

# Molecular quantum dynamics: Exercise series 4

## Problem 1: Energy of a coherent state

Consider the coherent state of the harmonic oscillator

$$\hat{H} = \frac{\hat{p}^2}{2m} + \frac{1}{2}m\omega^2\hat{x}^2,$$

i.e., a Gaussian wavepacket

$$\psi(x, t) = N \exp \left\{ \frac{i}{\hbar} \left[ \frac{1}{2} \alpha_t (x - q_t)^2 + p_t (x - q_t) + \gamma_t \right] \right\}$$

with a constant width parameter  $\alpha_t = \alpha_0 = im\omega$ . Show that the expectation value of energy in the state  $\psi(t)$  satisfies

$$E(t) := \langle \psi(t) | \hat{H} | \psi(t) \rangle = H(q_t, p_t) + c\hbar\omega$$

and find the constant  $c$ . Note:  $H(q_t, p_t)$  is the *classical* energy of the center of the wavepacket!

## Problem 2: Lagrangian and Hamiltonian for a diatomic molecule (or the hydrogen atom)

Consider a two-body problem, i.e., two particles of mass  $m_1$  and  $m_2$  interacting via a potential energy depending only on the distance,  $V(\mathbf{r}_1, \mathbf{r}_2) = V(|\mathbf{r}_2 - \mathbf{r}_1|)$ . An example is a diatomic molecule in the Born-Oppenheimer approximation, where the two particles are two nuclei and  $V$  is the Born-Oppenheimer potential energy surface. Another example is the hydrogen atom, where the two particles are the proton and electron,  $m_1 = m_p$  and  $m_2 = m_e$ , and  $V = -\frac{e^2}{4\pi\epsilon_0|\mathbf{r}_2 - \mathbf{r}_1|}$  is the Coulomb interaction. Because the potential energy depends only on the distance of the two particles, it is convenient to use new coordinates  $\mathbf{r} := \mathbf{r}_2 - \mathbf{r}_1$  (the relative coordinate) and  $\mathbf{R} := \frac{m_1\mathbf{r}_1 + m_2\mathbf{r}_2}{m_1 + m_2}$  (the *center-of-mass* coordinate). This is the simplest example of so-called *Jacobi coordinates*.

a) Kinetic energy in the original coordinates is  $T = \frac{1}{2}m_1\dot{\mathbf{r}}_1^2 + \frac{1}{2}m_2\dot{\mathbf{r}}_2^2$ . Express the kinetic energy in coordinates  $\dot{\mathbf{r}}$  and  $\dot{\mathbf{R}}$ . Simplify the result by using the total mass  $M := m_1 + m_2$  and the *reduced mass*  $\mu := \frac{m_1m_2}{m_1+m_2}$ . Express the Lagrangian  $L$  in coordinates  $\mathbf{r}, \mathbf{R}, \dot{\mathbf{r}}$ , and  $\dot{\mathbf{R}}$ .

b) Write down the Lagrange equations of motion for  $\mathbf{r}$  and  $\mathbf{R}$ . Solve the equation for the center of mass  $\mathbf{R}$ . Does it correspond to your intuition? (Equation for  $\mathbf{r}$  will be solved in the lecture.)

c) Using the definition from class, what are the momenta  $\mathbf{p}$  and  $\mathbf{P}$  conjugate to the coordinates  $\mathbf{r}$  and  $\mathbf{R}$ ? (Express them in terms of  $\dot{\mathbf{r}}$  and  $\dot{\mathbf{R}}$ .)

d) Write down the Hamiltonian  $H(\mathbf{r}, \mathbf{R}, \mathbf{p}, \mathbf{P})$ .

e) Write down Hamilton's equations of motion for  $\mathbf{r}, \mathbf{R}, \mathbf{p}$ , and  $\mathbf{P}$ . Solve the equations for  $\mathbf{R}$  and  $\mathbf{P}$ ; the solution should agree with part b). (Equations for  $\mathbf{r}$  and  $\mathbf{p}$  will be solved in the lecture.)

Note: In this homework, you have separated the center-of-mass motion from the relative motion. In the next lecture we will study the relative motion. In hydrogen atom, the reduced mass  $\mu \approx m_e$  since proton is much heavier than the electron, and in many texts this approximation is used and the center-of-mass motion is forgotten from the beginning. In more accurate calculations, one has to consider  $\mu$  instead of  $m_e$ .