



CHEMICAL BIOLOGY

- Moodle: <https://go.epfl.ch/CH-313>
 - Lecture slides (evening before the lecture)
 - Distributed presentation topics (assignments)
 - Forum (for questions and announcements)
- Examination (written, graded, detailed information will follow)
- Contact:
 - Moodle forum (for questions)
 - markus.jeschek@epfl.ch
- **“Concepts over details!”**
- **Interact! Ask! Discuss! Anytime!**

Group Presentations

- Critical discussion of primary literature
- Illustrative examples for topics from the lecture

- Why?
 - Repetition of core concepts, techniques etc.
 - Presentation skills and critical discussion of research
 - Insight into current research topics

- How?
 - Two students per group
 - Assignments distributed one week before delivery of presentation (via Moodle)
 - **Send slides: markus.jeschek@epfl.ch (Mon evening before presentation)**
 - **15 min presentation (both group members should present!) + Q&A**

EPFL Tipps for Group Presentations

- Rough structure
 - Short intro on general topic
 - Main presentation according to assignment
 - Brief outlook incl. points of criticism/open questions/personal opinion as kick-starter for the discussion
- Everybody should participate in the discussion, incl. constructive(!) feedback on presentation style
- Questionnaires with different points, feedback by peers
- Typical assignment:
 - You will receive a certain topic including a related publication
 - Introduce the topic using the publication
 - present the motivation behind the research, methodology, key results (not every graph!)
 - Additional questions will be provided hinting towards central points
 - Be encouraged to look/present beyond the questions and the provided paper

Group Presentations – Schedule

#	Name1	Name2	Presentation on...	Assignment on...
1	Winger Quentin	Jeremy	Sep 23, 2025	Sep 16, 2025
2	Ema	Ariane	Sep 30, 2025	Sep 23, 2025
3	Benjamin	Matthieu	Oct 7, 2025	Sep 30, 2025
4	Ivana	Ipek	Oct 14, 2025	Oct 7, 2025
5	Mridhula	Elodie	Oct 28, 2025	Oct 21, 2025
6	Abigail	Robin	Nov 4, 2025	Oct 28, 2025
7	Eva	Florian	Nov 11, 2025	Nov 4, 2025
8	Bastien	Axel	Nov 18, 2025	Nov 11, 2025
9	Melodie	Siolène	Nov 25, 2025	Nov 18, 2025
10	Nicole	Maria	Dec 2, 2025	Nov 25, 2025

Course Topics – Overview

- Week 1 | Introduction + DNA
- Week 2 | DNA
- Week 3 | DNA
- Week 4 | DNA
- Week 5 | DNA/RNA
- Week 6 | RNA/Translation
- Week 7 | Translation
- **Week 8 | Enzymes (Zoom)**
- Week 9 | Enzymes (Zoom)
- Week 10 | Enzymes/Metabolism (Zoom)
- Week 11 | Metabolism (Zoom)
- Week 12 | Engineering
- Week 13 | Engineering
- Week 14 | LSAM Intro + Exam Preparation

**!Due to paternity leave
the next lectures will be
delivered via Zoom!**

[tentative schedule]

- Week 1 | Introduction + DNA
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<https://epfl.zoom.us/j/68900732223>

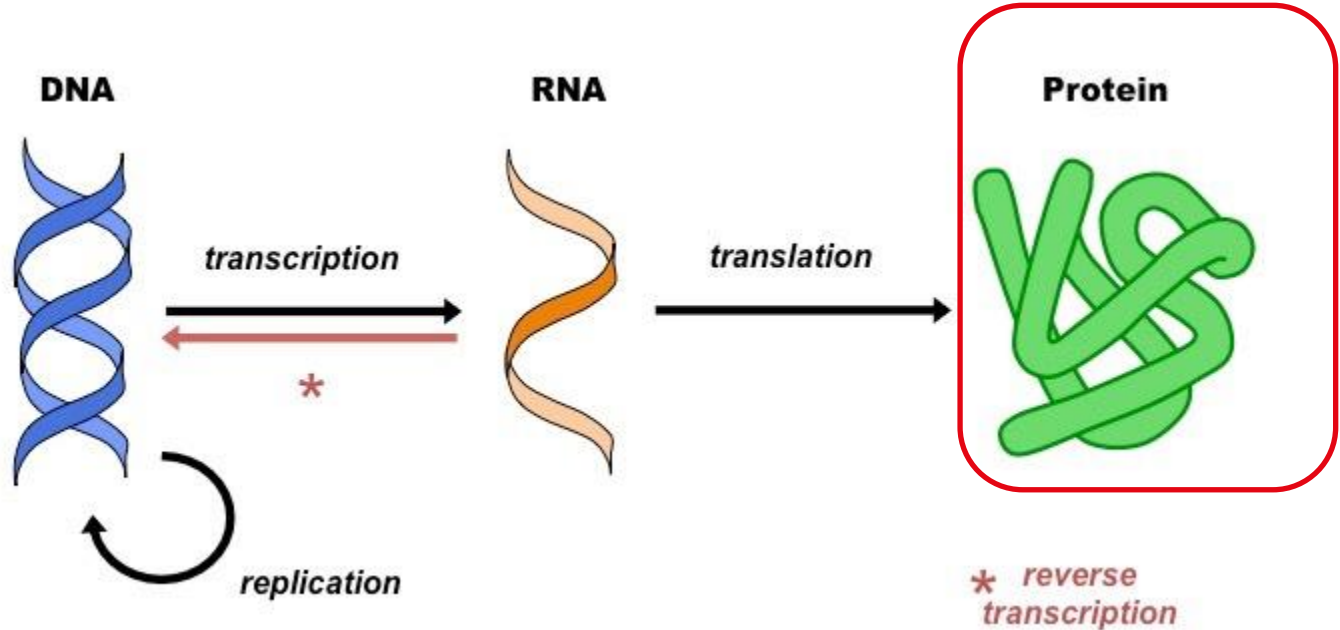
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[tentative schedule]

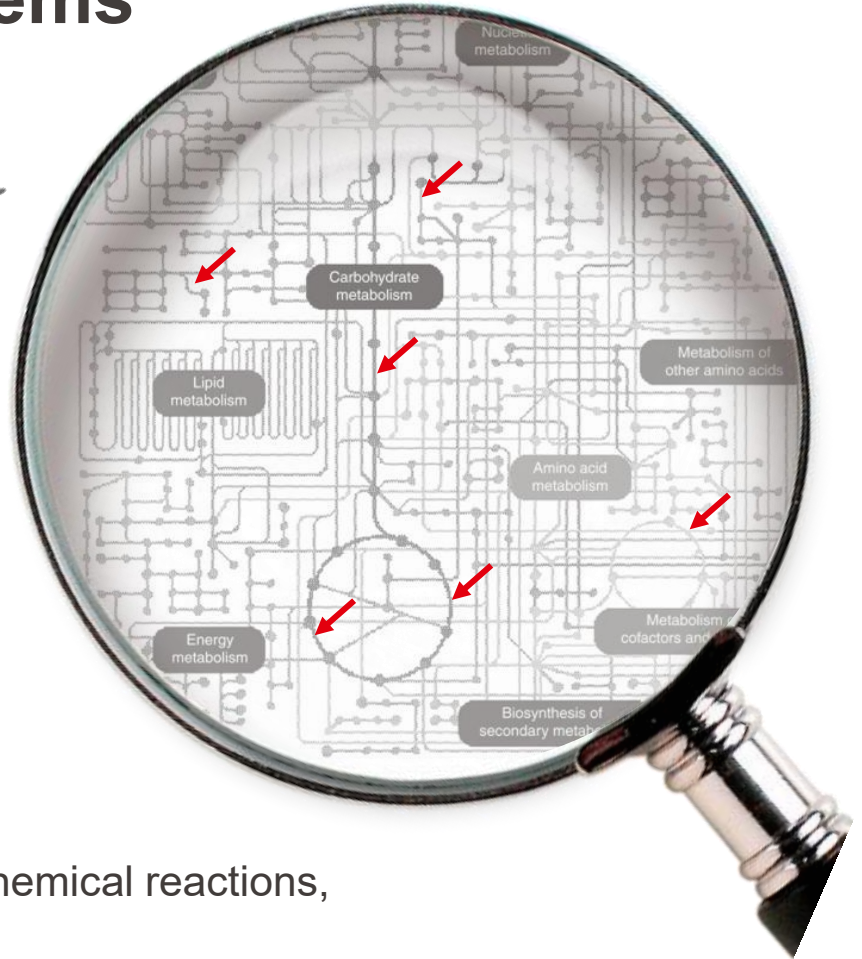
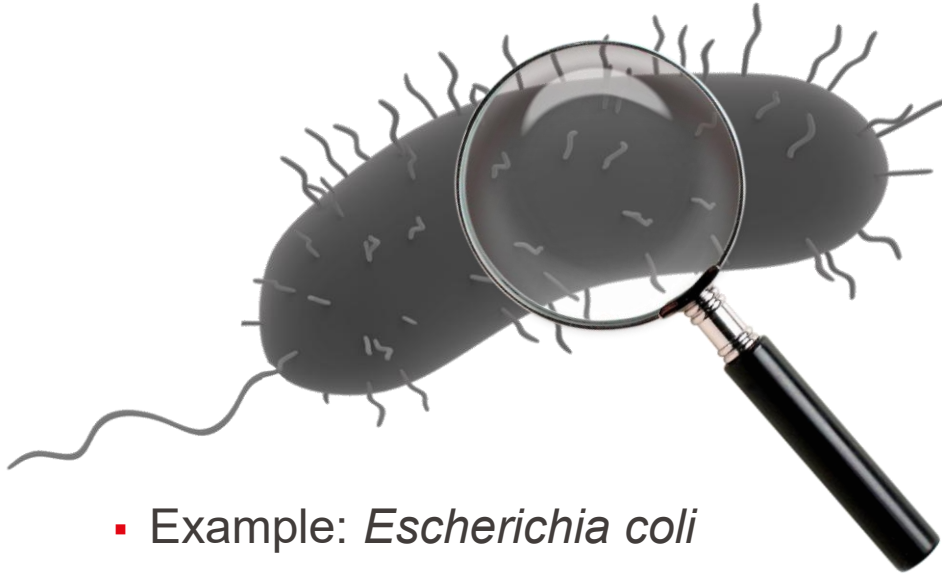
Protein – Enzymes

The Central Dogma

- ...of Molecular Biology



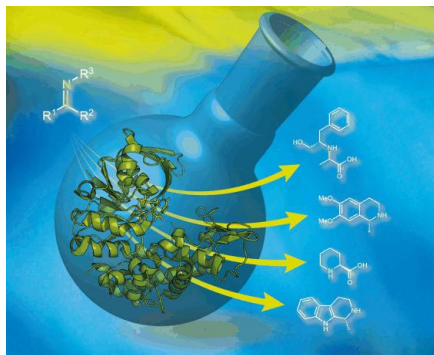
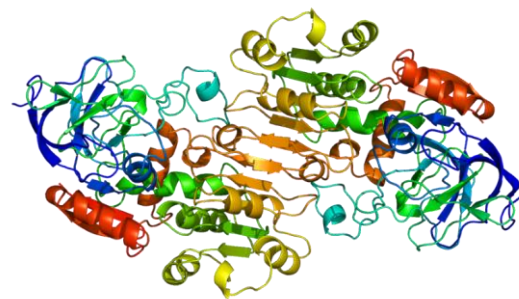
* reverse transcription



- Example: *Escherichia coli*

- approx. 4000 genes
- **600-800 enzymes**
- approx. 1000 metabolites
- Complex network of molecules, biochemical reactions, regulation and interaction

- catalysts built up of biological macromolecules
- “biocatalysts” (the catalysts of life)
- mostly proteins! (rarely nucleic acids, ribozymes etc.)
- “enzyme” from Greek “en zýmē” (“in yeast”/“in leaven”)
- Initially defined as sth. “dead”/ “outside of living cells”
- field of biocatalysis



primary structure

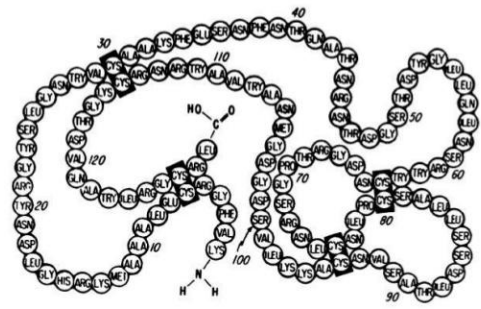
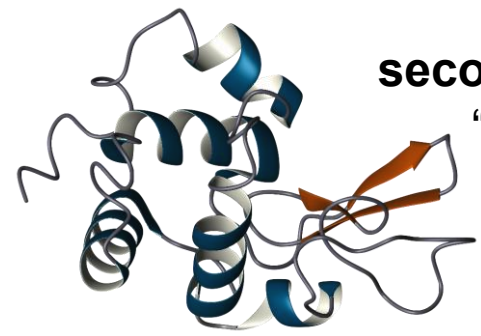
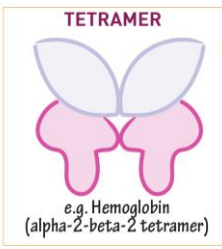
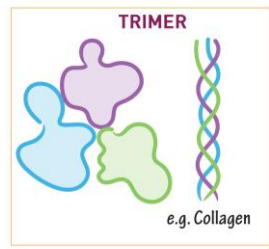
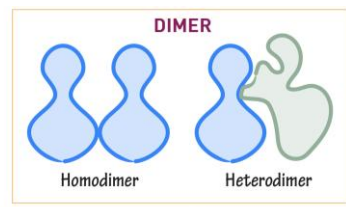
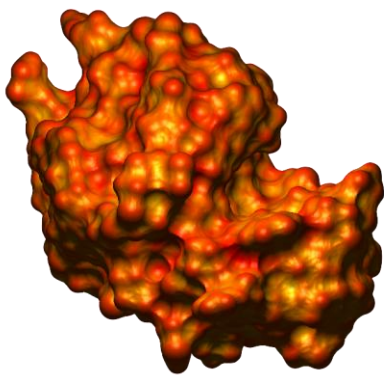


FIG. 1.—Amino acid sequence of hen egg-white lysozyme reproduced from Canfield and Liu, 1965.



secondary structure
"ribbon" view

tertiary structure
"surface" view

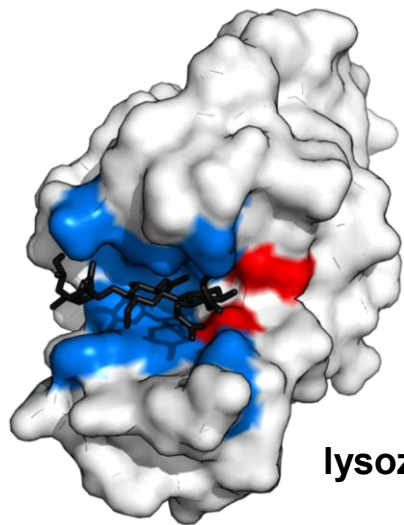


quaternary structure

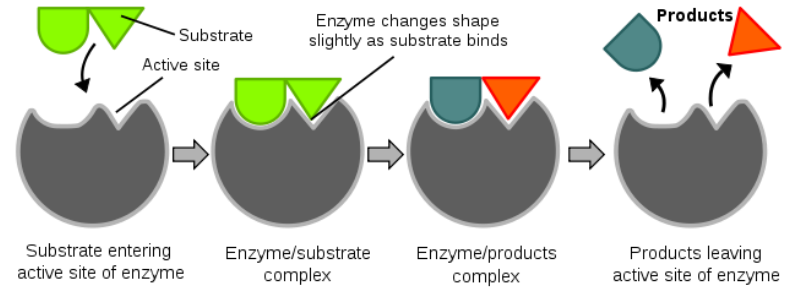
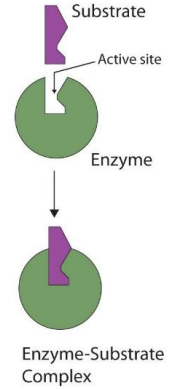
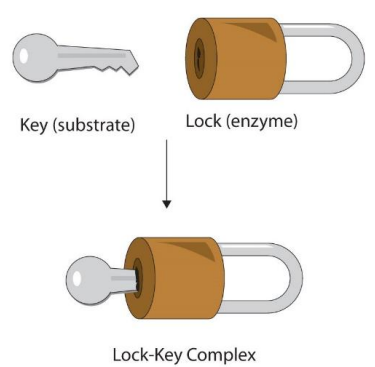
RECAP!

Q: What does lysozyme do? Where do you find it?

- (Main) functional site in the protein
- Few key amino acid residues directly responsible for catalysis
- Substrate binding and conversion to product
- “Lock and key” vs. “induced fit” mechanism



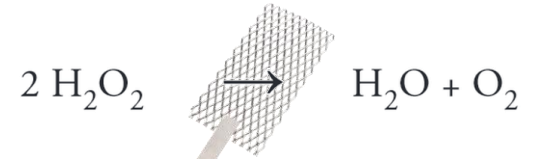
lysozyme active site



RECAP!

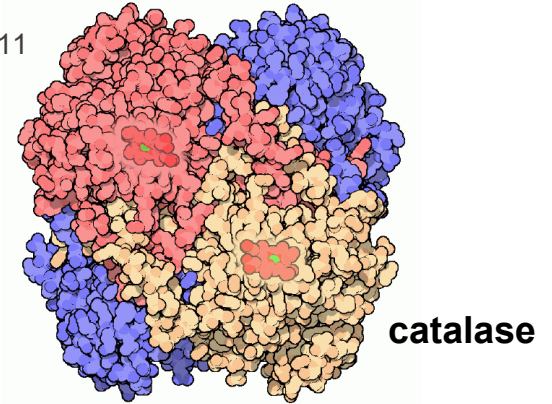
- 1783 Spallanzani digests meat with gastric juice
- 1836 Schwann calls the corresponding active substance „pepsin“
- 1876 Kühne coins the term „enzyme“ (=„in yeast“), in recognition of the role of yeast in alcoholic fermentation
- 1897 Buchner carries out alcoholic fermentation of with a cell-free yeast extract → end of "vis vitalis"
- 1908 Rosenthaler uses almond extracts to achieve chemical transformations
- 1926 Sumner crystallizes urease from jack bean: proteins!
- 1930 Northrop establishes firmly the protein character with a series of crystallization projects, starting with pepsin
- 1951 Sanger & Tuppy determine amino acid sequence of insulin beta-chain – start of modern era of molecular biology
- 1978 Noble prize for restriction enzymes (Arber, Smith)
- 1980's Directed evolution (enzyme engineering)
- 2008 First computer-designed enzyme
- 2018 Nobel Prize in Chemistry for Directed Evolution of Enzymes (F. H. Arnold)

- Example: decomposition of hydrogen peroxide

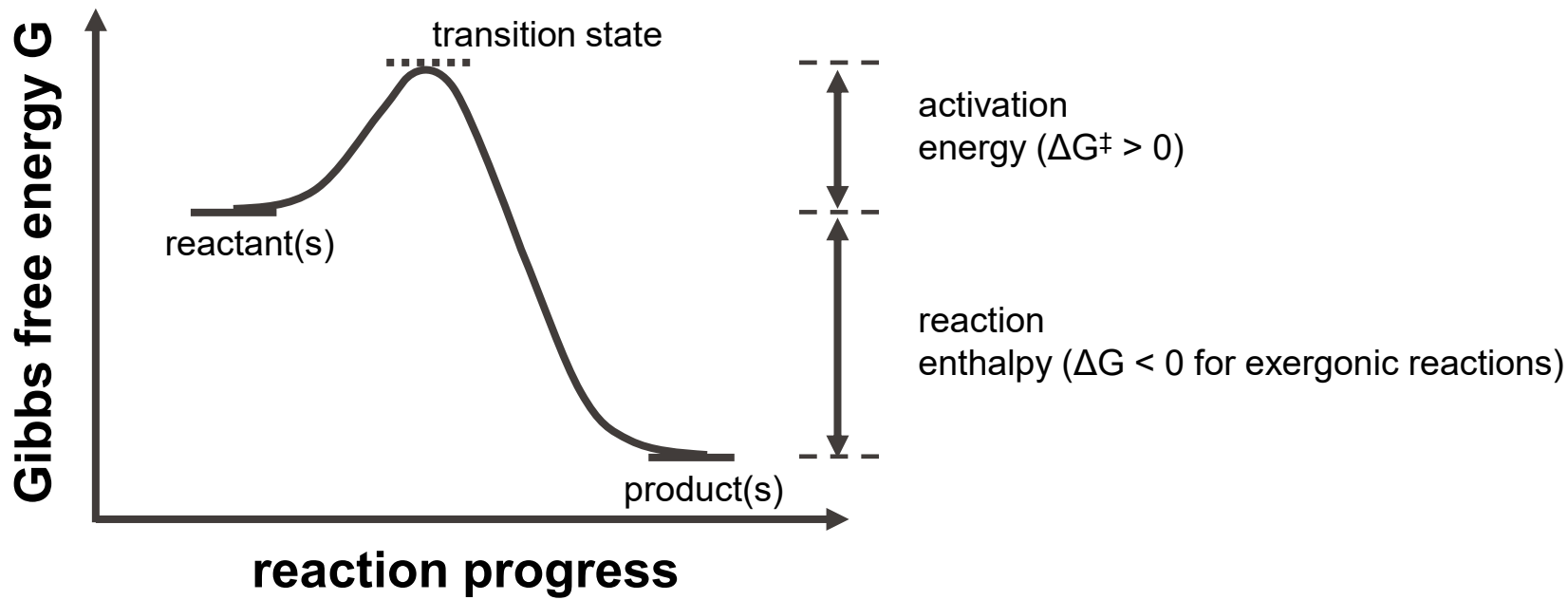


- without catalyst: relative reaction speed := 1
- with platinum: relative reaction speed = $\sim 2 \times 10^4$
- with catalase: relative reaction speed = $\sim 2 \times 10^{11}$

- How do enzymes achieve that?

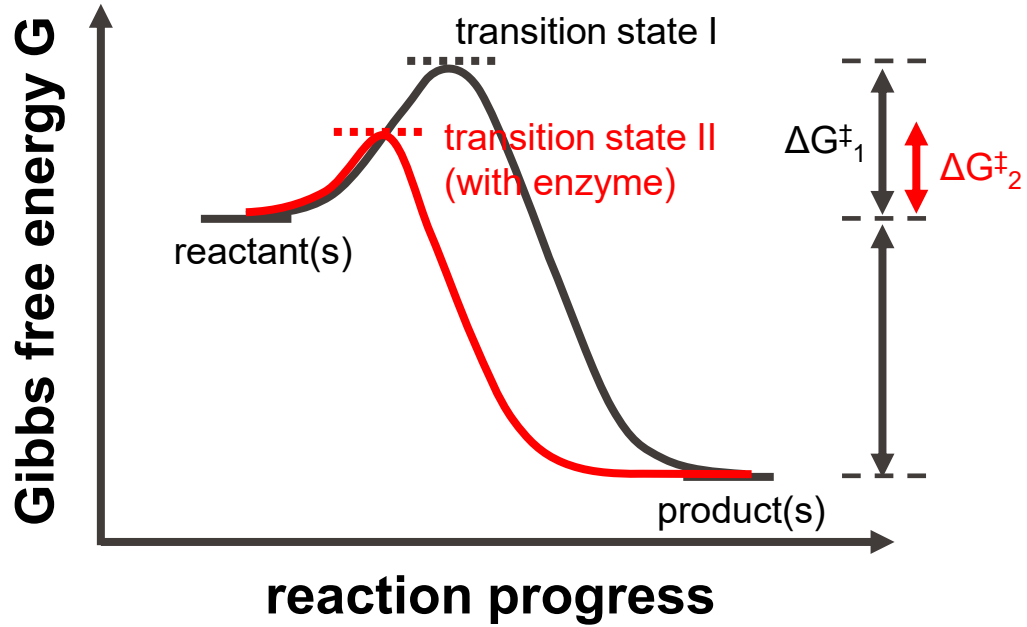


EPFL Energy Diagram of a Chemical Reaction



Transition state:
transient, short-lived
high energy state

EPFL How do Enzymes Work?



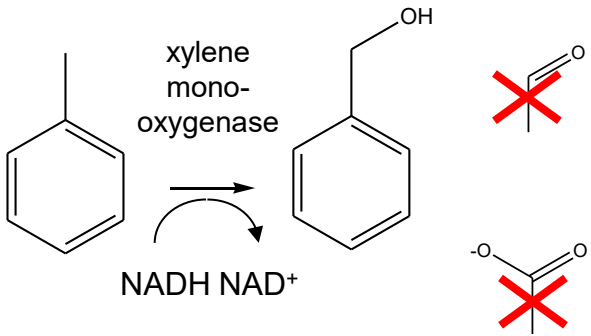
- Enzymes lower activation energies by stabilizing transition state structures (= different transition state)
- Faster reaction!
- Enzymes do not change the thermodynamic equilibrium!
- Dependence of reaction velocity (rate constant k) on $\Delta G^\ddagger \rightarrow$ Arrhenius equation

Arrhenius equation:

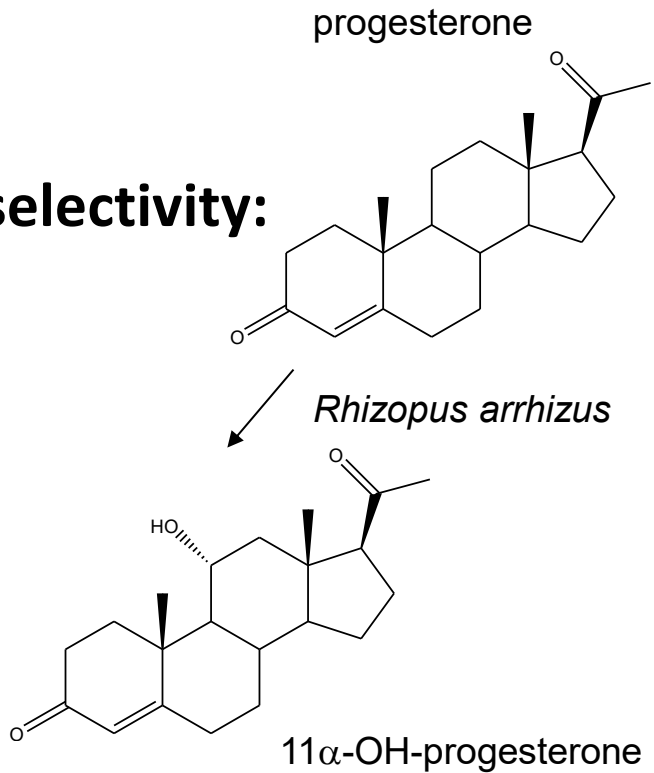
$$k = A \cdot e^{-\frac{\Delta G^\ddagger}{R \cdot T}}$$

A: frequency factor [1/s]
 ΔG^\ddagger : activation energy [J/mol]
R: gas constant, 8.314 [J/(mol*K)]
T: absolute temperature [K]

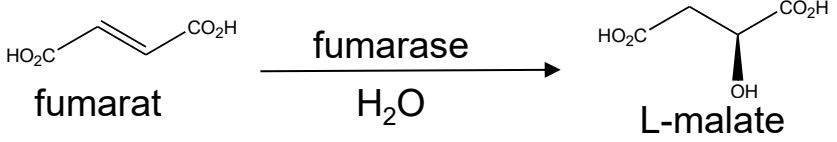
chemoselectivity:



regioselectivity:



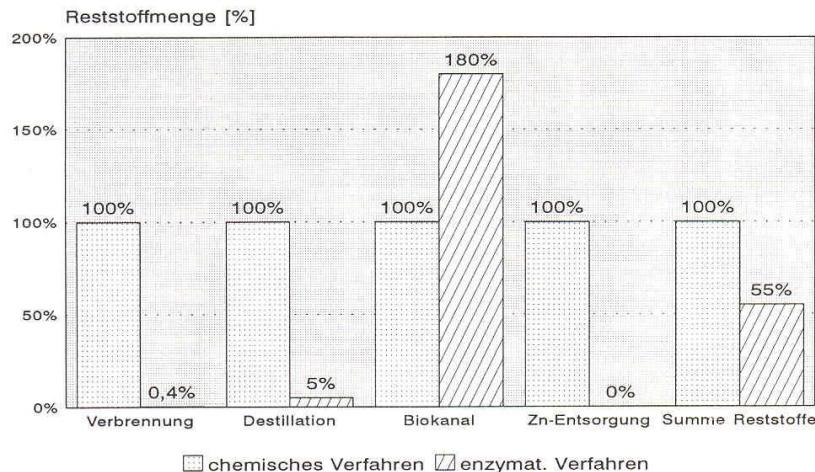
enantioselectivity:



E-factor = total waste / product

Table 1. E-Factors across the chemical industry

Industry sector	Annual production (t)	E-factor	Waste produced (t)
Oil refining	10^6 - 10^8	Ca. 0.1	10^5 - 10^7
Bulk chemicals	10^4 - 10^6	<1-5	10^4 - 5×10^6
Fine chemicals	10^2 - 10^4	5-50	5×10^2 - 5×10^5
Pharmaceuticals	10 - 10^3	25-100	2.5×10^2 - 10^5



Example: waste streams β -lactame production

Q: What are beta-lactams used for? How are they counteracted by e.g. bacteria?

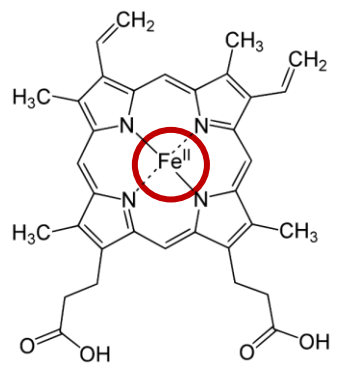
- Enzymes are excellent catalysts...
 - fast, efficient
 - selective (chemo-, regio-, stereo-)
 - work under “mild” reaction conditions (water, pH ~7, 10-40 °C, low salt etc.)
 - environmentally friendly (mostly)
 - can be easily combined! (cascade reactions)

- ...but...
 - can be expensive
 - cofactor dependence
 - reduced activity in organic solvents
 - inhibition
 - stability issues

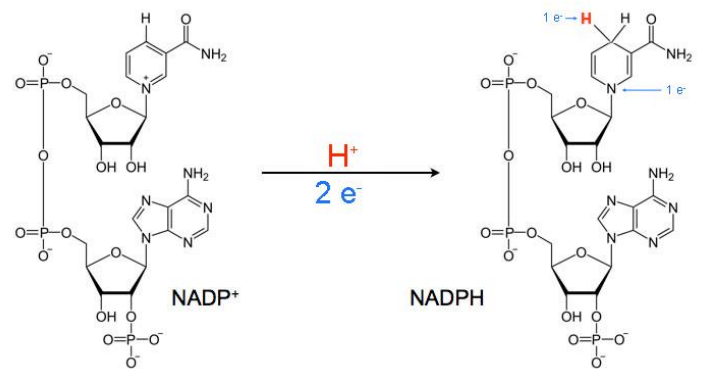
- **EC 1: Oxidoreductases** → redox reactions
- **EC 2: Transferases** → group transfer
- **EC 3: Hydrolases** → hydrolysis with H₂O
- **EC 4: Lyases** → cleavage (or joining) of molecules (non-hydrolytic!) forming double bonds or ring structures
- **EC 5: Isomerases** → isomerization (racemization, epimerization, rearrangement); mixtures (equilibrium!)
- **EC 6: Ligases** → joining of molecules at expense of ATP hydrolysis
- **EC 7: Translocases** → translocating molecules or ions across membranes (new class, since 2018)

- Small, non-protein molecules essential for enzymatic activity
- Holoenzyme = apoenzyme (protein part) + cofactor
- Types:
 - **coenzymes** (“cosubstrate”): organic molecules, non-covalent, dissociate after catalysis, donor/acceptors of groups/e⁻/H⁺ etc. (e.g. NAD(P)H, ATP, coenzyme A etc.)
 - **prosthetic groups**: organic molecules, covalently or non-covalently bound to protein, no dissociation (e.g. flavins, heme porphyrin, biotin etc.)
 - **metal ions**: in metalloenzymes (e.g. Fe → oxygenases, Ni → urease etc.); ~30% of enzymes (Mg, Zn, Fe, Cu, Co, Ni, Mn, Mo, Se, V etc.)

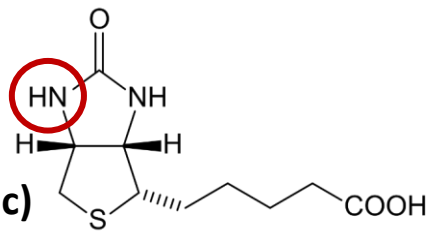
EPFL Cofactors – Examples



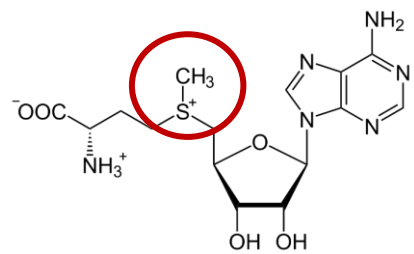
heme / iron porphyrins (prosthetic)
 → redox reactions, oxygen transfer



NAD(P)⁺ / NAD(P)H (coenzyme)
 → redox reactions, transfer of “H”



D-biotin (prosthetic)
 → carboxylations

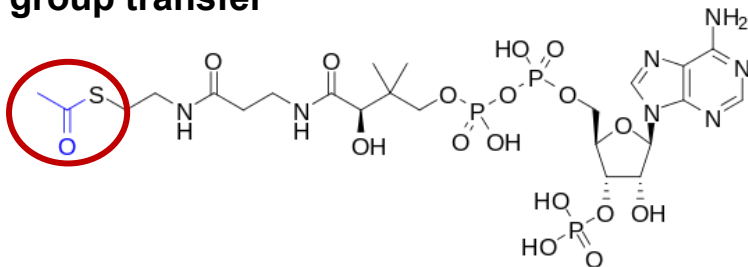


S-adenosyl methionine (SAM, coenzyme)
 → methyl (C₁) group transfer

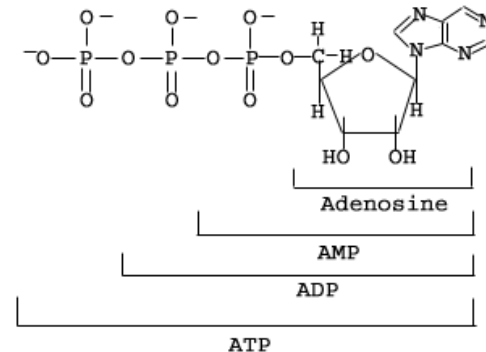
Q: In which enzyme classes do you expect these cofactors?

EPFL Cofactors – Examples

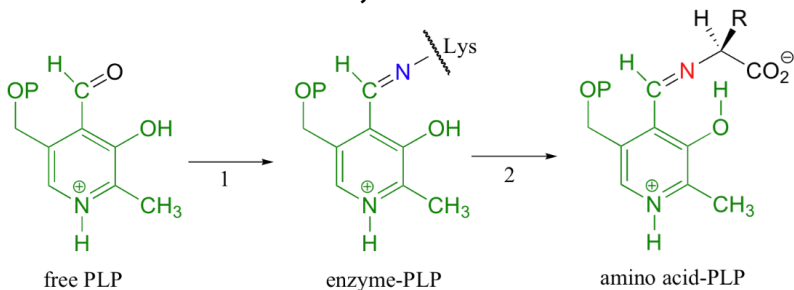
coenzyme A (CoA-SH) / acetyl-CoA
 → acyl group transfer



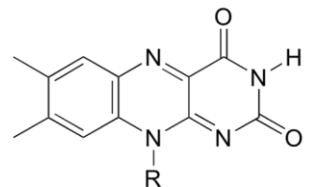
ATP / ADP / AMP (coenzyme)
 → phosphate transfer, “energy”



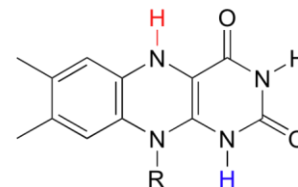
pyridoxal phosphate (PLP, coenzyme)
 → transamination, deamination etc.



riboflavins (prosthetic)
 → redox reactions, single electron transfer



FAD or FMN (oxidized flavin)



FADH₂ or FMNH₂ (reduced flavin)

Q: In which enzyme classes do you expect these cofactors?

- <https://www.brenda-enzymes.org/>
- Enzyme database
- Useful search tool:
 - substrates, products
 - main and side reactions
 - organisms
 - literature references
 - etc.



Please enter a search term

Enzyme, Ligand contains

add search field delete search field start search

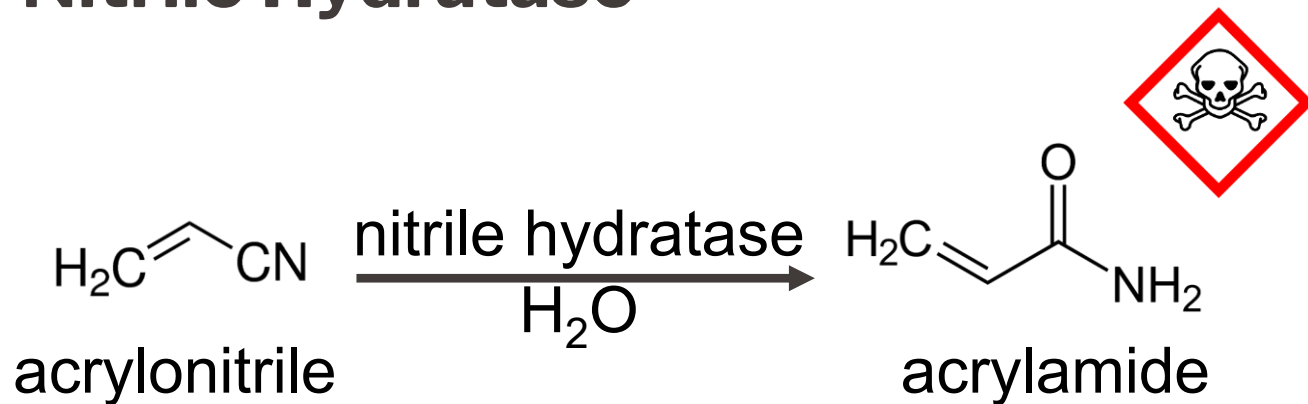
<p>Text-based queries</p> <ul style="list-style-type: none"> ▪ Full-text Search ▪ Advanced Search ▪ Enzyme & Disease 	<p>Structure-based queries</p> <ul style="list-style-type: none"> ▪ Ligand Substructure ▪ Metabolic Pathways ▪ Enzyme Structures 	<p>Explorer</p> <ul style="list-style-type: none"> ▪ Enzyme Classification ▪ TaxTree ▪ Protein folding: CATH / SCOPe ▪ Ontologies
<p>Visualization</p> <ul style="list-style-type: none"> ▪ Word Maps ▪ Genomes ▪ Functional Parameter Statistics ▪ Metabolic Pathways 	<p>Prediction</p> <ul style="list-style-type: none"> ▪ Membrane Helices ▪ EnzymeDetector 	<p>Supporting</p> <ul style="list-style-type: none"> ▪ BRENDA Tissue Ontology ▪ Biochemical Reactions

- as catalysts for synthetic chemical reactions (= biocatalysis)
- in detergents
- for bleaching
- food/feed processing or as additives
- pharmaceuticals
- etc.

→ enzymes appear both in industrial processes and in/as products!

Q: Which enzyme class is the most important one in industry?

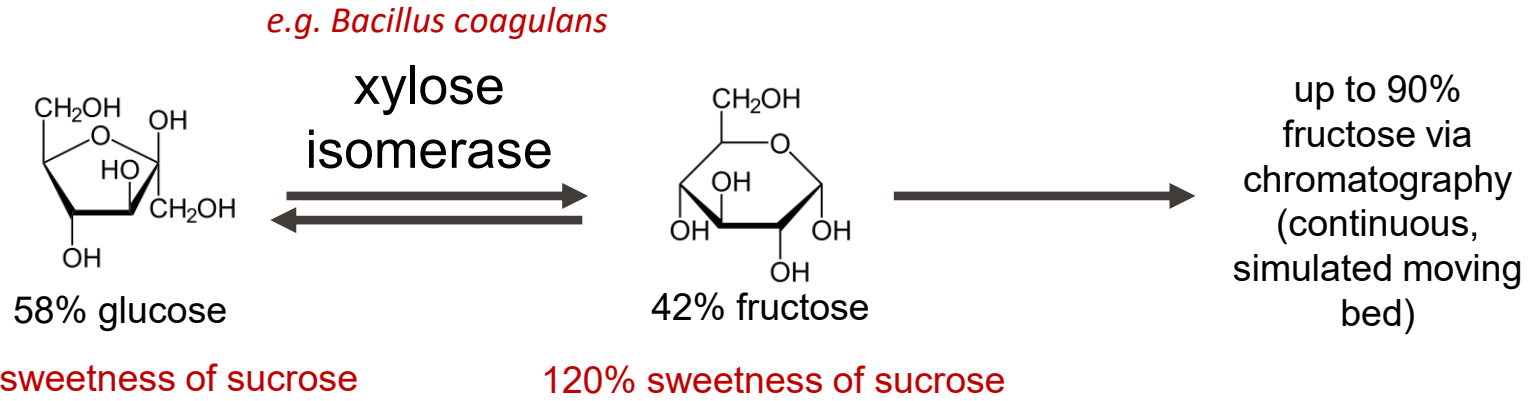
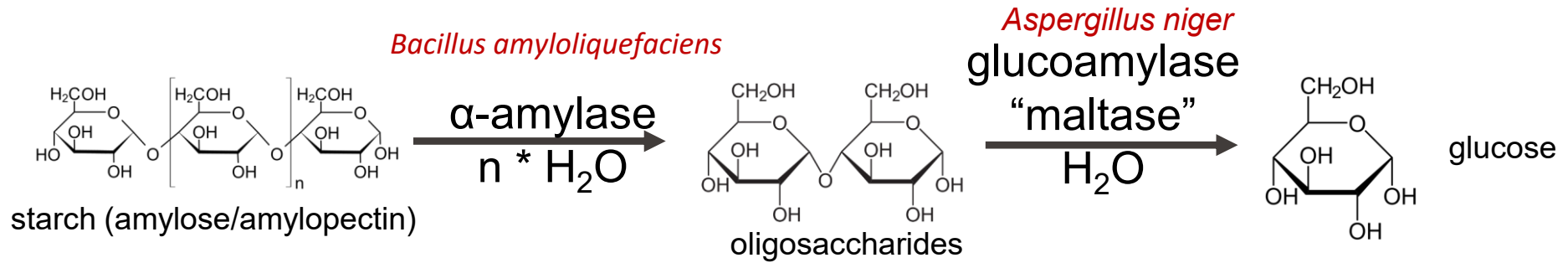
- β -lactame antibiotics (e.g. **penicillin acyclase**)
- acryl amide production with **nitrile hydratase**
- high-fructose corn syrup (**amylases, xylose isomerase**)
- **cellulases, laccases** for denim treatment, bleaching
- enzymes in detergents (**lipases, proteases, amylases, cellulases etc.**)
- **phytase** as feed additive
- enzymes as pharmaceuticals (**thrombolytic enzymes, uricase etc.**)
- etc. etc.



- polymer synthesis, flocculation agent (water treatment), dyes, glue, solvent etc.
- immobilized cells (*Rhodococcus* sp.), $T = 5\text{ }^\circ\text{C}$, 100% yield, > 30,000 t annually
- Nitto Chemical Industry (Japan)
- Chemical process: copper catalyst, side products (acrylic acid etc.), polymerization

Q: Which enzyme class does nitrile hydratase belong to?

High-fructose corn syrup (HFCS)



Q: Which enzyme classes are involved here?

novozymes



- laundry, dish washing
- hydrolytic enzyme cocktail:
 - lipases: oil and fat stains
 - amylases: starch remainders
 - proteases: e.g. egg, blood, grass
 - cellulases: smoothening of surface, stain removal
 - etc.
- washing at lower temperatures possible
- enzymes must be stable against heat, surfactants, bleach etc.!
→ thermophilic enzymes, enzyme engineering

