



CHEMICAL BIOLOGY

- Moodle: <https://go.epfl.ch/CH-313>
 - Lecture slides (evening before the lecture)
 - Distributed presentation topics (assignments)
 - Forum (for questions and announcements)
- Examination (written, graded, detailed information will follow)
- Contact:
 - Moodle forum (for questions)
 - markus.jeschek@epfl.ch
- **“Concepts over details!”**
- **Interact! Ask! Discuss! Anytime!**

Group Presentations

- Critical discussion of primary literature
- Illustrative examples for topics from the lecture

- Why?
 - Repetition of core concepts, techniques etc.
 - Presentation skills and critical discussion of research
 - Insight into current research topics

- How?
 - Two students per group
 - Assignments distributed one week before delivery of presentation (via Moodle)
 - **Send slides: markus.jeschek@epfl.ch (Mon evening before presentation)**
 - **15 min presentation (both group members should present!) + Q&A**

EPFL Tipps for Group Presentations

- Rough structure
 - Short intro on general topic
 - Main presentation according to assignment
 - Brief outlook incl. points of criticism/open questions/personal opinion as kick-starter for the discussion
- Everybody should participate in the discussion, incl. constructive(!) feedback on presentation style
- Questionnaires with different points, feedback by peers
- Typical assignment:
 - You will receive a certain topic including a related publication
 - Introduce the topic using the publication
 - present the motivation behind the research, methodology, key results (not every graph!)
 - Additional questions will be provided hinting towards central points
 - Be encouraged to look/present beyond the questions and the provided paper

Group Presentations – Schedule

#	Name1	Name2	Presentation on...	Assignment on...
1	Winger Quentin	Jeremy	Sep 23, 2025	Sep 16, 2025
2	Ema	Ariane	Sep 30, 2025	Sep 23, 2025
3	Benjamin	Matthieu	Oct 7, 2025	Sep 30, 2025
4	Ivana	Ipek	Oct 14, 2025	Oct 7, 2025
5	Mridhula	Elodie	Oct 28, 2025	Oct 21, 2025
6	Abigail	Robin	Nov 4, 2025	Oct 28, 2025
7	Eva	Florian	Nov 11, 2025	Nov 4, 2025
8	Bastien	Axel	Nov 18, 2025	Nov 11, 2025
9	Melodie	Siolène	Nov 25, 2025	Nov 18, 2025
10	Nicole	Maria	Dec 2, 2025	Nov 25, 2025

Course Topics – Overview

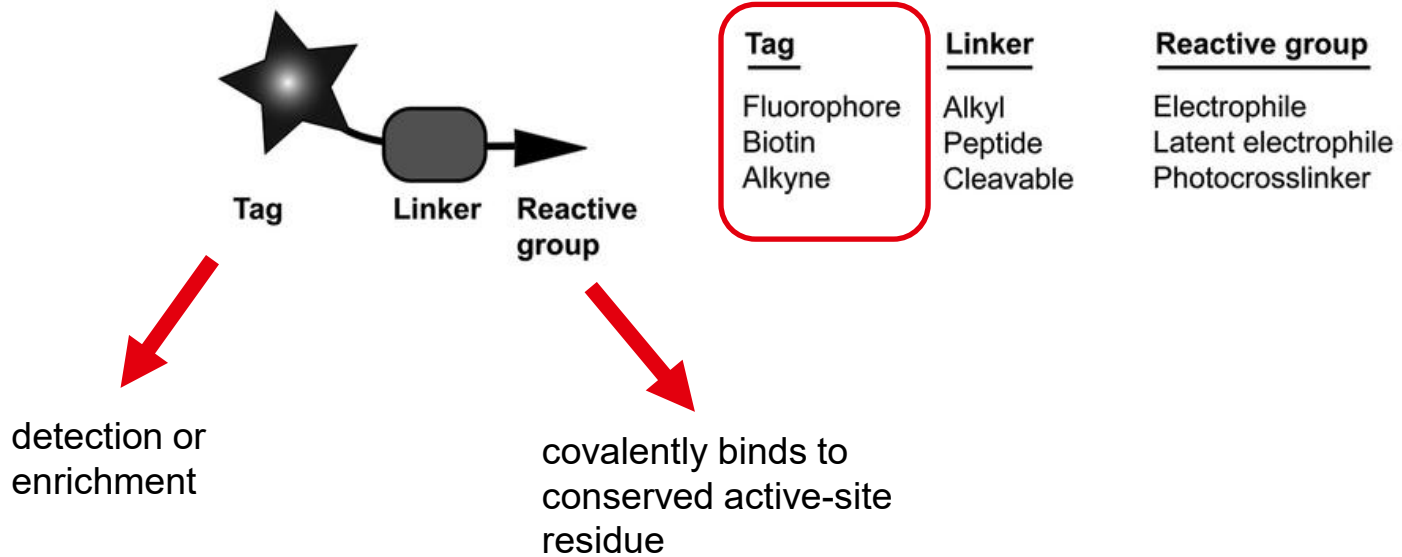
- Week 1 | Introduction + DNA
- Week 2 | DNA
- Week 3 | DNA
- Week 4 | DNA
- Week 5 | DNA/RNA
- Week 6 | RNA/Translation
- Week 7 | Translation
- Week 8 | Enzymes (Zoom)
- Week 9 | Enzymes (Zoom)
- Week 10 | Enzymes (Zoom)
- Week 11 | Enzymes (Zoom)
- Week 12 | Enzymes
- **Week 13 | Metabolism**
- Week 14 | LSAM Intro + Exam Preparation + Evaluation

[tentative schedule]

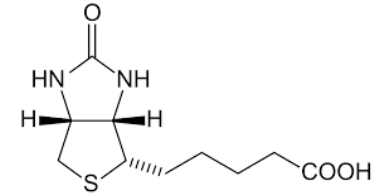
Protein – Enzymes (Kinetics)

EPFL Activity-Based Protein Profiling (ABPP)

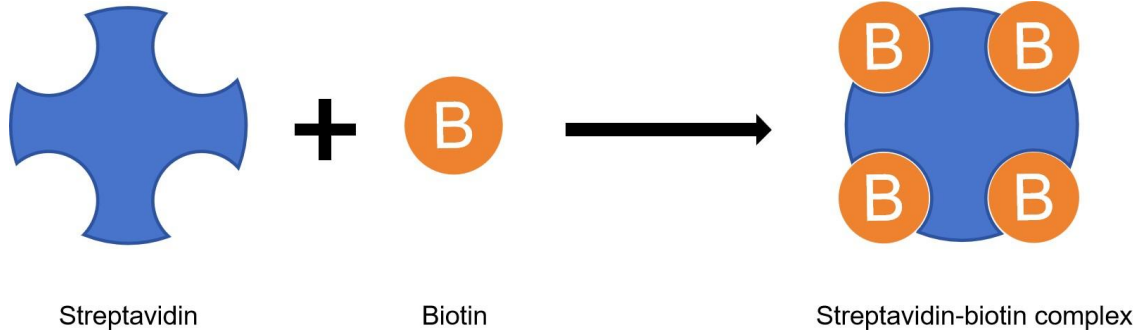
- Methods to identify drug targets, druggable sites etc.
- Testing for functional (catalytic) state of enzymes in complex proteomic mixtures
- Via activity-based probes (ABPs) that modify the active site



- Streptavidin: bacterial protein (4-mer) with high, non-covalent affinity to biotin
- Avidin = eukaryotic homologue
- $K_D \sim 10^{-15}$ M
- Extremely stable (T, solvents, pH etc.)
- Detection, labelling, purification, conjugation etc. etc.



D-biotin
("vitamin H")



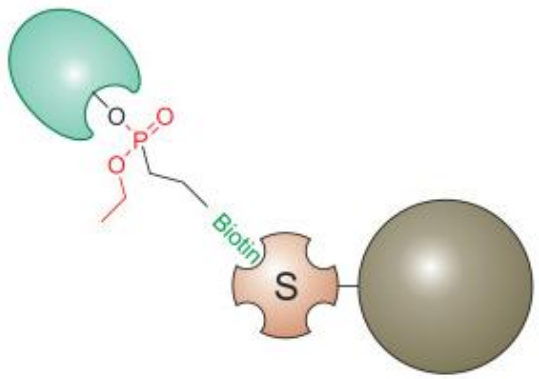
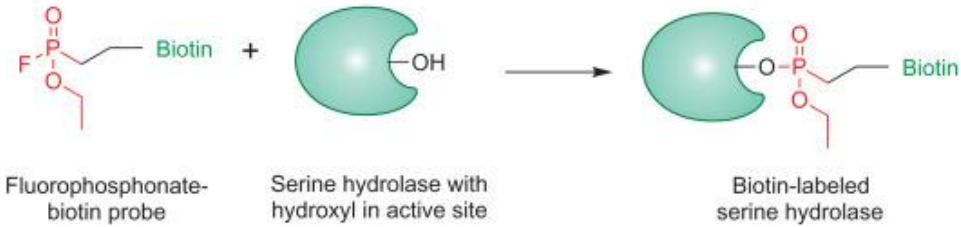
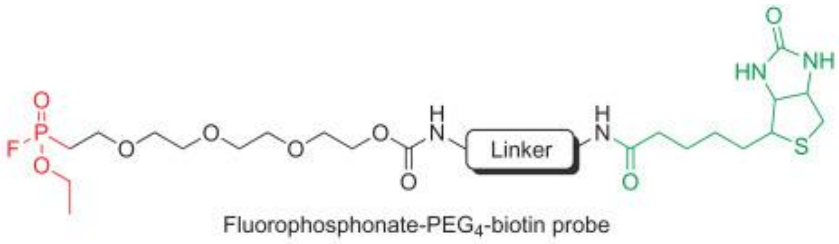
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e.g.:

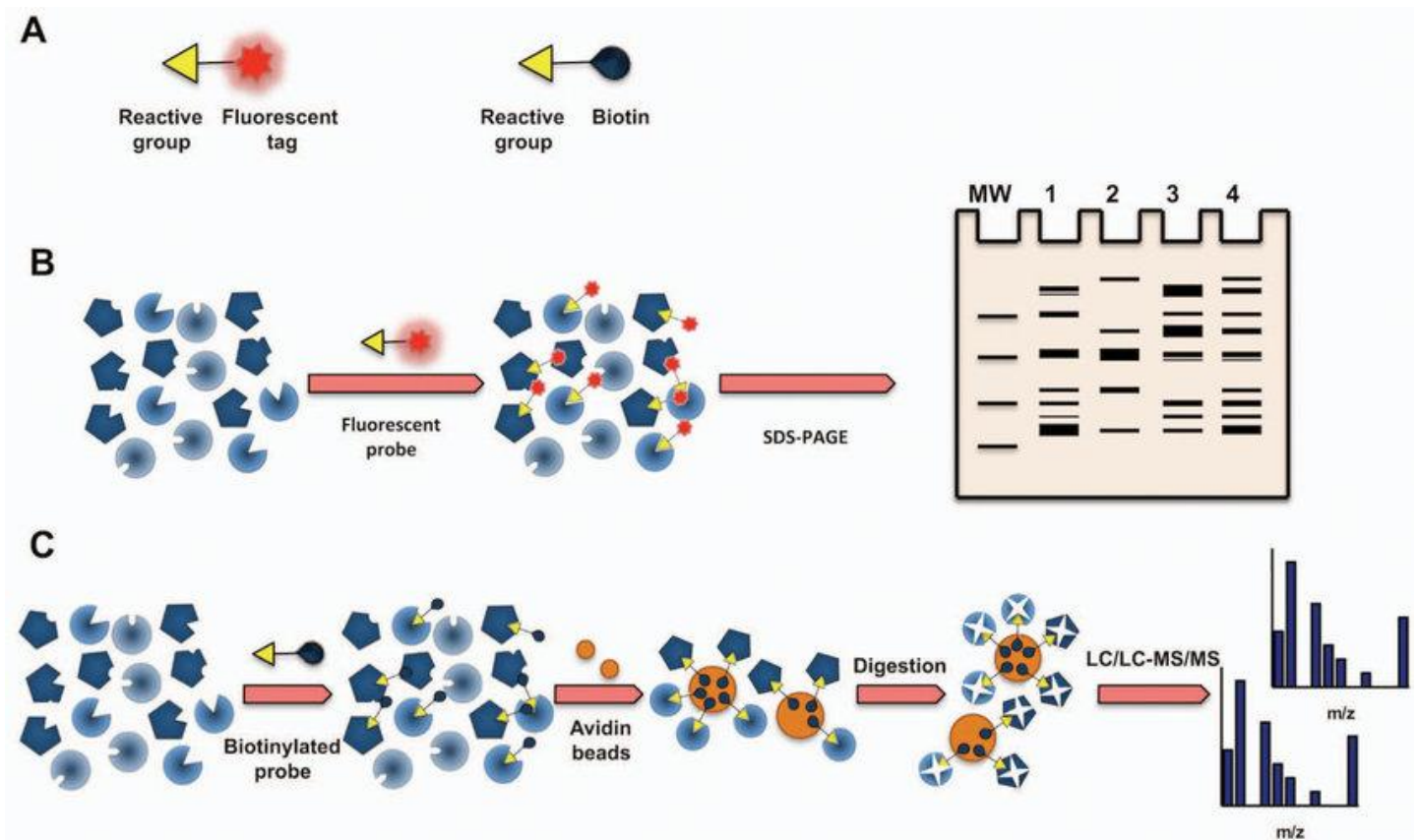
- fluorophosphonates → Ser hydrolases
- epoxides/vinyl sulfones → Cys proteases
- photoreactive groups

EPFL Fluorophosphonate ABPs

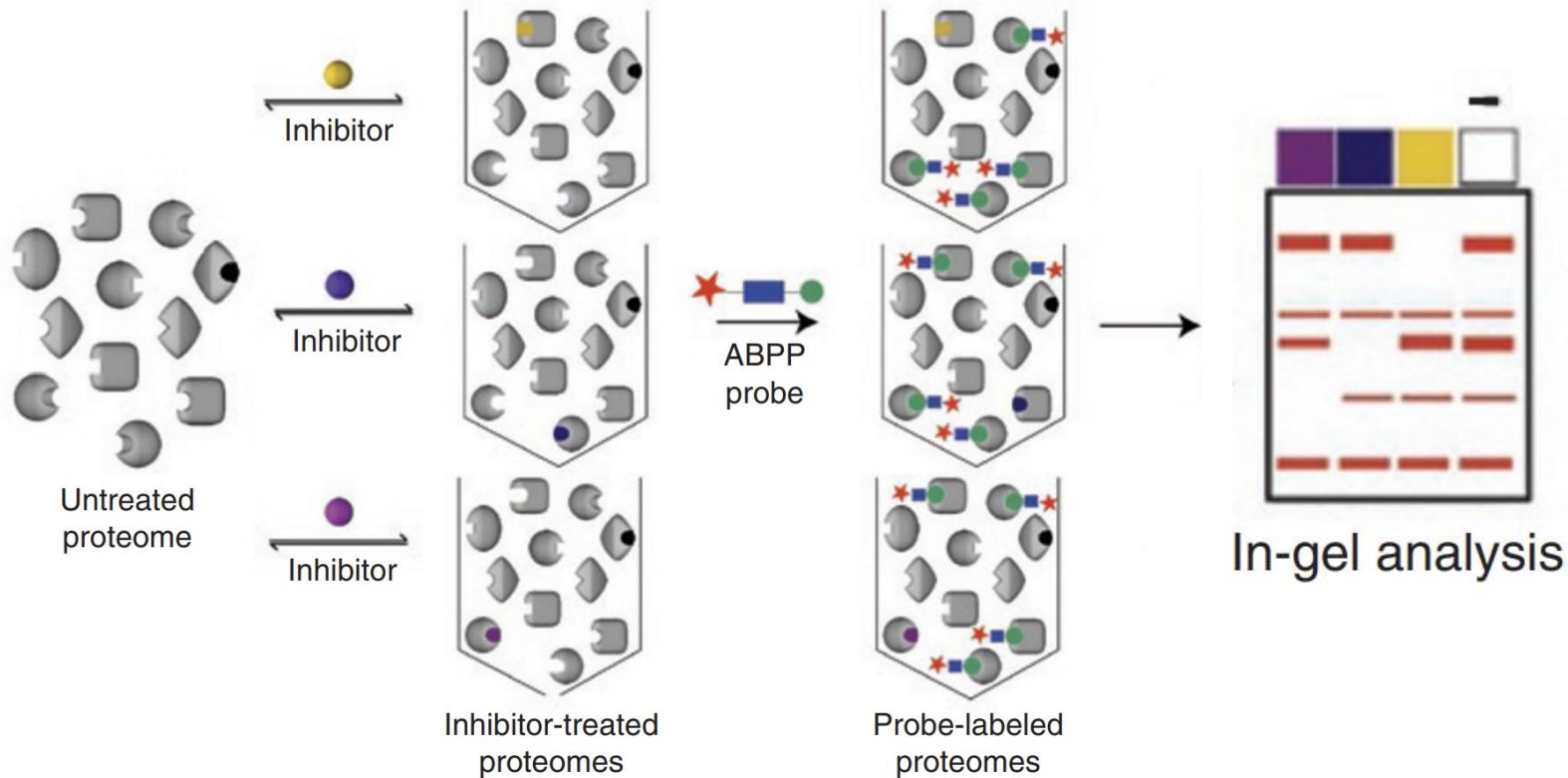


Isolation of serine hydrolases on immobilized streptavidin

EPFL ABPP – Direct Target Capture



EPFL Competitive ABPP



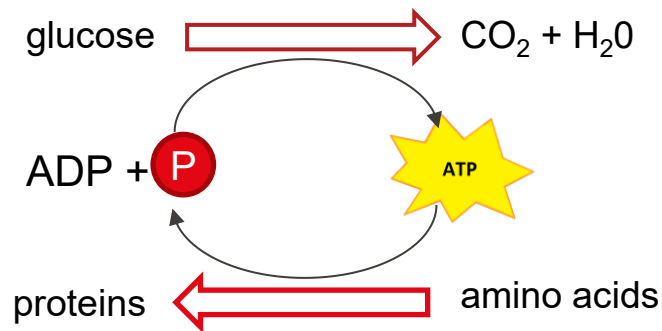
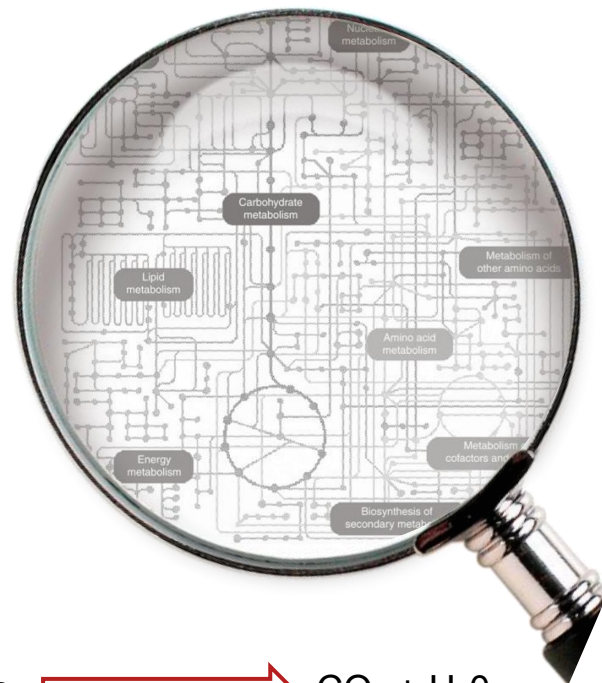
Q: How would this look like for an enrichment-based probe (e.g. biotinylated)?

Q: What are advantages of the competitive format (over direct capture)?

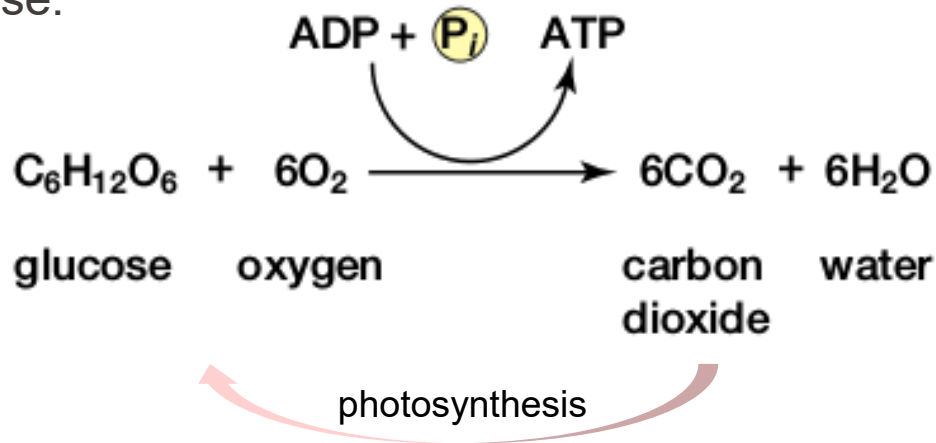
Metabolism

EPFL Metabolism - Overview

- “Sum of all (bio-)chemical reactions in a living organism.”
- E.g. *E. coli*:
 - 600-800 enzymes
 - approx. 1,000 metabolites
- Two main types:
 - Catabolism: Breakdown of larger molecules to “generate” chemical energy (ATP)
 - Anabolism: Building of larger molecules at the expense of chemical energy (ATP)



- = aerobic catabolism of sugars, amino acids, lipids etc.
- For glucose:



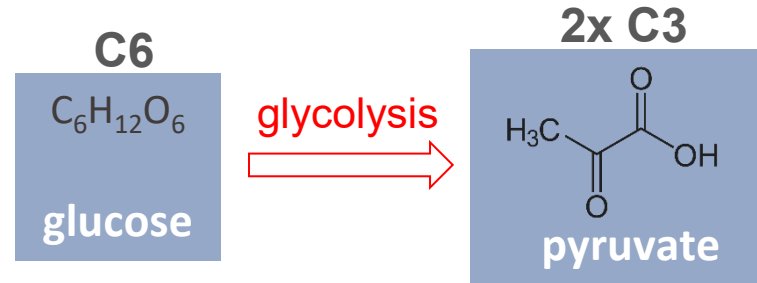
- Aerobic respiration: using molecular oxygen as terminal electron acceptor
- Complete oxidation of glucose: approx. 686 kcal/mol
- Energy stored in ATP: approx. 7 kcal/mol

Q: How much ATP is generated per molecule glucose through respiration?

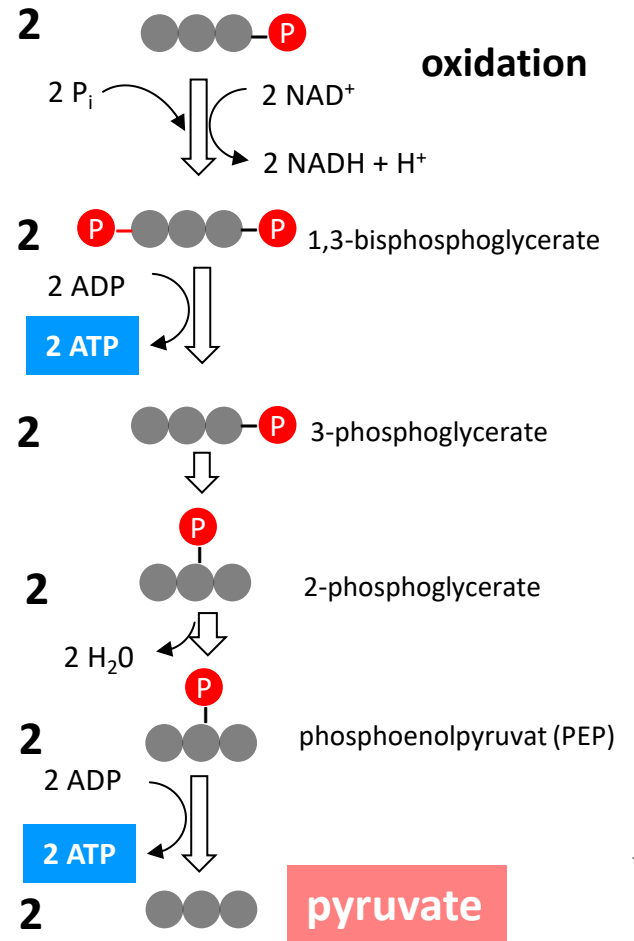
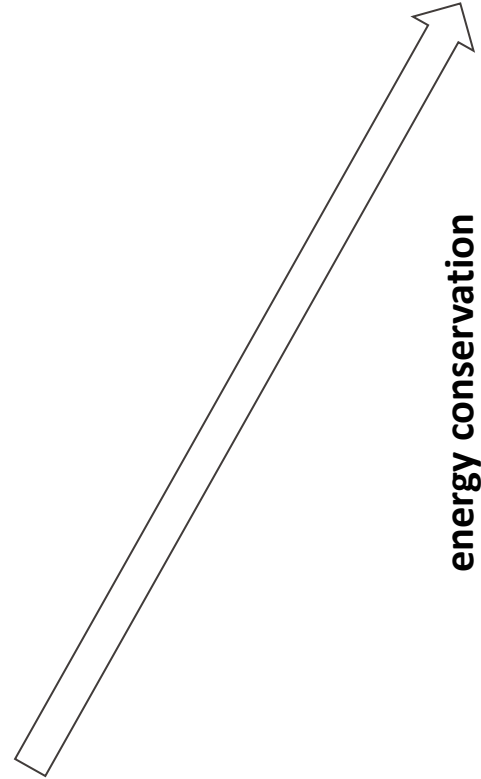
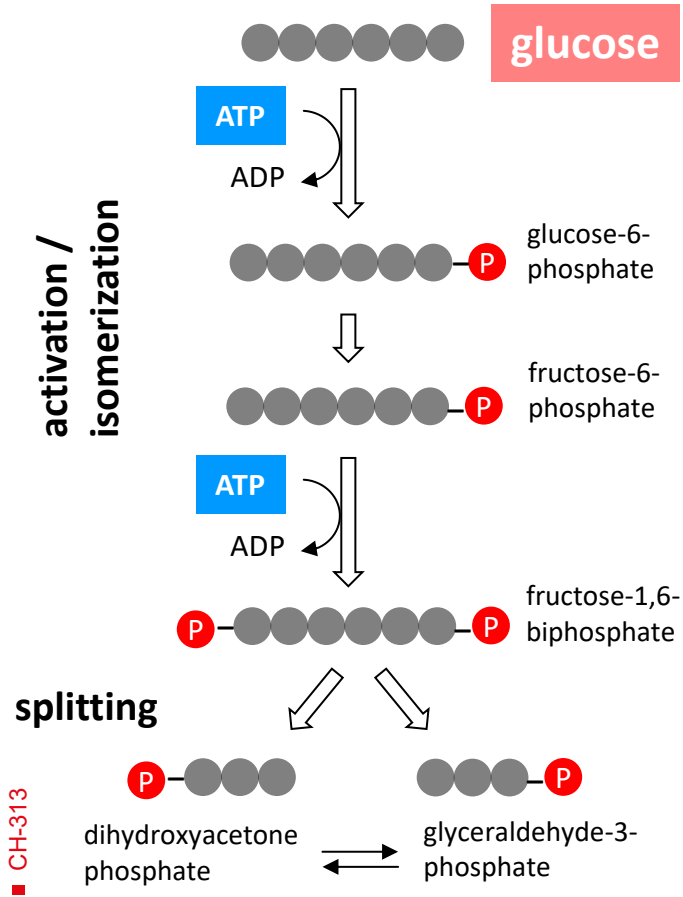
- Net reaction:



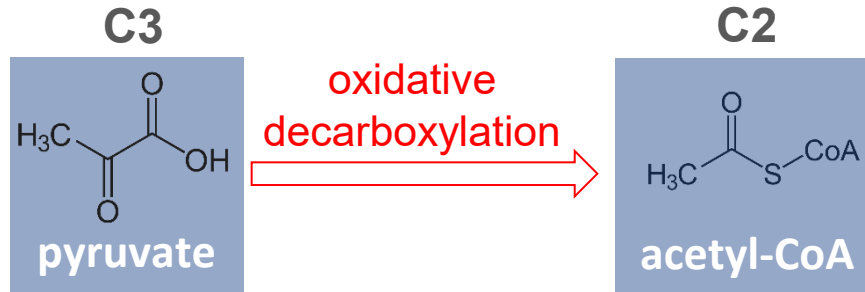
- 10 enzymatic steps in the cytosol
- no oxygen required
- product: pyruvate
 - further processed of respiration
 - alternative: fermentation (anaerobic catabolism)
 - important anabolic precursor (e.g. amino acids etc.)



Glycolysis – Conceptual Steps

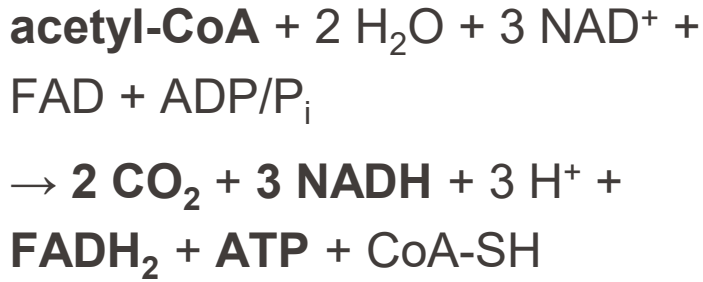


- Net reaction (2x per glucose molecule):

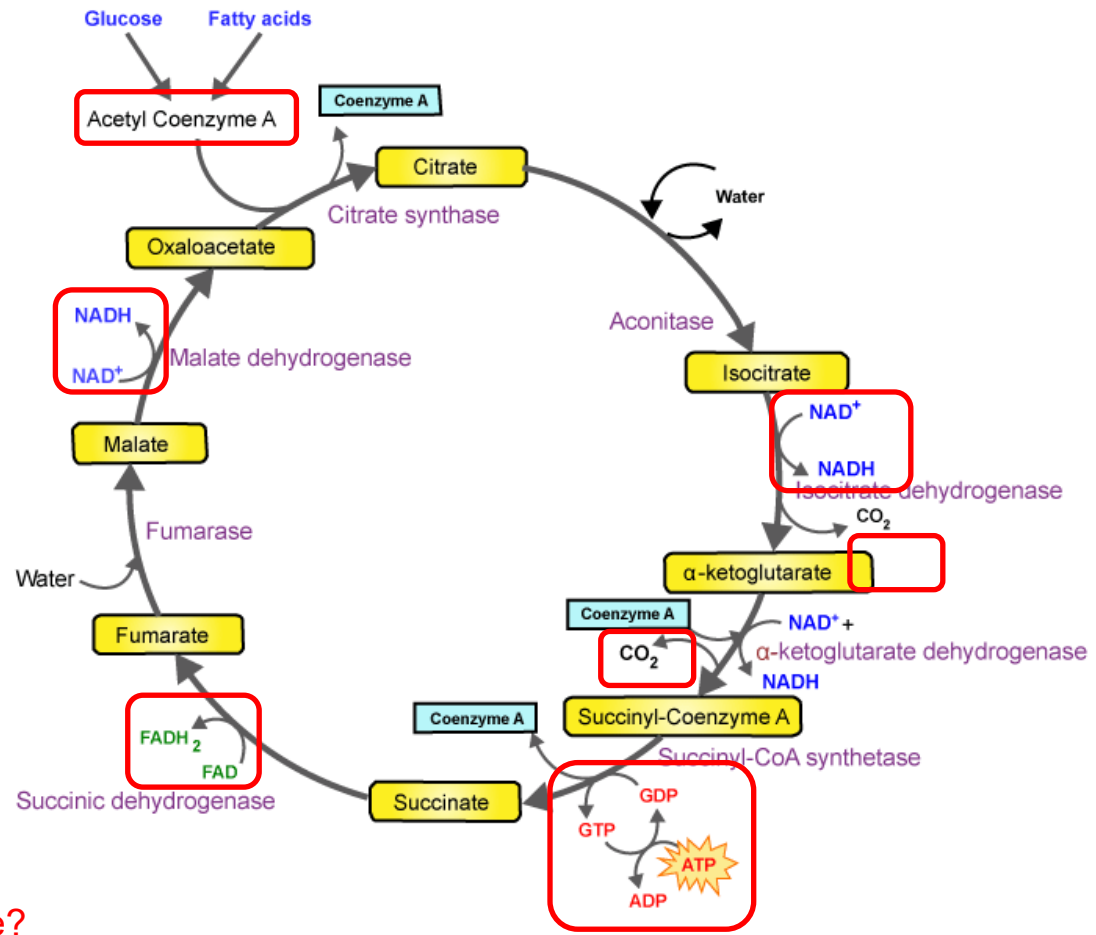


EPFL Citric Acid/Krebs/TCA Cycle

- Net reaction (2x per glucose molecule):

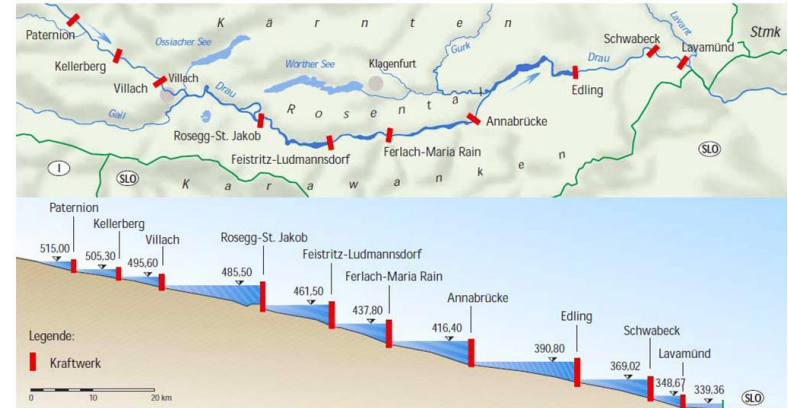


- Delivers various precursors for anabolism (amino acids, fatty acids, purins, pyrimidins etc.)

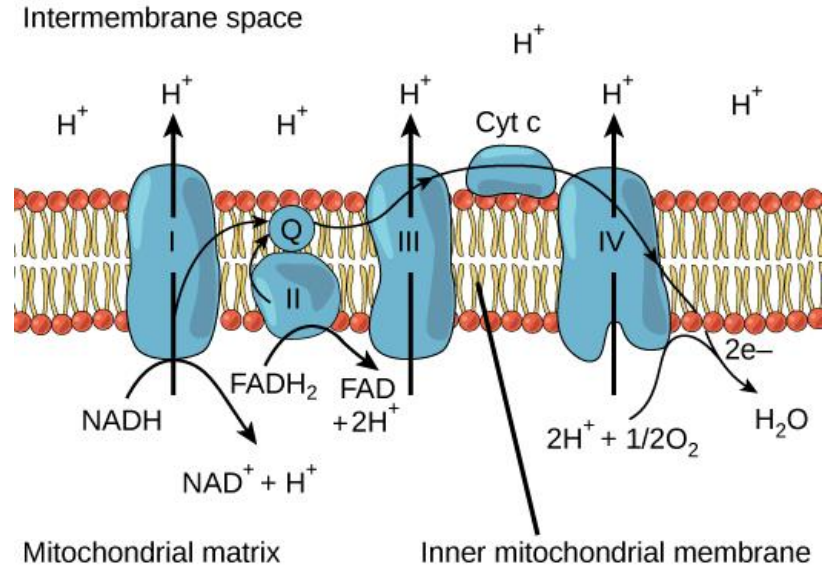


Q: Where does the TCA cycle take place?

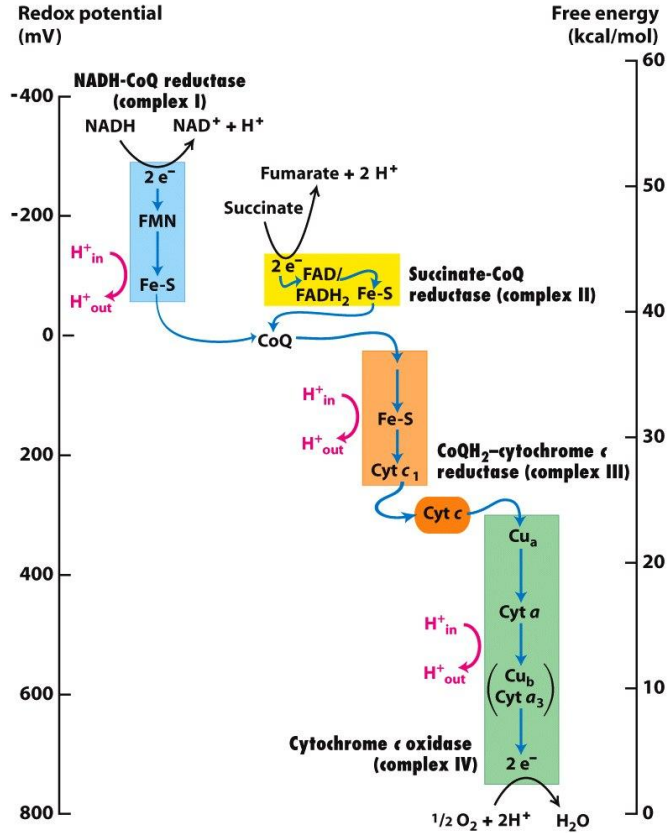
- Energy and redox balance after TCA cycle (per glucose molecule)
 - 4 ATP (2x glycolysis, 2x TCA)
 - **10 NADH** (2x glycolysis, 2x oxid. decarb., 6x TCA)
 - **2 FADH₂** (TCA)
- ETC: controlled, step-wise transfer of electrons to the terminal acceptor O₂
 - regeneration of NAD⁺/FAD⁺
 - establishment of trans-membrane proton gradient
(→ later: ATP generation)
 - ΔG (NADH→NAD⁺) ~ -218 kJ/mol



- Intricate, membrane-bound system of redox enzymes, cofactors and other electron carriers (quinones)
- Four complexes (I – IV)
 - I: NADH dehydrogenase:
 - II: succinate dehydrogenase
 - III: cytochrome c reductase
 - IV: cytochrome c oxidase
- Gradual export of protons
→ proton motive force (PMV) across the membrane



Electron Transport Chain (ETC)



- Complex I: 4 H⁺
 - Complex III: 4 H⁺
 - Complex IV: 2 H⁺
- 10 H⁺ per NADH
- 6 H⁺ per FADH₂

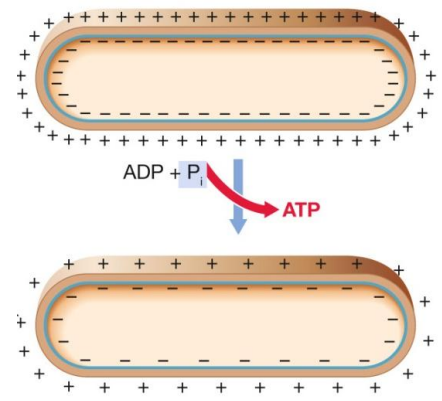
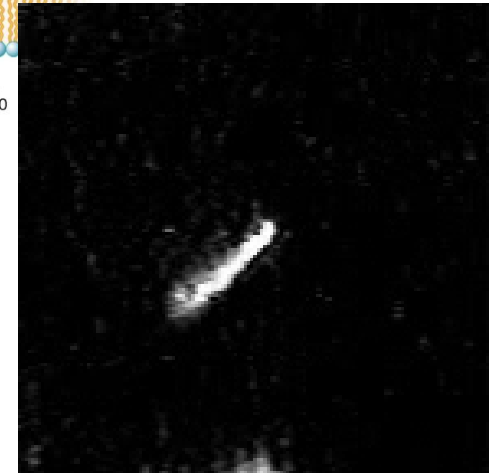
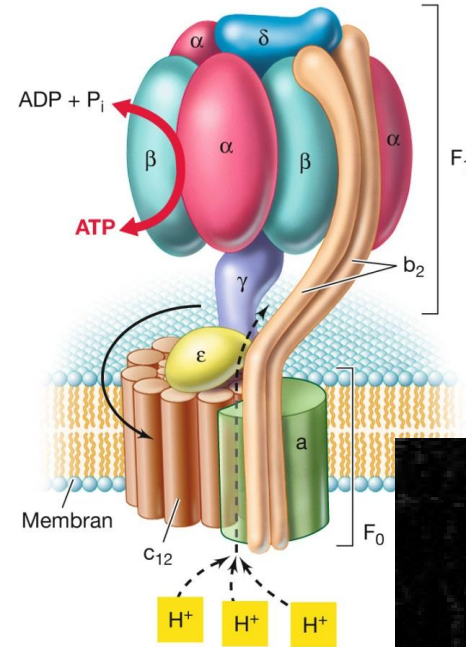


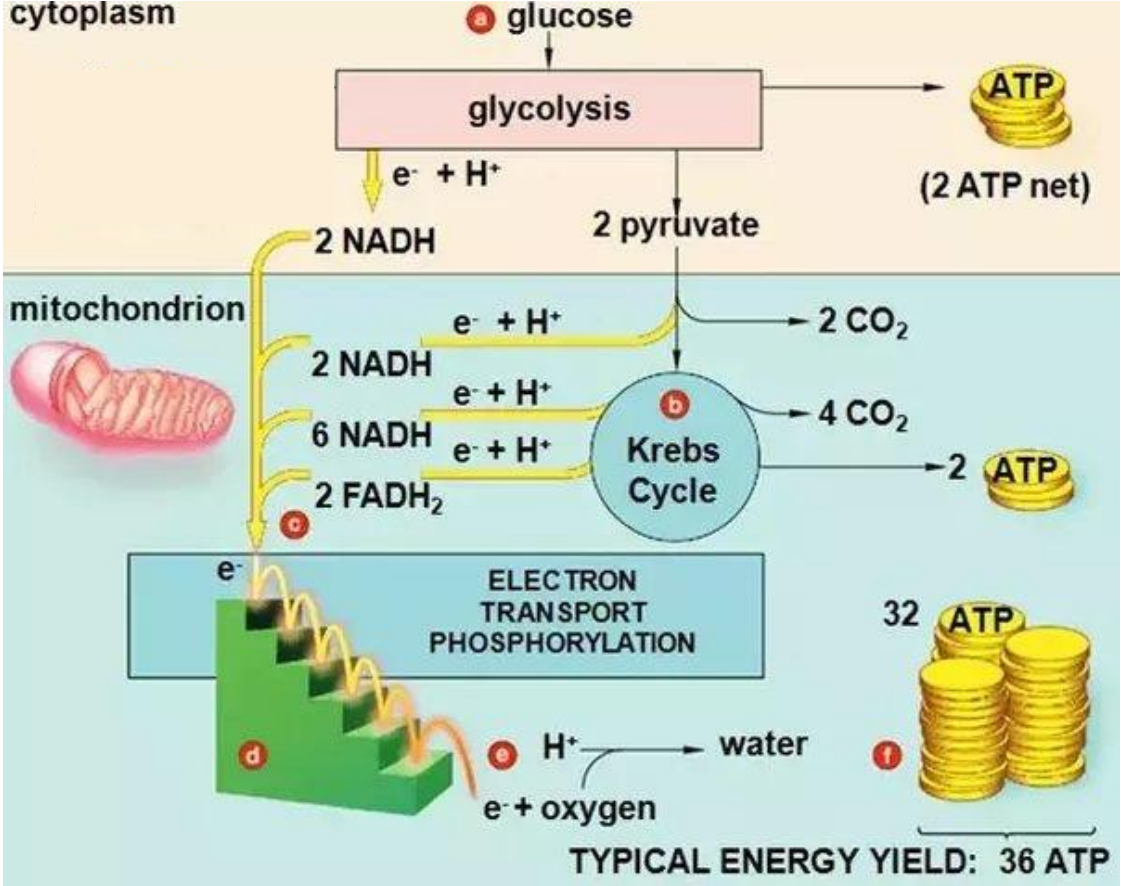
Figure 12-18
Molecular Cell Biology, Sixth Edition
 © 2008 W. H. Freeman and Company

- PMV is used to generate ATP
- ATP-synthase (“ATPase”) complex
 - F_0 : protons are channelled to the inside leading to rotation of c subunits
 - F_1 : rotation leads to conformational changes that lead to ATP formation
- 3-4 protons per ATP
- ATPase reaction is reversible!



Q: What could the reverse reaction of the ATPase be used for in nature?

EPFL Balance of Aerobic ATP Generation

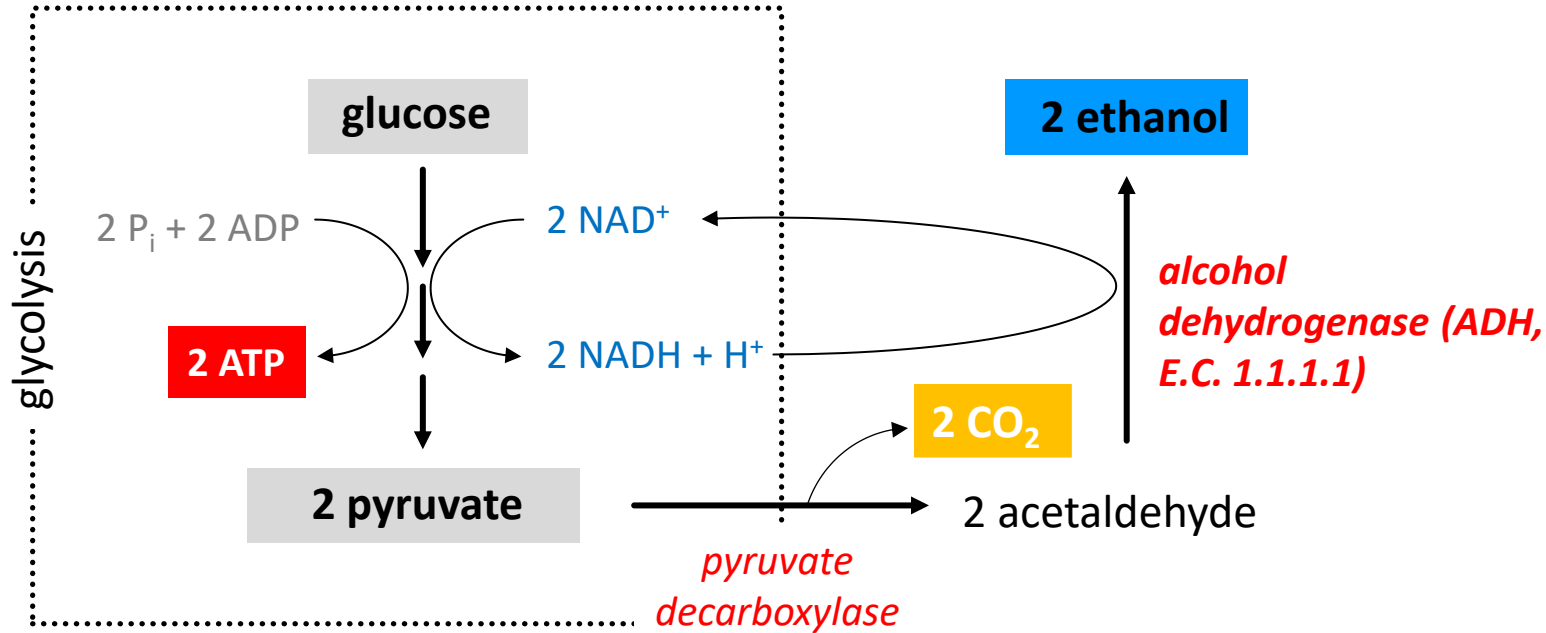


- Catabolism of organic substrates (e.g. sugars) without using oxygen
- Terminal electron acceptors are organic metabolites (substrate-derived)
- Goals:
 - Regeneration of NAD^+ !
 - ATP generation (in addition to glycolysis) → not all types of fermentation



Q: Which types of fermentations do you know? Which are associated with these pictures?

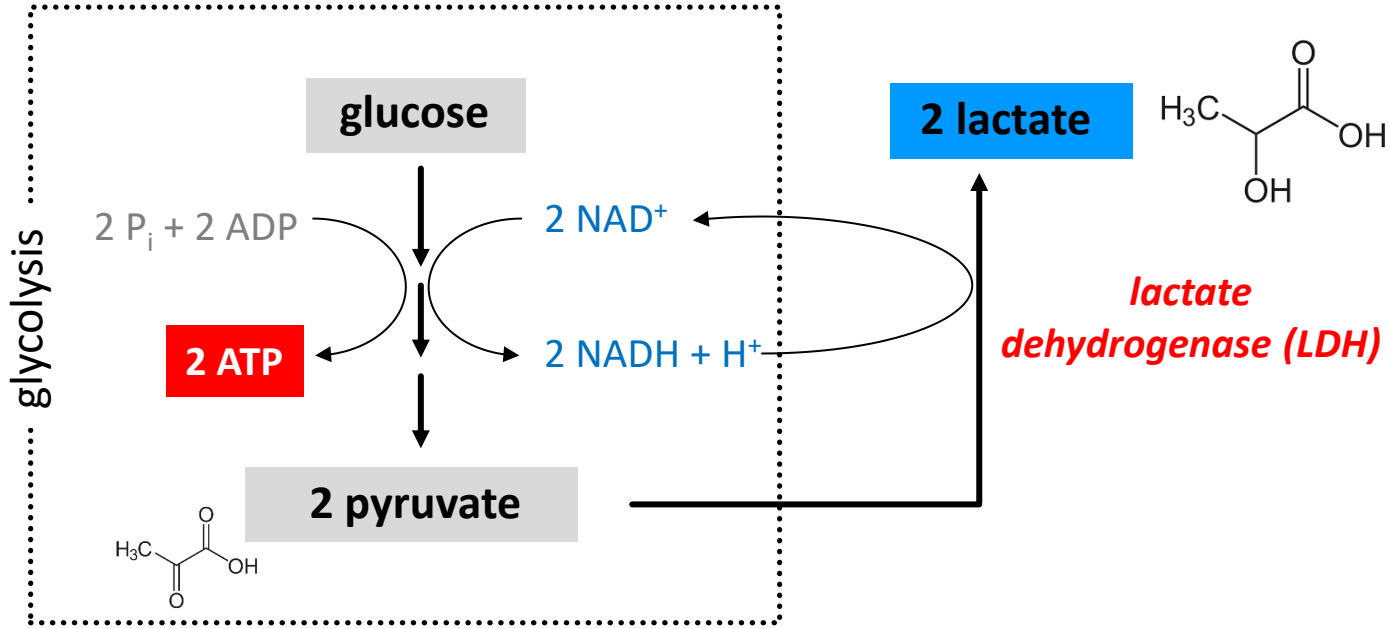
EPFL Alcoholic/Ethanol Fermentation



- different yeasts (facultative anaerobes), few bacteria
- no additional ATP production

Q: What are facultative anaerobes?

EPFL Lactic Acid Fermentation (Homolactic)



- different bacteria, human cells (e.g. muscles)
- no additional ATP production

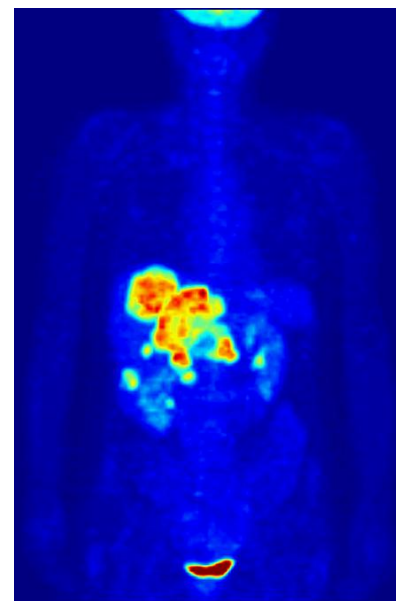
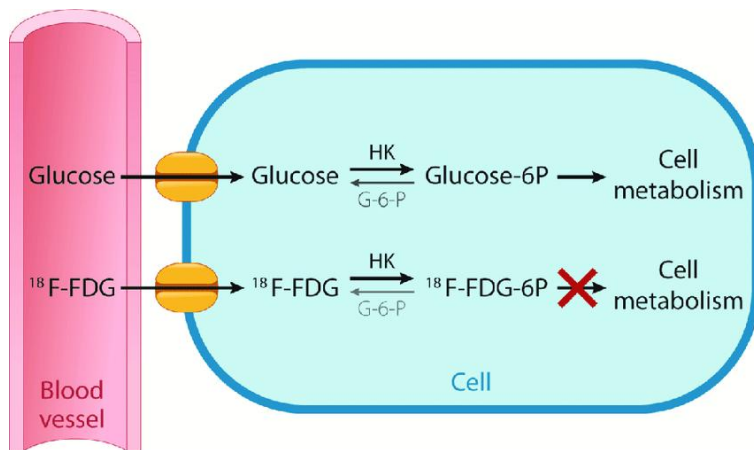
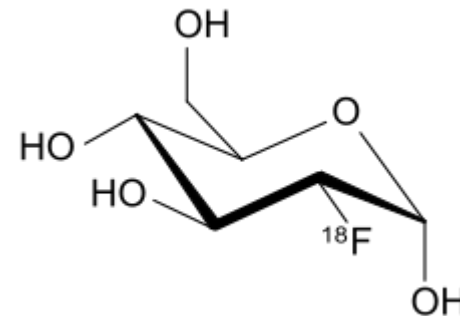
- Otto Heinrich Warburg, Nobel Prize 1931
- most cancers use “aerobic fermentation” and lactic acid fermentation for energy generation (not respiration via TCA!)
 - low ATP yield
 - no complete break down of metabolites to CO_2

→ increased precursor supply for anabolism/biomass formation

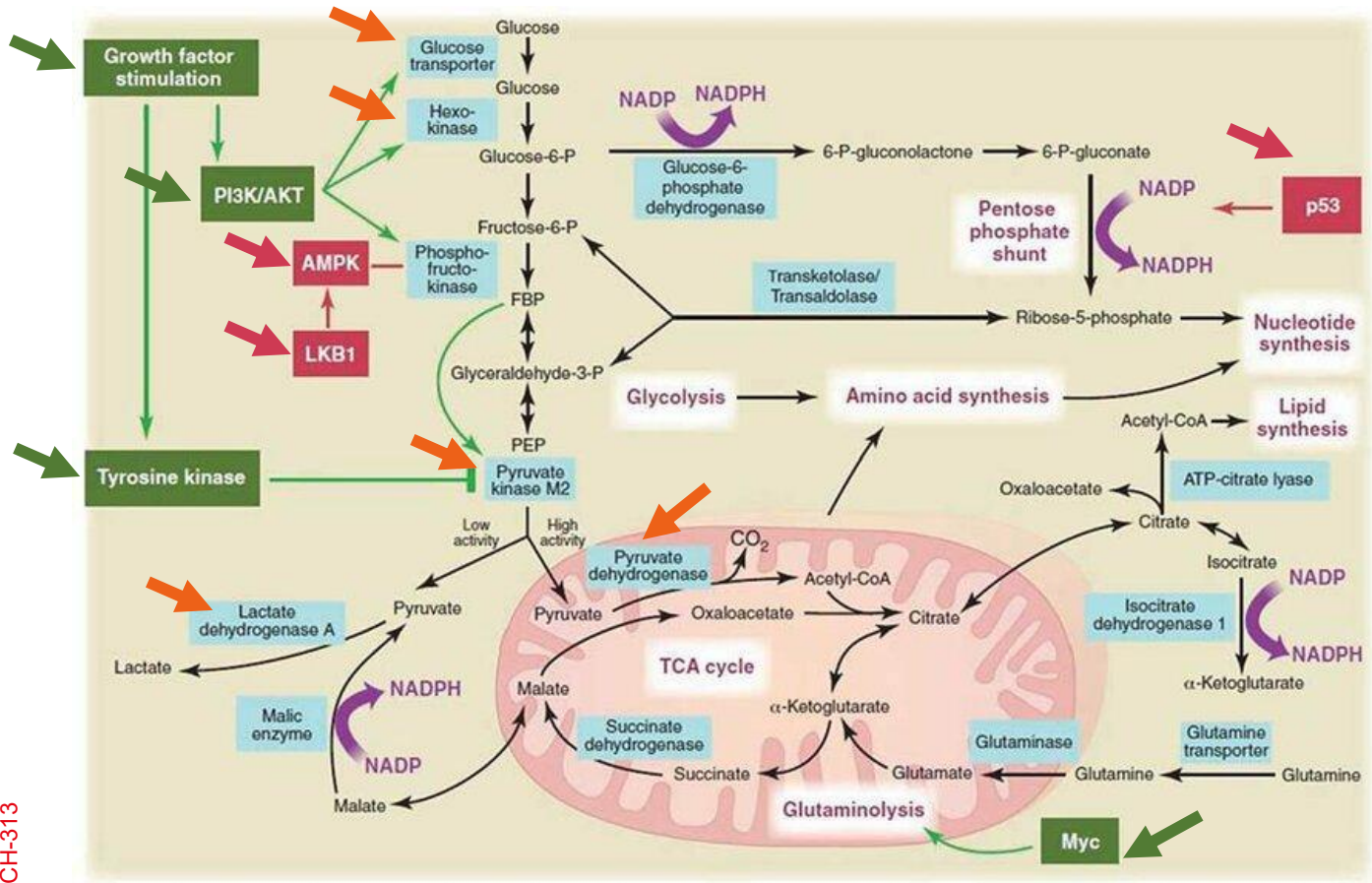
 - fast glucose consumption! (glycolysis rate is approx. 200-fold compared to “normal” cells)
- Related: so-called “Crabtree effect” yeast



- Fluorodeoxyglucose (^{18}F)
- Radiopharmaceutical for positron emission tomography
- Taken up by cells and phosphorylated by hexokinase
- Not further metabolized \rightarrow accumulation in glucose-consuming cells (e.g. tumors)



EPFL Cancer Metabolism and Therapy



- ➔ oncogenes
- ➔ tumor suppressor genes
- ➔ potential drug targets