

RECAP:

① Combined first and second laws of thermo:

$$\boxed{dU = TdS + \sum_i Y_i dx_i}$$

$Y_i \rightarrow$ intensive STATE variable

$x_i \rightarrow$ extensive STATE variable

$$\delta W_{\text{reversible}} = Y dx \quad \text{General expression for reversible Work.}$$

② Legendre transforms:

$$dU = TdS + \sum_i Y_i dx_i \Rightarrow U(S, X_1, X_2, \dots)$$

NATURAL VARIABLES OF U

if we control some intensive variables in experiment we need to perform Legendre transforms to compute the CHARACTERISTIC POTENTIAL under the experimental boundary conditions.

$$\underline{\Lambda} = U - \sum_i Y_i x_i$$

⚡
sum over the experimentally
controlled intensive
variables.

EXAMPLES:

* Constant V, T : $F = U - TS$;

$$dF = -SdT - pdV$$

$$\left(\frac{\partial F}{\partial T}\right)_V = -S \quad \left(\frac{\partial F}{\partial V}\right)_T = -p$$

(eg) IDEAL GAS: from statistical mechanics.

$$F(T, V) = -nRT \left[\log V - \frac{3}{2} \log T + \frac{3}{2} \log \left(\frac{m}{2\pi \hbar^2 k_B} \right) \right]$$

$$p = - \left(\frac{\partial F}{\partial V} \right)_T = \frac{nRT}{V}$$

$$S = - \left(\frac{\partial F}{\partial T} \right)_V = nR \left[\log V - \frac{3}{2} \log T + \frac{3}{2} \log \left(\frac{m}{2\pi \hbar^2 k_B} \right) \right] - \frac{3}{2} nR$$

$$U = pV + TS = \frac{3}{2} nRT = U(T)$$

Whenever an INTENSIVE variable is experimentally controlled, we have to perform a LEGENDRE TRANSFORM

In general: $\phi = U - Y_i X_i$ $\left\{ \begin{array}{l} X_i = \text{extensive} \\ Y_i = \text{intensive} \end{array} \right.$

↖
char. potential under the experimental conditions.

EXAMPLES:

* Constant T, V :



* start with fundamental equation:

$$dU = Tds - pdV \longrightarrow U(S, V)$$

need to replace this with conjugate T
good ✓

Legendre transform:

$$F = U - TS$$

Legendre transform.

$$dF = dU - Tds - SdT$$

$$= Tds - pdV - Tds - SdT$$

$$\boxed{dF = -SdT - pdV} \longrightarrow F(T, V)$$

$$dF = \left(\frac{\partial F}{\partial T} \right)_V dT + \left(\frac{\partial F}{\partial V} \right)_T dV \Rightarrow$$

$$\boxed{\begin{array}{l} \left(\frac{\partial F}{\partial T} \right)_V = -S \quad \left(\frac{\partial F}{\partial V} \right)_T = -p \\ \text{EQUATIONS OF STATE} \end{array}}$$

2. constant T, p



fundamental equation: $dU = TdS - pdV$

$U(S, V) \rightarrow$ replace both variables with their conjugates

$$G = U - TS + pV$$

$\hookrightarrow$ sign because $Y_i = -p$

$$dG = -SdT + Vdp$$

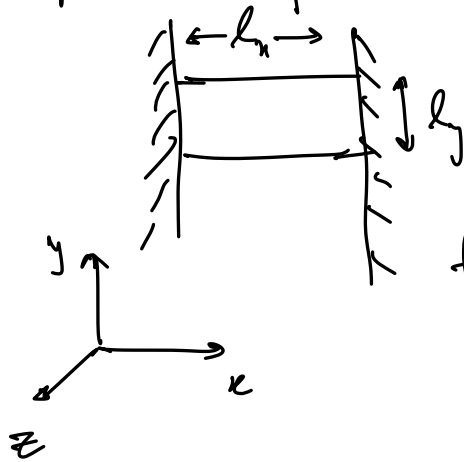
$G(T, p) \rightarrow$ GIBBS FREE ENERGY

$$dG = \left(\frac{\partial G}{\partial T} \right)_p dT + \left(\frac{\partial G}{\partial p} \right)_T dp$$

EQUATIONS OF STATE:

$$S = - \left(\frac{\partial G}{\partial T} \right)_p \quad V = \left(\frac{\partial G}{\partial p} \right)_T$$

4. Clamped block of material @ constant T, p



generalized stress/strain work

$$\sum_i \sum_j V_0 \sigma_{ij} d\epsilon_{ij}$$

for simplicity, assume no shear stress/strain

$$\delta W = V_0 (\sigma_{xx} d\epsilon_{xx} + \sigma_{yy} d\epsilon_{yy} + \sigma_{zz} d\epsilon_{zz})$$

fundamental equation:

$$dU = TdS + V\sigma_{xx}d\epsilon_{xx} + V\sigma_{zz}d\epsilon_{zz} + V\sigma_{yy}d\epsilon_{yy}$$

$$U(S, V\epsilon_{xx}, V\epsilon_{zz}, V\epsilon_{yy})$$

we control:

$$\left(T, \underbrace{V\epsilon_{xx}}_{\text{extensive}}, \underbrace{V\epsilon_{zz} + V\epsilon_{yy}}_{-p} \right)$$

$$\Delta = U - TS - V\sigma_{yy}\epsilon_{yy} - V\sigma_{zz}\epsilon_{zz} = U - TS + pV(\epsilon_{yy} + \epsilon_{zz})$$

$$d\Delta = -SdT + \sigma_{xx}Vd\epsilon_{xx} - V(\epsilon_{yy} + \epsilon_{zz})dp$$

EQUATIONS OF STATE:

$$S = - \left(\frac{\partial \Delta}{\partial T} \right)_{V\epsilon_{xx}, p}$$

$$\sigma_{xx} = \left(\frac{\partial \Delta}{\partial \epsilon_{xx}} \right)_{T, p} \frac{1}{V}$$

$$V(\epsilon_{xx} + \epsilon_{yy}) = - \left(\frac{\partial \Delta}{\partial p} \right)_{T, V\epsilon_{xx}}$$

SECOND DERIVATIVES OF CHARACTERISTIC POTENTIAL

— Related to "RESPONSE FUNCTION"

Response functions:

(eg) Control T, p experimentally

* fix p , change $T \rightarrow$ look at response of the system.

$$1. \delta Q \rightsquigarrow C_p = \left(\frac{\delta Q}{dT} \right)_p = T \left(\frac{\partial S}{\partial T} \right)_p \quad [\text{II Law}]$$

$$2. \Delta V \rightsquigarrow \beta = \frac{1}{V} \left(\frac{\partial V}{\partial T} \right)_p$$

* fix T , change $p \rightarrow$ look at response

$$1. \Delta V \rightsquigarrow \bar{K} = \frac{-1}{V} \left(\frac{\partial V}{\partial p} \right)_T$$

In general: response function $\rightsquigarrow \frac{\partial(\text{freely varying state variable})}{\partial(\text{expt. controlled state variable})}$

$$\text{Usually } \rightsquigarrow \frac{\partial(\text{extensive})}{\partial(\text{intensive})}$$

* Coming back to char. potentials:

(example) constant $T, p \rightsquigarrow$ char. potential $G(T, p)$

$$G = U - TS + pV$$

$$dG = -SdT + Vdp$$

* EOS:

$$S = -\left(\frac{\partial G}{\partial T} \right)_p \quad V = \left(\frac{\partial G}{\partial p} \right)_T$$

S, V are allowed to freely vary in experiment

* Second derivatives of G (Hessian):

$$\left(\frac{\partial^2 G}{\partial T^2} \right) = \frac{\partial}{\partial T} \left(\frac{\partial G}{\partial T} \right) = -\left(\frac{\partial S}{\partial T} \right)_p = -\frac{1}{T} T \left(\frac{\partial S}{\partial T} \right)_p = -\frac{C_p}{T}$$

$$\left(\frac{\partial^2 G}{\partial p^2} \right) = -V\bar{K}$$

$$\frac{\partial^2 G}{\partial T \partial p} = \frac{\partial^2 G}{\partial p \partial T} = \left(\frac{\partial V}{\partial T} \right)_p = -\left(\frac{\partial S}{\partial p} \right)_T = V\beta$$

↑
Maxwell's relation.

$$\text{Hessian: } \begin{bmatrix} \frac{\partial^2 G}{\partial T^2} & \frac{\partial^2 G}{\partial T \partial p} \\ \frac{\partial^2 G}{\partial T \partial p} & \frac{\partial^2 G}{\partial p^2} \end{bmatrix} = \begin{bmatrix} -\left(\frac{C_p}{T}\right) & \frac{V\beta}{T} \\ \frac{V\beta}{T} & -\frac{V\alpha}{T} \end{bmatrix}$$

(eg) control $T, V \longrightarrow F = U - TS = F(T, V)$

$$dF = -SdT - pdV$$

* EOS: $\left(\frac{\partial F}{\partial T}\right)_V = -S \quad \left(\frac{\partial F}{\partial V}\right)_T = -p$

* second derivative: $\begin{bmatrix} \frac{\partial^2 F}{\partial T^2} & \frac{\partial^2 F}{\partial T \partial V} \\ \frac{\partial^2 F}{\partial T \partial V} & \frac{\partial^2 F}{\partial V^2} \end{bmatrix} = \begin{bmatrix} -\frac{C_p}{T} & -\frac{p}{T} \\ -\frac{p}{T} & \frac{1}{V\alpha} \end{bmatrix}$

EQUILIBRIUM (CRITERIA:

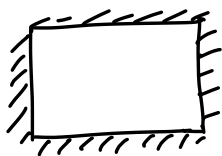
Second Law of Thermodynamics

$$dS \geq \frac{\delta Q}{T}$$

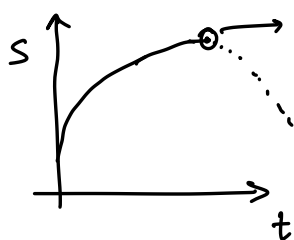
Consider an isolated system:

Consider an isolated system $\rightarrow$ all extensive variables are constant

(eg) Volume can't change $\Rightarrow$ no p-V work



II Law: $dS \geq 0$



stop equilibrium $\rightarrow$ cannot evolve since entropy would decrease

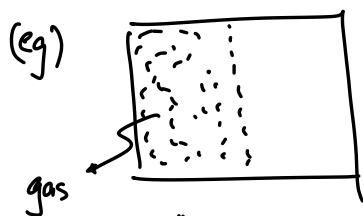
$$dS = \frac{dU}{T} + \frac{P}{T} dV$$

$U \rightarrow$ constant

$V \rightarrow$ constant

what are we maximizing over?

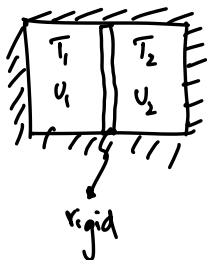
FIND EQUILIBRIUM STATE BY MAXIMIZING S OVER SOME INTERNAL EXTENSIVE VARIABLES



"internal" degree of freedom $\rightarrow$ volume of the gas

$S \rightarrow$ maximized when gas fills the whole volume.

① THERMAL EQUILIBRIUM:



$U = U_1 + U_2 = \text{constant}$ since no work or heat exchange with the environment

$$dU_1 = dU_2 = 0$$

Let U_1 be the independent internal extensive variable

$$dU_1 = T dS_1 - P dV_1 \rightarrow 0$$

$$\boxed{dU_1 = T_1 dS_1}$$

$$dU = 0$$

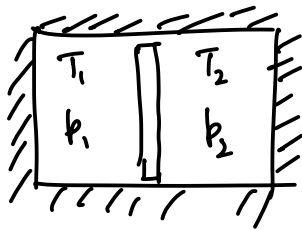
$$\Rightarrow dU_1 = -dU_2$$

$$S = S_1 + S_2 \Rightarrow dS = dS_1 + dS_2 = \frac{dU_1}{T_1} + \frac{dU_2}{T_2} = dU_1 \left(\frac{1}{T_1} - \frac{1}{T_2} \right)$$

$$S(U_1) \rightarrow \left[\frac{dS}{dU_1} = \frac{1}{T_1} - \frac{1}{T_2} \right]$$

$$\text{only if } T_1 = T_2 \Rightarrow \boxed{dS = 0} \rightarrow \text{ENTROPY MAXIMIZED}$$

② THERMAL & MECHANICAL ^{AND THERMAL} EQUILIBRIUM:



$$U = \text{constant}$$

$$= U_1 + U_2$$

$$dU = 0 \Rightarrow dU_1 = -dU_2$$

$$V = \text{constant}$$

$$dV = 0 \Rightarrow dV_1 = -dV_2$$

fundamental equation:

$$dS_1 = \frac{dU_1}{T_1} + \frac{p_1}{T_1} dV_1 \quad \& \quad dS_2 = \frac{dU_2}{T_2} + \frac{p_2}{T_2} dV_2$$

2 independent internal extensive variables:

$$U_1 \quad \& \quad V_1$$

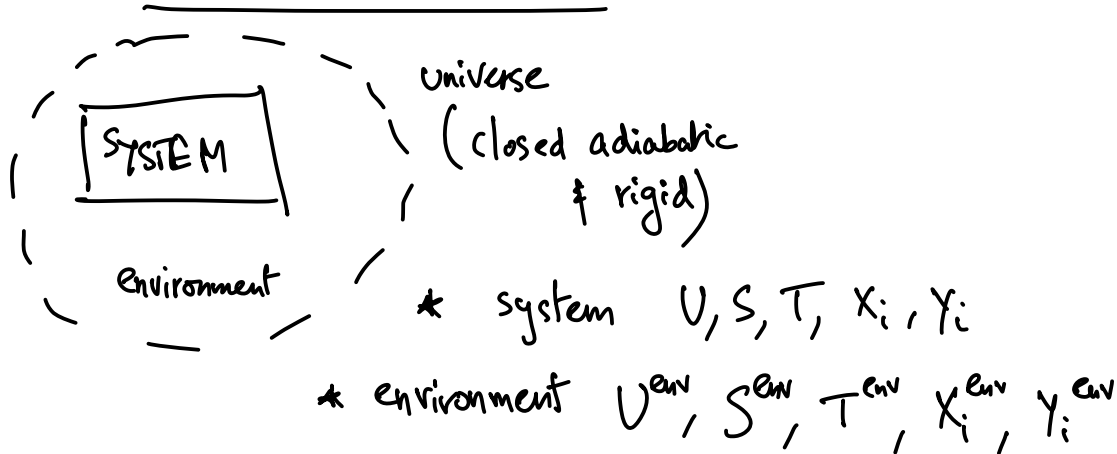
$$dS = \left(\frac{1}{T_1} - \frac{1}{T_2} \right) dU + \left(\frac{p_1}{T_1} - \frac{p_2}{T_2} \right) dV_1$$

$$dS = 0 \rightarrow \text{max entropy.}$$

$$\Rightarrow \left(\frac{\partial S}{\partial U} \right)_{V_1} = 0 \Rightarrow T_1 = T_2 \quad ; \quad \left(\frac{\partial S}{\partial V_1} \right)_{U_1} = 0 \Rightarrow p_1 = p_2$$

INTENSIVE VARIABLES CONJUGATE TO FREELY VARYING INTERNAL EXTENSIVE VARIABLES ARE EQUAL BETWEEN ALL SUBSYSTEMS

* GENERAL EQUILIBRIUM CRITERIA:



$$U^{env} + U = \text{constant}$$

$$X_i^{env} + X_i = \text{Constant} \quad \text{--- Conserved extensive variables}$$

initial state:

$$T = T^{env}$$

$$Y_i = Y_i^{env}$$

find eq^m state

$$T = T^{env}$$

$$Y_i = Y_i^{env}$$

Environment is large enough such that transformation in the system does not affect intensive variables.

CHANGE OF STATE OF THE SYSTEM

$$\Delta U, \Delta X_i, \Delta S, \dots$$

for the environment: (EULER)

$$\Delta U^{env} = T^{env} \Delta S^{env} + \sum_i Y_i^{env} \Delta X_i^{env}$$

$$\Delta U^{env} = -\Delta U$$

$$\Delta X^{env} = -\Delta X$$

$$\Rightarrow -\Delta U = T^{\text{env}} \Delta S^{\text{env}} - \sum_i \gamma_i^{\text{env}} \Delta x_i$$

$$\underline{\text{II Law}} \quad \Delta S^{\text{env}} + \Delta S \geq 0$$

$$\Rightarrow \Delta S^{\text{env}} \geq -\Delta S$$

$$T^{\text{env}} \Delta S^{\text{env}} \geq -T^{\text{env}} \Delta S$$

$$-\Delta U + \sum_i \gamma_i^{\text{env}} \Delta x_i \geq -T^{\text{env}} \Delta S$$

$$\Rightarrow \boxed{0 \geq \Delta U - T^{\text{env}} \Delta S - \sum_i \gamma_i^{\text{env}} \Delta x_i}$$

- all freely varying extensive variables
are system variables

- intensive variables are set by the
environment

$$\Rightarrow \boxed{0 \geq \Delta U - T \Delta S - \sum_i \gamma_i \Delta x_i}$$

change of state can be irreversible
only end points need to be in eq^m

EQ^m UNDER DIFFERENT BOUNDARY CONDITIONS:

① fix all extensive variables

$$S_i, x_i = \text{constant}$$

$$\Rightarrow \Delta U \leq 0 \longrightarrow U \text{ has to be minimal}$$

U is the
characteristic
potential.

② fix T and all extensive mechanical variables

$$\Delta x_i = 0, \text{ since } x_i \text{ is fixed.}$$

$$\Delta U - T \Delta S \leq 0$$

$$\Delta F \leq 0 \longrightarrow \text{minimize the characteristic potential } F(T, v)$$

③ fix T & p:

$$\Delta U - T \Delta S + p \Delta V \leq 0$$

$$\boxed{\Delta G \leq 0} \longrightarrow \text{minimize the characteristic potential } G(T, p)$$

INGENERAL: Given boundary conditions $(\gamma_i, x_i) \rightarrow \text{minimize } \boxed{\Delta \mathcal{L} = U - \sum_i \gamma_i x_i}$

PHASE TRANSITIONS & ONE COMPONENT SYSTEMS:

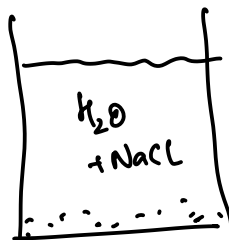
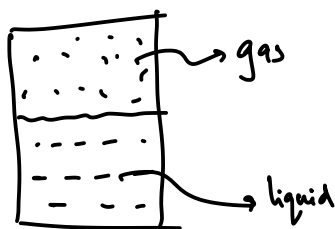
We will control $T, p \rightarrow$ look for states (phases) with lowest G

To find the lowest energy state we need to know how G changes with T, p .

$$\Rightarrow \boxed{dG = -SdT + Vdp}$$

* Reminder:

What is a phase?



Both containers have "distinct portions" with properties that are markedly different

(eg) ρ_{liquid} v/s $\rho_{\text{solid}} \rightarrow$ densities

each "distinct portion" $\rightarrow$ PHASE

* The shape of the Gibbs free energies:

$$dG = Vdp - SdT$$

$$\left(\frac{\partial G}{\partial p}\right)_T = V > 0$$

$$\left(\frac{\partial G}{\partial T}\right)_p = -S < 0$$

$$\left(\frac{\partial^2 G}{\partial T^2}\right)_p = -\left(\frac{\partial S}{\partial T}\right)_p = -C_p/T < 0$$

$$\left(\frac{\partial^2 G}{\partial p^2}\right)_T = \left(\frac{\partial V}{\partial p}\right)_T = -V\bar{\kappa}$$

$$< 0$$