

### Part 1: The Branching Fraction

We must determine the probability that an inhaled atom will decay *inside* the lung rather than being cleared biologically.

**1. Calculate Decay Constants ( $\lambda$ )** The decay constant is defined as  $\lambda = \ln(2) / T_{\text{half}}$ .

- **Physical Constant:**  $\lambda_{\text{phys}} = 0.693 / 138.4 \text{ days} = 0.00501 / \text{day}$
- **Biological Constant:**  $\lambda_{\text{bio}} = 0.693 / 50 \text{ days} = 0.01386 / \text{day}$

**2. Calculate the Fraction of Decays ( $F_{\text{decay}}$ )** The fraction of atoms that decay in the lung is the ratio of the physical decay rate to the total removal rate.

$$F_{\text{decay}} = \lambda_{\text{phys}} / (\lambda_{\text{phys}} + \lambda_{\text{bio}})$$

$$F_{\text{decay}} = 0.00501 / (0.00501 + 0.01386)$$

$$F_{\text{decay}} = 0.265$$

Only **26.5%** of the Polonium trapped in the lungs actually emits radiation there, the remaining  $\sim 73.5\%$  is cleared by the body before it can decay.

### Part 2: Source Term Estimation

**1. Daily Activity Inhaled** (1 pCi = 0.037 Bq).

$$A_{\text{daily}} = (30 \text{ cigs}) * (0.516 \text{ pCi}) * (0.037 \text{ Bq/pCi}) * 0.50 * A_{\text{daily}} = 0.286 \text{ Bq/day}$$

**2. Convert Activity to Number of Atoms ( $N_{\text{daily}}$ )**

- $T_{\text{phys}}$  (in seconds) =  $138.4 * 86,400 = 11,957,760 \text{ s}$
- $\lambda_{\text{phys}}$  (in seconds) =  $5.795 \times 10^{-8} / \text{s}$

$$N_{\text{daily}} = 0.286 / (5.795 \times 10^{-8}) \quad N_{\text{daily}} = 4.935 \times 10^6 \text{ atoms/day}$$

**1. Total Atoms Deposited in 1 Year ( $N_{\text{total}}$ )**

$$N_{\text{total}} = (4.935 \times 10^6) * 365 \text{ days}$$

$$N_{\text{total}} = 1.801 \times 10^9 \text{ atoms}$$

### Part 3: Energy Deposition

We calculate the energy only for the fraction of atoms that actually decay ( $F_{\text{decay}}$ ).

$$E_{\text{total}} = N_{\text{total}} * F_{\text{decay}} * E_{\text{alpha}}$$

$$E_{\text{total}} = (1.801 \times 10^9) * 0.265 * 5.307 \text{ MeV}$$

$$E_{\text{total}} = 2.53 \times 10^9 \text{ MeV}$$

**Convert to Joules (1 MeV =  $1.602 \times 10^{-13} \text{ J}$ ):**

$$E_{\text{total}} (\text{J}) = (2.53 \times 10^9) * (1.602 \times 10^{-13})$$

$$\mathbf{E_{\text{total}} (\text{J}) = 4.05 \times 10^{-4} \text{ J}}$$

#### *Part 4: Dose Calculation*

**1. Absorbed Dose (D)** Using the specific mass of the bronchial epithelium (0.02 kg).

$$D = \text{Energy (J)} / \text{Mass (kg)}$$

$$D = (4.05 \times 10^{-4}) / 0.02$$

$$\mathbf{D = 20.2 \text{ mGy}}$$

**2. Dose Equivalent (H)**

Adjusting for radiation type (Alpha particles,  $w_R = 20$ ).

$$H = D * w_R$$

$$\mathbf{H = 404 \text{ mSv}}$$

**3. Effective Dose (E)**

Adjusting for tissue sensitivity ( $w_T = 0.12$ ).

$$E = H * w_T$$

$$\mathbf{E = 48.5 \text{ mSv}}$$