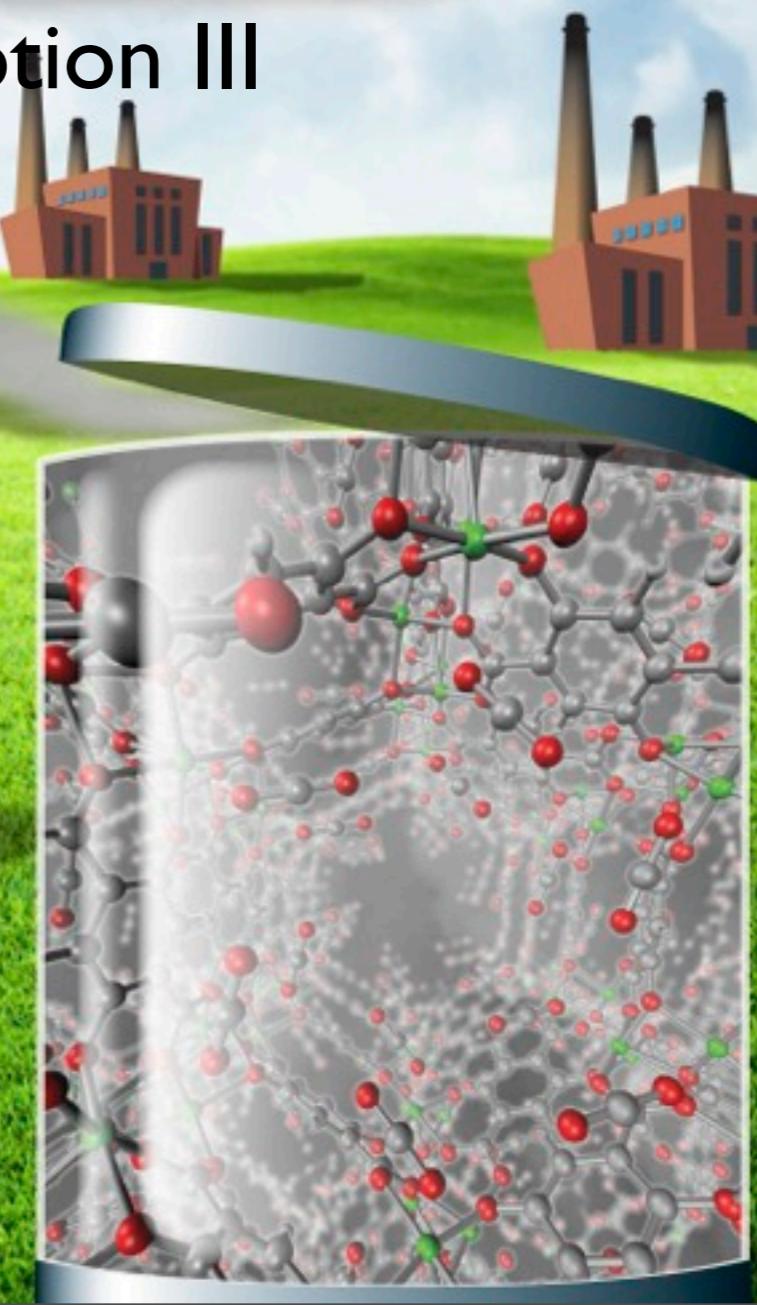


# Introduction to Carbon Capture and Sequestration

## Chapter 6: Adsorption

Thermodynamics of adsorption III  
Henry Coefficients



# Chapter 6: Adsorption

## Thermodynamics of Adsorption III

### (Henry Coefficients)

- Calculate the Henry coefficient of some model materials

## Henry coefficient:

$$K_H = \frac{1}{k_B T} \exp\left(-\frac{\mu^{ex}}{k_B T}\right)$$

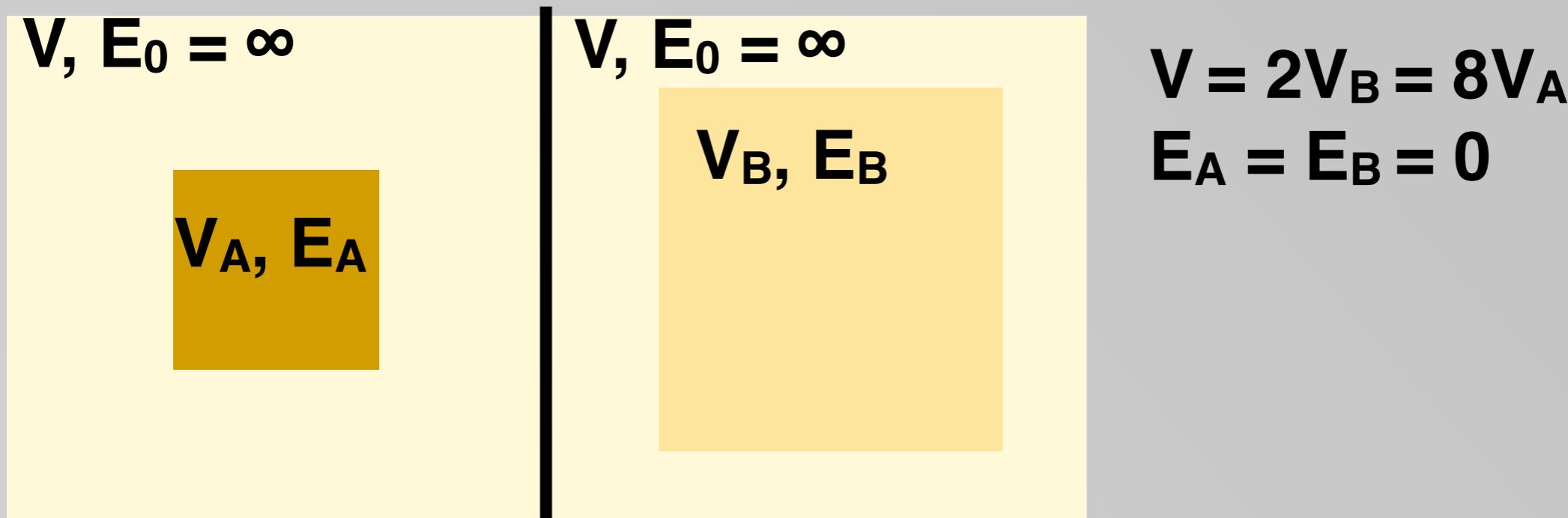
## Statistical Mechanics:

The **average energy** of a CO<sub>2</sub> inserted in a random position in the adsorbent

$$\exp\left(-\frac{\mu^{ex}}{k_B T}\right) = \left\langle e^{-U_{CO_2}/k_B T} \right\rangle_{ads}$$

$$K_H = \frac{1}{k_B T} \left\langle e^{-U_{CO_2}/k_B T} \right\rangle_{ads}$$

# Model adsorbent:



**Model adsorbate:** Ideal gas molecules that feel energy  $E_i$  in the pores

## Question:

- Which material has the highest Henry coefficient?

$V, E_0 = \infty$

$V_A,$   
 $E_A = 0$

$$K_H = \frac{1}{k_B T} \left\langle e^{-U_{CO_2}/k_B T} \right\rangle_{ads}$$

$$K_H^A = \frac{1}{k_B T} \left[ \frac{V_A}{V} e^0 + \frac{V - V_A}{V} e^{-\infty} \right] = \frac{V_A}{k_B T V}$$

Let us look at the limit  $V_A \rightarrow V$ :

$$K_H^A = \frac{1}{k_B T}$$

ideal gas

Loading:

$$\sigma = K_H^A p = \frac{p}{k_B T}$$

For  $V_A < V$ , the loading is:

Entropy loss

$$\sigma = K_H^A p = \frac{p}{k_B T} \frac{V_A}{V}$$

$V, E_0 = \infty$

$V_A,$   
 $E_A = 0$

$V, E_0 = \infty$

$V_B,$   
 $E_B = 0$

Pore A:

$$K_H^A = \frac{1}{k_B T} \left[ \frac{V_A}{V} e^0 + \frac{V - V_A}{V} e^{-\infty} \right] = \frac{V_A}{k_B T V}$$

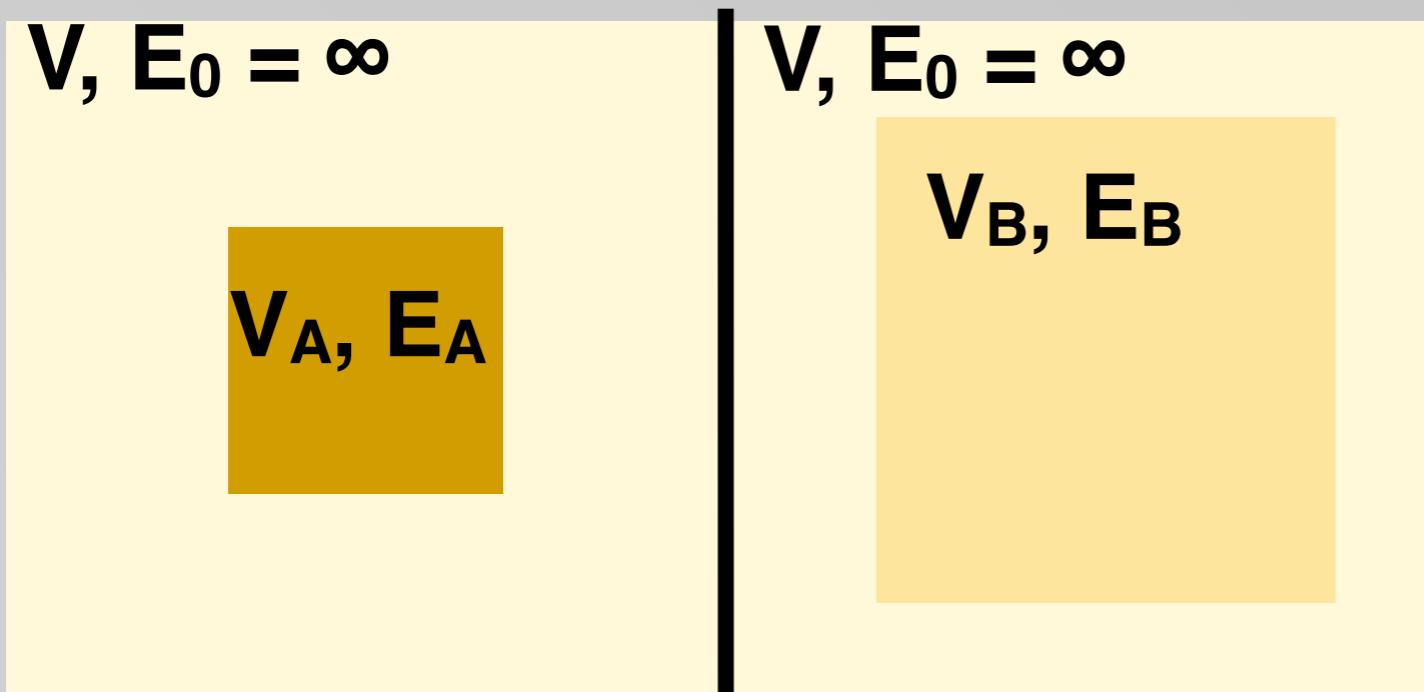
Pore B:

$$K_H^B = \frac{1}{k_B T} \left[ \frac{V_B}{V} e^0 + \frac{V - V_B}{V} e^{-\infty} \right] = \frac{V_B}{k_B T V}$$

Ratio of the Henry coefficients:

$$\frac{K_H^B}{K_H^A} = \frac{V_B}{V_A}$$

# Model adsorbent:



$$V = 2V_B = 8V_A$$
$$E_A = 2E_B = -20 \text{ kJ/mole}$$

## Questions:

- At  $T=0$  which material has the highest Henry coefficient?
- At  $T \gg$  which material has the highest Henry coefficient
- For which temperature the Henry coefficients of the two materials are equal?

$V, E_0 = \infty$

$V_A, E_A$

$V = 2V_B = 8V_A$   
 $E_A = 2E_B = -20 \text{ kJ/mole}$

$$K_H = \frac{1}{k_B T} \left\langle e^{-U_{CO_2}/k_B T} \right\rangle_{ads}$$

**Material A:**

$$K_H = \frac{1}{k_B T} \left[ \frac{V_A}{V} e^{-\frac{E_A}{k_B T}} + \frac{(V - V_A)}{V} e^{-\frac{\infty}{k_B T}} \right]$$

$$K_H = \frac{1}{k_B T} \frac{V_A}{V_0} e^{-\frac{E_A}{k_B T}}$$

Energy gain      Entropy loss

$$k_B T \ln(k_B T K_H) = -E_A + k_B T \ln \frac{V_A}{V}$$

# Summary

Henry coefficient: balance between the entropy loss caused by the confinement and the energy gain caused by the interactions with the wall