

1 a) Which is the rate-limiting step in an electrophilic aromatic substitution reaction?

b) Draw the corresponding mesomeric structures of the sigma-complex of benzene.

c) Substituted aromatics can react faster or slower than benzene ( $R = H$ ). How is  $R = \text{MeO}$  and  $R = \text{NO}_2$  generally influencing the reactivity?



2 A mixture of styrene (vinylbenzene) and 1,3,5-trimethoxybenzene is treated with a strong non-nucleophilic acid ( $H^+$ , or for a real example tetrafluoroboric acid). Please draw the reactants, intermediates and the product(s) that you would expect.

3 Propose a synthetic route to prepare compound **A** **selectively**. Only toluene is the only allowed aromatic starting material (more than one step is required).

