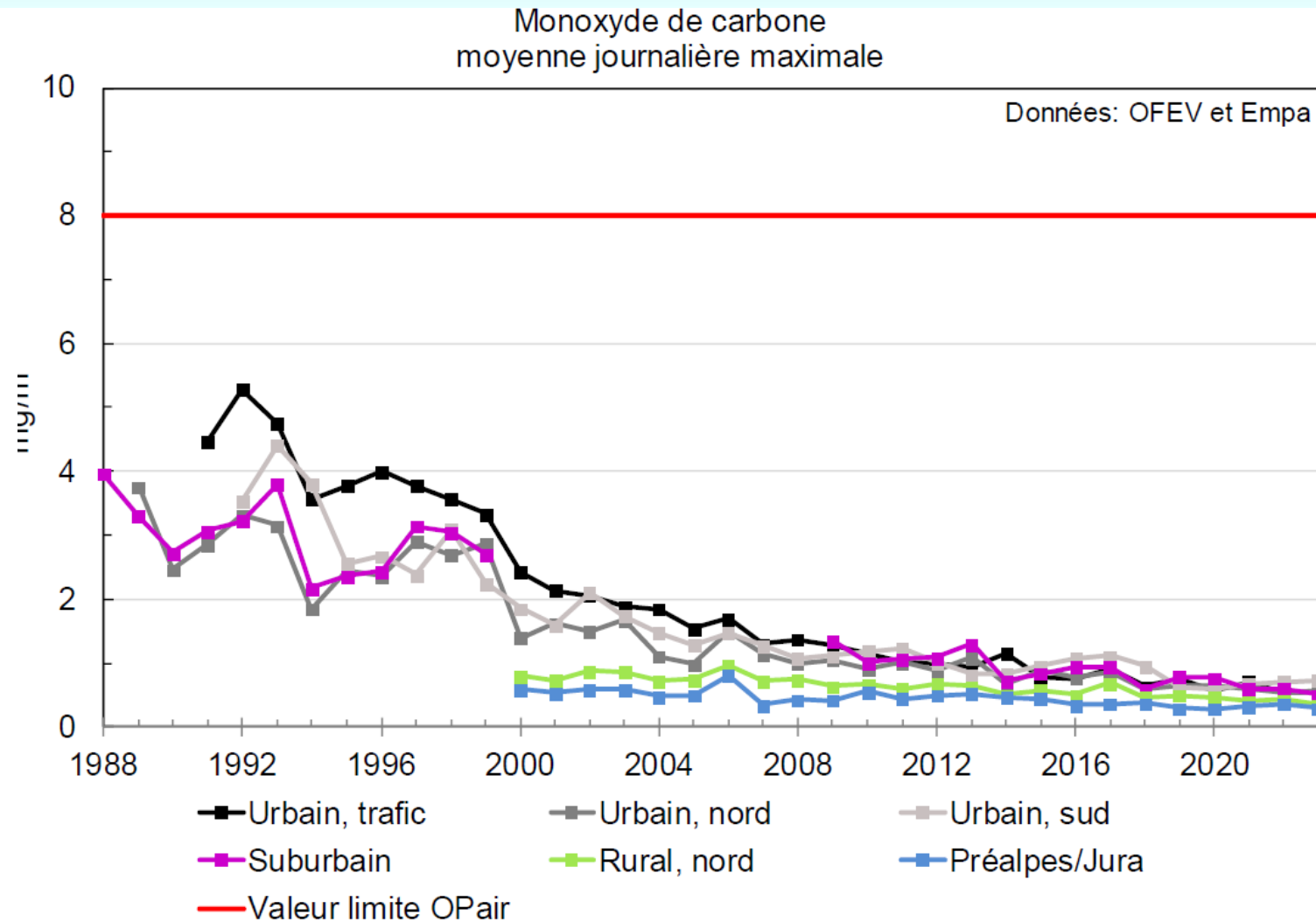
Lifetime:
2 months



CO oxidation by OH



CO: Switzerland (yearly averages)



Limit value: 24-h average should not exceed 8 $\mu\text{g}/\text{m}^3$

No exceedance in the last decades

Overview of measurement-techniques for tropospheric trace compounds

	Compound	Monitor	Gas Chrom.	Laser	Satellite
air pollutants in OPair	SO ₂	XX			X
	NO _x	XX		X	X
	CO	XX		XX	X
	O ₃ (Troposph.)	XX			(X)
	Particles	X		X	(X)
greenhouse gases in Kyoto Protocol	VOCs	(X)	XX		
	CO ₂	XX		XX	X
	CH ₄		X	XX	X
	N ₂ O		X	XX	(X)
	Halocarbons		XX		(X)

XX: best available method
X: good method
(X): possible

CO: measurement principles

- 1. NDIR Spectrometry** (Non-dispersive infrared)
The absorbed amount of light is direct proportional to the CO mixing ratio
- 2. Quantum cascade laser/cavity ring-down laser**

CO and CO₂ measurement principles

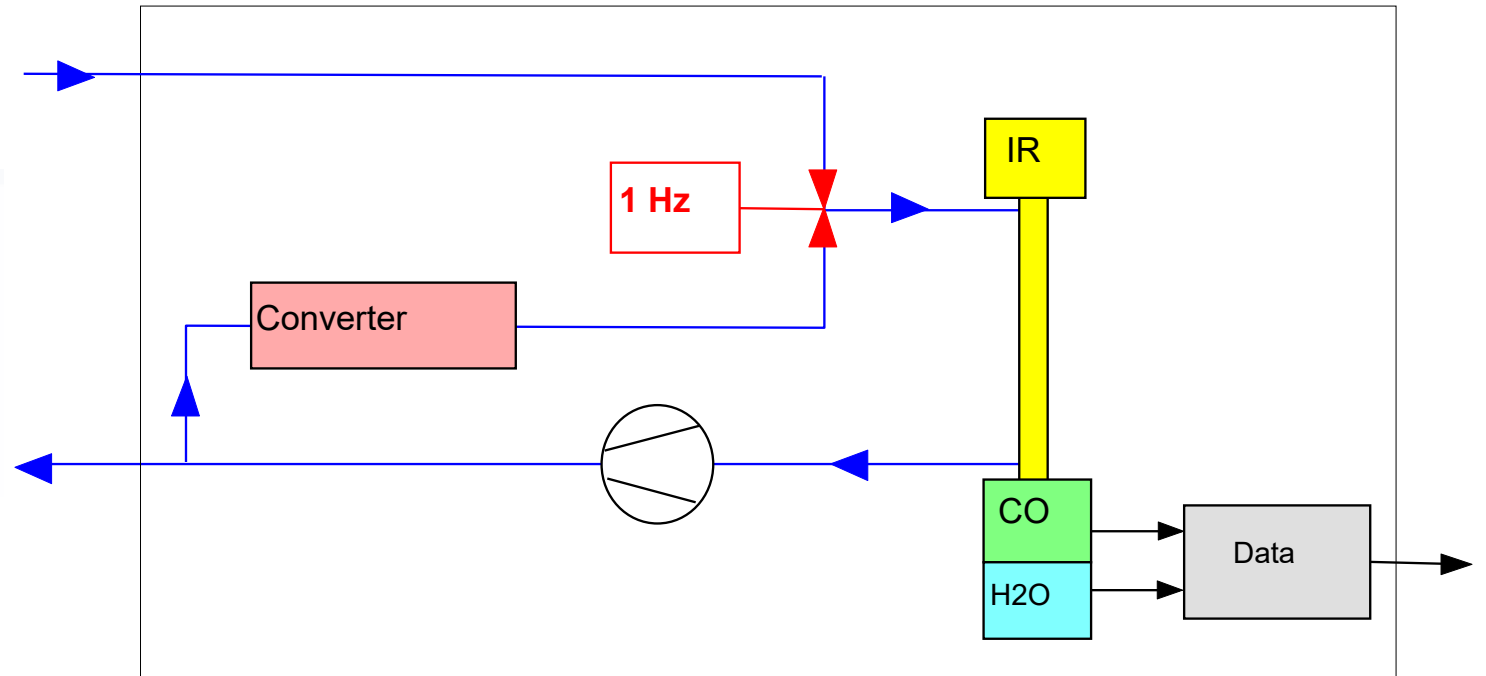
1. NDIR: Non-dispersive infra red



Horiba APMA-360

Linear, robust

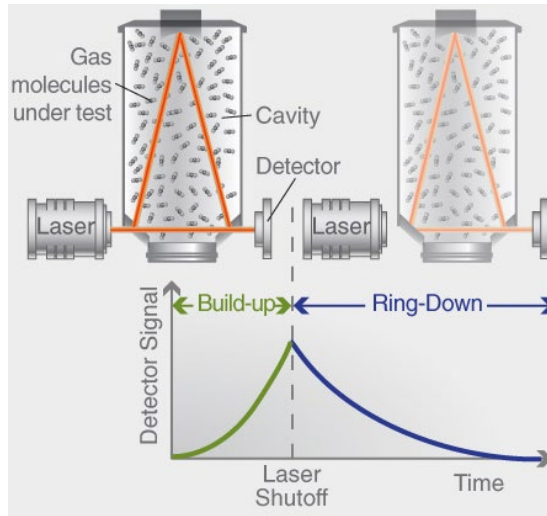
Linearity: 0-100 ppm
Detection limit: ~10 ppb
needs zeroing about every week



2. Laser-based: Cavity-ring down laser (CRD)



Picarro G1302

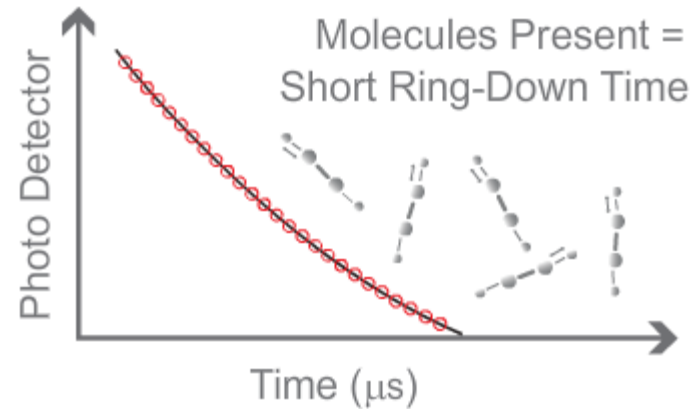
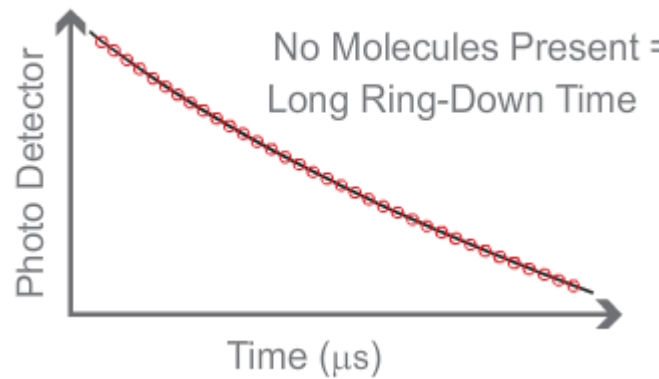


detection limit (1s) <2 ppb

linearity 0-1000 ppm

time resolution 5s – 5 min.

time coverage integral



CO before and after Corona

