

#### **Master in Electrical and Electronics Engineering**

**EE-517**: Bio-Nano-Chip Design

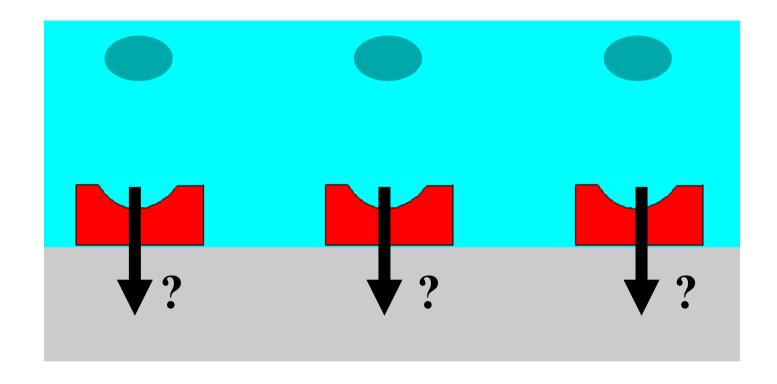
# Lecture #5 Probe Detection Principles (with Antibodies and DNA)

#### **Lecture Outline**

(Book Bio/CMOS: Chapter' paragraphs §6.1-4 & 6.8)

- Uptake Ab/Ag @ Bio/CMOS interface
- DNA hybridization @ Bio/CMOS
- Layering effects with DNA or Antibodies
- Helmholtz Planes & Debye Length

#### **CMOS/Sample sensing interface**



The interface between the CMOS circuit and the bio-sample needs to be deeply investigated and organized

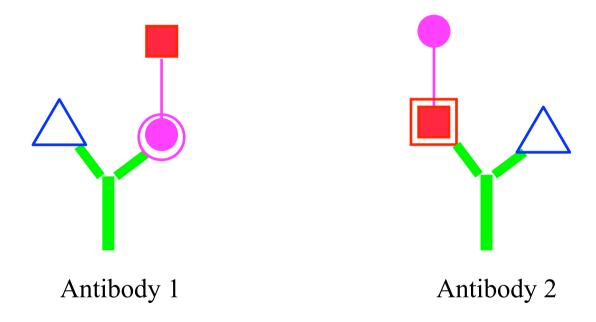
## Different Kinds of Antibody

Dealing with real cases results in a bit more complex situation than just adding antigens to antibodies with a unique perfect match. Monoclonal antibodies are, then, all antibodies that have exactly the same specificity because they are from the same cloned single cell. However, antibodies are in general secreted in blood plasma by cells that are from different cell lines. Therefore, it is easy to obtain antibodies that are all against the same antigen but that do not have exactly the same specificity: these are polyclonal antibodies. Different kinds of antibodies means different kinetics on the same antigen.

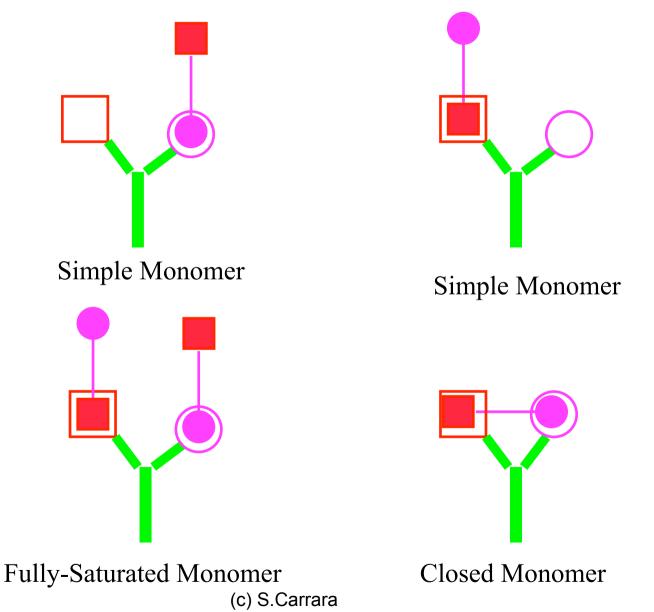
We may also obtain different kinetics by involving the same antibody. It happens when the secreted antibody possesses two different paratopes to address two different epitopes of the same antigen

(c) S.Carrara

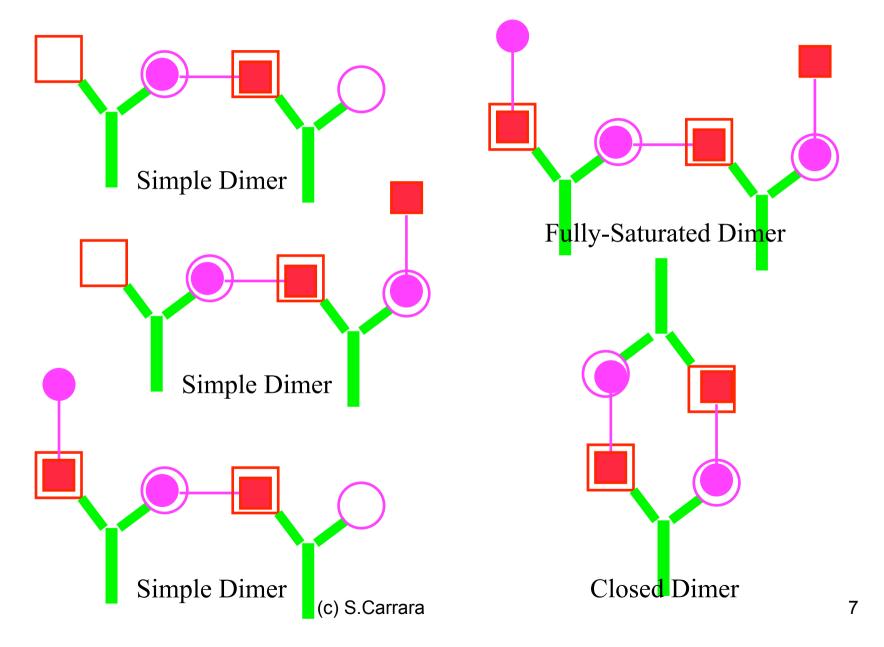
#### **Polyclonal Antibodies**



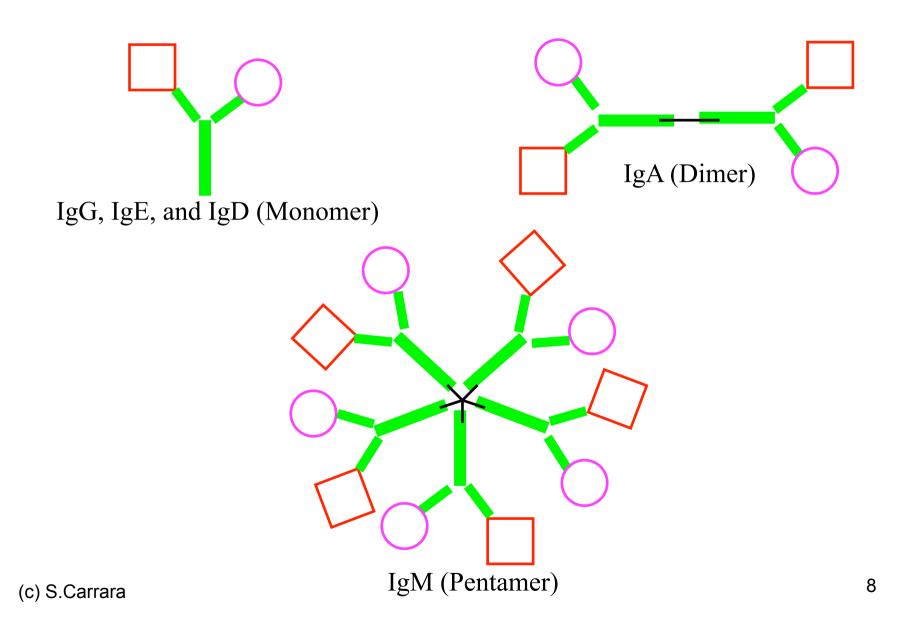
#### Different monomers with same antibody



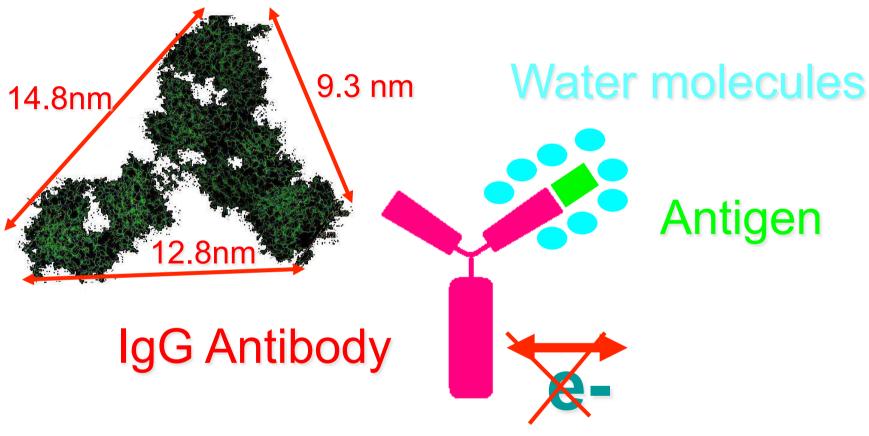
#### Different dimers with two antibodies



#### All possible classes of immunoglobulins

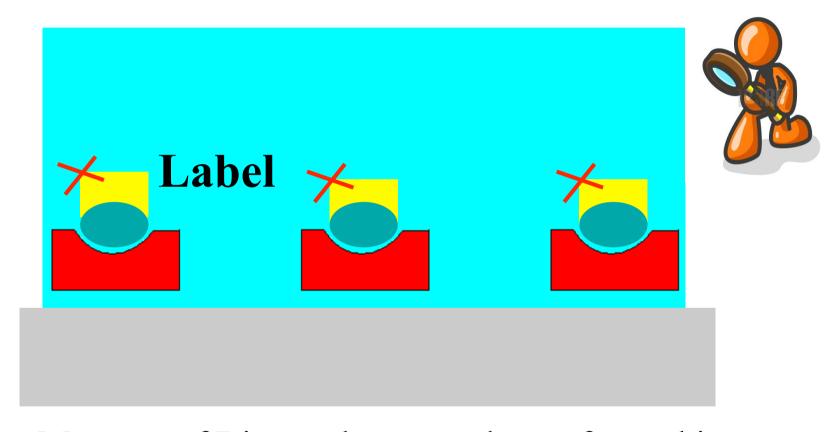


### The interaction Ab/Ag

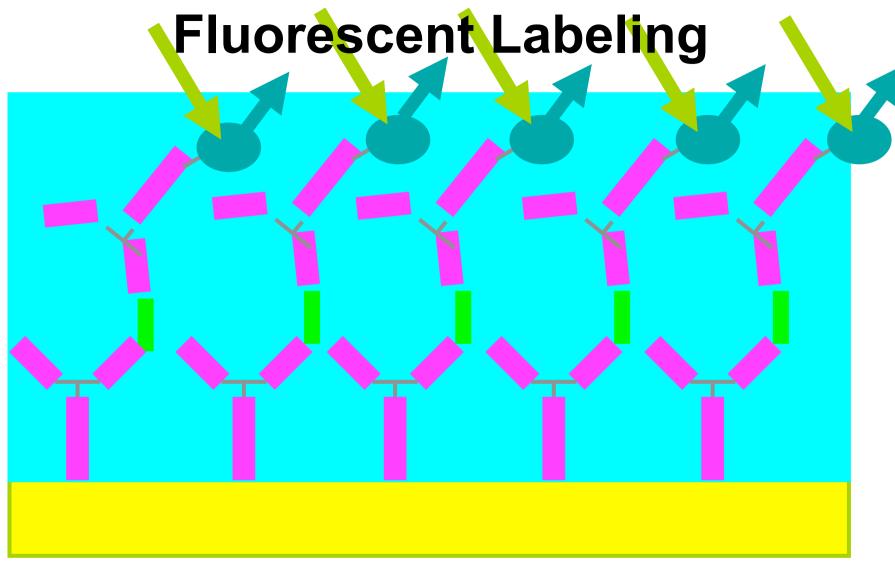


The antigen rests in a very tight binding pocket which is exactly the right size and shape to receive it. Other important factors include enthalpic contributions from van der Waals interactions and hydrogen bonds, and entropic contributions from the release of bound water upon antigen binding

### **Measuring Bio-Markers**



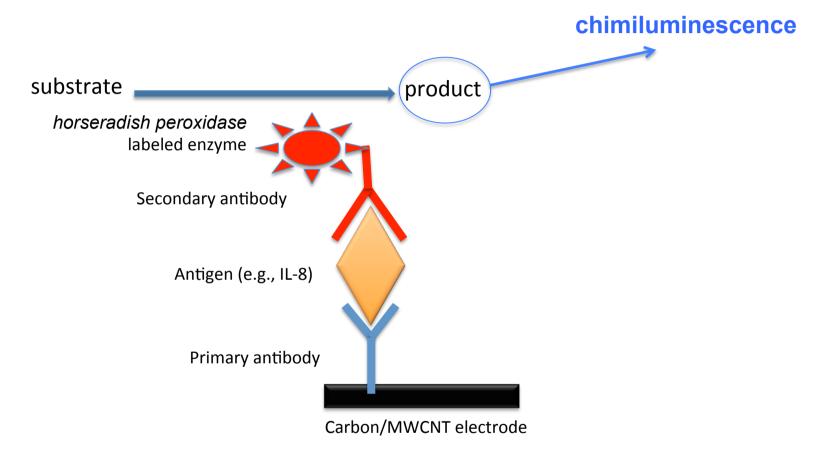
The Measure of Bio-markers may be performed in a labeled manner or in label-free mode



Antigens are specific detected by Secondary Antibodies with Fluorescent Labels

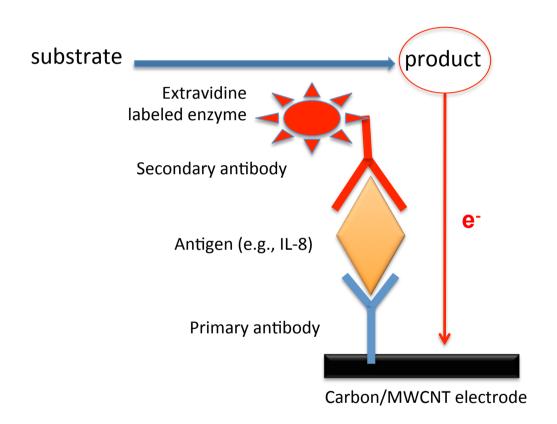
(c) S.Carrara

#### **Chemilunescence Labeling**



Antigens are specific detected by Secondary Antibodies with Chemiluninescence Labels

#### **Electrochemical Labeling**



Antigens are specific detected by Secondary Antibodies with Electrochemical Labels

#### **Electrochemical Labels**

p-Aminophenol

#### a) Process at the label

$$NH_{2} \longrightarrow OPO_{3}^{2-} \xrightarrow{-PO_{4}^{3-}+OH^{-}}$$

$$p-Aminophenylphosphate$$

$$\underline{Substrate}$$

$$\underline{Substrate}$$

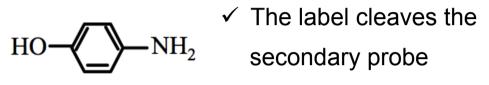
$$\underline{Alkaline}$$

$$\underline{Phosphatase}$$

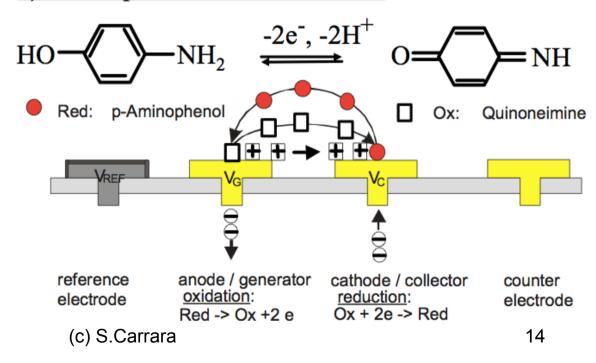
$$(ALP)$$

$$\underline{Label}$$

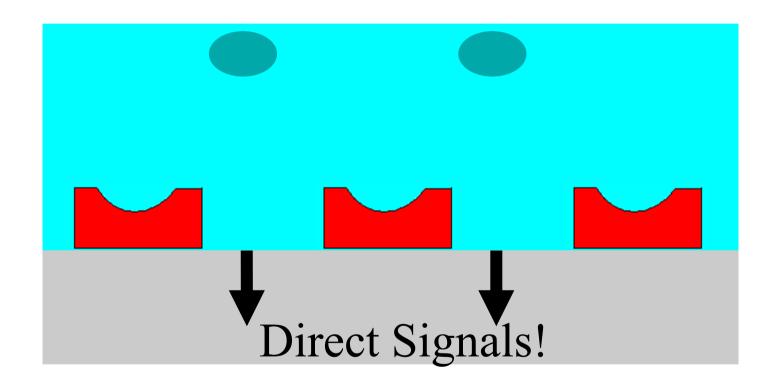
✓ The product of the cleavage is generating an oxidation process at the anode and, once oxidized, a reduction at the cathode



b) Redox process at the electrodes

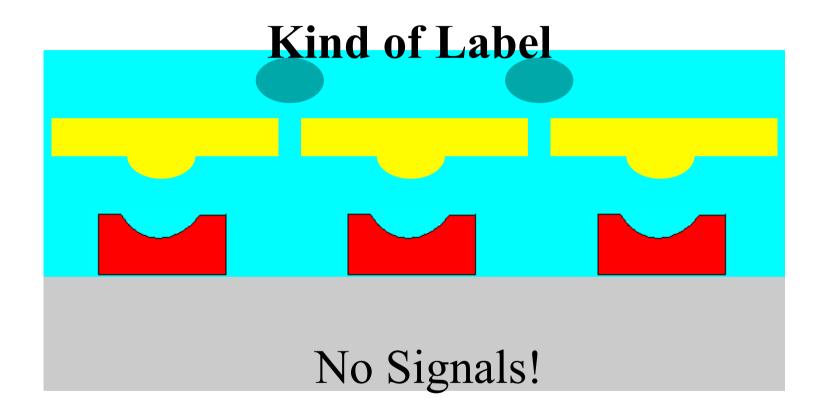


#### **CMOS/Sample interface**



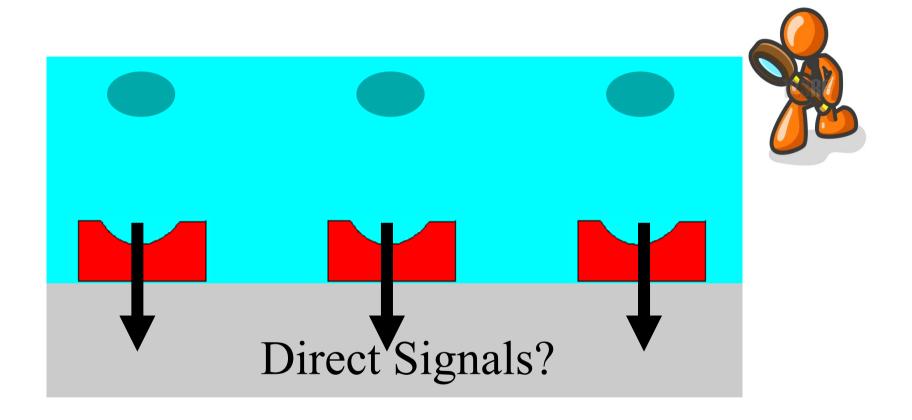
How to get direct signals of probe/target interactions?

## **CMOS/Sample interface**



How to get direct signals of probe/target interactions?

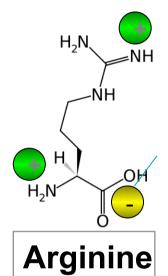
#### **CMOS/Sample interface**

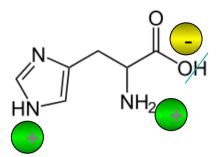


How to get direct signals of probe/target interactions?

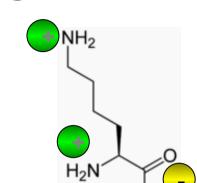
## Charged Residues

**Positively Charged** 



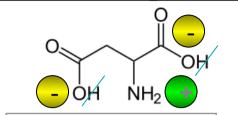


Histidine

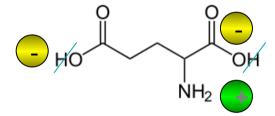


Lysine

#### Neg. Charged

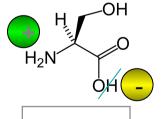


#### **Aspartic Acid**

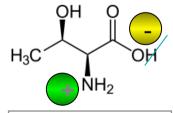


**Glutamic Acid** 

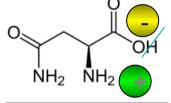
#### **Polar Uncharged**



Serine

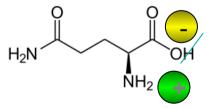


**Threonine** 



**Aspargine** 

(c) S.Carrara



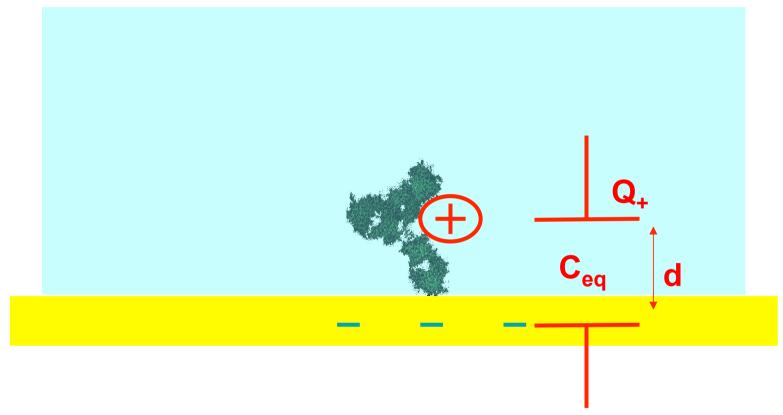
**Glutamine** 

#### The charges of an Antibody



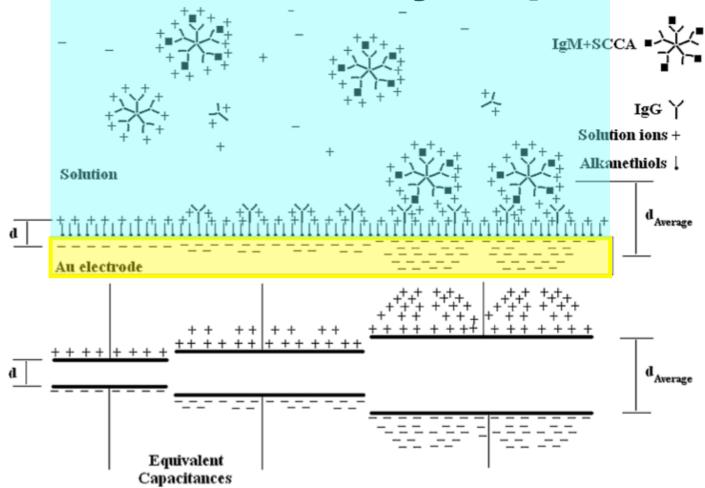
The crystallographic structure of an antibody

#### **Capacitive detection**



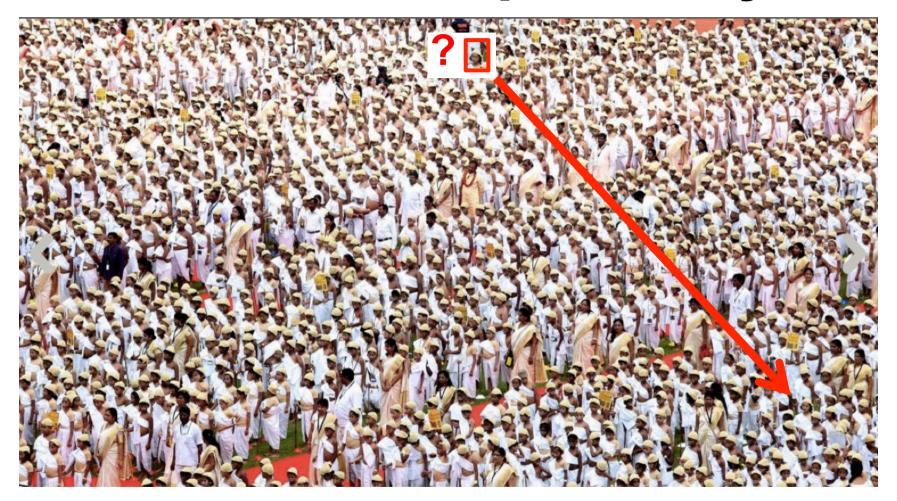
Charged residues of the antibody may affect charge carriers in the electrode

Cancer Detection by capacitance



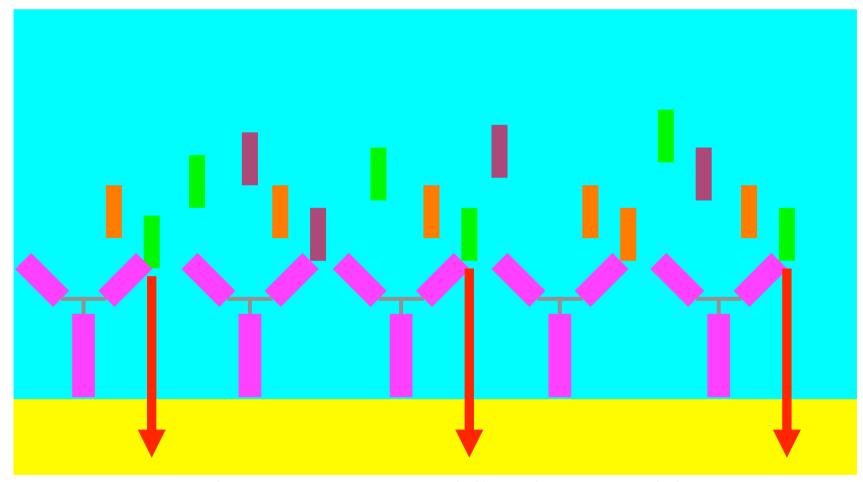
Schematic of the capacitive detection principle

## **Problem of Specificity**



Find your friend in a very crowded square!

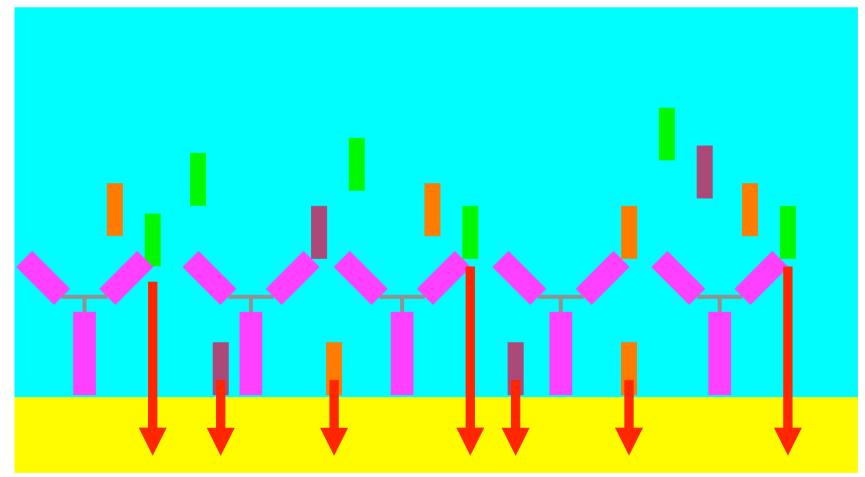
#### **Specificity of the Probe**



Antigens are specific detected by immobilizing the right antibodies

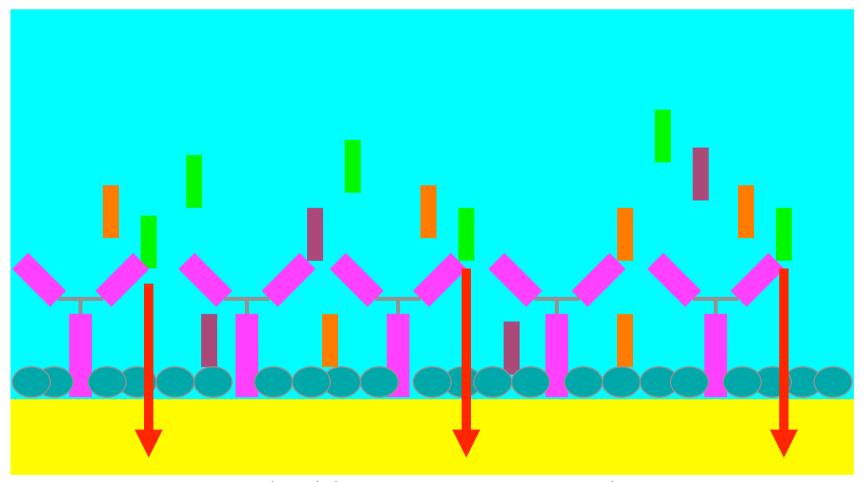
(c) S.Carrara

#### **Specificity of the Surface**



Antibody are specific but the resulting surface might not be specific enough (c) S.Carrara

#### **Specificity of the Surface**



Blocking agents are used to improve surface specificity

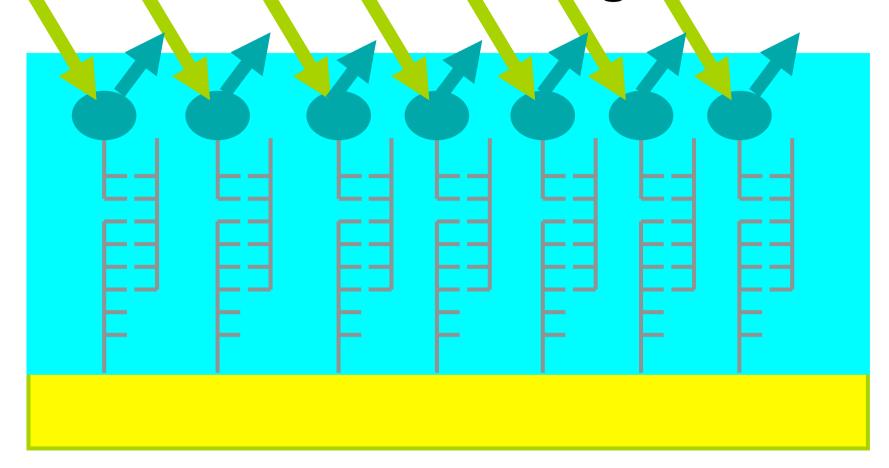
(c) S.Carrara

## Specificity of the Probe

Duplex	<b>∆G</b> [kJ/mol]
GGTTATTGG	-26.8
CCAATAACC	
GGTT <mark>A</mark> TTGG	-12.0
CCAAAAACC	
GGTT <mark>C</mark> TTGG	-12.4
CCAATAACC	

Specificity of the DNA probe: the Gibbs free energy of non-matching duplexes is not zero!

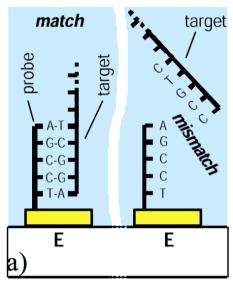
#### Fluorescent Labeling of DNA

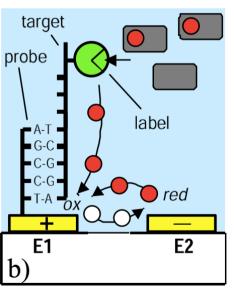


DNA targets are specific detected by Secondary DNA-probes with Fluorescent Labels

(c) S.Carrara

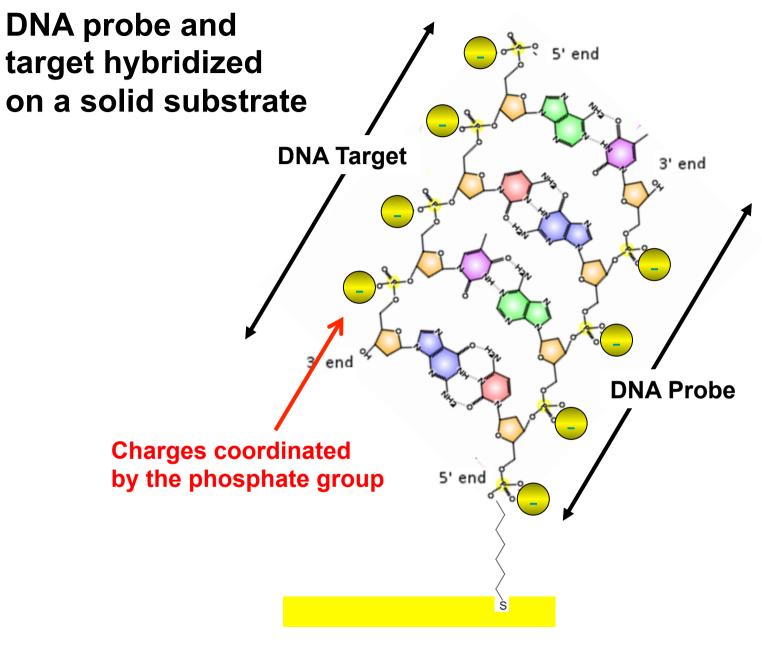
#### **Amperometric Labeling of DNA**

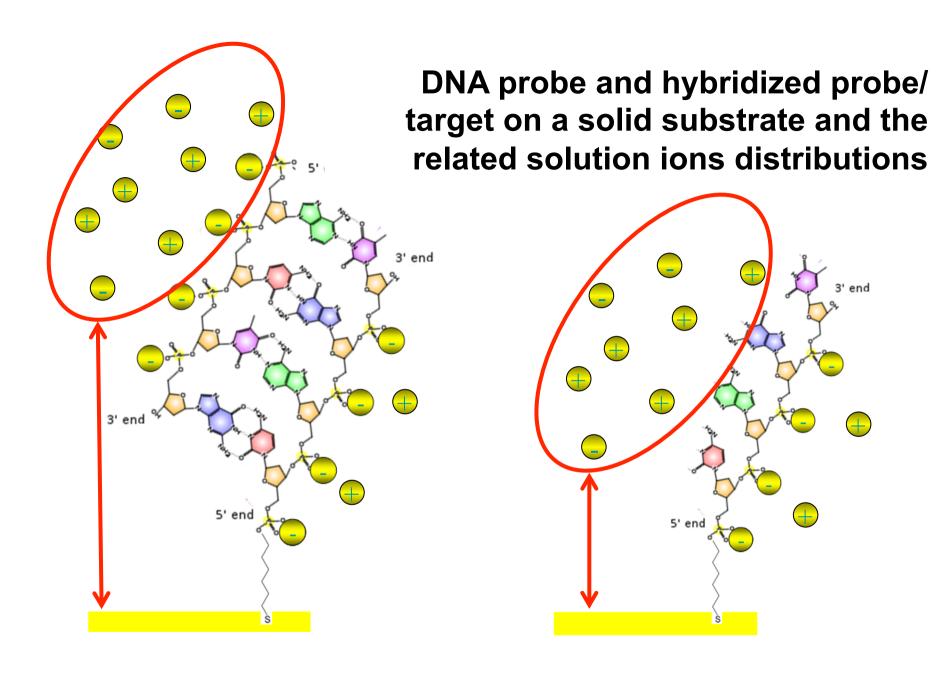




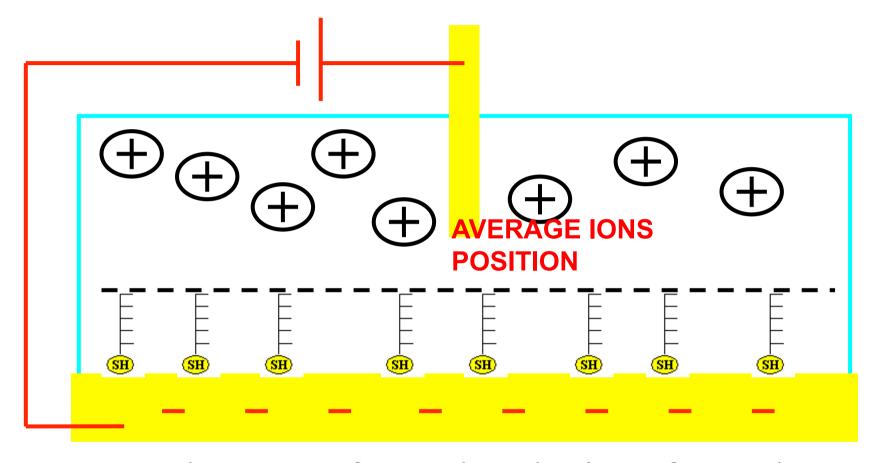
#### **How it works:**

- ✓ First, single stranded DNA molecules (about 20 bases) are immobilized by using a spotting machine on top of the gold electrodes due to gold-thiol coupling.
- ✓ Then, the chip is flooded with an analyte containing labeled target DNA ss: hybridization takes place in case of matching.
- ✓ A suitable substrate is applied to the buffer solution and it is enzymatically cleaved by the label.
- ✓ Resulting species starts an electrochemical redox process at the electrodes.
- ✓ Faradaic currents generated by the related redox process is detected and transduces DNA hybridization



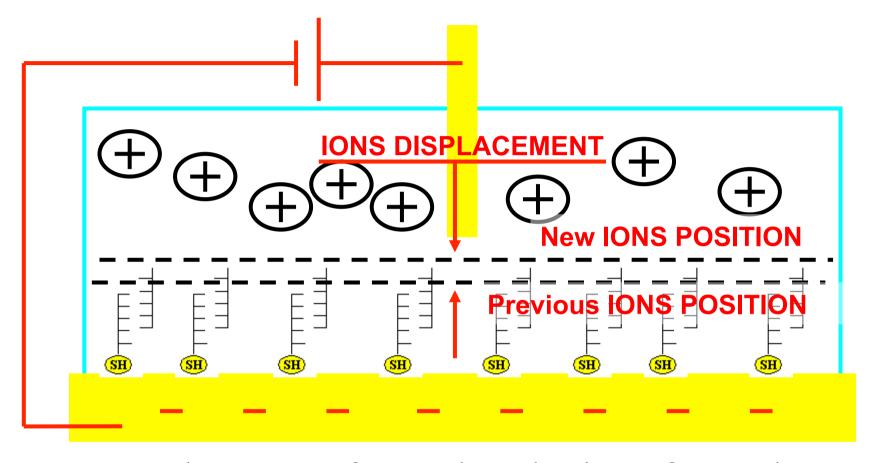


#### **Electrochemical Interface**



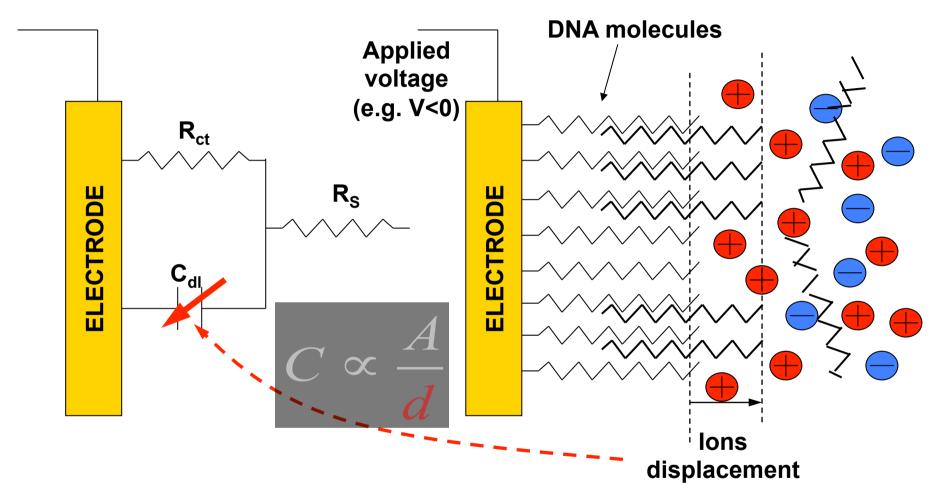
Ion planes are formed at the interface when electrodes immersed in solution are polarized

#### **Electrochemical Interface**



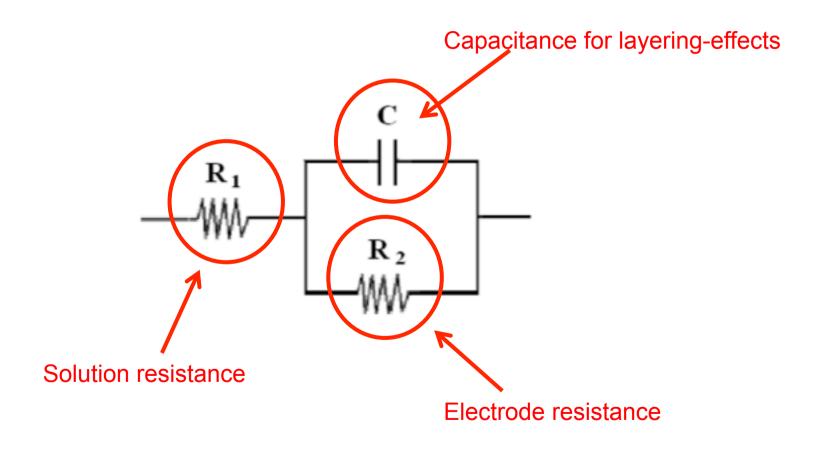
Ion planes are formed at the interface when electrodes immersed in solution are polarized

#### The Capacitance DNA Detection



Unlabeled ssDNA may be detected with capacitance measurements as due to charge displacement

## Equivalent Circuit with Layering effects (Randle Model)



#### Equivalent C of sensing electrodes

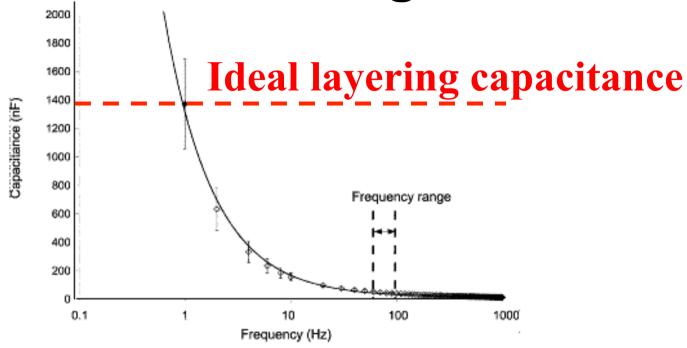
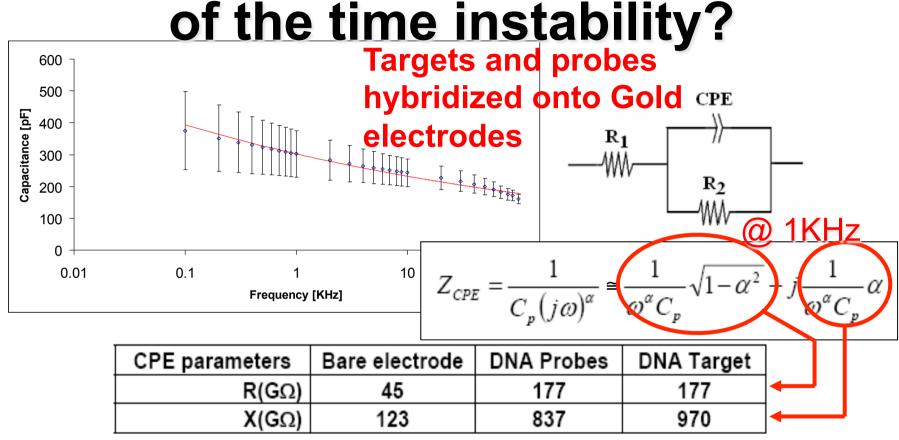


Fig. 9. Measured capacitance versus charge/discharge frequency on clean gold electrodes. The continuous line shows the fitting.

STAGNI et al.: FULLY ELECTRONIC LABEL-FREE DNA SENSOR CHIP IEEE SENSORS JOURNAL, VOL. 7, NO. 4, APRIL 2007

The equivalent capacitance of Helmholtz ion planes on bare electrodes is frequency-dependent

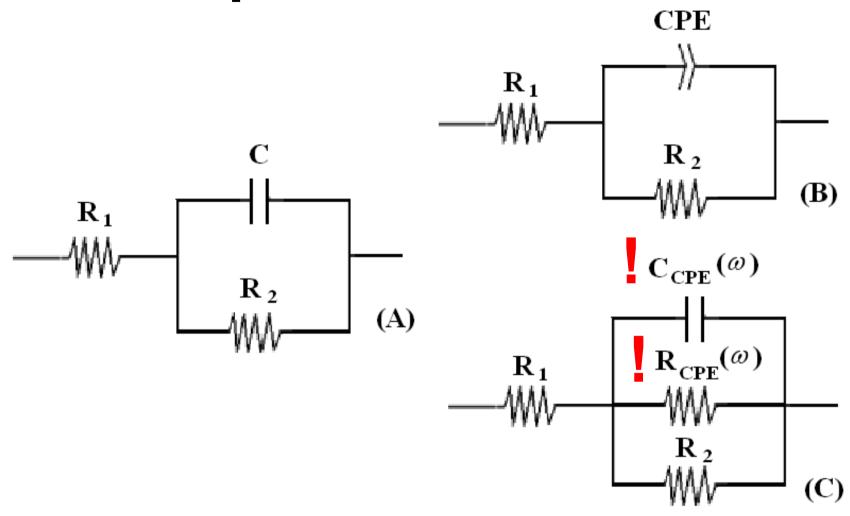
## How to understand the reason of the time instability?



S. Carrara et al., Sensors and Transducer Journal 76 (2007) 969-977

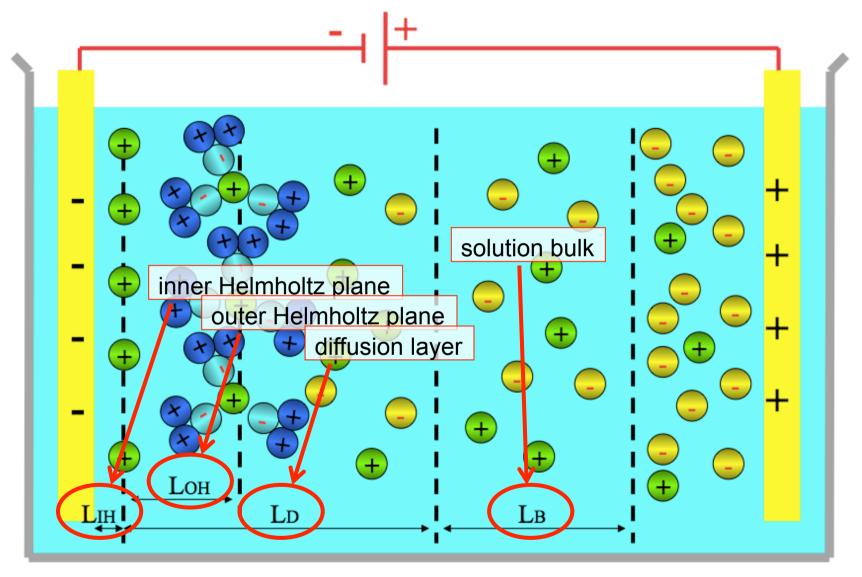
Charge transfer pathways through the DNA layer affect the ideal Capacitance behavior of the interface with the solution sample

#### **Equivalent circuits**



Equivalent circuits of DNA Bio/CMOS interface

#### **Helmholtz Planes**



## Debye Length

Charge density: 
$$\rho_e = \sum_i z_i e n_i$$

 $z_i$  = charge of species i (e.g. +2, -1, etc.)

 $n_i$  = concentration of species *i* (number per volume)

$$\nabla f = \frac{\partial f}{\partial x}\hat{\mathbf{x}} + \frac{\partial f}{\partial y}\hat{\mathbf{y}} + \frac{\partial f}{\partial z}\hat{\mathbf{z}}$$

$$\nabla^2 \phi = 0 \qquad \text{In the bulk}$$

$$\nabla \cdot \nabla f = \nabla^2 f$$

$$\nabla^2 \phi = -\frac{\rho_e}{\kappa \varepsilon_0} \qquad \text{Close to electrodes}$$

$$\nabla^2 \phi = 0$$
 In the bulk

$$\nabla^2 \phi = -\frac{\rho_e}{\kappa \varepsilon_0}$$
 Close to electrodes

For perturbation away from equilibrium at finite temperature

$$\hat{\phi} = \phi - \phi_0 \qquad \rho_e = \sum_i z_i e n_{i0} \exp\left(-\frac{z_i e \hat{\phi}}{k_B T}\right)$$

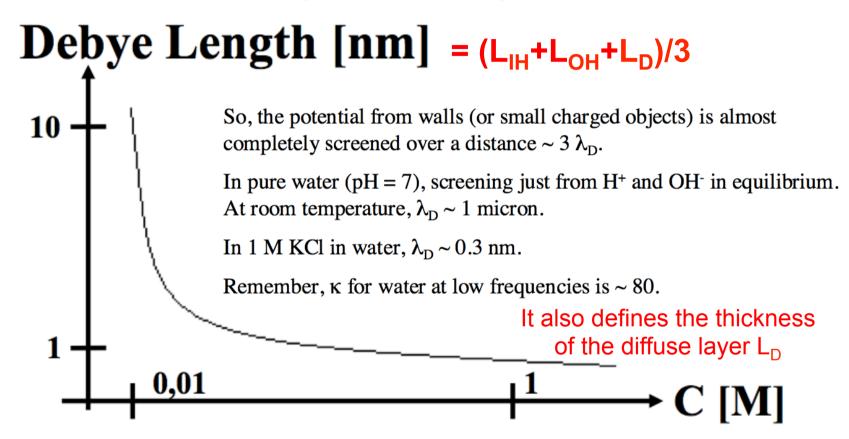
#### **Debye Length**

$$\nabla^2 \hat{\phi} = -\frac{1}{\kappa \varepsilon_0} \sum_i z_i e n_{i0} \exp \left( -\frac{z_i e \hat{\phi}}{k_B T} \right) \approx -\frac{1}{\kappa \varepsilon_0} \sum_i z_i e n_{i0} + \frac{e^2}{\kappa \varepsilon_0 k_B T} \sum_i z_i^2 n_{i0} \hat{\phi} = \frac{1}{\lambda_D^2} \hat{\phi}$$

~ 0 for equilibrium neutrality

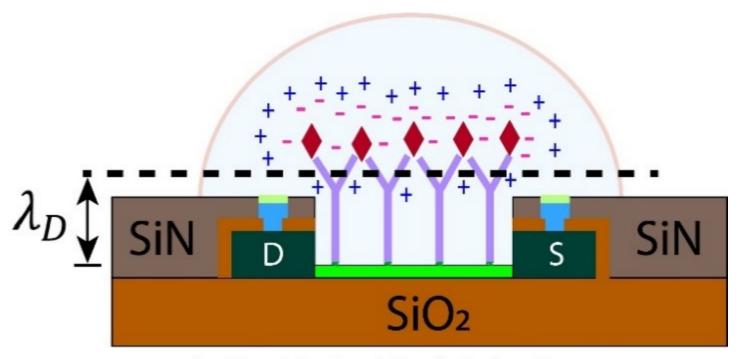
$$\lambda_D = \left(\frac{e^2}{\kappa \varepsilon_0 k_B T} \sum_i z_i^2 n_{i0}\right)^{-1/2}$$

## **Debye Length**



The Debye Length is defined as the region of charge carrier's net electrostatic effect in solution

## Capacitance Detection & Debye Length



In liquid electrical detection

The Debye Length is top important to establish the limits of charge-based detections in solution