

# **CH-413 Nanobiotechnology**

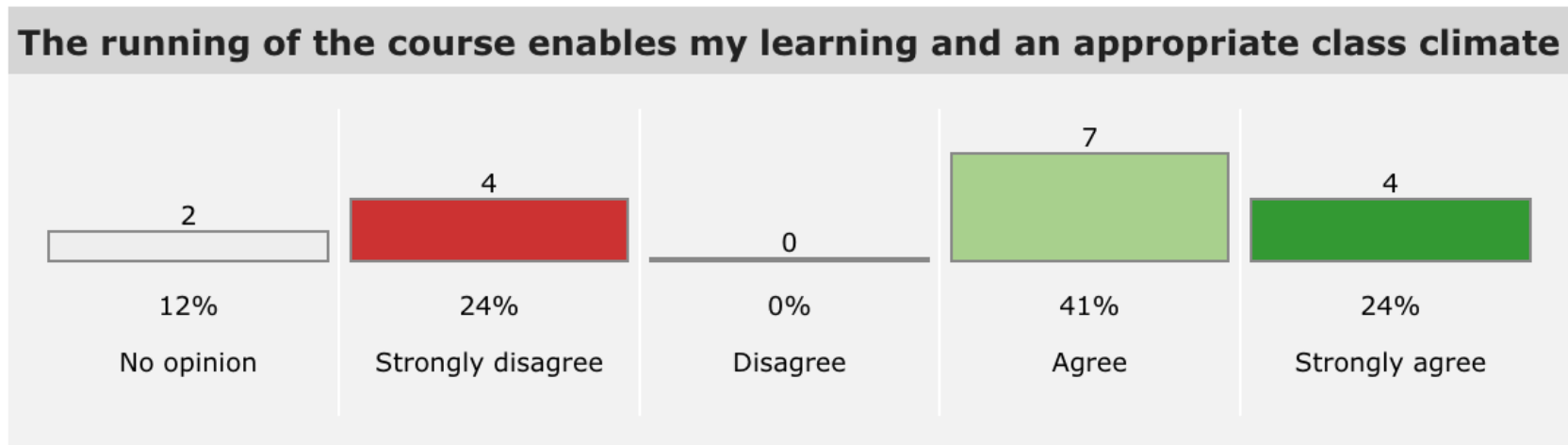
## **Self-assembly of nanoscale objects**

Angela Steinauer

April 17, 2025

# Indicative feedback

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- Research proposal: more help desired
- Release exercise sheets earlier
- More time between exercises and deadline to submit
- Workload is too high

Thank you for your feedback, it helps me improve the course.

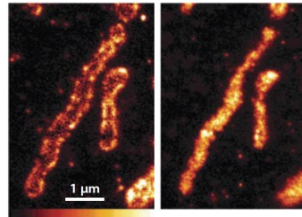
# Quick poll

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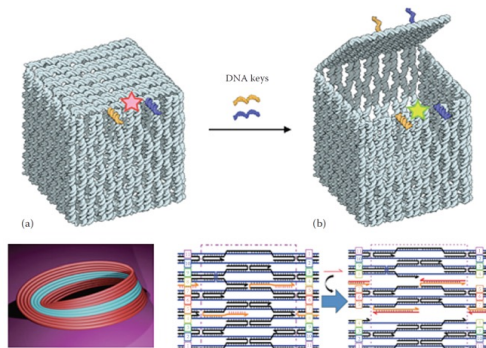
- Option 1: Exercises and final written exam (0)
- Option 2: Debates and final written exam (2)
- Option 3: Exercises and proposal (7)
- Option 4: Debates and proposal (11)
- (Option 5: Exercises and debates)
- (Option 6: Proposal and final written exam)

# Engineering of micro- and nanoscale objects

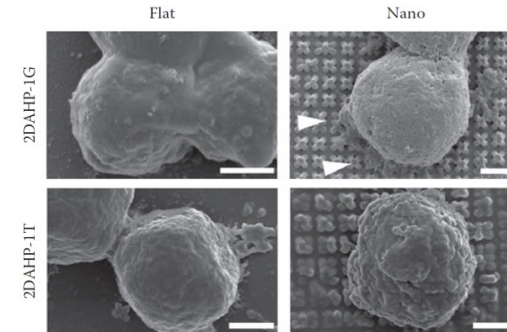
**Watch** molecular biology happen  
and **manipulate**  
the processes



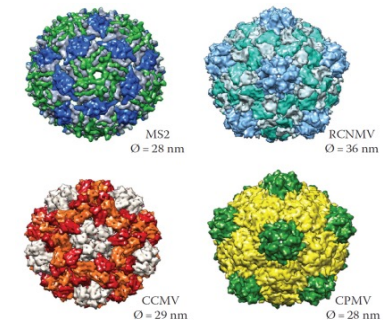
**Craft** new  
biomaterials



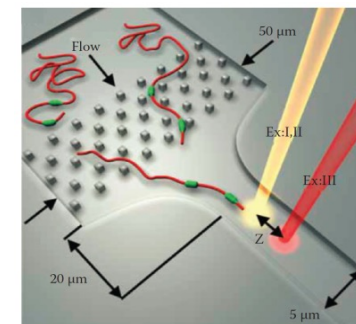
**Grow** cells and complex tissues  
*in vitro*



**Target** drugs to  
individual  
cancer cells



**Diagnose**  
diseases from  
single molecules  
or cells



# What is self-assembly

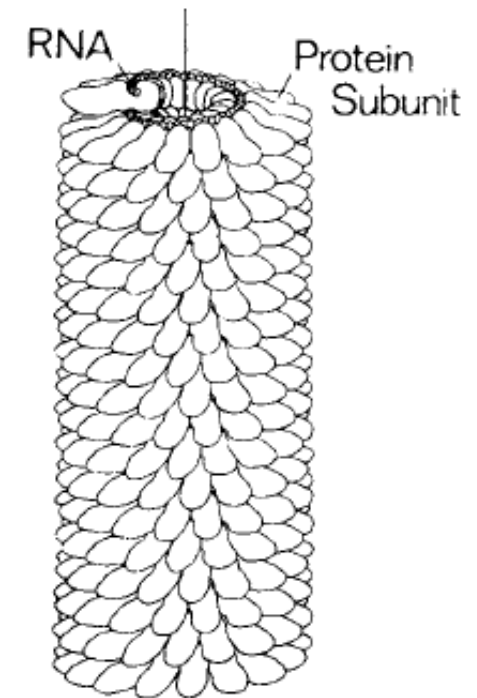
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## Definition of a self-assembly process

"**Spontaneous** association of molecules under **equilibrium conditions** into stable, **structurally well-defined** aggregates joined by **noncovalent bonds**."

(Whitesides, Mathias and Seto)

- Molecules adjust their position to reach a **thermodynamic minimum**
- Self-organization of complex systems, **basis of life**



# Model of self-assembling virus

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<https://youtu.be/X-8MP7g8XOE>

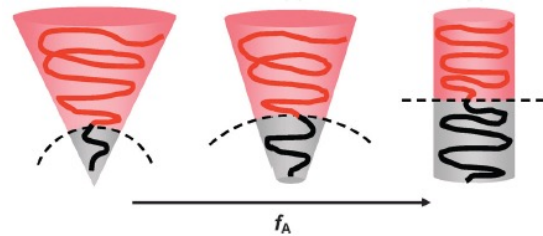
# Self-assembly is nature's solution to the nanoengineering gap

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## Engineering of materials with 10-100 nm sized features:

- Difficult / inaccessible size scale: **no-man's land between synthetic chemistry and top-down fabrication**
- Larger than single molecules -> synthetic chemistry cannot help
- Too small for lithography

### Self-assembly or controlled synthesis:



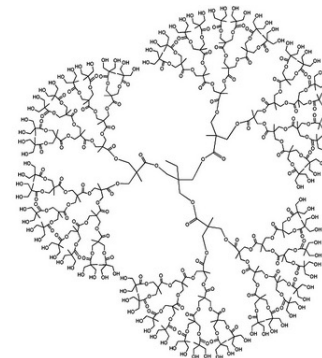
*Mai & Eisenberg,  
Chem Soc Rev  
2012*

Block  
copolymers

Self-assembled  
monolayers

Inorganic nanoparticle  
(arrays)

### Dendrimers



*Feliu et al.,  
Biomaterials 2012*

However, all result in repetitive structures

# Self-assembly is governed by free energy minimization

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$$\Delta G = \Delta H - T\Delta S$$

$\Delta G$  = change in Gibbs free energy

$\Delta H$  = enthalpy change (heat absorbed or released)

$\Delta S$  = entropy change (degree of disorder)

$T$  = absolute temperature

For self-assembly to occur spontaneously,  $\Delta G$  must be negative.



# Enthalpic contributions ( $\Delta H$ )

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**Favorable enthalpy ( $\Delta H < 0$ )** arises from:

- **Non-covalent interactions:**
  - Hydrogen bonding
  - Electrostatics
  - van der Waals forces
  - $\pi$ - $\pi$  stacking
  - Metal coordination
- These interactions **release energy** when formed, contributing a **negative  $\Delta H$** .
- Example: DNA double helix formation is stabilized by **base pairing (hydrogen bonds)** and **base stacking (vdW/ $\pi$ - $\pi$ )**.

# Entropic contributions ( $\Delta S$ )

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Entropy tends to **oppose ordering**, because:

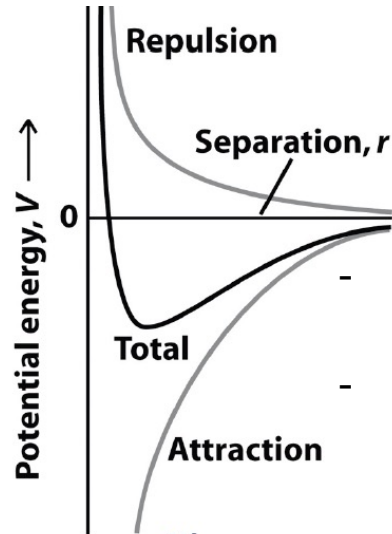
- **Assembly reduces the number of independent particles** (e.g., 100 monomers  $\rightarrow$  1 micelle = fewer microstates = lower entropy).
- **Translational and rotational entropy** are lost upon assembly.

However, in some cases **entropy can drive self-assembly**:

- **Hydrophobic effect**: Water molecules around nonpolar groups become more disordered when hydrophobes cluster  $\rightarrow$  **entropy of water increases**.
- **Depletion forces / crowding**: Entropic forces can push components together to maximize space for other molecules.

# Forces governing molecular self-assembly

## Van der Waals interaction

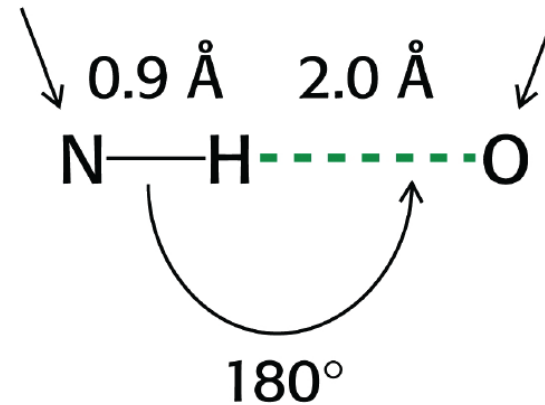


- Individually weak, cumulatively strong
- Molecules like to sit at the minimum of the Lennard-Jones potential

## Hydrogen bonding

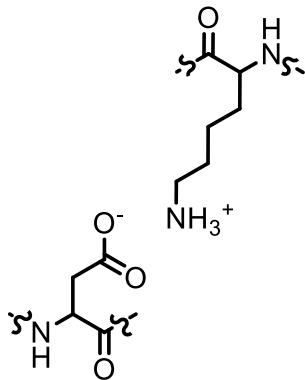
Hydrogen donor

Hydrogen acceptor



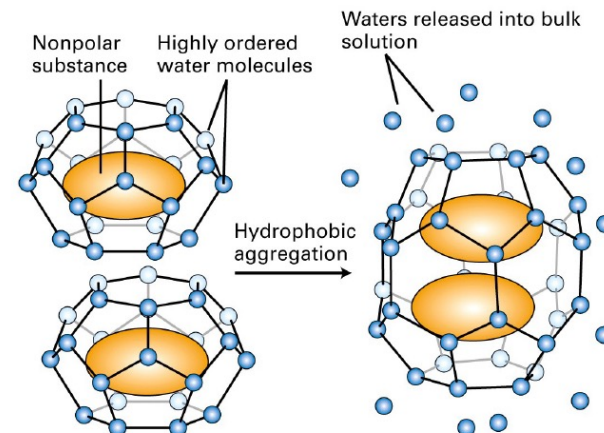
- Directional
- Hydrogen donor/acceptor interaction

## Electrostatic interactions



- Coulomb attraction/repulsion between charged species
- Dampened in high salt, polar solvents

## Hydrophobic effect



- **Entropic effect**
- Mainly due to disruption of water hydrogen bonding network

# Forces governing molecular self-assembly

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Force	Strength (kcal/mol)	Range	Directionality	System Examples
Hydrophobic	~0.6–3	Short	Low	Micelles, proteins
Electrostatic	~0.6–6	Long	Low	DNA–protein, colloids
Hydrogen Bonding	~1–6	Short	High	Base pairing, $\beta$ -sheets
van der Waals	~0.1–0.6	Very short	None	Nanoparticles, SAMs
$\pi$ – $\pi$ Stacking	~1–3	Short	Moderate	Aromatic molecules, DNA stacking
Metal–Ligand Coordination	~6–60	Short	High	MOFs, metallocages

# Balancing enthalpy and entropy

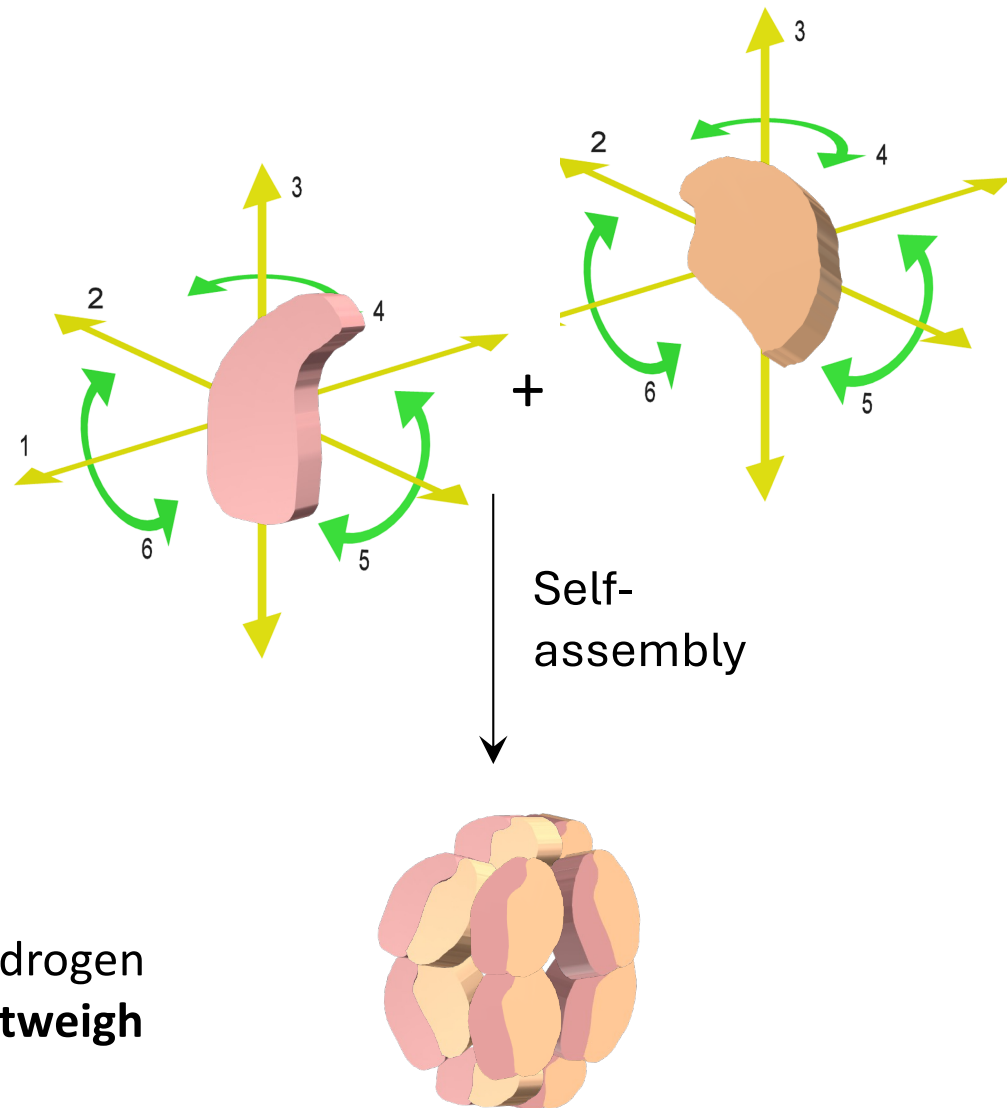
Monomer in solution:

**Translational degrees of freedom:** ability to move freely along different axes in the solution

**Rotational degrees of freedom:** ability to rotate around different axes

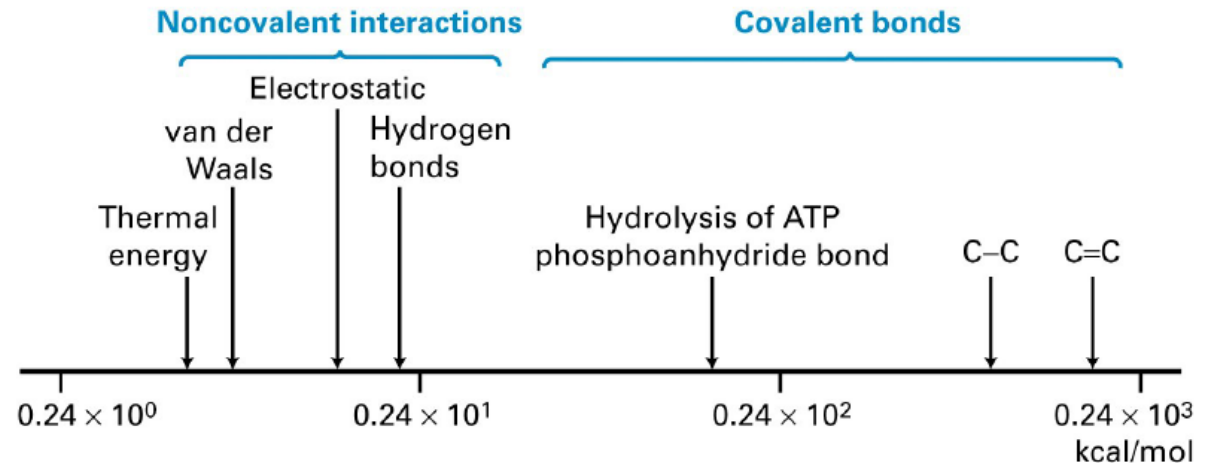
Upon assembly, monomers lose much of their translational and rotational freedom, **resulting in loss of entropy.**

**Gain in enthalpy** (from VdW forces, hydrogen bonds, ionic interactions etc.) **must outweigh entropic penalty.**



# Forces governing molecular self-assembly

- Non-covalent bonds are weak:
  - 0.1 – 5 kcal/mol (vs. 40-100 kcal/mol for covalent bonds)
- Many bonds** are required



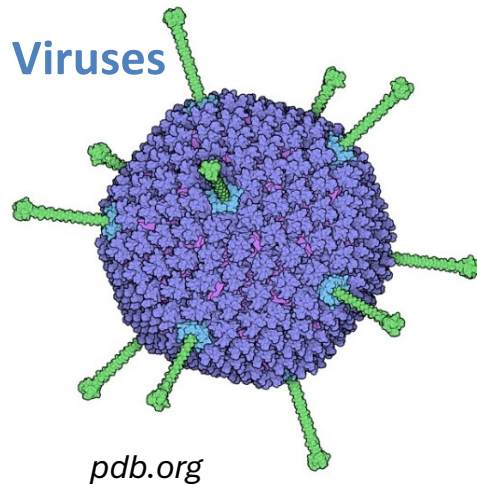
- Interactions between molecules must be **more favorable than solvent interactions**
- Must overwhelm **entropic** advantage of dissolving the complex

# Biological self-assembly

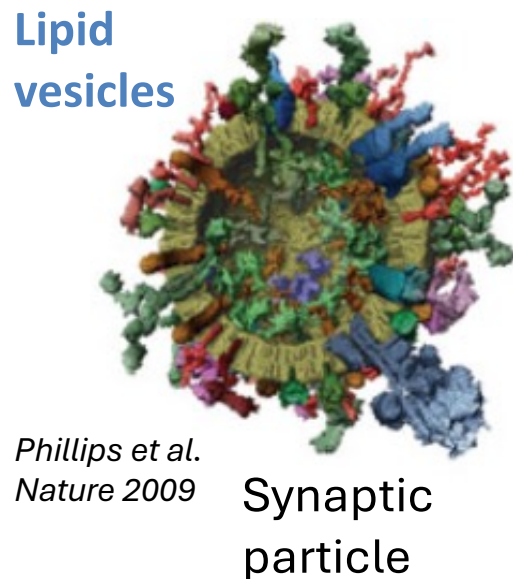
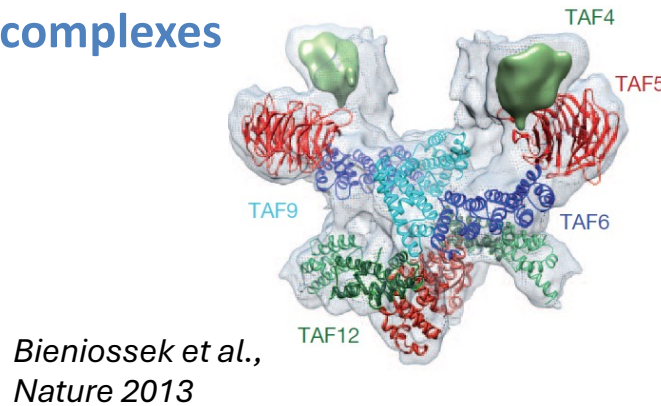
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- Many weak reversible interactions to obtain final structure → thermodynamic minimum
- Modular process through stable sub-assemblies
- Often small number of molecule types involved
- Positive cooperativity
- Complementarity in molecular shape through VdW and hydrophobic interactions

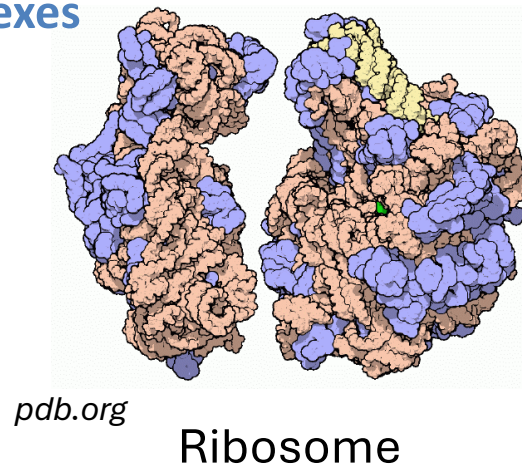
# Self-assembly in nature



## Protein complexes



## Protein-RNA complexes



- Dimer- multimerization
- Homo- vs. heterooligomers
- RNA-based association
- Membrane-supported self-assembly
- **Cooperativity:** the modification of the conformation of the individual particles that increases the affinity for the other components
- **All-or-none transition** (e.g. nucleation-and-growth model for viruses)

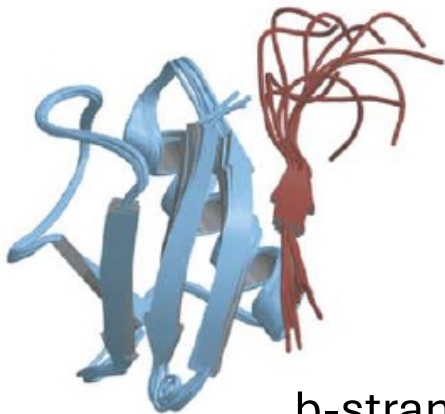
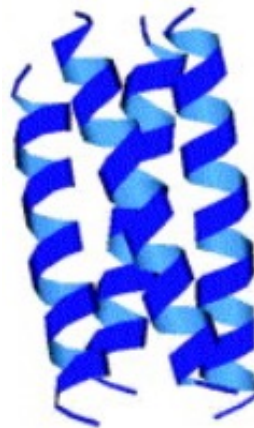


# Protein motifs for self-assembly

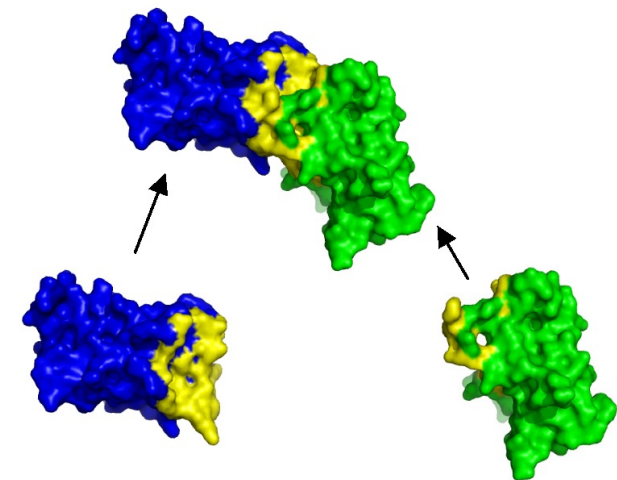
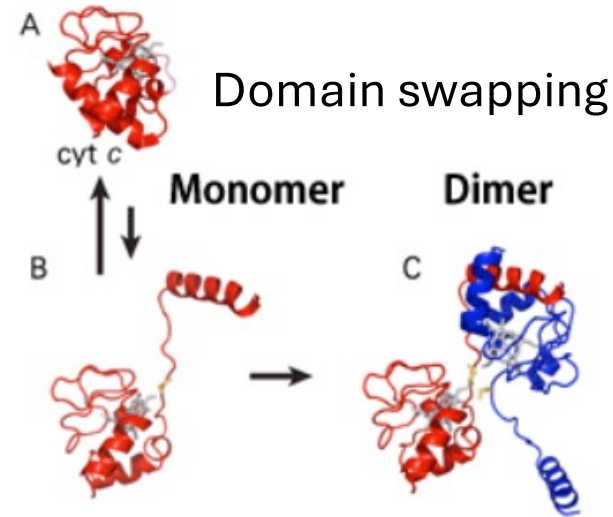


Coiled coil  
of  $\alpha$ -helices

Helix bundle



$\beta$ -strand  
addition



Large highly complementary  
protein-protein interfaces

# Engineered protein self-assembly

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## Aim:

The generation of new materials / particles with functionality based on proteins (mimicking nature).

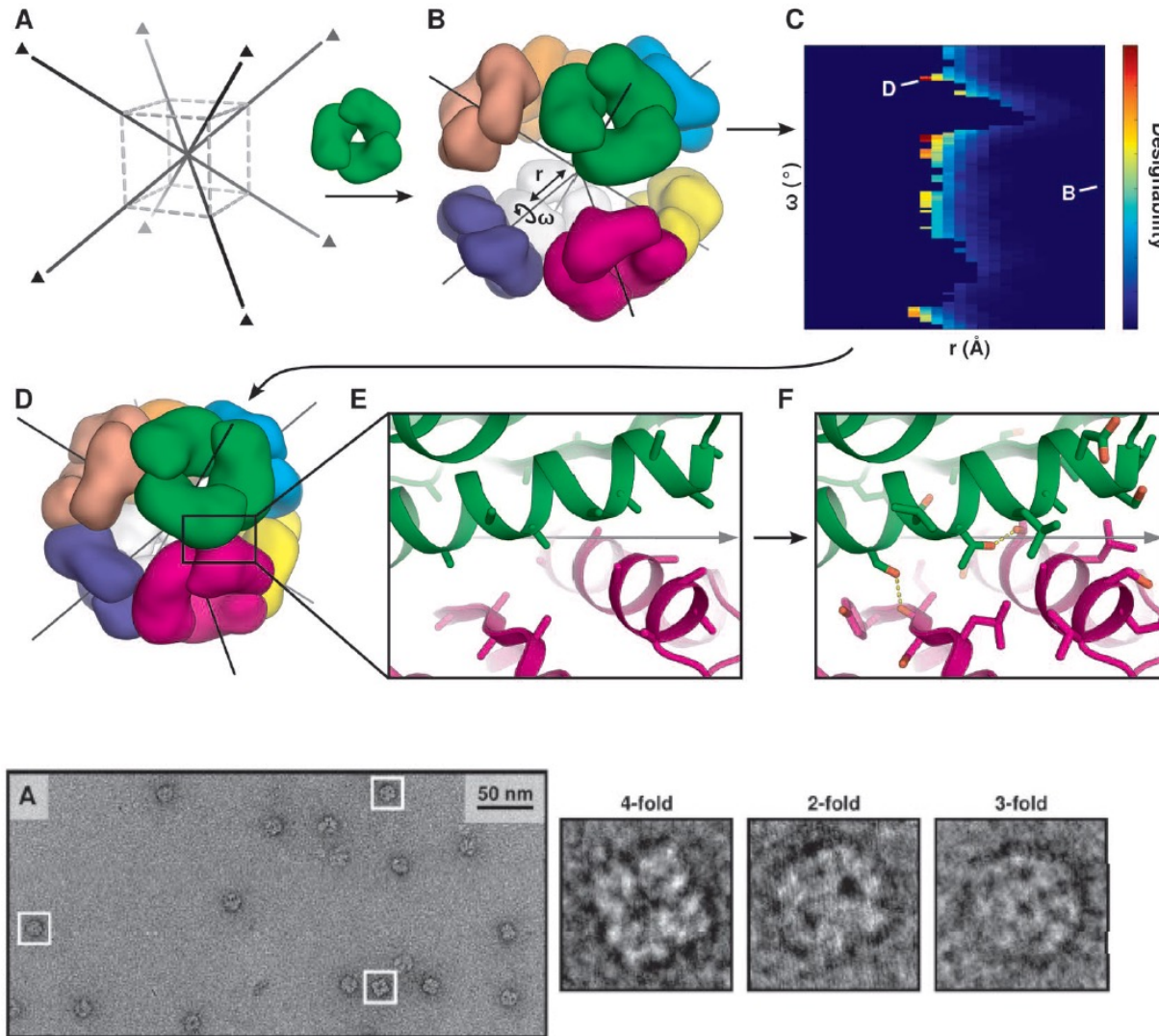
## Problems:

- Engineering of a multitude of interactions
  - Protein folding
  - Protein association thermodynamics
  - Control of topography / stoichiometry

## Methods:

- Directed evolution from a natural starting point
- Design from a natural starting point
- De-novo design

# De novo design: Programmed protein self-assembly

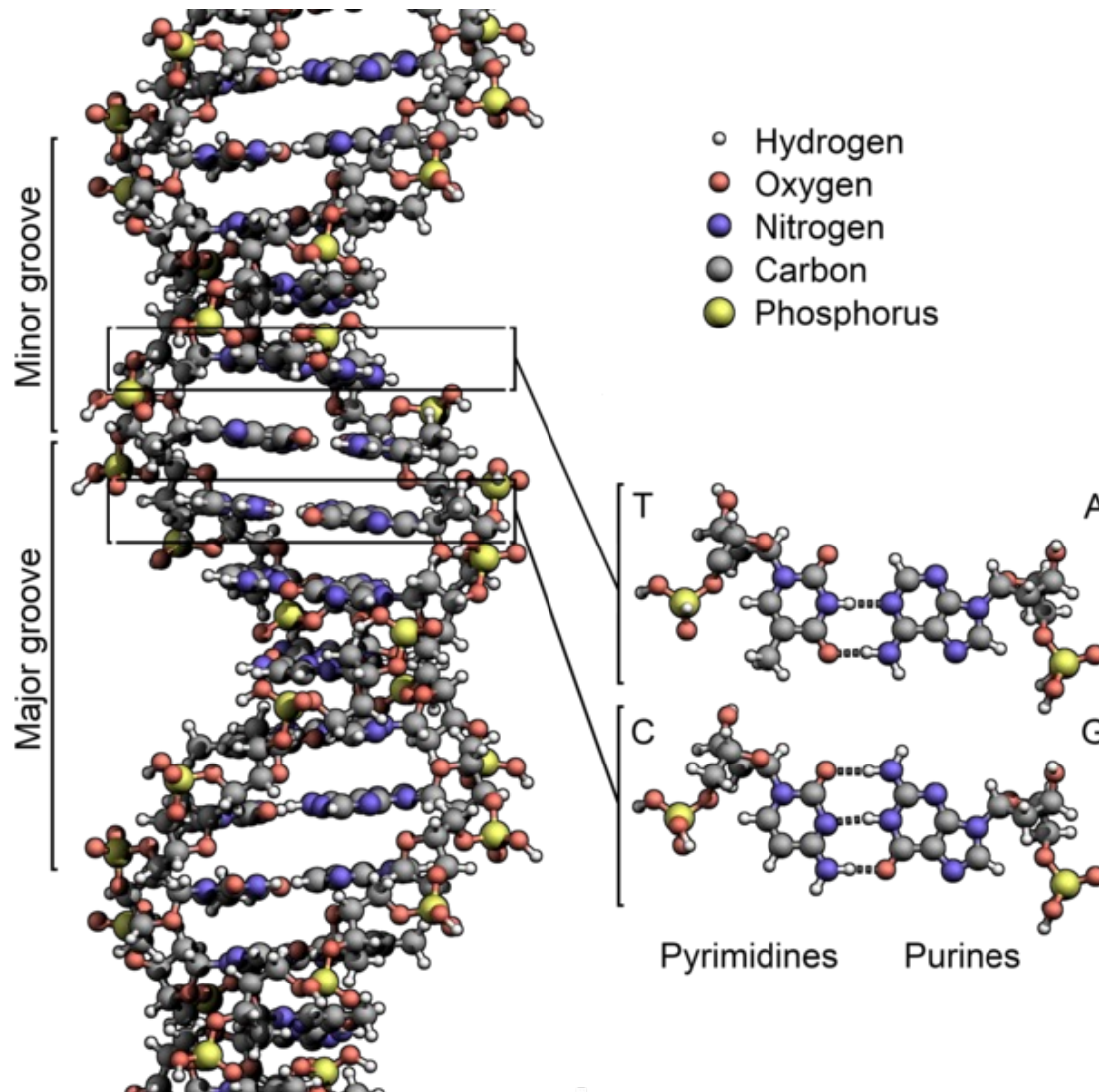


## Computational *de novo* design

- Symmetrical docking, optimizing of degrees of freedom (RosettaDesign)
- Interface design:
  - Optimized shape complementarity, hydrophobic packing, hydrogen bonding, and electrostatics
- Naturally occurring trimeric proteins as building blocks
- Reconstitution of objects with tetrahedral, octahedral, and icosahedral architecture

Baker lab: King et al.,  
*Science* 2012

# DNA as a nanomaterial



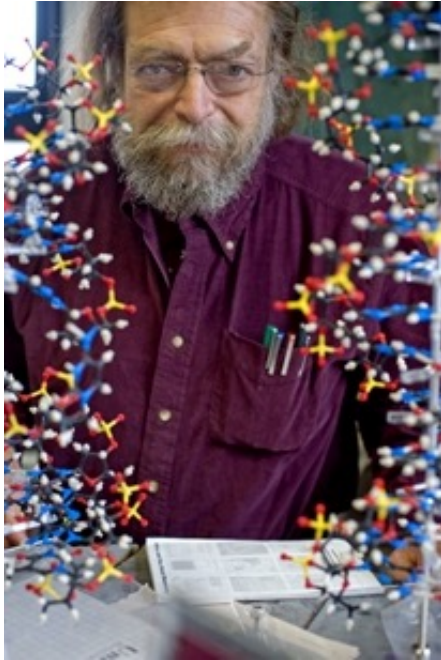
[www.wikipedia.com](http://www.wikipedia.com)

- **Basepairing:** information content
- **Non-repeating polymer** (unique DNA sequences)
- **Stiff** structure
- Chemically very **stable**
- Tolerant to **high temperatures** (thermal cycling possible)
- Defined programming of basepair sequence possible
- Chemical synthesis cheap and efficient
- **Programmable biomaterial!**



# The foundations of DNA nanotechnology

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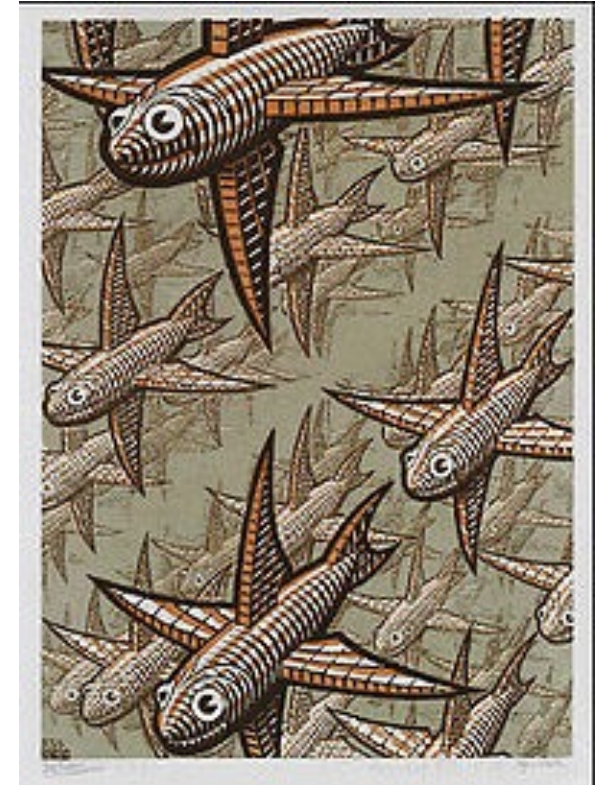


**Nadrian Seeman**

X-ray crystallographer

“One day I went over to the pub to think about what six-armed [Holliday] junctions might look like when I realized that they’d be just like the flying fish in Escher’s woodcut *Depth* [...] And they’re arranged like the molecules in a crystal.”

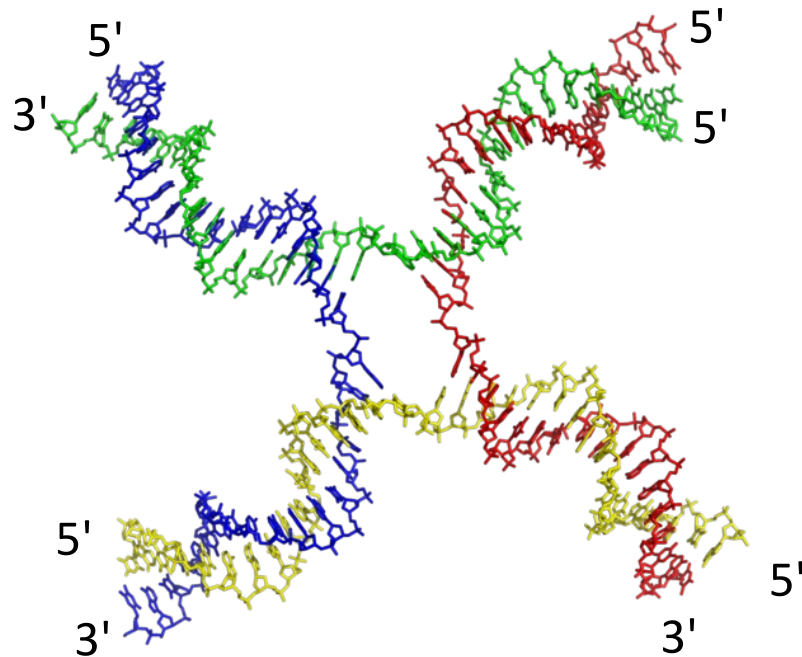
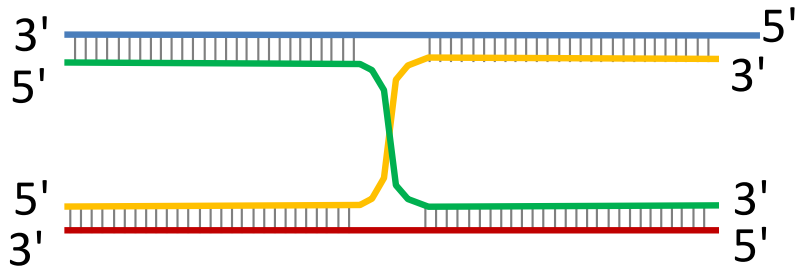
*the-scientist.com*



M. C. Escher - *Depth*

→ DNA nanotechnology

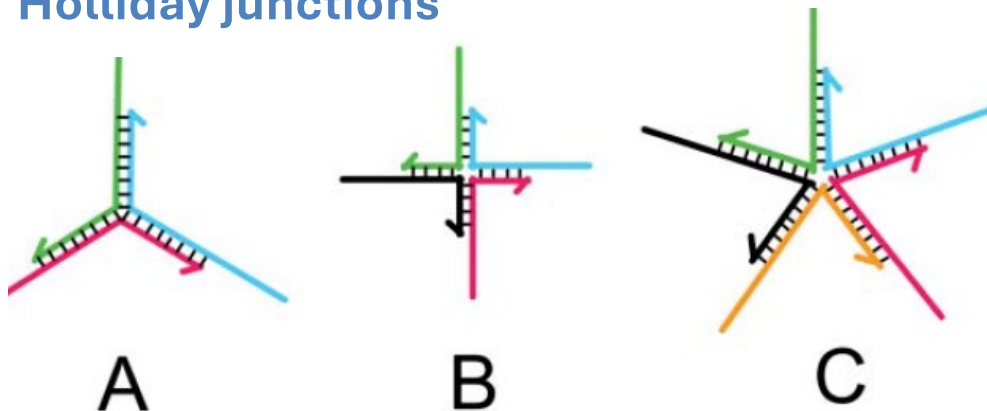
# The Holliday junction



- Structure first described in the 1960s by Robin Holliday
- Forms during crossing-over / strand invasion process during meiosis
- Vital for genetic diversity: allows mixing of parental alleles!
- Occurs during homologous recombination processes
- Observed by microscopy in 1970s
- Can be resolved by cuts and religations into different products, resulting in strand exchange
- **This crossover forms the basis of most DNA nanostructures**

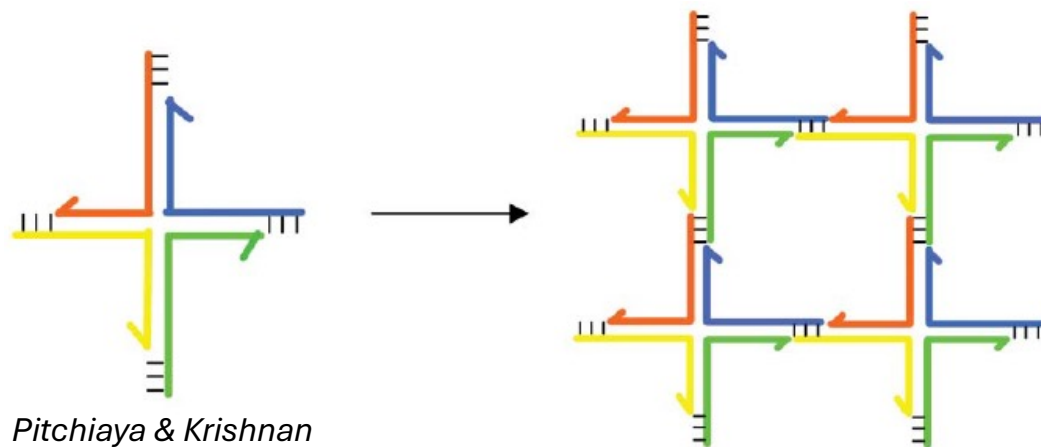
# Self-assembly with Holliday-junctions

## Topologies of multi-armed Holliday junctions



- Multi-armed DNA constructs are possible

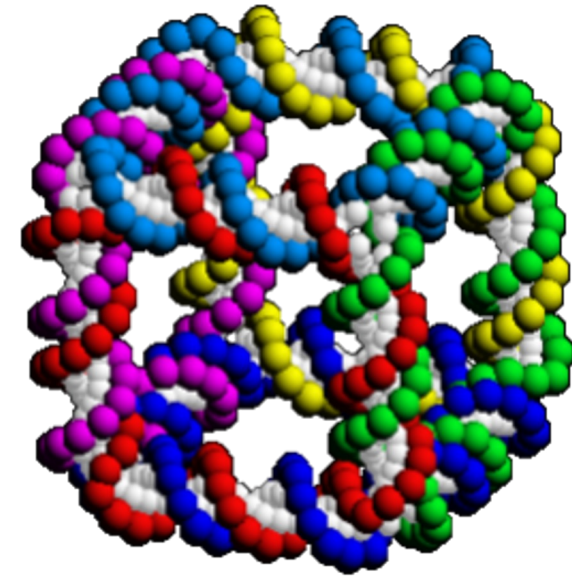
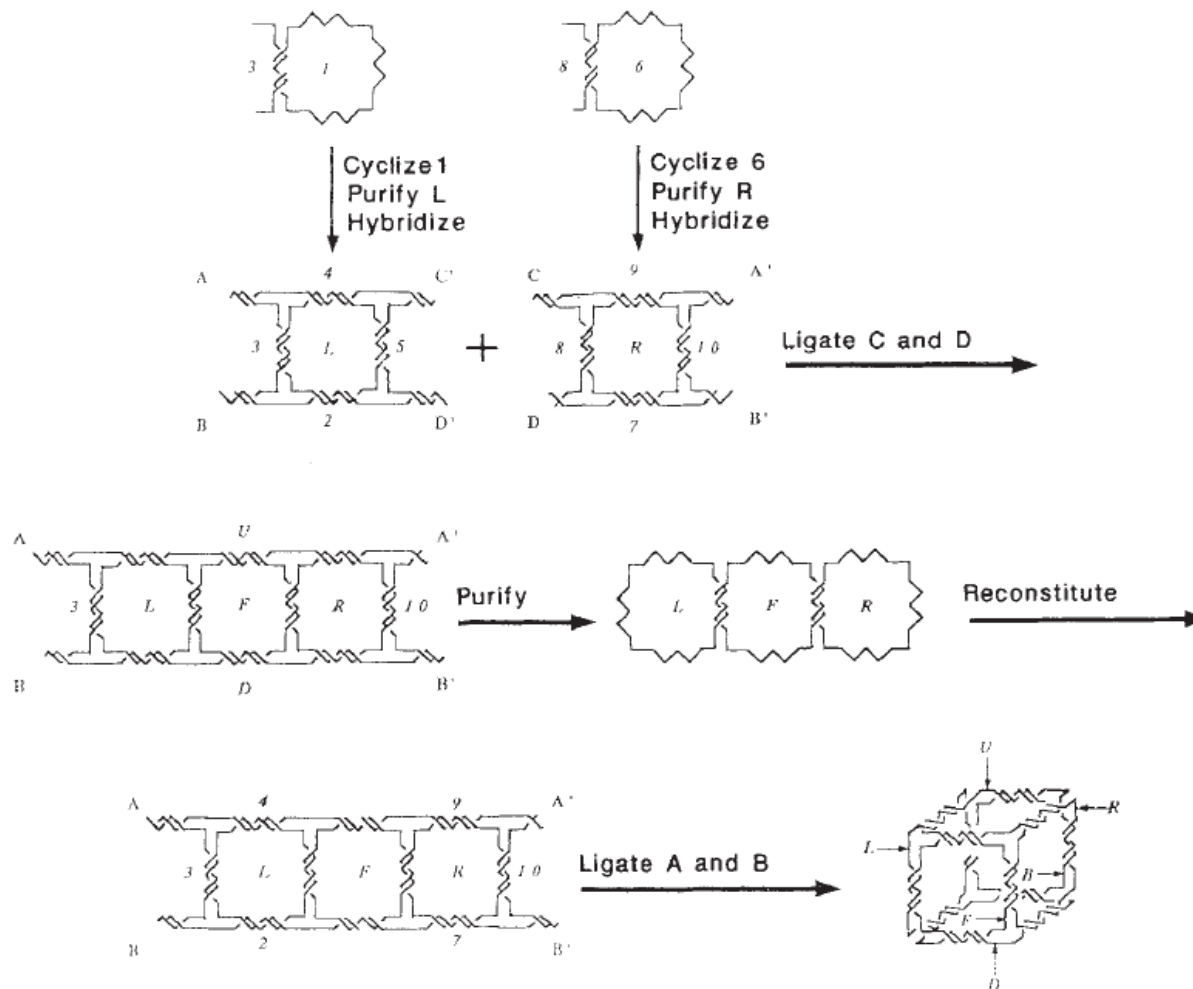
## 2D-array formation



*Pitchiaya & Krishnan  
ChemSocRev 2006*

- Sticky ends allow self assembly into 2D or 3D elements

# DNA cube: First rationally designed DNA polyhedron structure

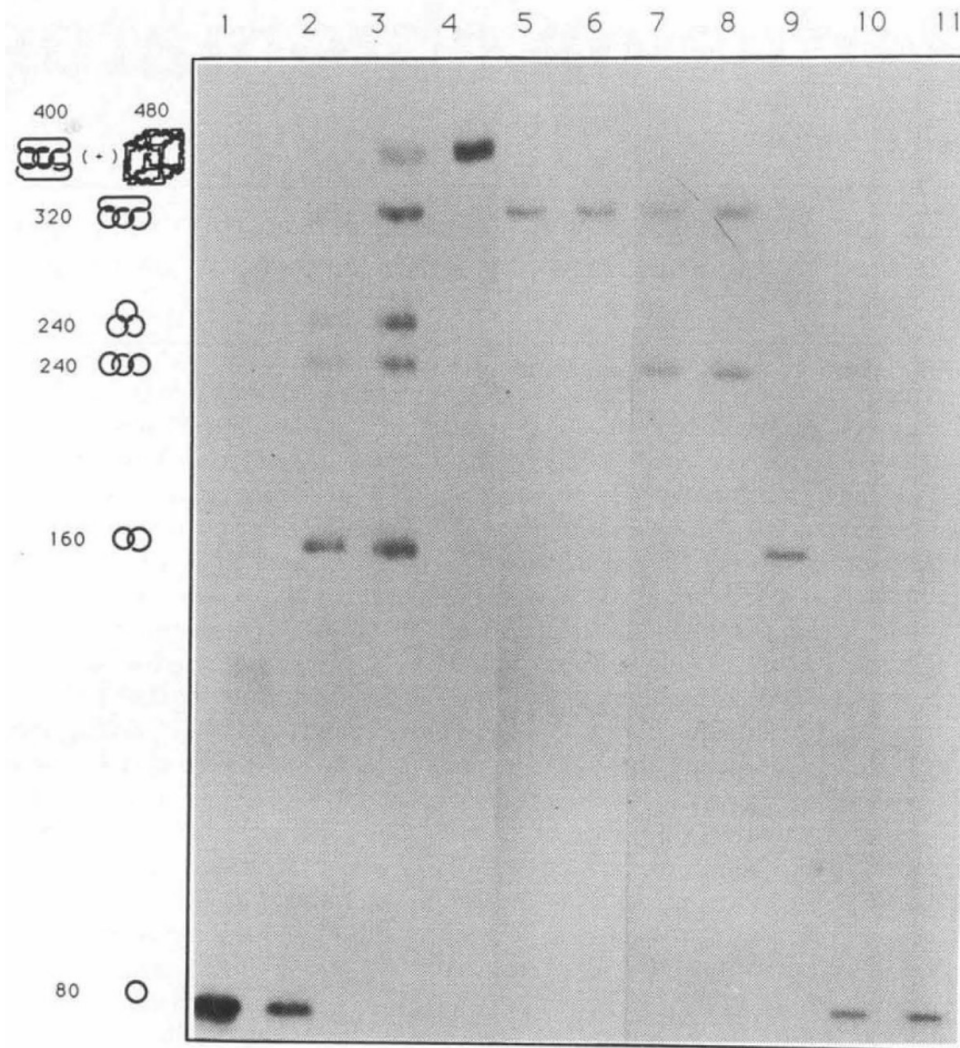


- Very small object!
- 12 edges of equal length
- Multistep synthesis
- Low yield (<10%)
- **First demonstration of DNA as an architectural material**

Chen & Seeman, *Nature* 1991



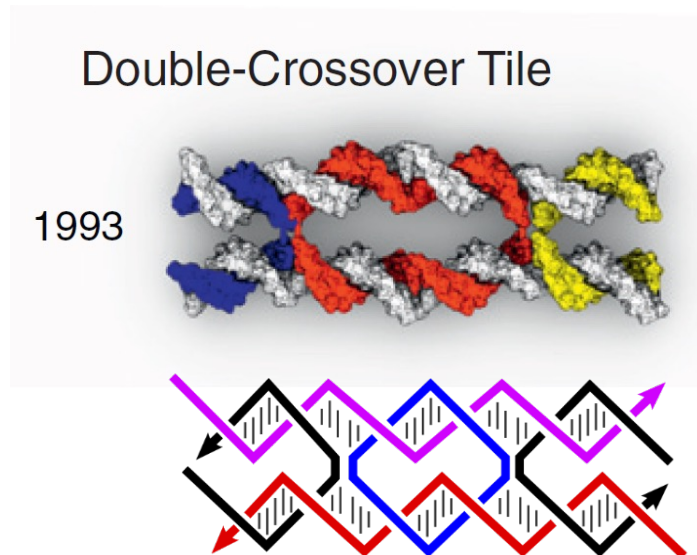
# DNA cube: products of the final ligation step



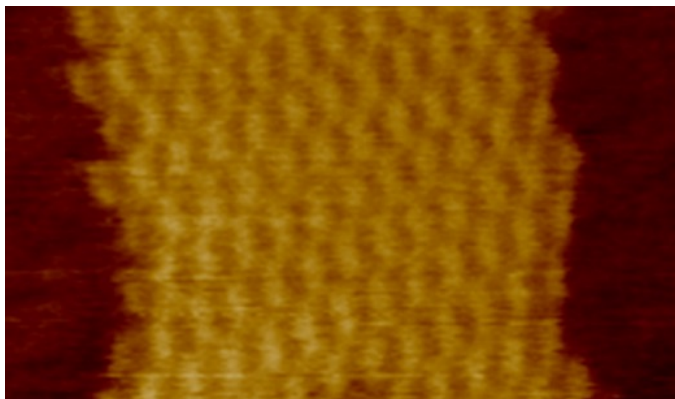
Chen & Seeman, Nature 1991

- Autoradiodiagram of denaturing gel
  - Lane 1: cyclic 80-mer marker
  - Lane 2: markers for intermediate products (up to four cycles)
  - Lane 3: Products of final ligation (80-mer lost during manipulation)
  - Lane 4: purified product
- Digestion analysis
  - Lanes 5 and 6: FR and LF digestions produce 4-cycle product
  - Lanes 7 and 8: double digests for BL and RB, and BL and LF, respectively, produce double belt
  - Lane 9: BL and RD digest produces 2-circle catenane
  - Lanes 10 and 11: Digest for BL, RB, FR, and LF produces single 80-mer circle

# Flat double-crossover tile allows construction of more complex objects



*Fu & Seeman*  
*Biochemistry* 1993

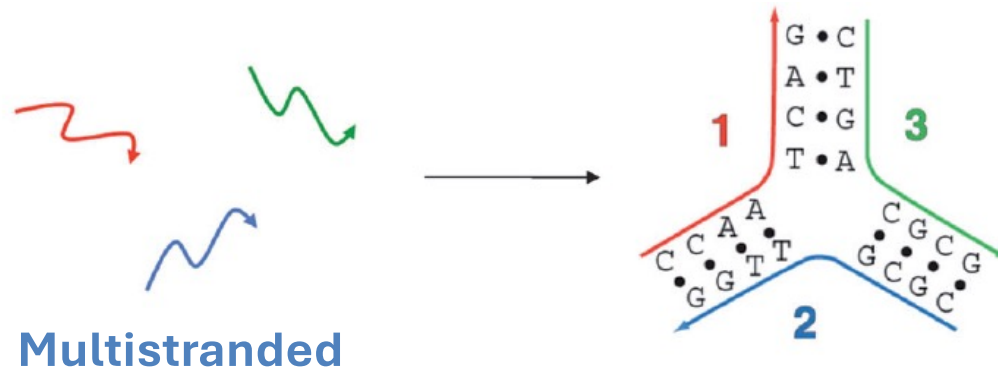


*Image source: Wikipedia*

- **Two parallel DNA double helices linked by crossover**
- Many conformation / topologies possible
- They differ in mechanical and chemical stability (protection of the junction sited in the interior)
- All topologies were mapped and the most stable one was determined -> some prone to aggregation
- Winfree et al. produced large tile arrays from tiles with sticky ends
- Even DNA-based computation possible (tile is a molecular logic gate, specific sequences represent binary values (0 or 1))

# DNA nanostructures: Origami approach

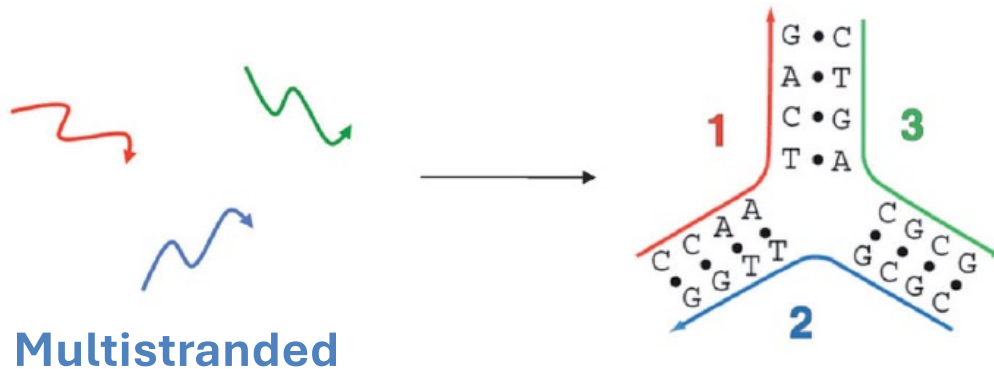
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## Problem:

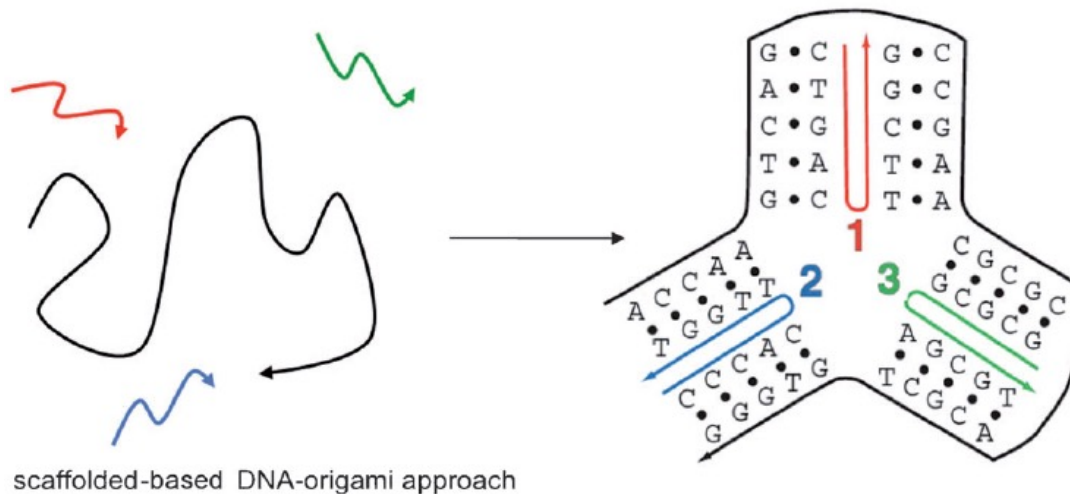
- Stoichiometry
- Entropy
- Only simple shapes

# DNA nanostructures: Origami approach



## Problem:

- Stoichiometry
- Entropy
- Only simple shapes



## Origami:

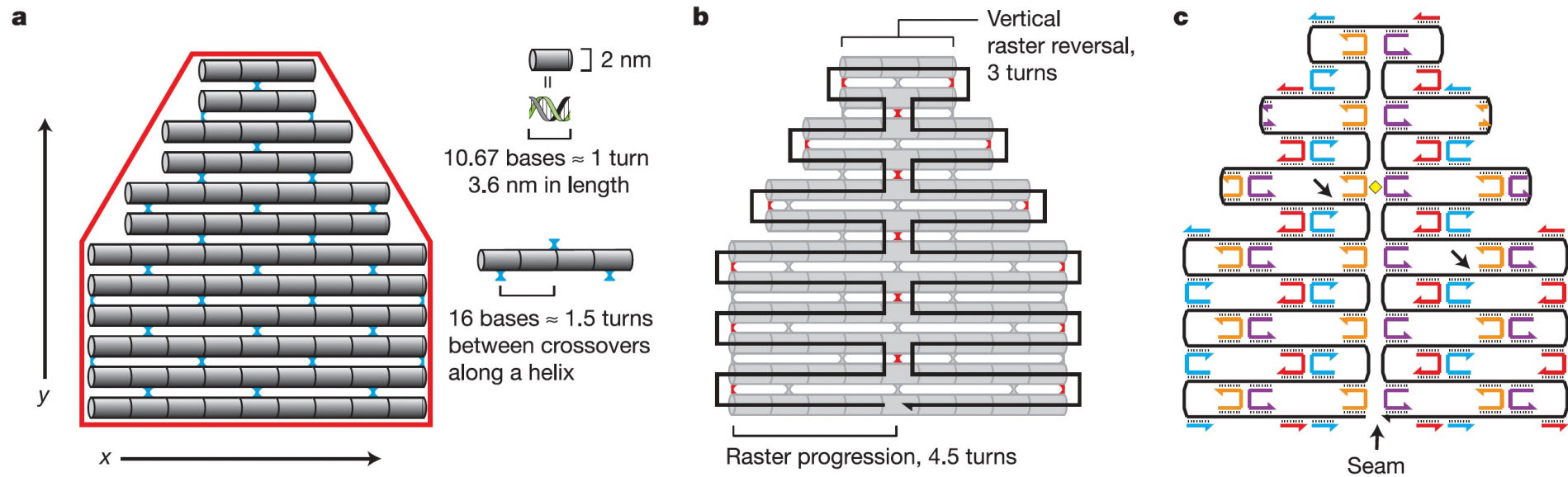
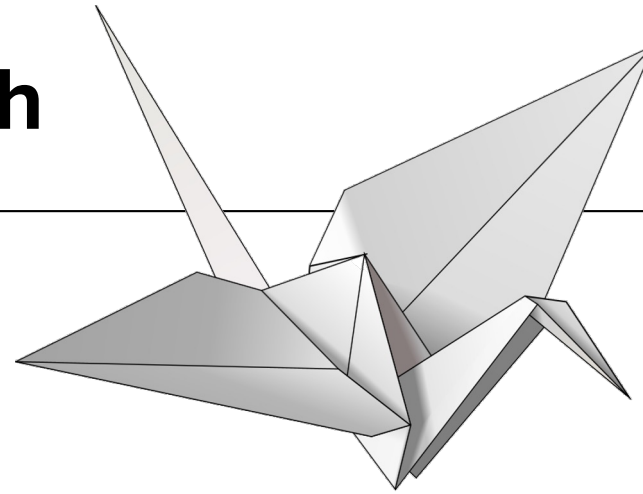
- Scaffolds and staples
- One guiding strand → entropic advantage

## Scaffolded DNA origami approach

Barbara Sacc & Christof M. Niemeyer  
*Angew Chem* 2012

# DNA origami approach

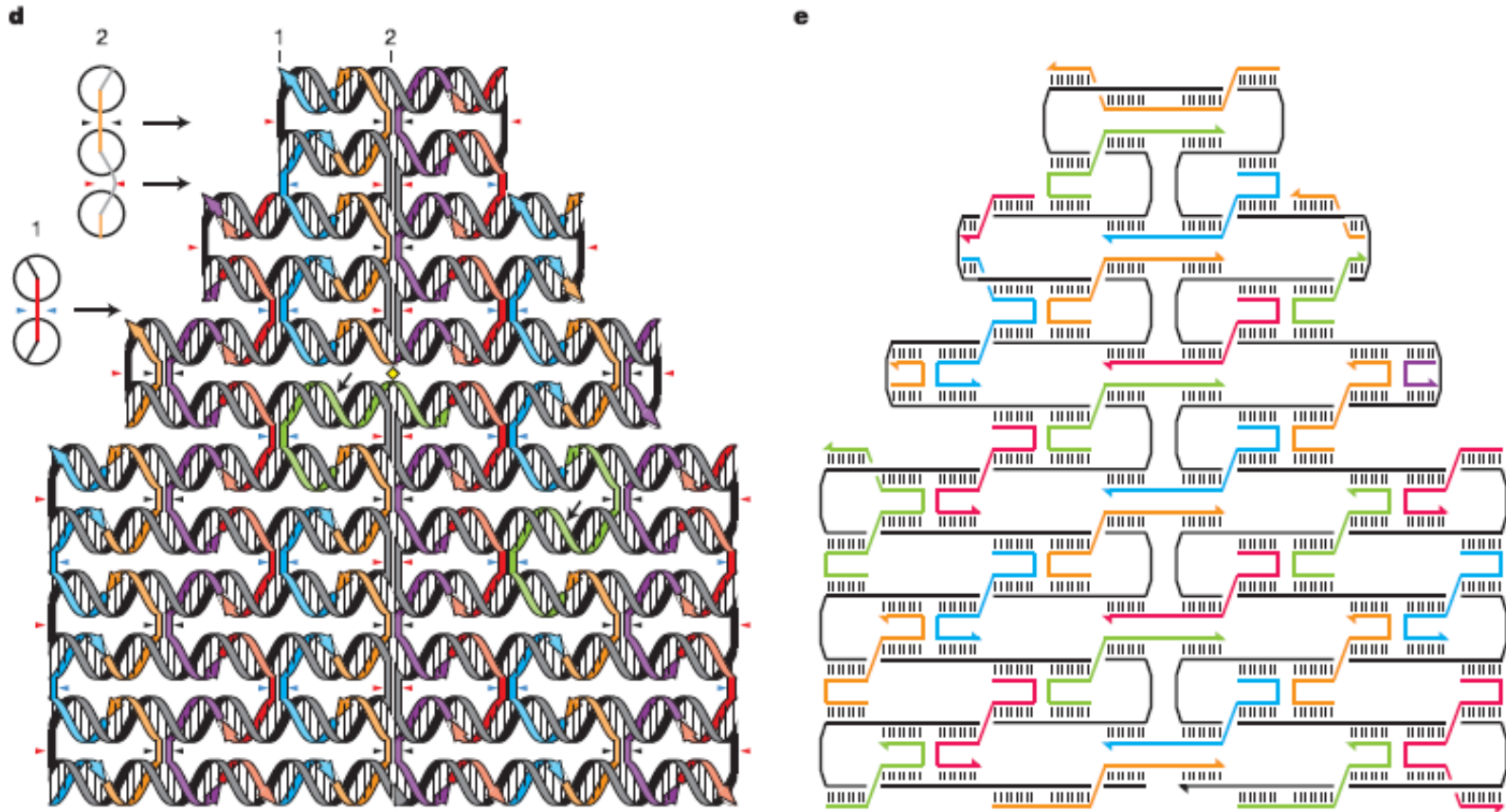
Paul Rothemund, Nature 2006



Staples bind two helices and are 16-mers

# DNA origami approach

Paul Rothemund, Nature 2006

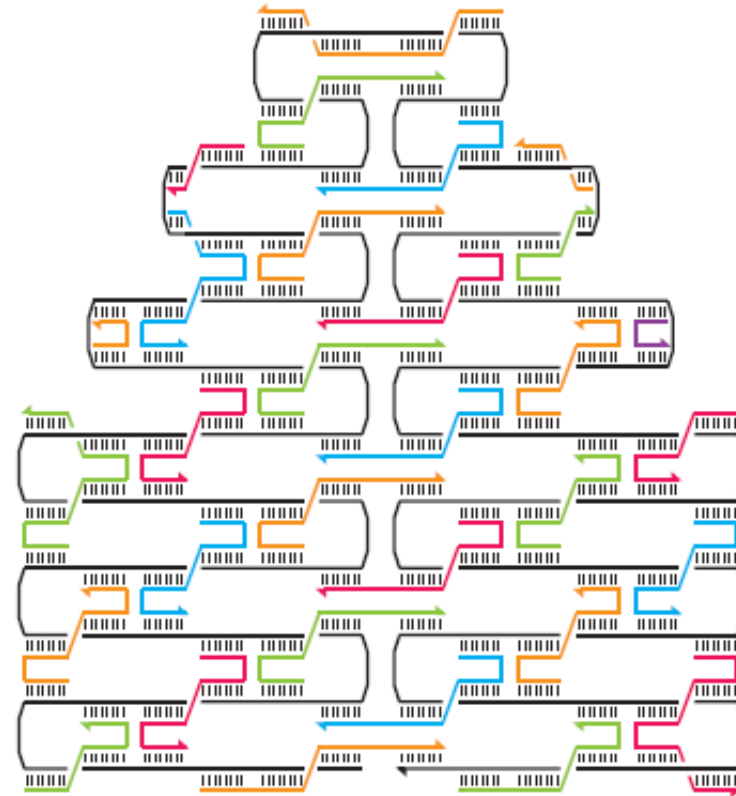




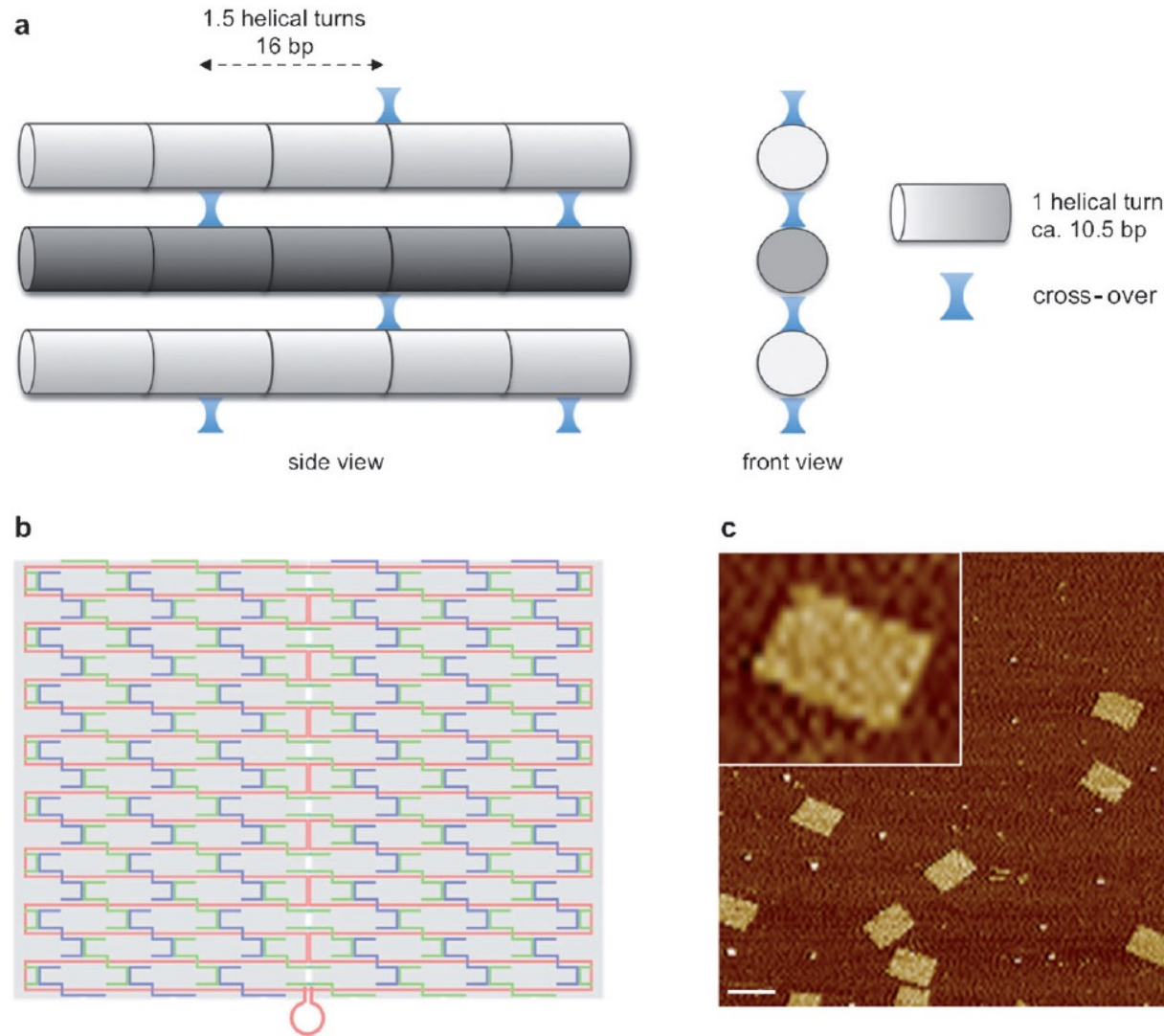
# DNA folding procedure

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- 7.25 kilobase long **M13mp18** genome (circular)
- Folding with aid of ca. 200 staple-strand
- Array of antiparallel helices through periodic cross-overs
- Self-assembly process:
  - One pot, requires counterions ( $\text{Mg}^{2+}$ ,  $\text{Na}^+$ )
  - Heat to 90 °C
  - Cool to room temperature



# General origami approach



Design of appropriate  
staple strands:

Arrangement of DNA  
double helices

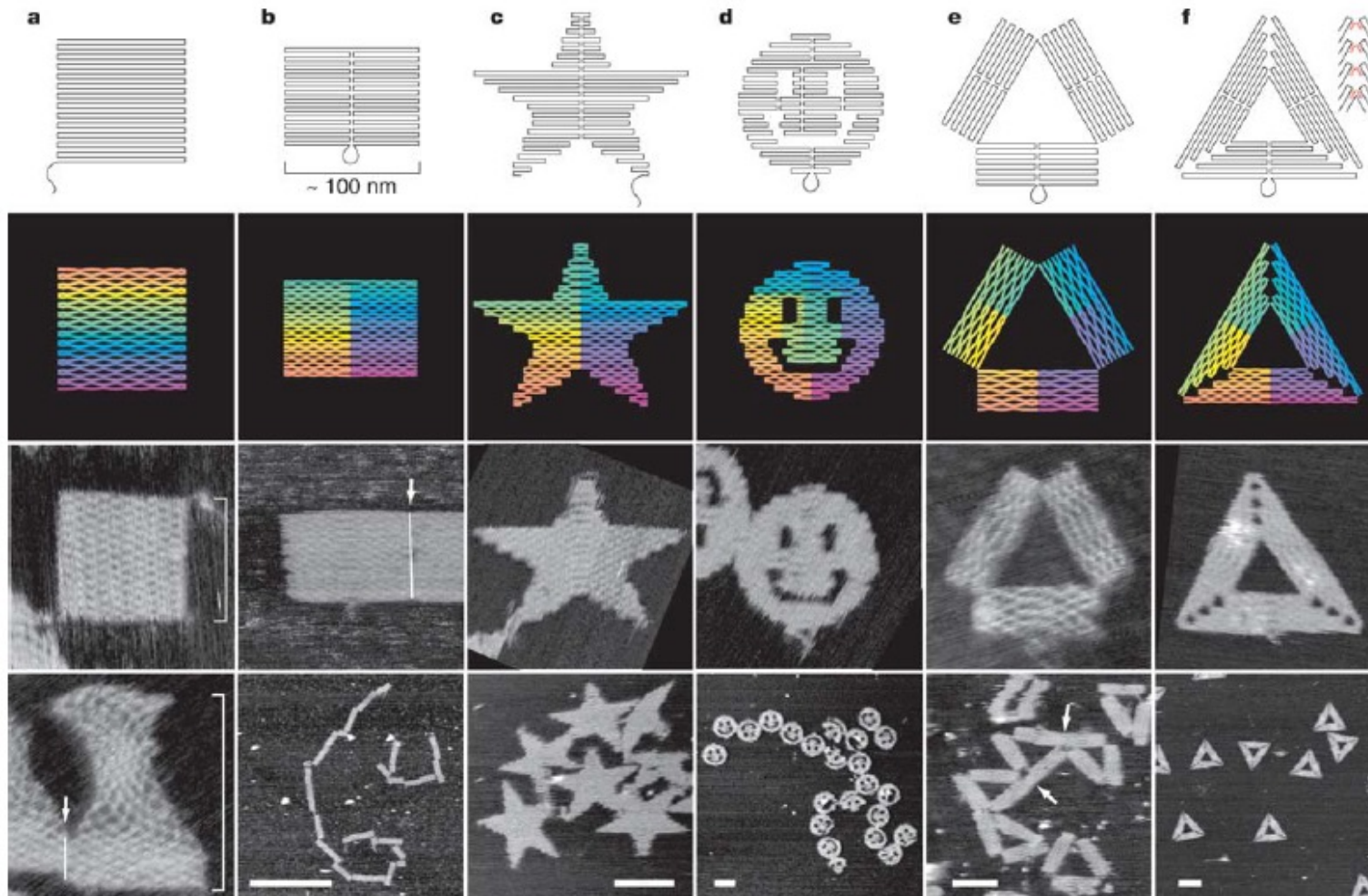
High yield,  
homogenous  
distribution of particles

$10^{14}$  similar structures  
achievable in one  
assembly



# DNA origami: Arbitrary structures achievable

Paul Rothemund, Nature  
2006



# Paul Rothemund: DNA origami

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- [https://www.youtube.com/watch?v=WhGG\\_boRxU&list=WL&index=2&t=595s&ab\\_channel=TED](https://www.youtube.com/watch?v=WhGG_boRxU&list=WL&index=2&t=595s&ab_channel=TED)
- 5:09

# Exercise

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You have designed a DNA origami structure to form a specific 3D shape using a scaffold strand derived from the M13 bacteriophage and 200 staple strands. After running your assembly experiment, you observe that the yield of correctly formed structures is significantly lower than expected. Instead of the desired shape, you find a mixture of incomplete and misfolded structures when analyzed by atomic force microscopy (AFM).

**Discuss the potential reasons for the low yield and incorrect formation of the DNA origami structures. Propose a systematic troubleshooting plan to identify and address the issues.**

# Potential Causes and Troubleshooting Plan for DNA Origami Yield Issues

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## Potential Causes and Troubleshooting Plan for DNA Origami Yield Issues

### 1. Staple Strand Concentrations

Potential Cause: Incorrect stoichiometry of staple strands can lead to incomplete or incorrect folding of the scaffold.

#### Troubleshooting Steps:

Quantify and Adjust Staple Concentrations: Use UV-Vis spectrophotometry to accurately measure the concentration of each staple strand. Adjust the concentrations to ensure they match the desired stoichiometric ratios. Prepare a master mix with equimolar concentrations of all staple strands to avoid pipetting errors.

Control Experiments: Prepare a series of samples with varying staple-to-scaffold ratios (e.g., 10:1, 20:1, 50:1) to determine the optimal concentration for correct folding.

### 2. Sequence Design Errors

Potential Cause: Errors in the design of staple sequences, such as insufficient complementarity or unintended secondary structures.

#### Troubleshooting Steps:

Re-examine Computational Design: Use software tools to re-evaluate the designed sequences for potential errors and unintended secondary structures.

Secondary Structure Prediction: Employ computational tools to predict secondary structures of individual staple strands. Modify sequences if strong secondary structures are predicted.

Redesign Problematic Sequences: Synthesize and test modified staple strands if errors or strong secondary structures are identified.

### 3. Thermal Cycling Protocol

Potential Cause: Suboptimal heating and cooling rates during thermal cycling can affect the assembly process.

#### Troubleshooting Steps:

Optimize Thermal Cycling Profile: Experiment with different thermal cycling protocols. For example, use a slower cooling rate (e.g., 1 °C per minute) from 65 °C to room temperature to allow more time for proper folding.

Incremental Temperature Drops: Introduce steps where the temperature is held constant for a period (e.g., hold at 50 °C for 1 hour) before continuing to cool, to help intermediate structures stabilize.

# Potential Causes and Troubleshooting Plan for DNA Origami Yield Issues

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## 4. Scaffold Quality and Purity

Potential Cause: Contaminants or degradation products in the scaffold strand can interfere with the assembly.

### Troubleshooting Steps:

Assess Scaffold Integrity: Use agarose gel electrophoresis to check the integrity of the scaffold strand. High-quality, intact scaffolds will appear as a single, sharp band.

Purify Scaffold Strand: If impurities or degradation products are detected, purify the scaffold using size-exclusion chromatography or additional gel extraction steps.

Mass Spectrometry Analysis: Employ mass spectrometry to confirm the molecular weight and purity of the scaffold strand.

## 5. Ionic Conditions

Potential Cause: Suboptimal buffer composition and inadequate concentrations of stabilizing ions can affect the stability of the DNA origami structure.

### Troubleshooting Steps:

Buffer Optimization: Test different buffer compositions by varying the concentration of magnesium ions (e.g., 5 mM, 10 mM, 20 mM  $\text{MgCl}_2$ ). Include additives such as NaCl or KCl to stabilize the structure.

Systematic Variation: Prepare a series of folding reactions with different ionic conditions to determine the optimal buffer composition for your DNA origami structure.

Real-time Monitoring: Use dynamic light scattering (DLS) or small-angle X-ray scattering (SAXS) to monitor the effect of different ionic conditions on the folding process in real-time.

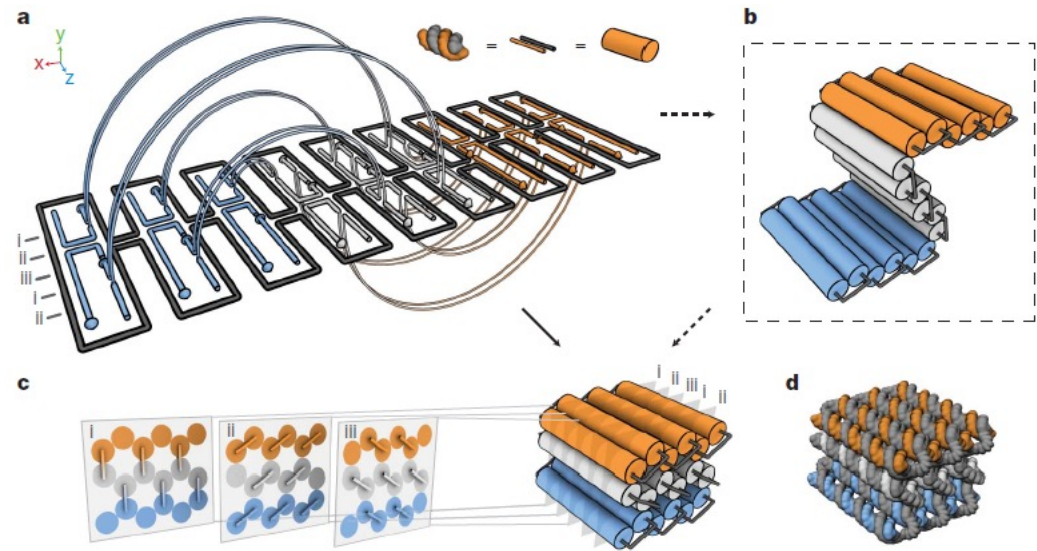
By systematically addressing these potential causes and implementing the suggested troubleshooting steps, you can identify the specific issues affecting the yield and correctness of your DNA origami structures and make the necessary adjustments to improve the assembly process.

# Extension to 3D structures

- Single layer origami: not stable for shear stress
- More rigid 3D objects are required
- Strategy: packing multiple helices into space-filling structure



Diez & Simmel lab: Langecker et al. Science 2012



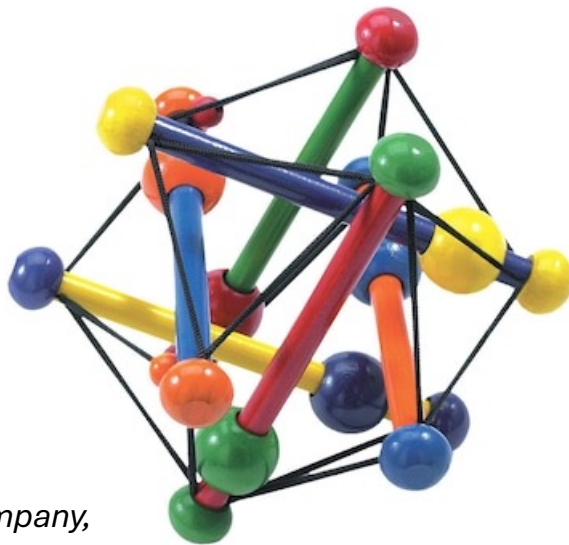
Shih lab: Douglas et al. Nature 2009

# Tensegrity rules

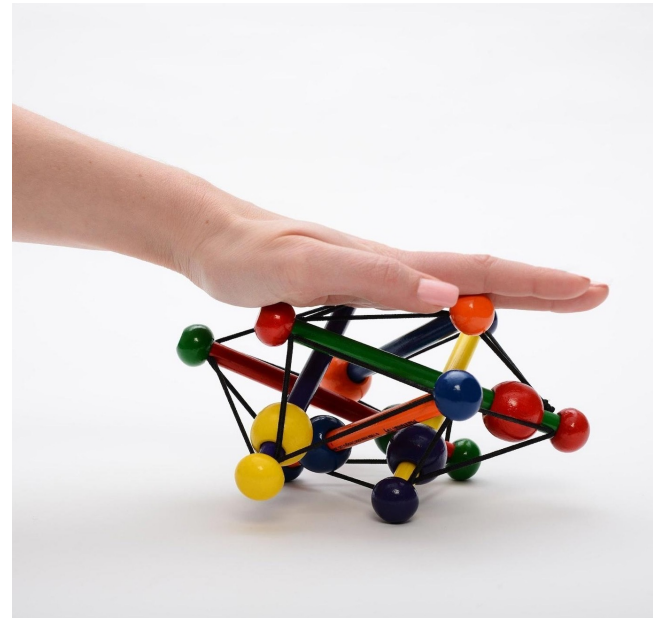
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*Tensegrity: “The property of a skeletal structure having continuous tension members (such as wires) and discontinuous compression members (such as metal tubes) so that each member performs efficiently in producing a rigid form.”*  
(Merriam Webster dictionary)

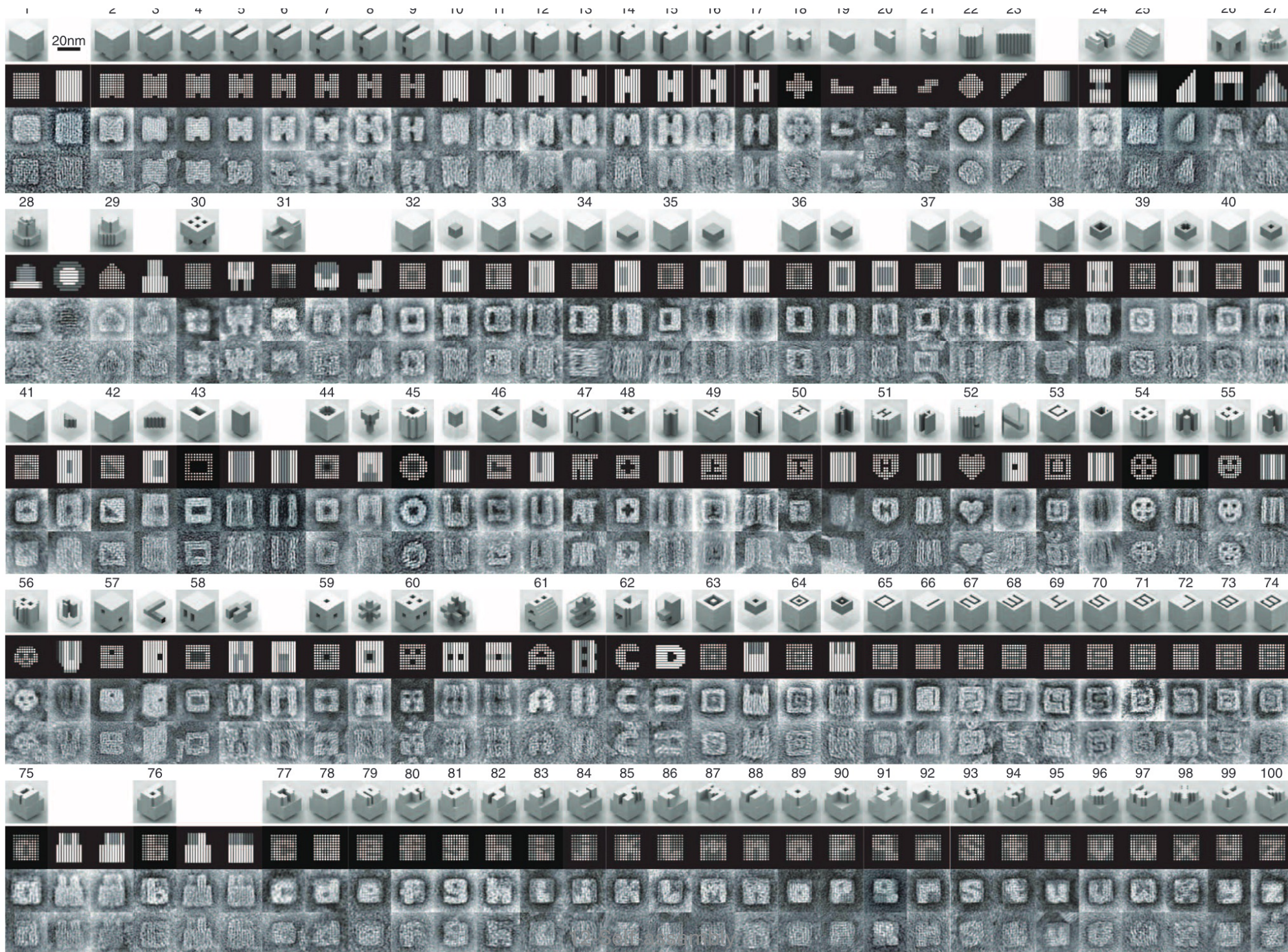
- Stiff sequences (struts) → push outward
- Flexible linkers (tendons) → pull inward



Manhattan Toy Company,  
Skwish Classic









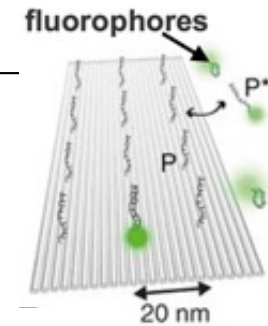
# Use of DNA origami

Prepared by computational design

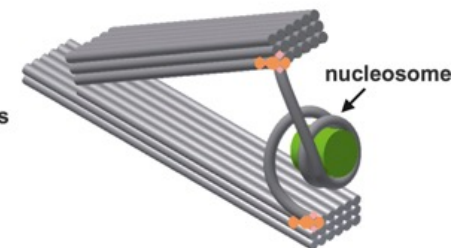
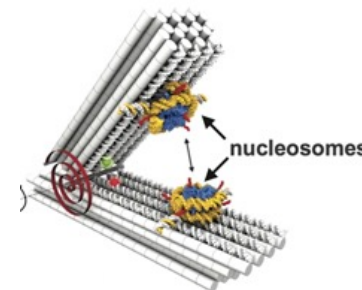
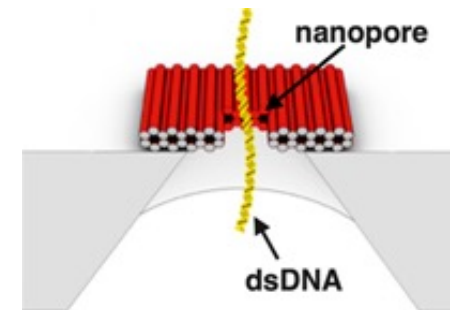
Analyzed by single molecule imaging (AFM, EM)

Uses:

- Molecular pegboards to arrange arbitrary objects in 2/3D space, for arrayed objects such as:
  - Nucleic acids
  - Small molecules
  - Proteins
  - Nanoparticles
- Functional devices:
  - Nano-calipers: force probes
  - Pores and channels
  - Encapsulation of function
  - DNA encoding of function (aptamers)



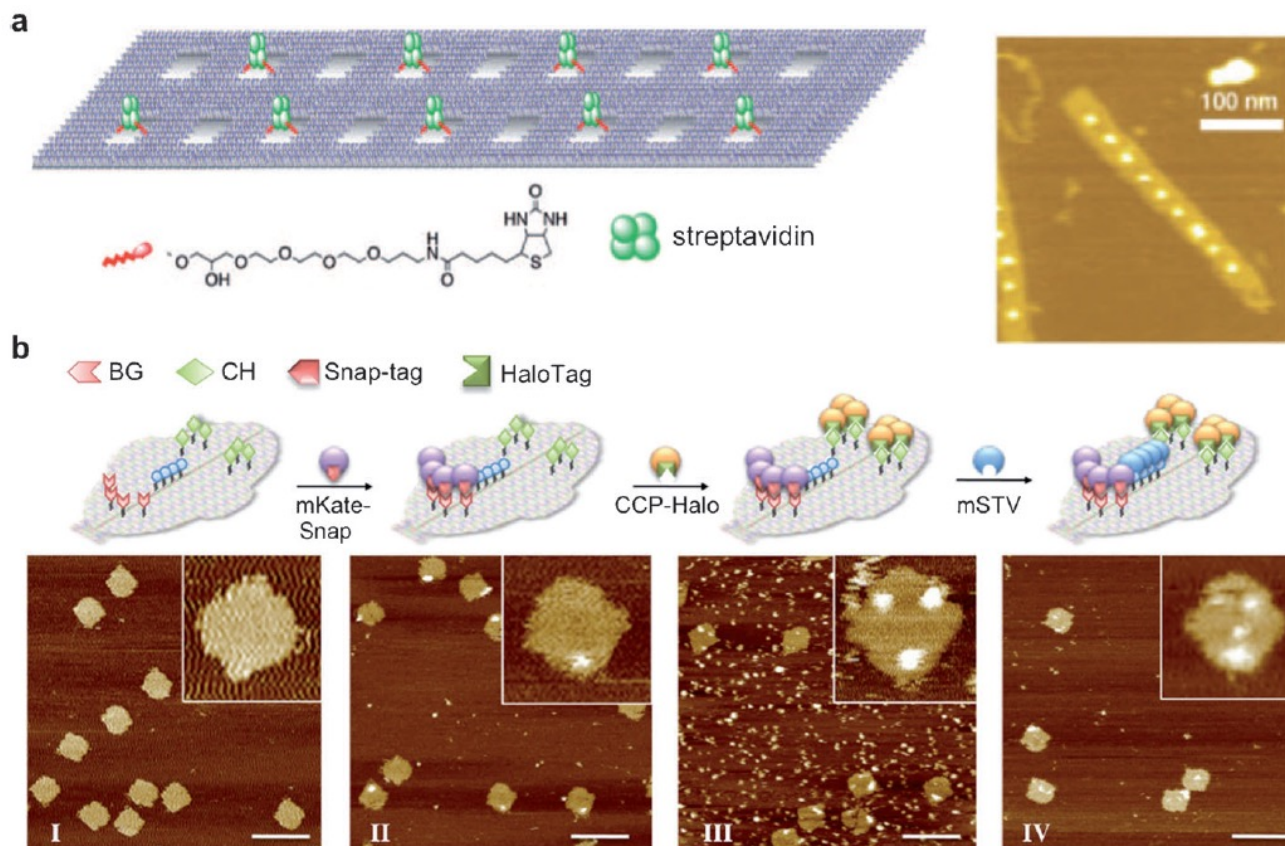
Pegboard:



DNA origami devices for molecular-scale precision measurements, 2017

[Carlos E. Castro](#), [Hendrik Dietz](#) and [Björn Högberg](#)

# Functionalization of DNA objects



- Bulky motifs, dumbbell hairpins
- Hybridization of DNA tagged components to terminal extensions
- Biotinylated DNA insertion

Barbara Sacca & Christof M. Niemeyer  
Angew Chem 2012

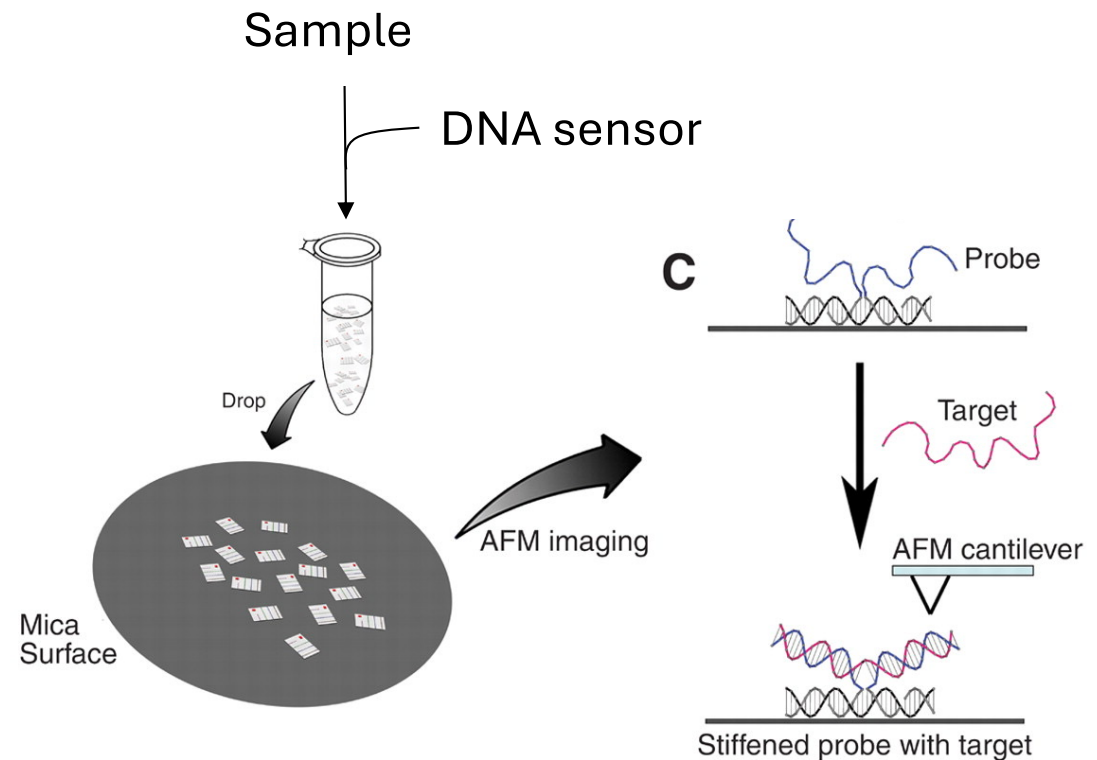
# Case study: DNA origami sensors

## Aim:

Generation of a DNA microarray on the nanoscale

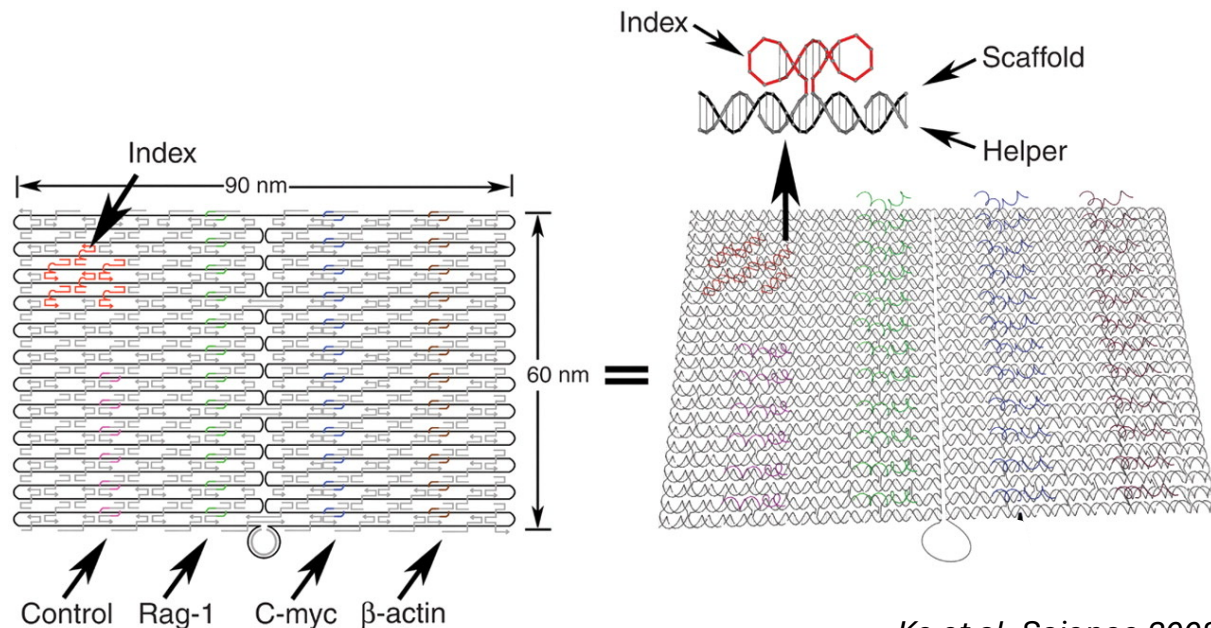
**Multiplexed detection** of several gene products on a single molecule level

Spatial arrangement on a DNA origami tile → readout by AFM



*Ke et al. Science 2008*

# Sensor design: DNA origami tile with index



*Ke et al. Science 2008*

DNA tile: circular M13 viral DNA, held together with 200 helper strands

**DNA dumbbell,**  
asymmetrically placed  
for index readout

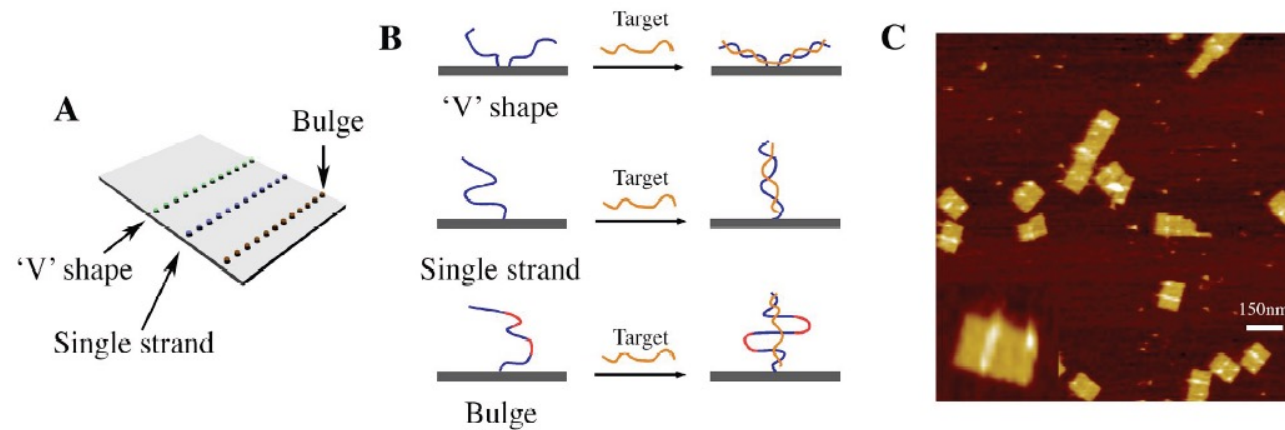
Four test strips for  
detection of RNA

12 copies of the probe  
strand, separated by 5  
nm

Lines separated by 20 nm

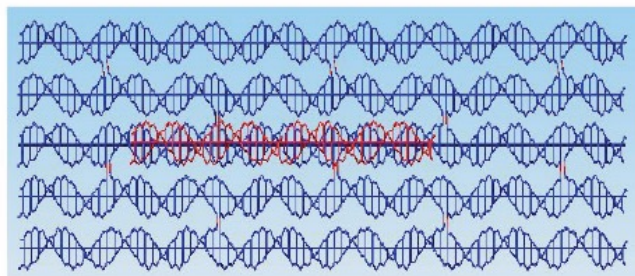


# Design of the probe architecture

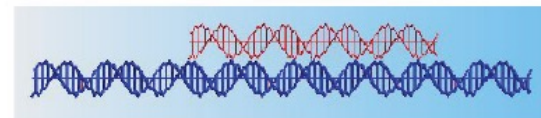


Testing different detection methods

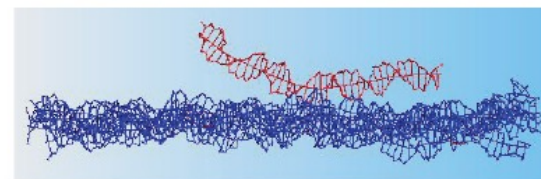
V-shape results in best signal by AFM



Top view before simulation



Side view before simulation

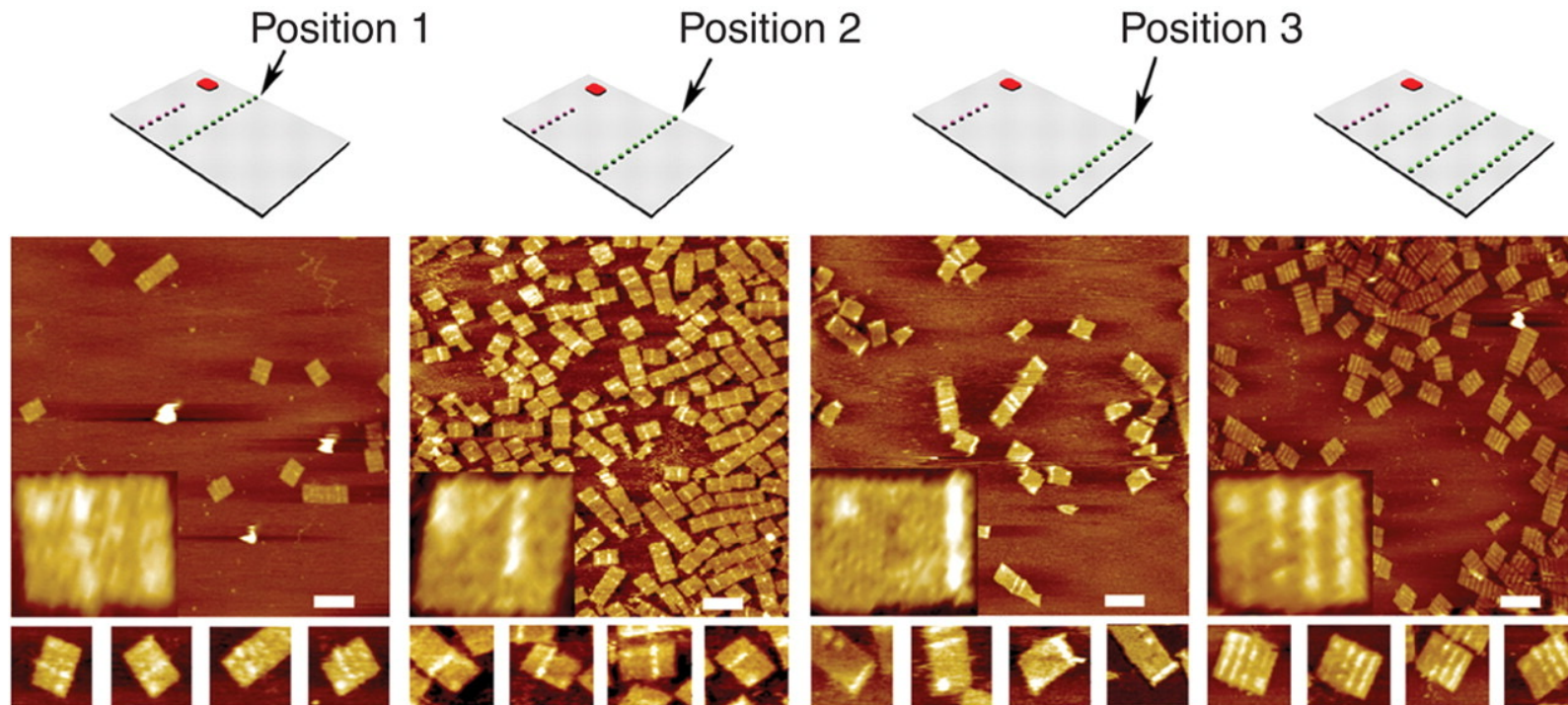


Side view after simulation

Molecular dynamics simulations exhibit V-shape of bound DNA-RNA hybrid strands

*Ke et al. Science 2008*

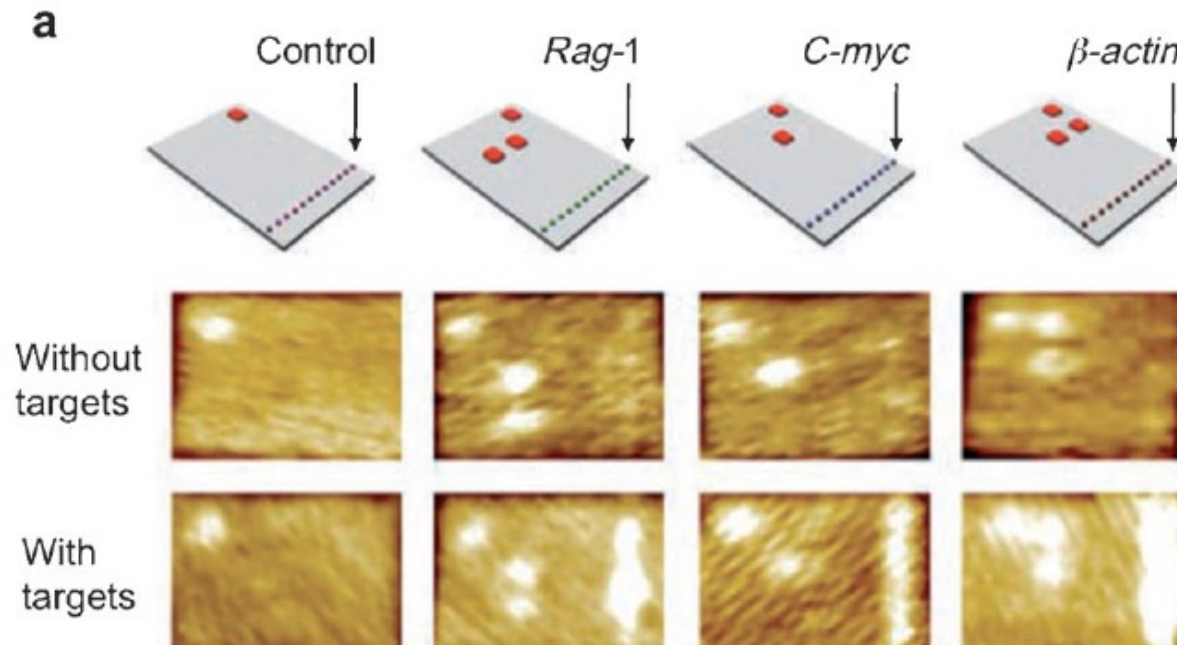
# Probe position and binding efficiency



*Ke et al.  
Science  
2008*

- All probes are for RNA of Rag-1 protein, 10 nM of tiles mixed with 600 nM of target DNA
- Interestingly, **probe position influences binding efficiency**
- Edge position results in best binding, why?

# DNA origami sensors

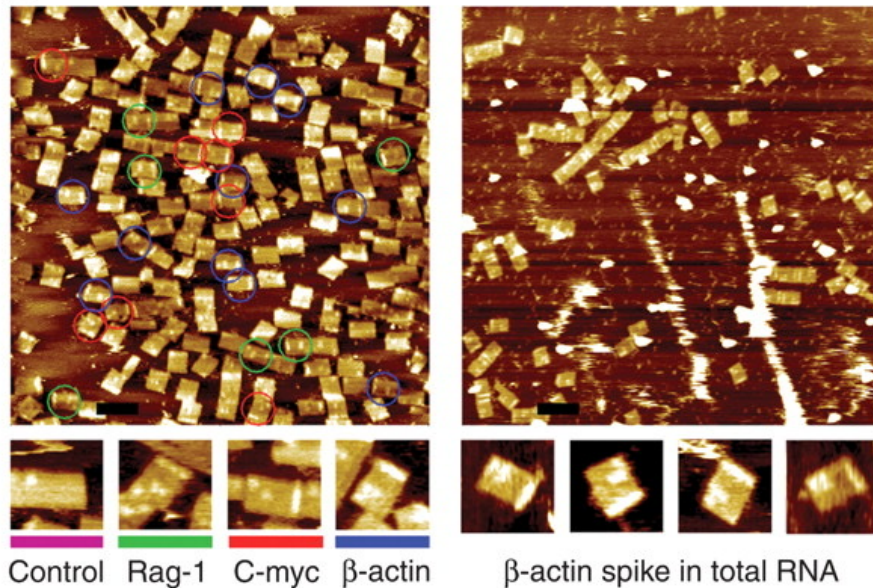


- Only the edge positions are employed for sensing
- Barcode in the index region is used for tile discrimination
- AFM images can show distinct pattern for all tiles

Ke, S. Lindsay, Y. Chang, Y. Liu, H. Yan, *Science* 2008, 319, 180



# Tiles function as high specificity sensors

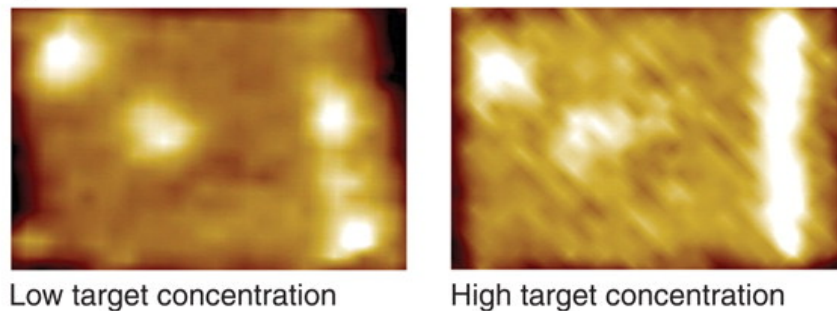


Four tiles (for the four model RNAs) are readily discriminated by their barcodes in a mixture

Spike-in of  $\beta$ -actin RNA into high background of cellular RNA: specific detection

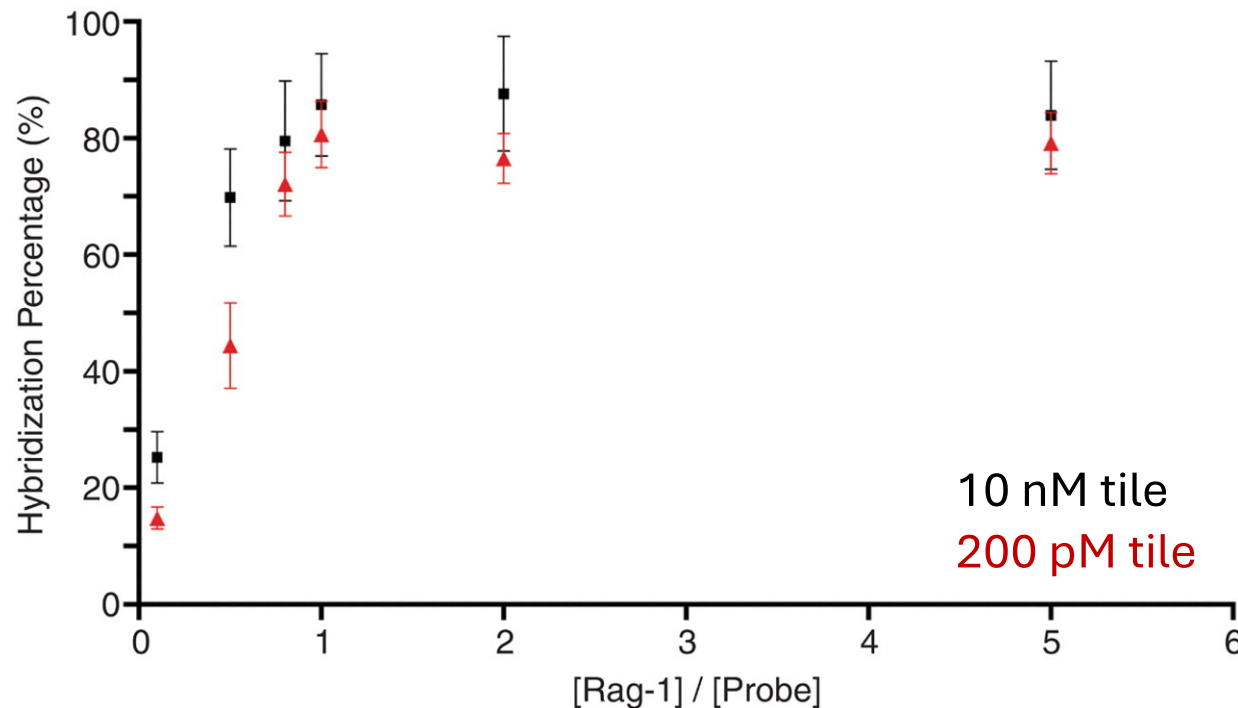
No cross-reactivity observed (only  $\beta$ -actin tiles show binding)

**C**



*Ke et al. Science*  
2008

# Binding is highly specific and stoichiometric



10 nM tile  
200 pM tile

*Ke et al.  
Science  
2008*

- Nearly linear increase at  $[\text{target}]/[\text{probe}] < 1$  observed
- Saturation at  $[\text{target}]/[\text{probe}] > 1$
- Non-Michaelis-Menten binding
- Detection is only limited by concentration of tiles
- Every target molecule is bound at low concentrations.
- This is due to the very high energy of the binding (-50 kcal/mol)

## Conclusion:

These sensor tiles can detect RNA at single molecule resolution

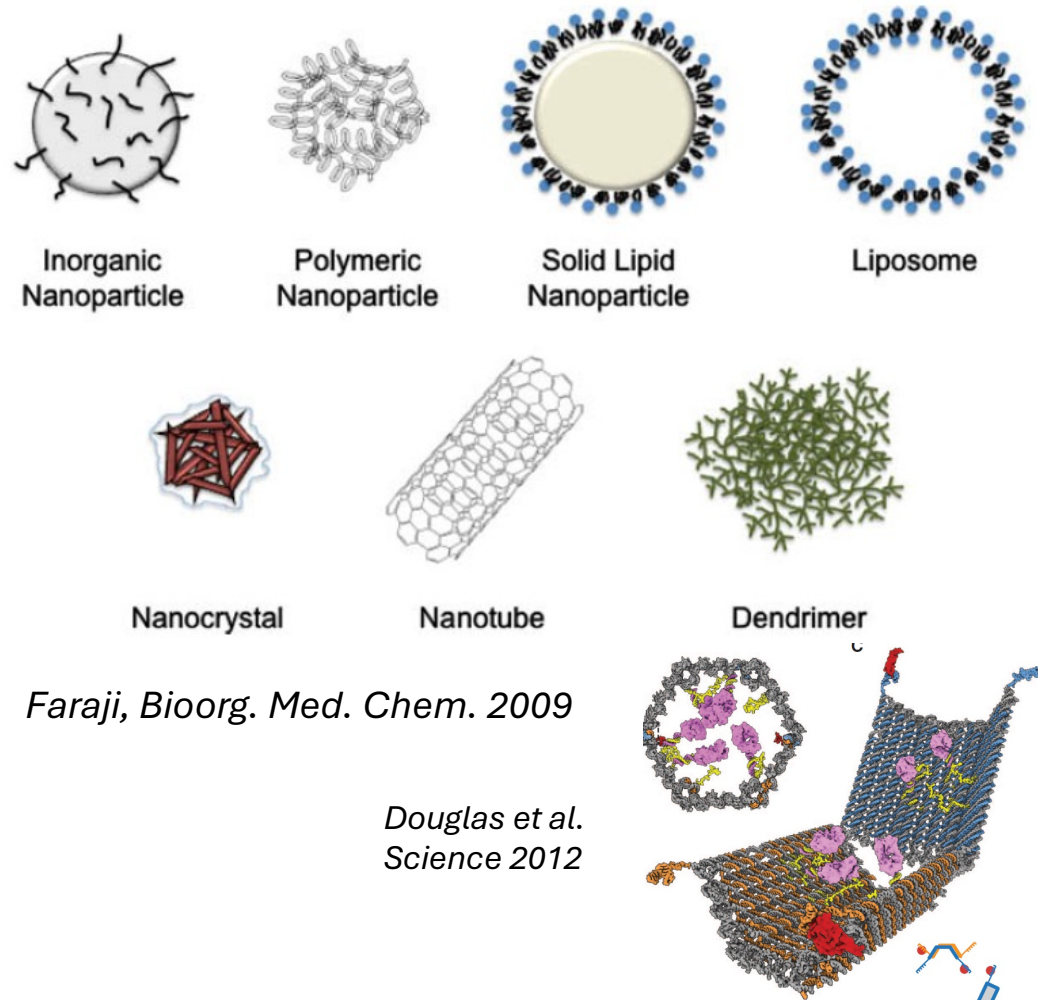
→ Challenge: difficult to estimate the  $K_d$  of this reaction

# Medical Applications: Drug Encapsulation and Controlled Delivery

Nanoparticles employed in drug delivery

## Possible applications:

- Encapsulation of drug molecules
- Retention in cells, improved pharmacokinetics (slow release, long plasma lifetimes)
- Exact targeting of toxic molecules (reduction of side effects)



# Activity

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- Discuss: what are the limitations of DNA origami structures?

# Activity

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Limitation	Details
Size constraint	Limited by scaffold length (~7 kb)
Mechanical flexibility	Susceptible to bending, deformation
Ionic sensitivity	Needs high $Mg^{2+}$ ; unstable in low-salt or in vivo
Yield and folding errors	Can misfold or form aggregates
Functionalization complexity	Difficult to attach and control cargos or proteins
Biological instability	Degradation by nucleases in physiological fluids
Cost and scalability	Expensive at large scales
Limited 3D precision	Less compact/functionally rich than proteins

# Connections

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- <https://connections.swellgarfo.com/game/-Ny0-36i13UCj5NWiwnk>