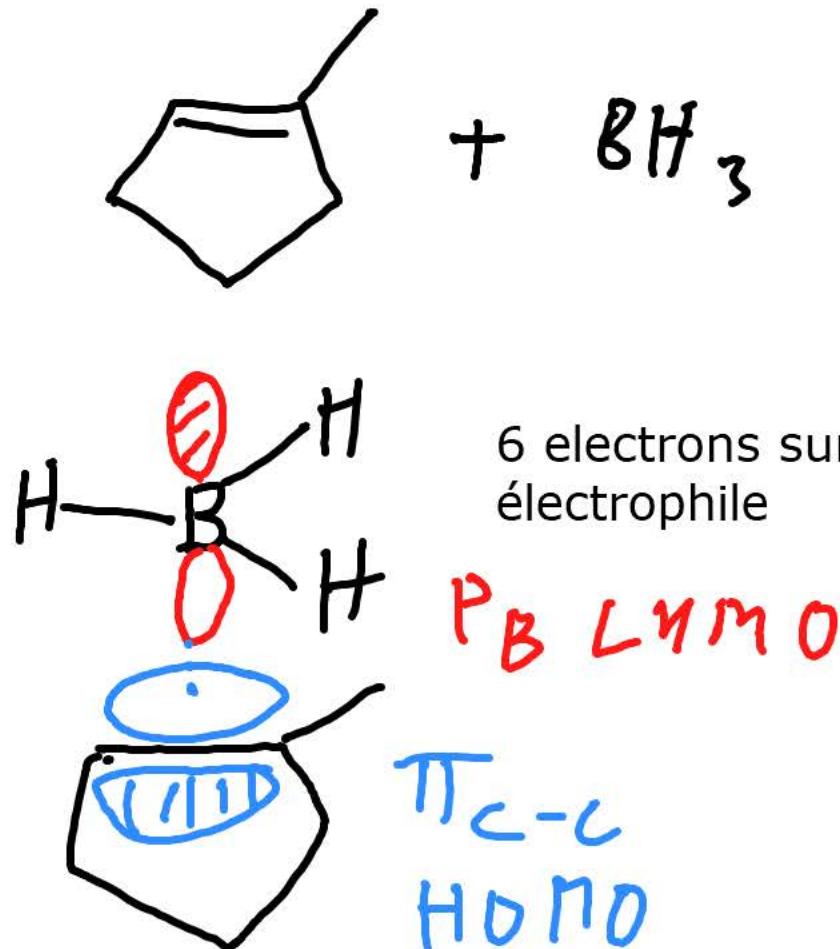


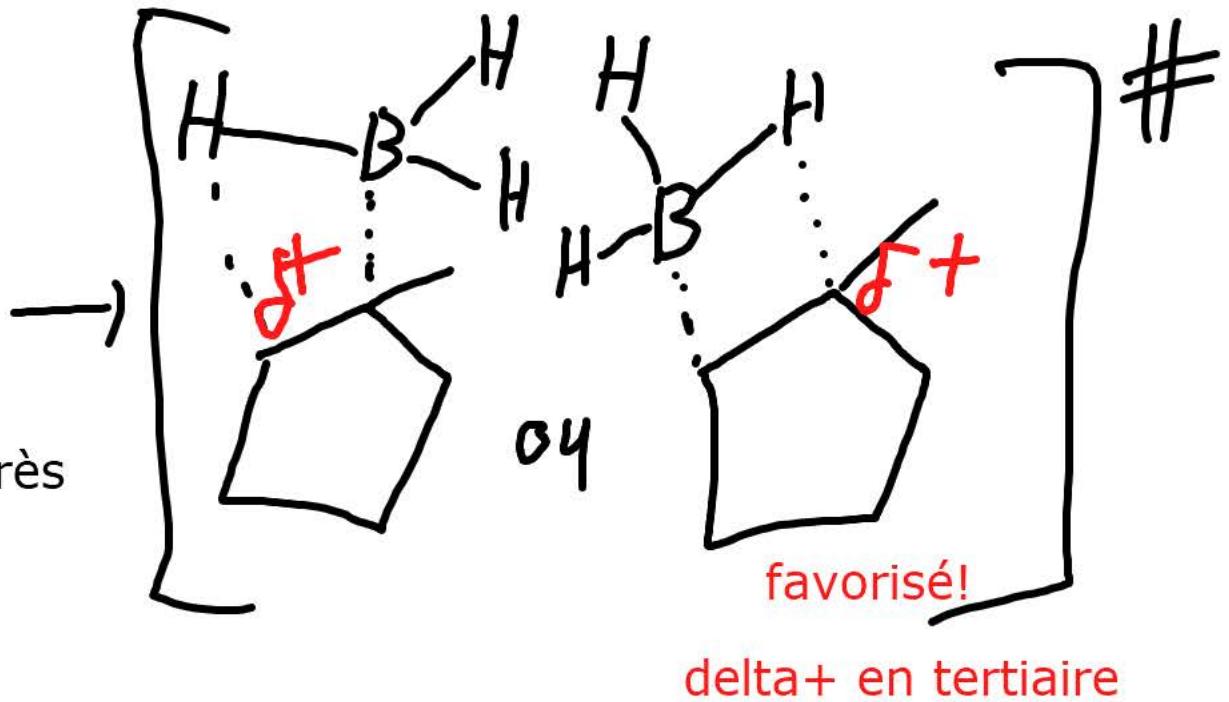
hydroboration



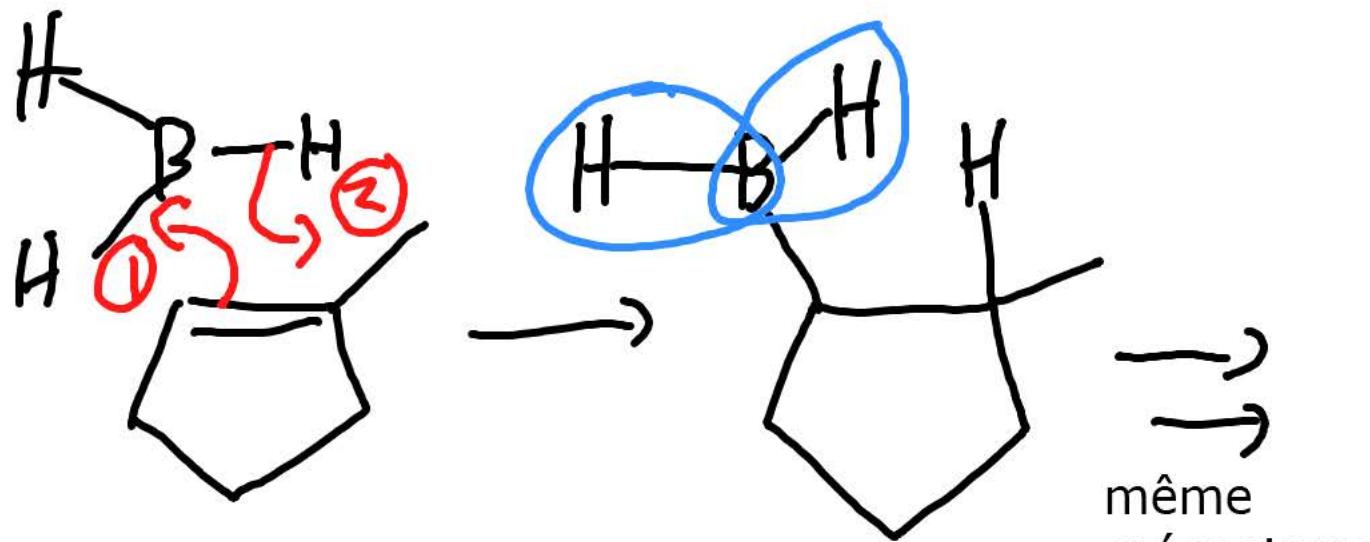
6 électrons sur B, très électrophile

P B L Y M O

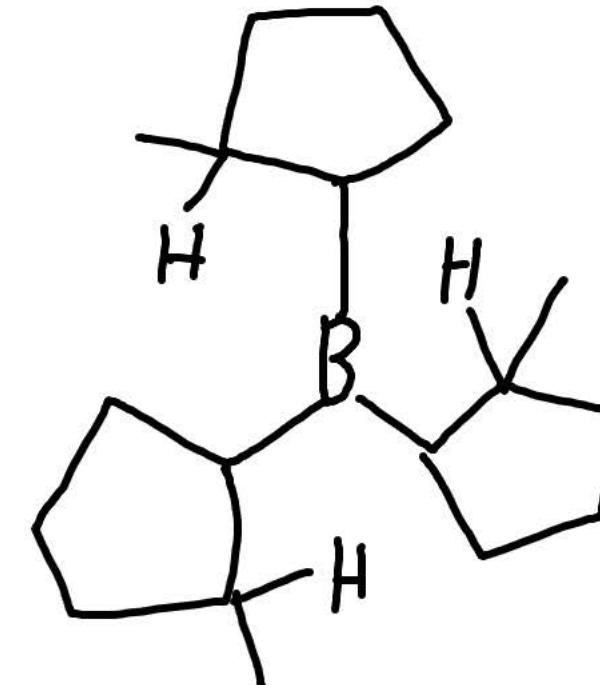
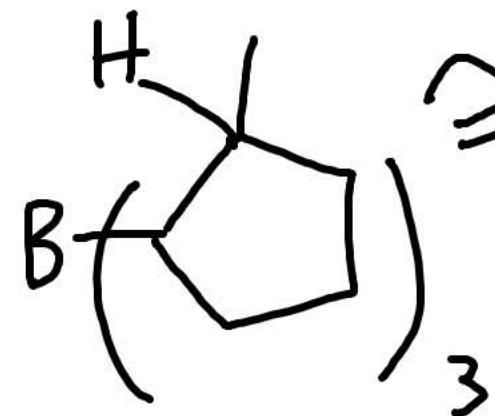
π_{c-c}
HONO



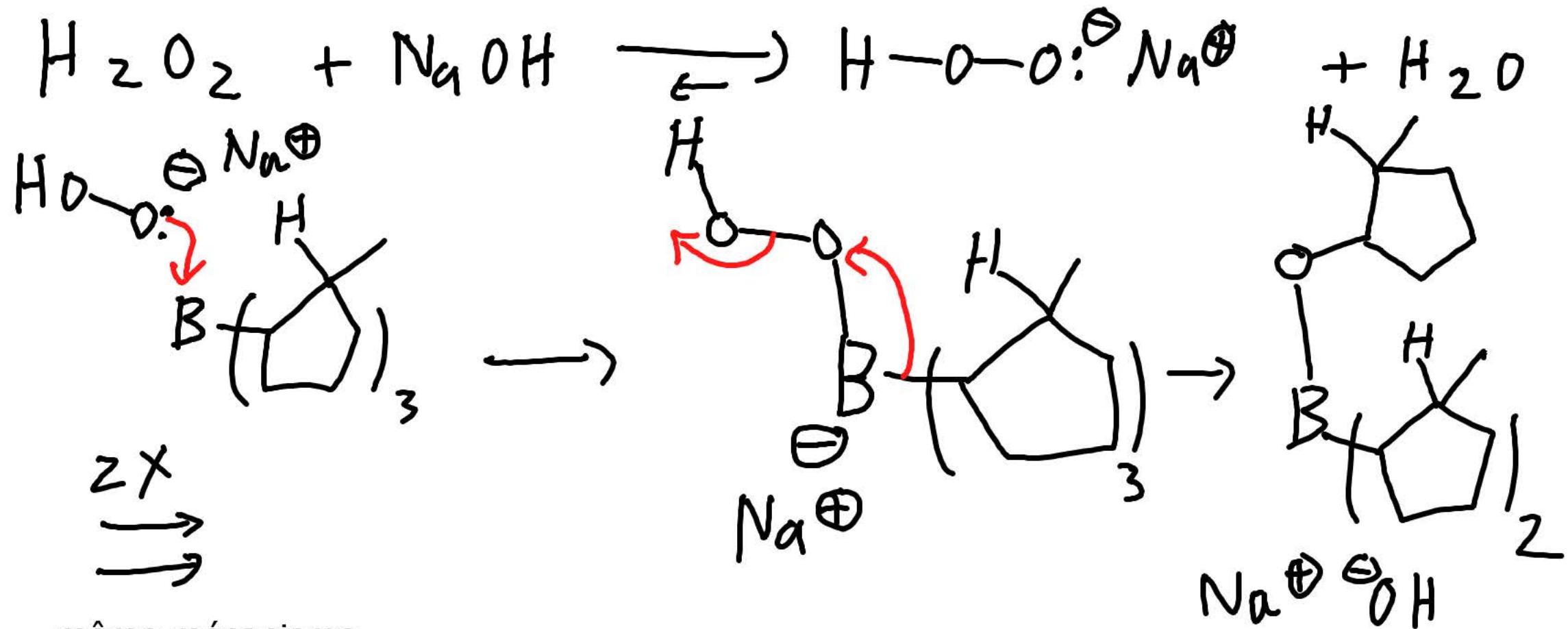
état de transition à 4 atomes
La distance C-B est plus
courte que la distance C-H
dans l'état de transition

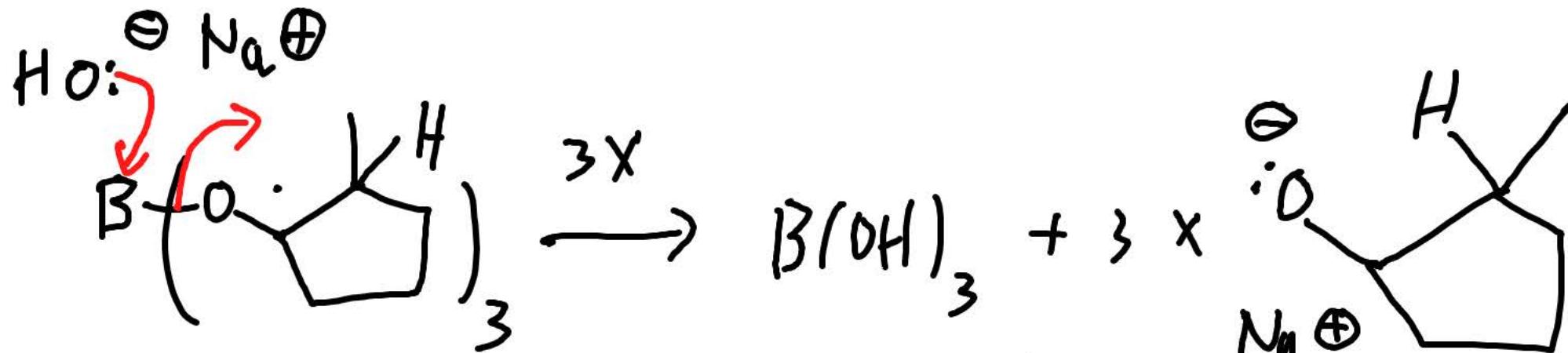


les 3 liaisons B-H peuvent réagir avec l'alcène



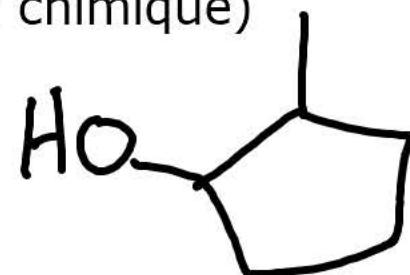
étape d'oxydation



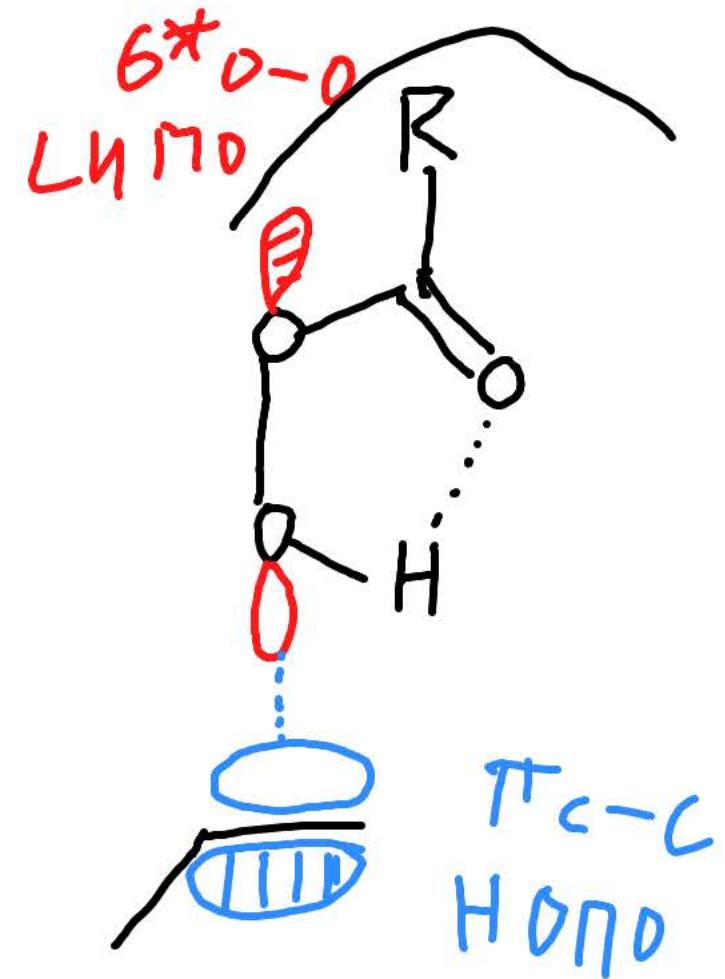
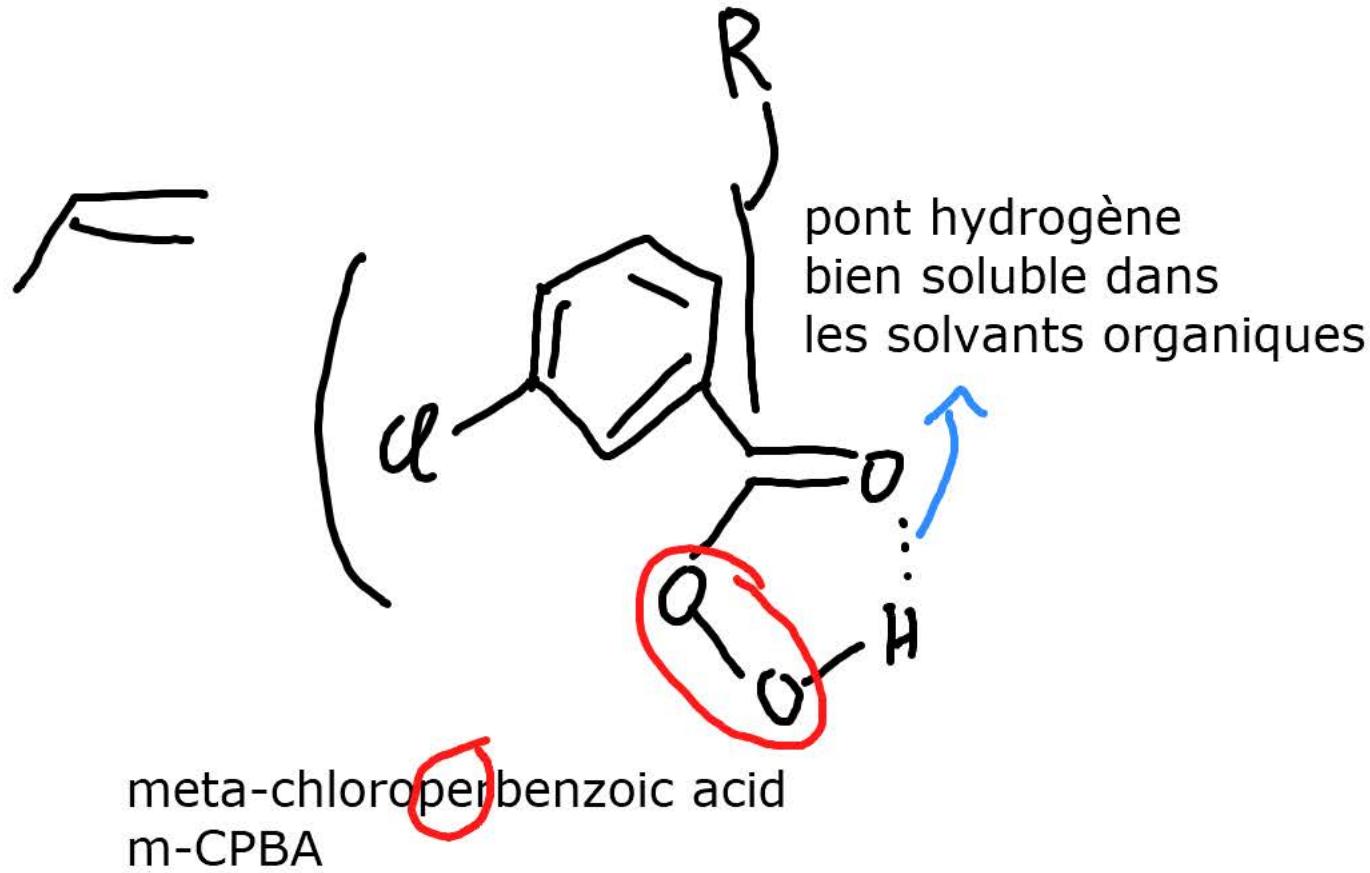


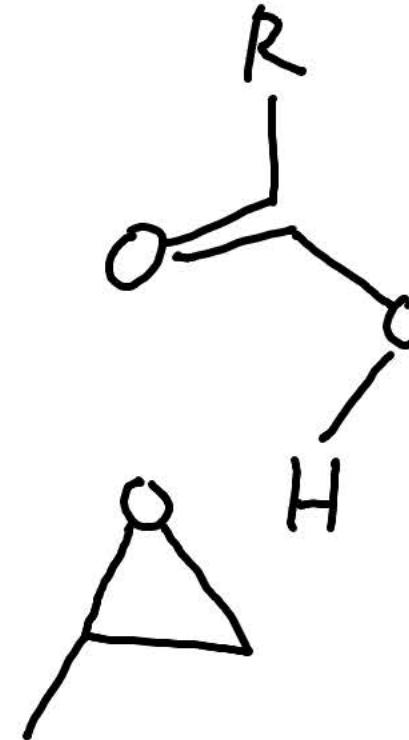
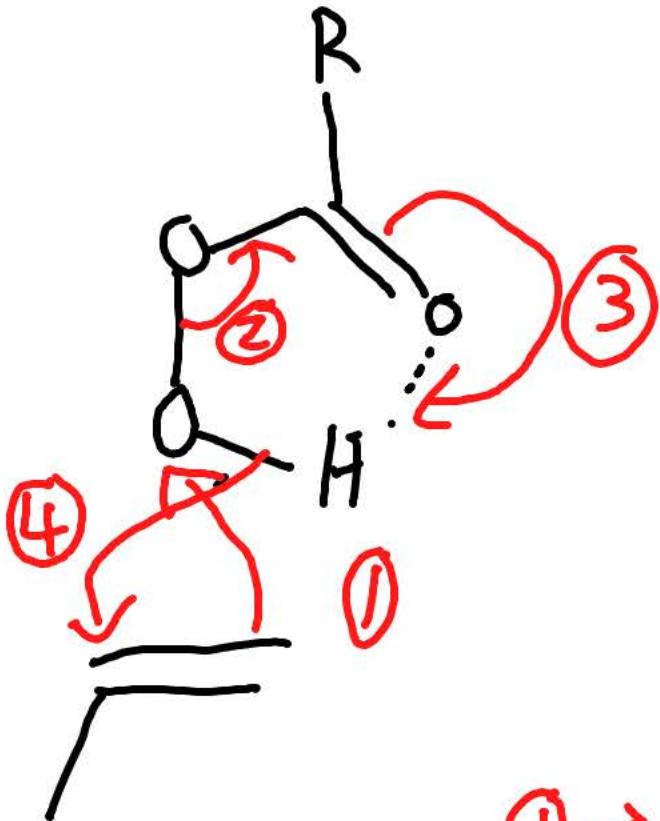
work-up acidic
(souvent pas indiqué dans
l'équation chimique)

au total: addition de H_2O sur
alcène, mais anti-Markovnikov



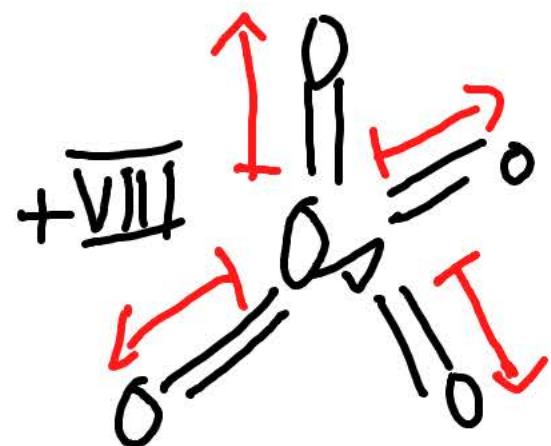
époxidation avec les peroxydes





acide: pas de pont hydrogène
moins soluble, précipite

réaction de dihydroxylation avec OsO₄



4 substituants, tétraédrique

dipole globale: 0

lipophiliqe: passer la barrière
sang-cerveau
sublime facilement