

# Environmental transport phenomena: Lecture II

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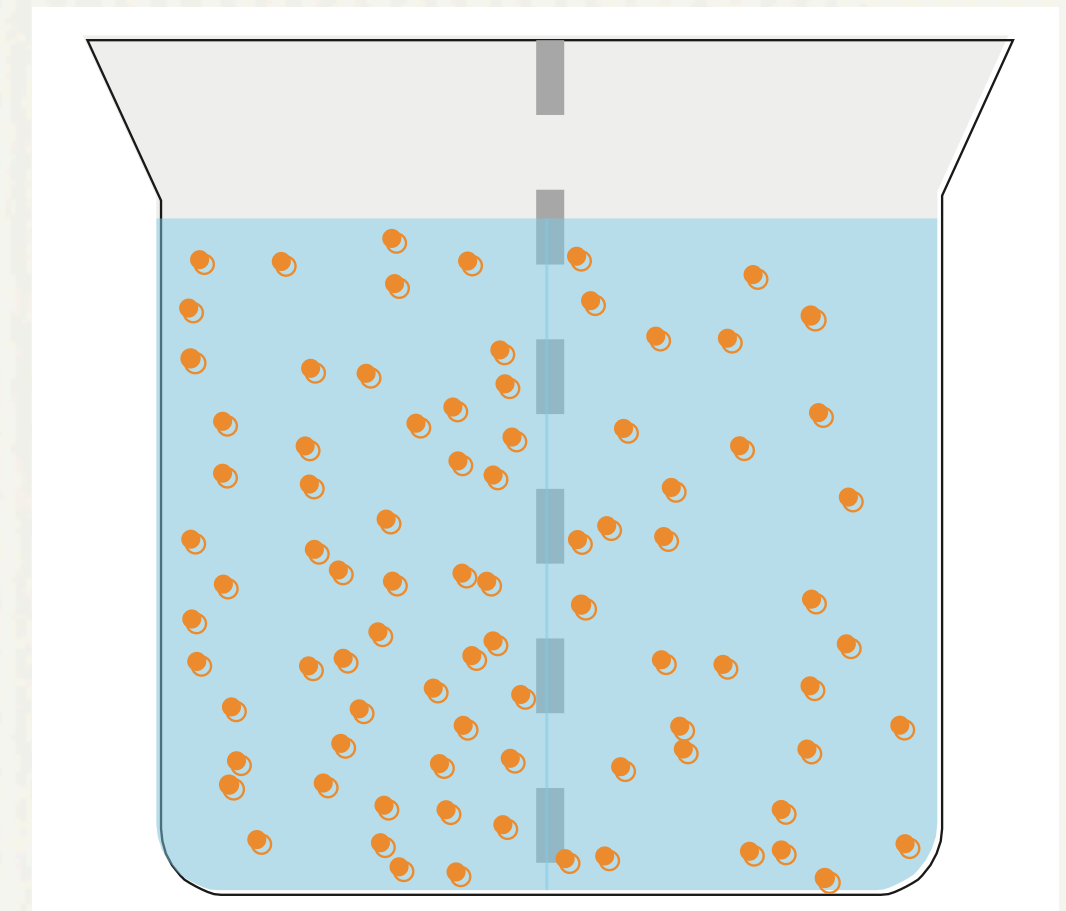
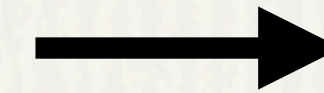
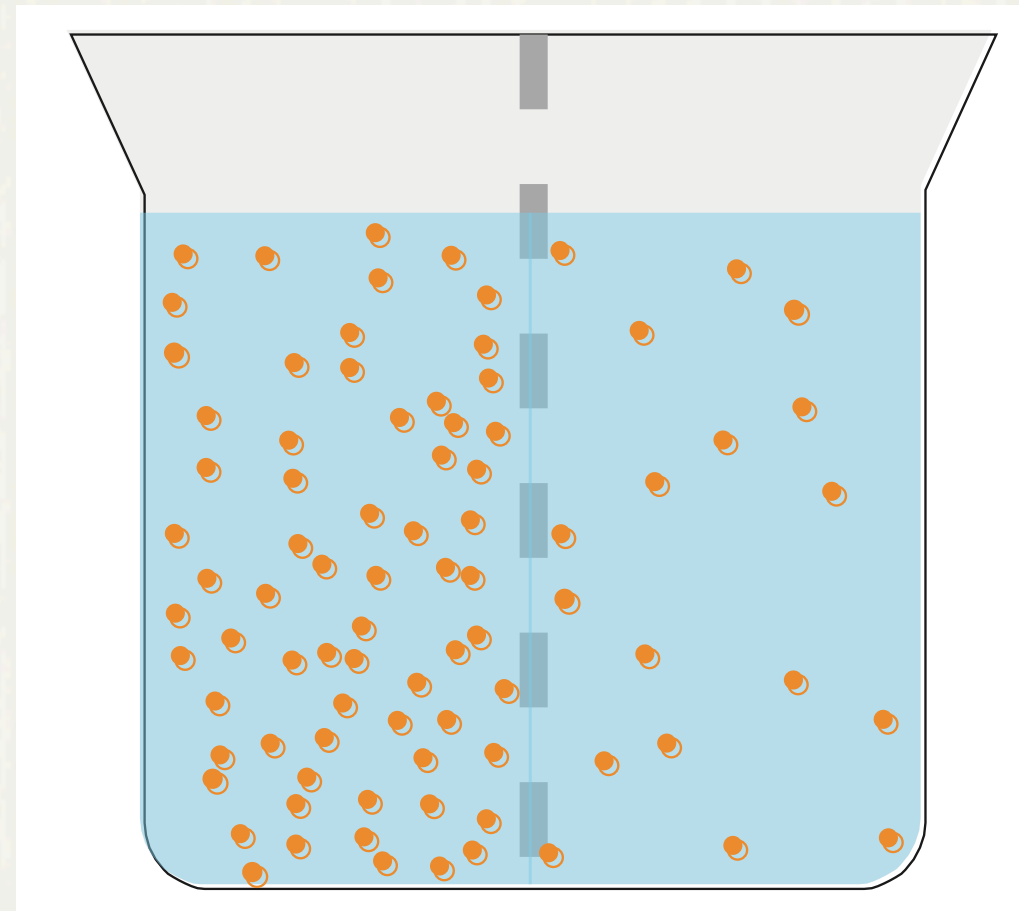
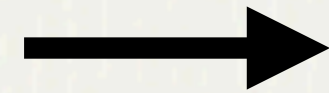
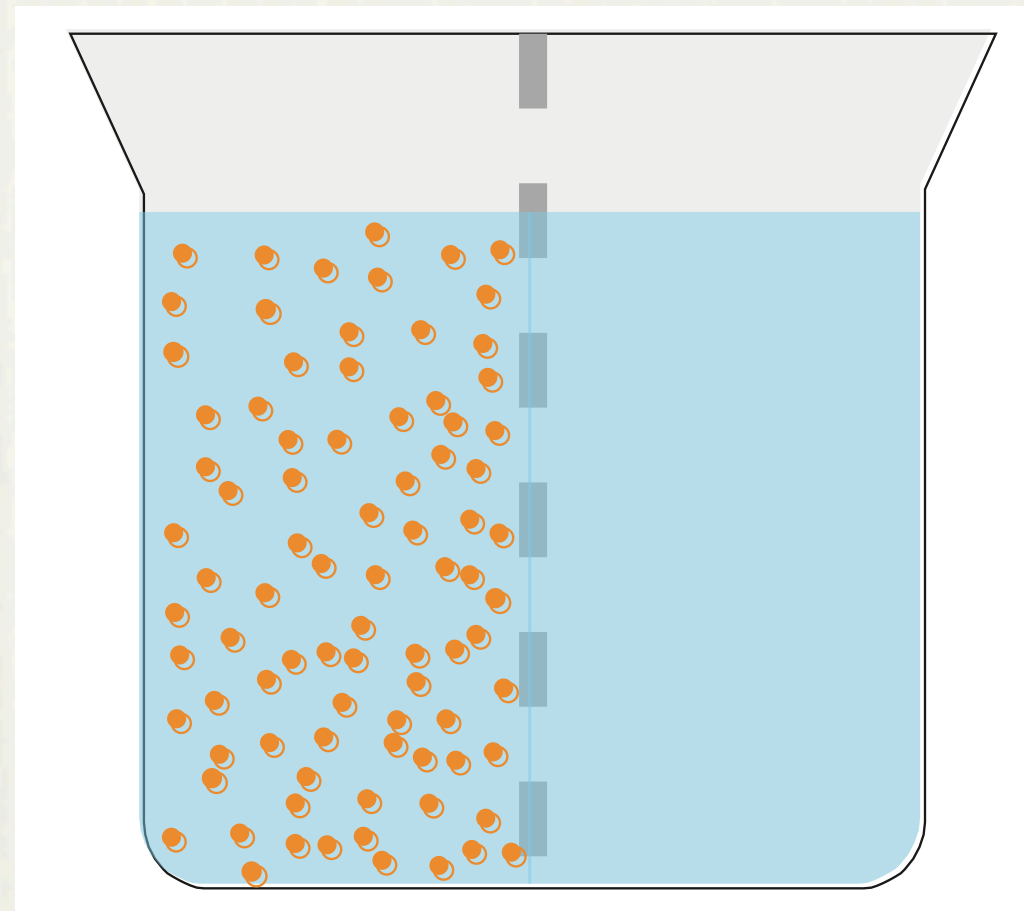


# Diffusion equation

Diffusion-like equations can be found in **different contexts**:

► diffusion of substances

$$\frac{\partial C(x, t)}{\partial t} = D \frac{\partial^2 C(x, t)}{\partial x^2}$$

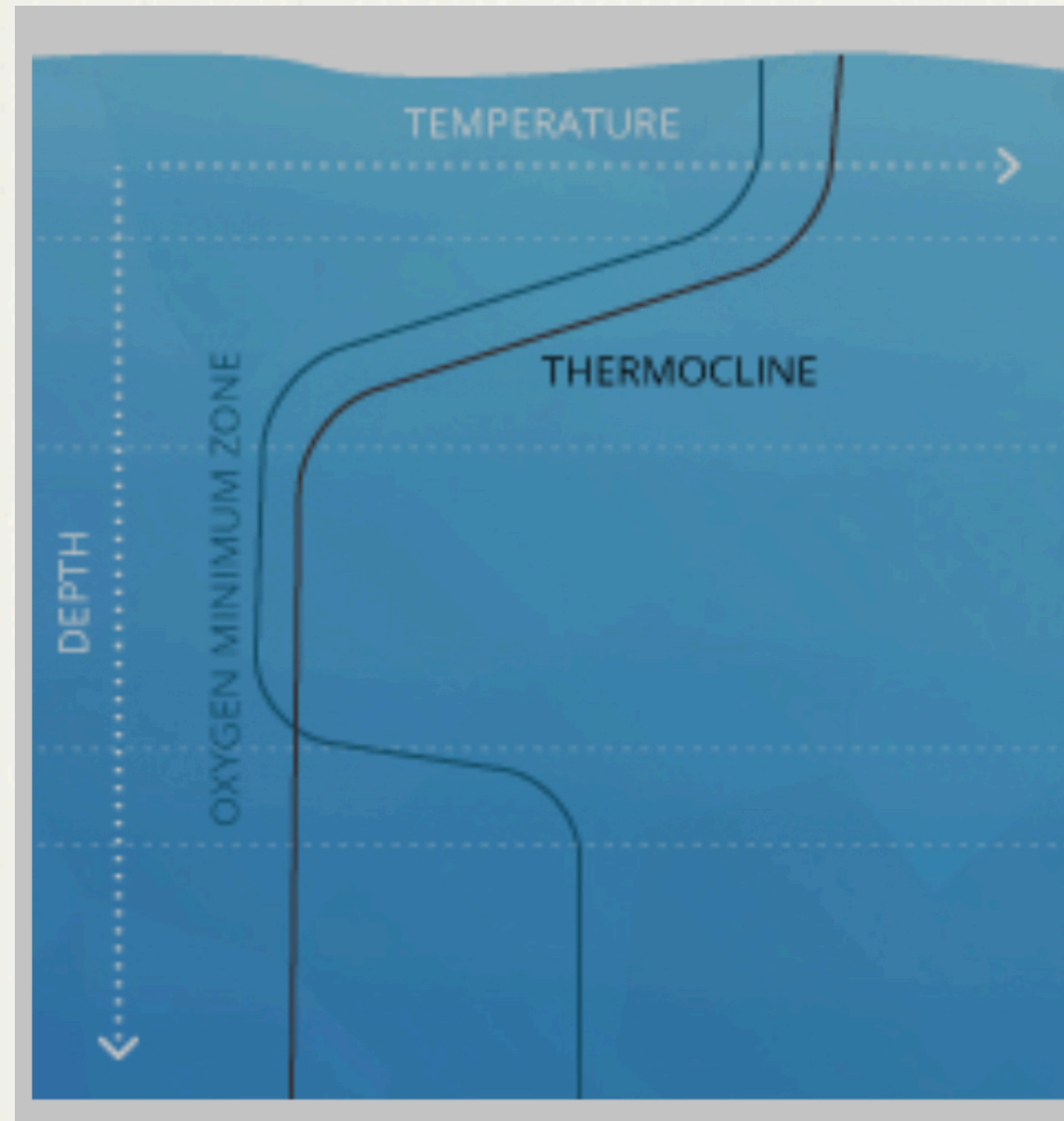




# Diffusion equation

Diffusion-like equations can be found in **different contexts**:

- diffusion of substances: oxygen diffusion in a water body

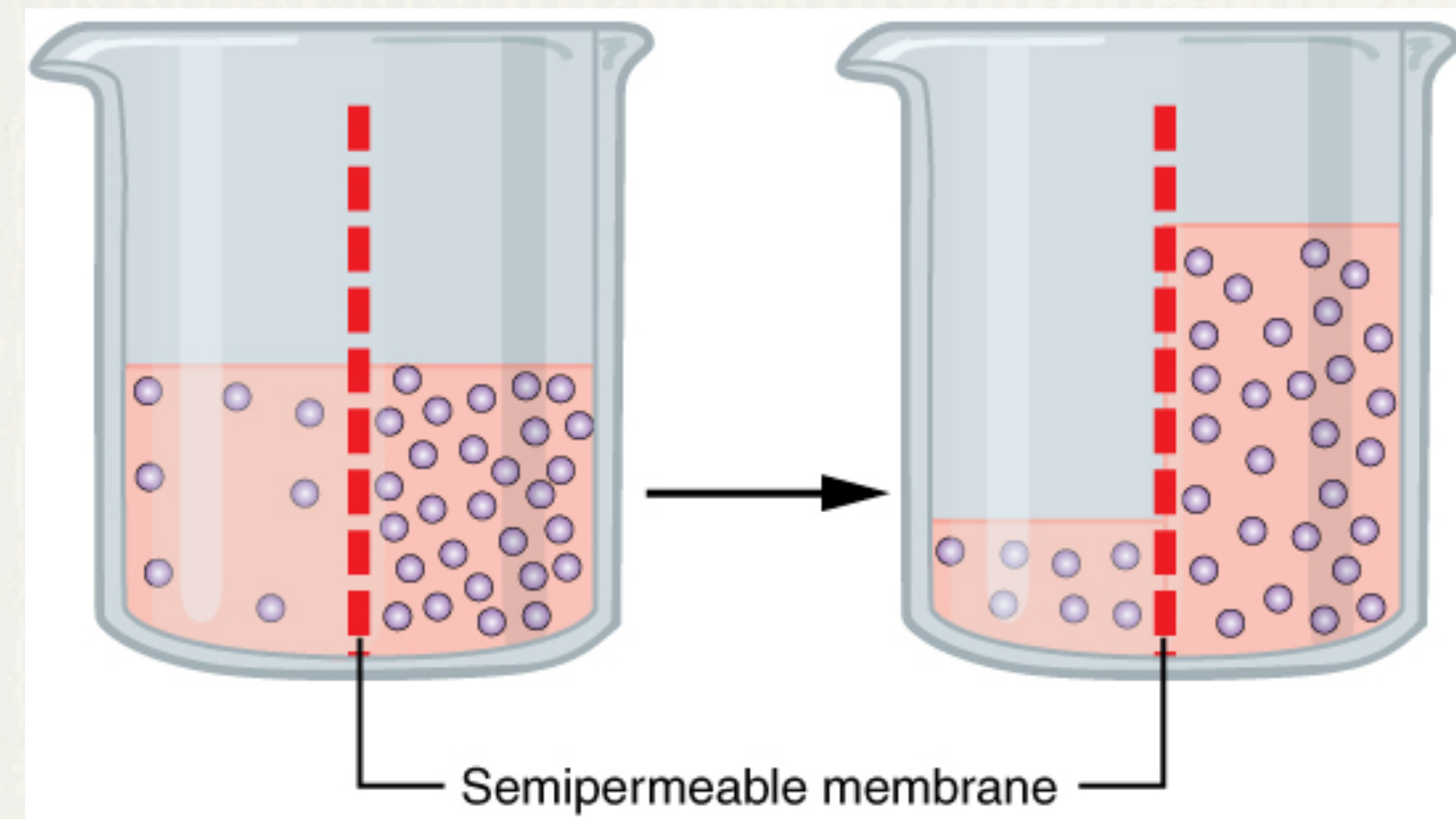
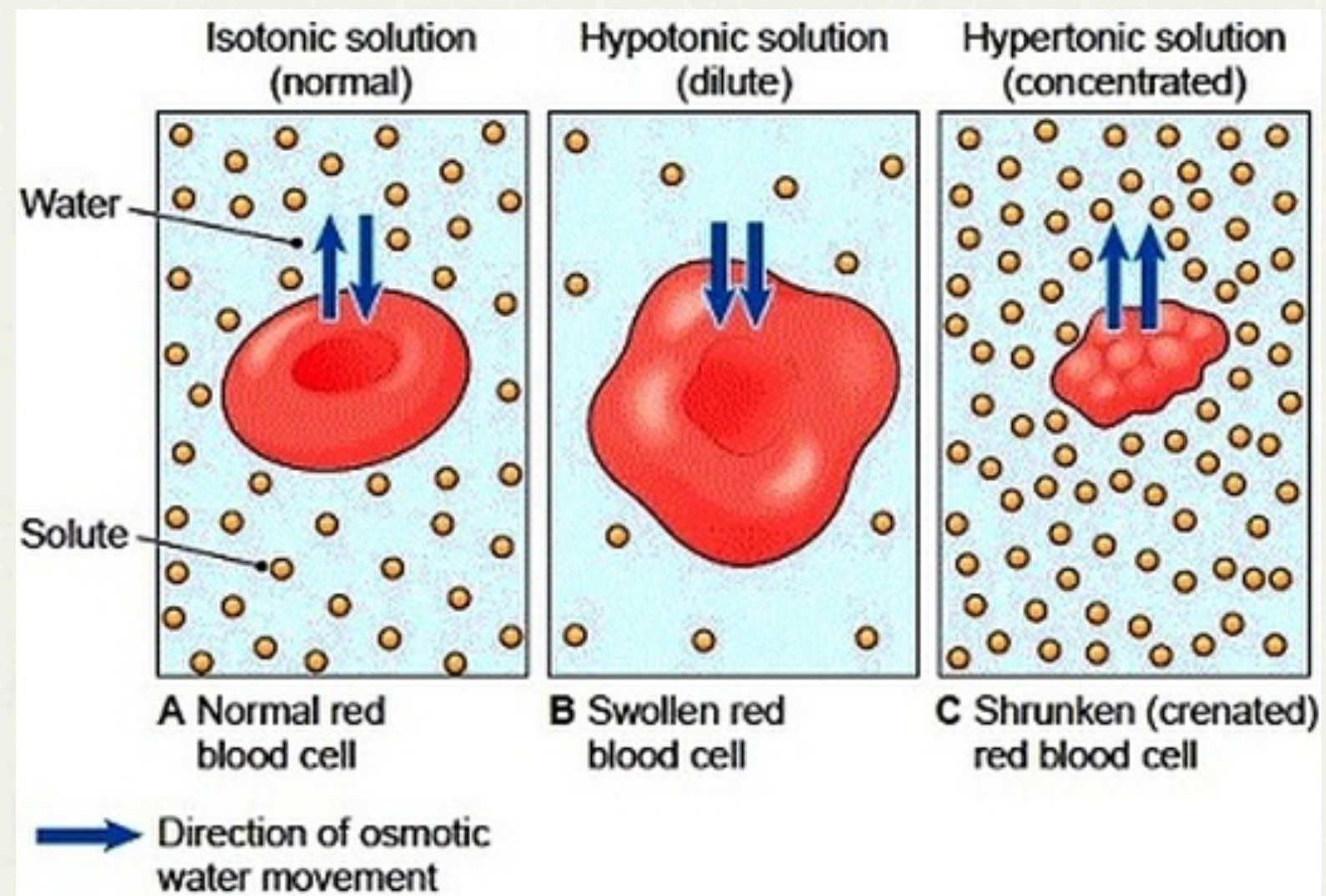




# Diffusion equation

Diffusion-like equations can be found in **different contexts**:

► diffusion of substances: osmosis





# Diffusion equation

Diffusion-like equations can be found in **different contexts**:

► Navier-Stokes: **momentum** diffusion

$$\rho \left( \frac{\partial \mathbf{v}}{\partial t} + (\mathbf{v} \cdot \nabla) \mathbf{v} \right) = -\nabla p + \boxed{\mu \nabla^2 \mathbf{v}} + \mathbf{f}$$

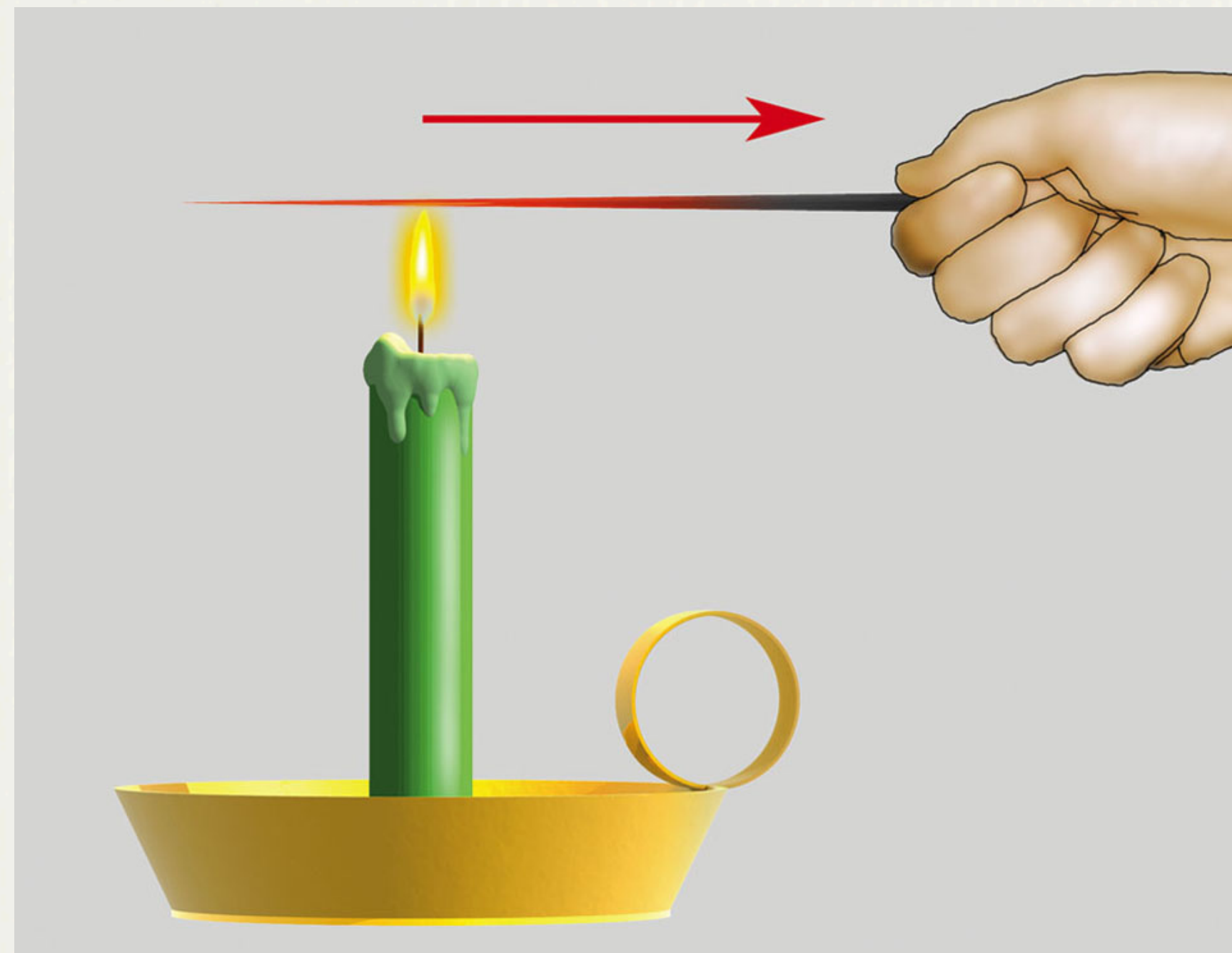




# Diffusion equation

Diffusion-like equations can be found in **different contexts**:

► **Heat** equation  $\frac{\partial u}{\partial t} = \alpha \nabla^2 u$      $u$  is the temperature and  $\alpha$  the thermal diffusivity

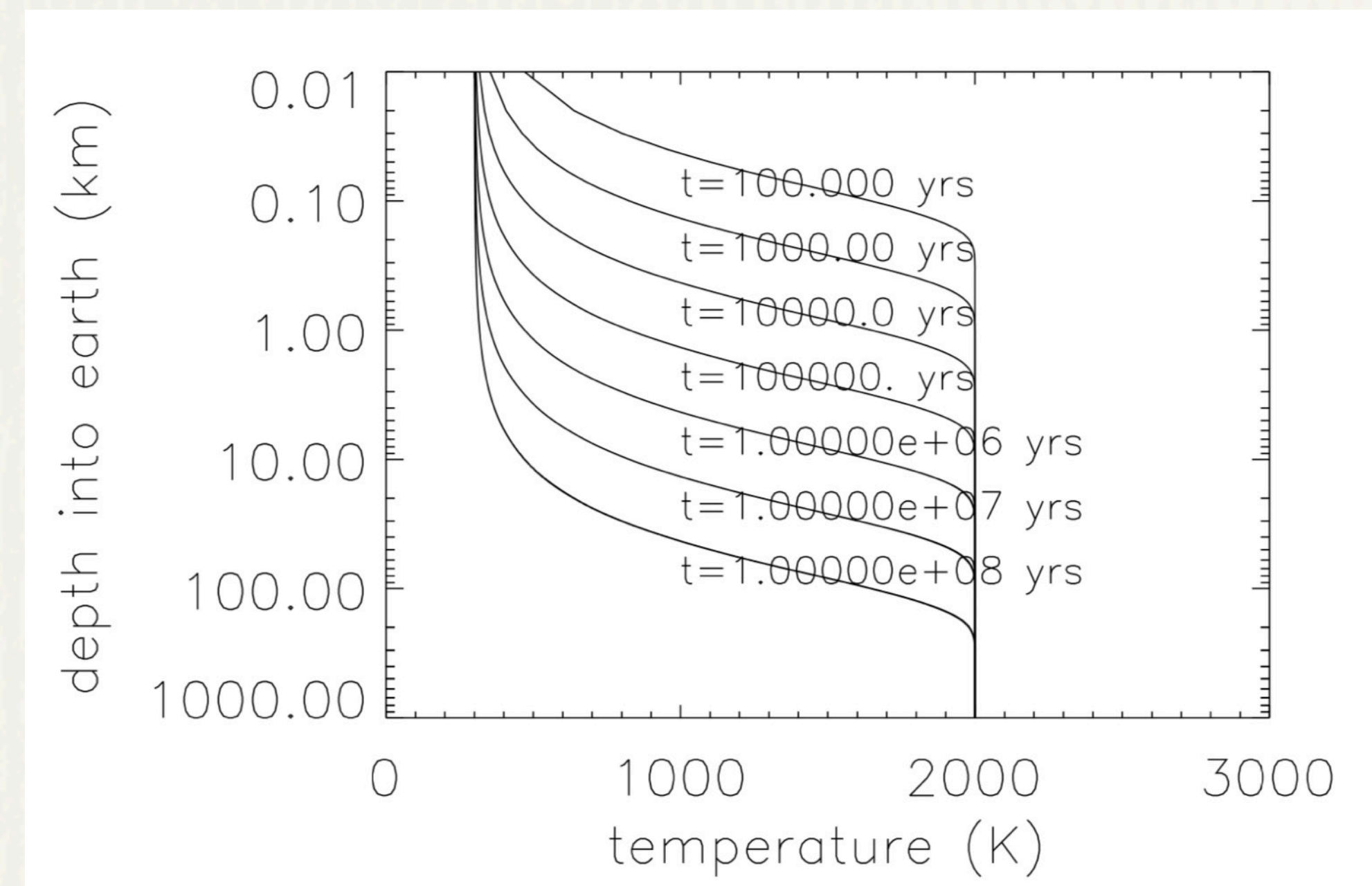
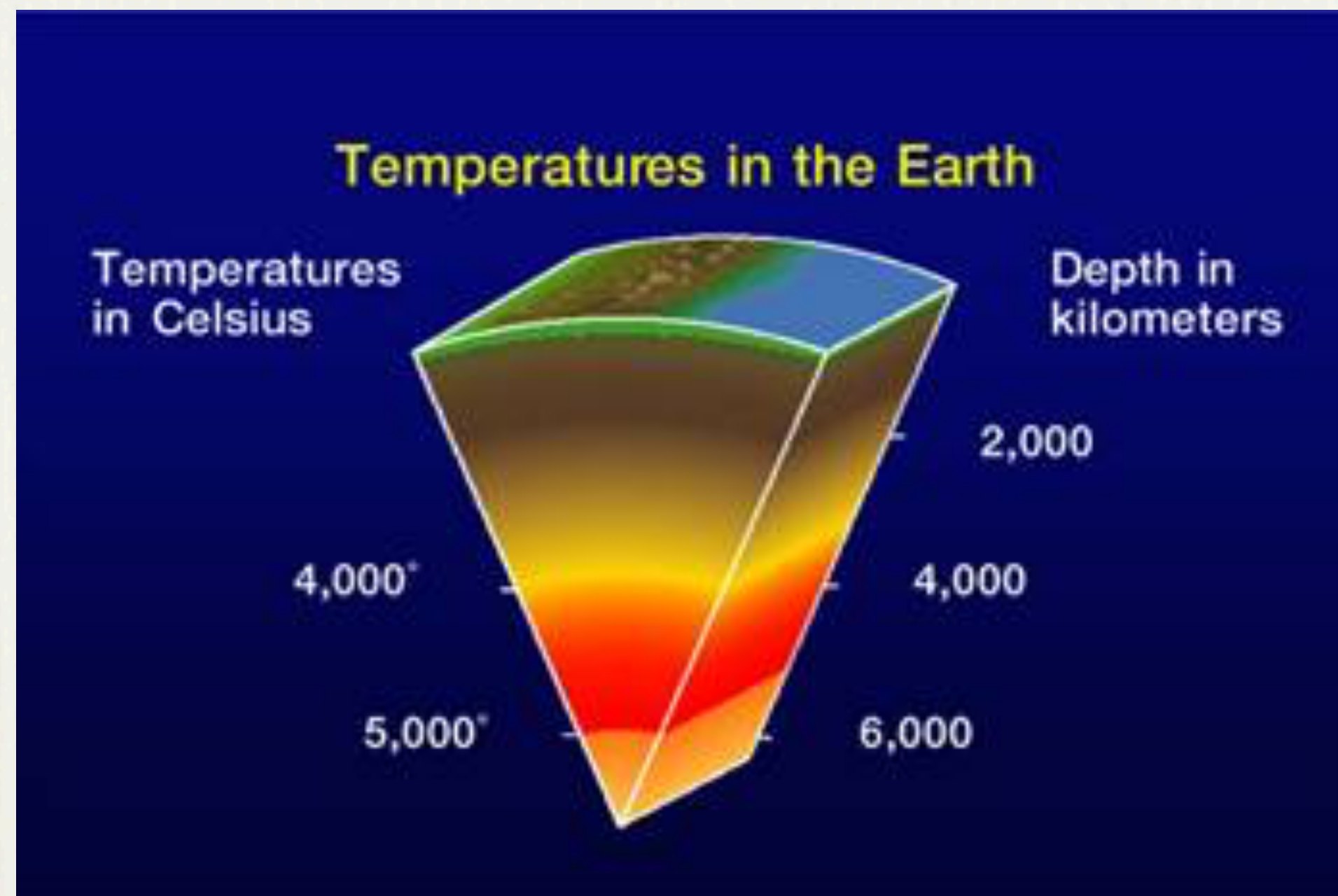




# Diffusion equation

Diffusion-like equations can be found in **different contexts**:

► **Heat** equation: cooling of the earth





# Diffusion equation

Diffusion-like equations can be found in **different contexts**:

► Schrödinger equation (free particle)  $i\hbar \frac{\partial \psi}{\partial t} = -\frac{\hbar^2}{2m} \nabla^2 \psi$       Diffusion in imaginary time





# Diffusion equation (1D)

The diffusion equation needs to be supplemented with one **initial** condition and two **boundary** conditions. The complete problem takes the form:

► evolution equation

$$\frac{\partial C(x, t)}{\partial t} = D \frac{\partial^2 C(x, t)}{\partial x^2}$$

► domain boundary concentrations

$$C(x_{1,2}, t) = C_{1,2} \quad \forall t$$

**or** fluxes

$$\left. \frac{\partial C(x, t)}{\partial x} \right|_{x=x_{1,2}} = C'_{1,2} \quad \forall t$$

domain boundaries:  $x_{1,2}$

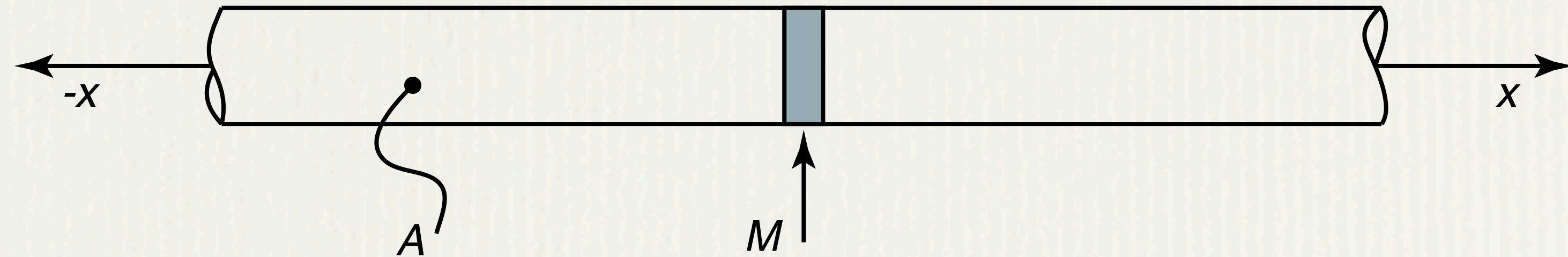
► Initial concentration distribution

$$C(x, 0) = f(x)$$



# 1D diffusion equation: point source solution

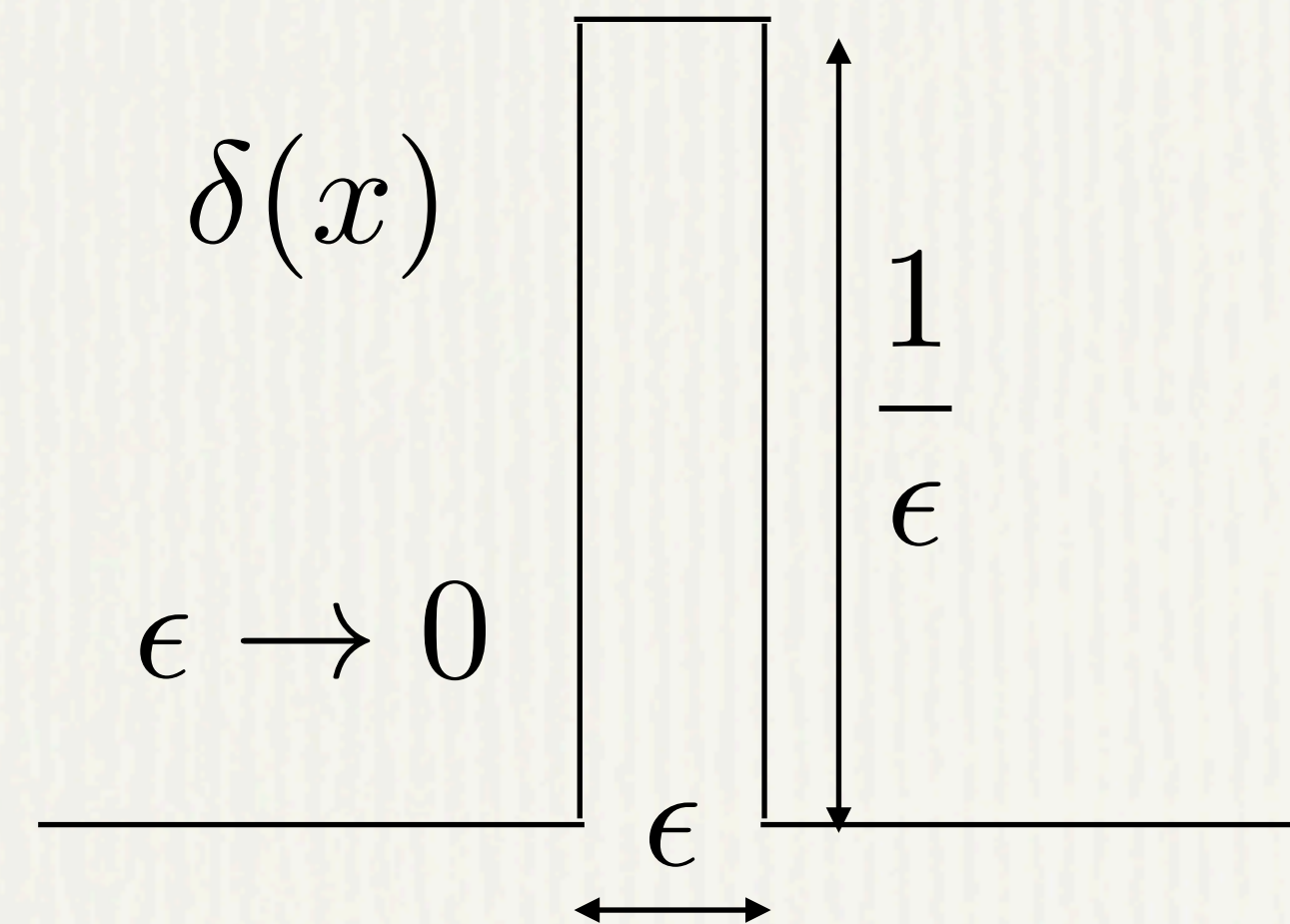
- We assume an **infinite** 1D domain and a **point** release (e.g. pipe with uniform concentration over the section A)



$$\frac{\partial C(x, t)}{\partial t} = D \frac{\partial^2 C(x, t)}{\partial x^2}$$

$$C(x, t = 0) = \frac{M}{A} \delta(x)$$

$$C(x = \pm\infty, t) = 0$$



- **Dirac** distribution  $\delta(x)$  satisfies (intuition: finite mass in a point):

$$\int \delta(x) dx = 1$$

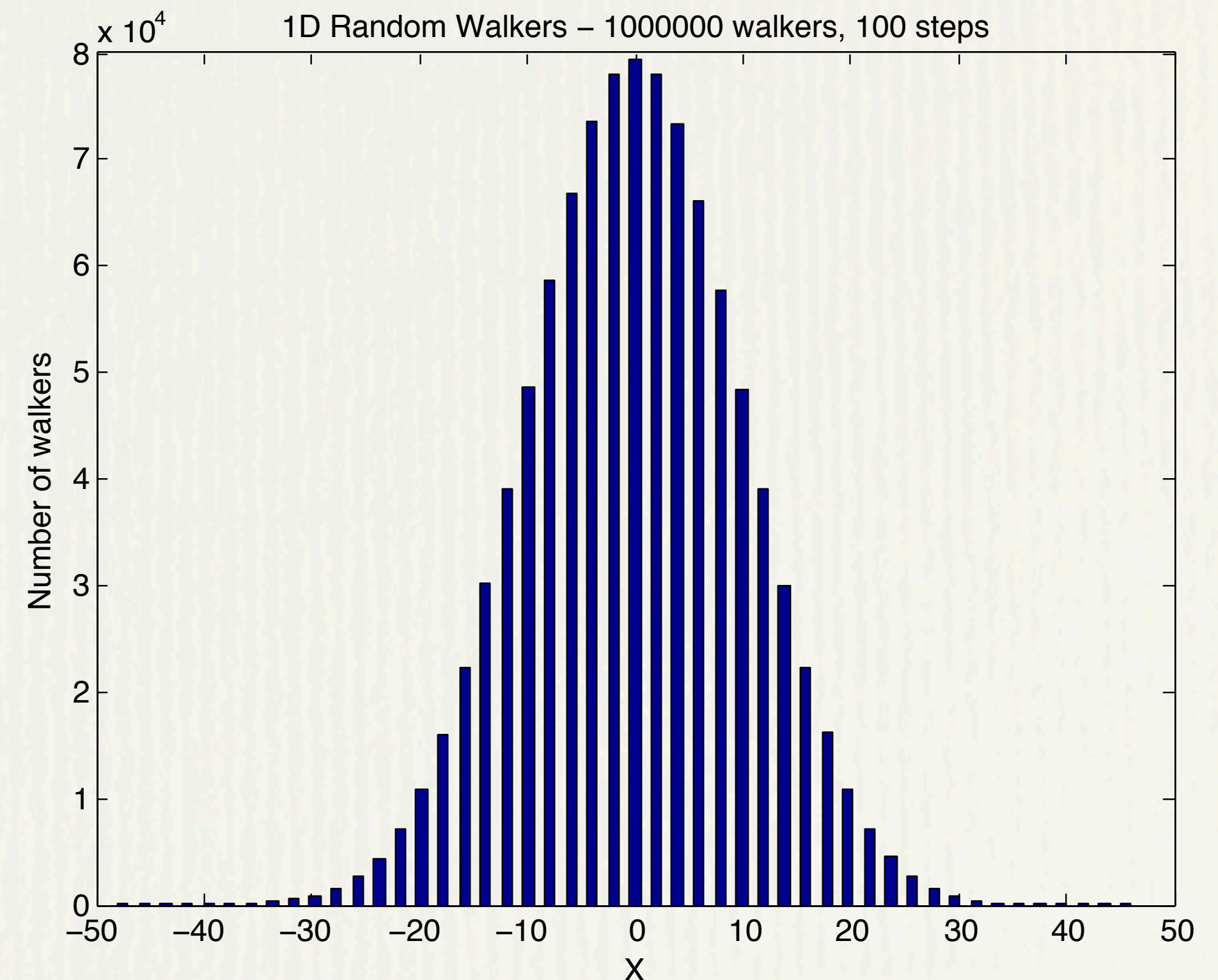
$$\int \delta(x - x_0) f(x) dx = f(x_0)$$



# 1D diffusion equation: point source solution

$$C(x, t) = \frac{M}{A\sqrt{4\pi Dt}} e^{-\frac{x^2}{4Dt}} \quad \sigma(t) = \sqrt{2Dt}$$

- Results known from normal distribution can be used: 68% of the particles found within one standard deviation, 95% within two...



- Generalisation to a point source located at position  $x=\mu$  is immediate

$$C(x, t) = \frac{M}{A\sqrt{4\pi Dt}} e^{-\frac{(x-\mu)^2}{4Dt}}$$