

# Homework 3

## 03/10/2013

### (Gaussian) Freely Jointed Chain with Varying Step Length

Consider a freely jointed chain in  $d$  dimensions, but let each bond vector  $\vec{r}_i$  be drawn from the Gaussian probability distribution

$$p(\vec{r}) = \frac{1}{(2\pi a^2)^{\frac{d}{2}}} e^{-\frac{\vec{r}^2}{2a^2}} .$$

Find the probability distribution of the end to end distance  $\vec{R}_e = \sum_{i=1}^N \vec{r}_i$  and calculate  $\langle \vec{R}_e^2 \rangle$ . **Hint:** remember that the Gaussian integral in  $d$  dimensions is  $\int d\vec{r} e^{-\alpha \vec{r}^2} = (\pi/\alpha)^{d/2}$ .

### Equivalent Freely Jointed Chain

A common feature of all ideal chains is that the mean square end-to-end distance is proportional to the number of monomers, i.e.  $\langle R_{end}^2 \rangle \propto n$ .

The simplest ideal chain is the *freely jointed chain model* with constant bond length  $l$ ; it has  $\langle R_{end}^2 \rangle = nl^2$  and its maximum end-to-end distance is  $R_{max} = nl$ .

Any ideal polymer made of  $n$  bonds of length  $l$  can be “macroscopically” described by an **equivalent freely jointed chain** made by  $N$  freely-jointed bonds of effective length  $b$  and with the same  $\langle R_{end}^2 \rangle$  and  $R_{max}$  of the actual polymer. The length of the effective bond  $b$  is called the **Kuhn length** of the polymer.

Calculate the Kuhn length of:

1. the model of the homework 2, i.e.  $\langle R_{end}^2 \rangle = nl^2 \frac{1+\epsilon}{1-\epsilon}$ ,  $R_{max} = ?$ ;
2. the freely rotating chain with fixed bond angle  $\theta$ , i.e. a chain where the angle  $\theta_i$  formed by segments  $\vec{r}_i$  and  $\vec{r}_{i+1}$  is fixed to a certain value  $\theta$ , while the angle  $\phi_i$  is taken uniformly in  $[0, 2\pi]$ . In this case, it is easy to show (do it!) that  $\langle R_{end}^2 \rangle = nl^2 \frac{1+\cos\theta}{1-\cos\theta}$ . Who is  $R_{max} = ?$ ;
3. a polyethylene chain with a fixed bond angle  $\theta = 68^\circ$ , (actual) bond length  $l = 1.54 \text{ \AA}$ ,  $\frac{\langle R_{end}^2 \rangle}{nl^2} = 7.4$ , and the same  $R_{max}$  as for the freely rotating chain.

The double-stranded DNA has a bond length of about  $l \approx 3 \text{ \AA}$  and a Kuhn length  $b \approx 100 \text{ nm}$ . If we model it as a freely rotating chain, what is the approximate value of the bond angle  $\theta$ ?

**Hint:** is the double-stranded DNA a stiff polymer? What does this tell us about  $\theta$ ?

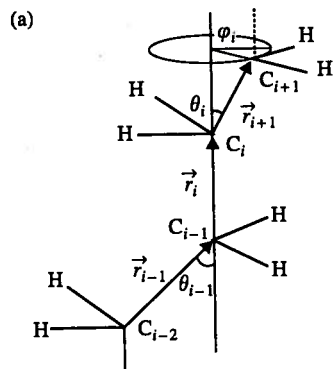


Figure 1: In the freely rotating chain model the angle  $\theta$  is fixed once and for all (from *Polymer Physics*, M.Rubinstein and R.H.Colby).