

Quantum Chemistry

Exercises 2C

1. The Schrödinger equation for a particle of mass m constrained to move on a circle of radius a is

$$-\frac{\hbar^2}{2I} \frac{d^2\psi(\theta)}{d\theta^2} = E\psi(\theta) \quad 0 \leq \theta \leq 2\pi \quad \text{where } I = ma^2$$

This problem is sometimes called the *particle-on-a-ring* and represents the rotation of a linear molecule in a plane.

Find the solutions to this equation (that is, find the eigenfunctions and eigenvalues), including the normalization constant for the wave function.

Hint: Use the fact that the wave function must be continuous to determine the appropriate boundary conditions needed to solve the problem.

2. Recall the particle on a ring problem from above. The angular momentum operator for a particle on a ring is

$$\hat{L} = -i\hbar \frac{d}{d\theta}$$

- a. Assume that the system is in an eigenfunction corresponding to $n = 3$. What is the angular momentum of the particle with this wave function?
- b. The problem of the particle-on-a-ring is different from the particle-in-a-box in that the quantum numbers n can take on both positive and negative values. Physically, how would you interpret the difference between a particle on a ring described by wavefunctions with $n = +3$ and $n = -3$?

3. Consider the situation in which a particle on a ring can be described by the wave function

$$\Psi(\theta, t) = \frac{\sqrt{3}}{3} \frac{1}{\sqrt{2\pi}} e^{i3\theta} e^{-\frac{iE_3}{\hbar}t} + \frac{\sqrt{3}}{3} \frac{1}{\sqrt{2\pi}} e^{i6\theta} e^{-\frac{iE_6}{\hbar}t} + \frac{\sqrt{3}}{3} \frac{1}{\sqrt{2\pi}} e^{-i6\theta} e^{-\frac{iE_6}{\hbar}t}$$

where the E_n are the energies of the stationary states for a particle on a ring.

- a. Is the probability of finding the particle at a particular angle θ independent of time if the state is described by the wave function $\Psi(\theta, t)$ given above?
- b. If you were to measure the energy of a particle in this state $\Psi(\theta, t)$, what value or range of values could you obtain? If your answer includes a range of values, what is the probability for obtaining each value in that range?
- c. Calculate $\langle E \rangle$, the expectation value of the energy that you would obtain if you were to make repeated measurements on a number of identical systems. (Hint: You can do this without explicitly evaluating any integrals)

- d. Let's say you make a measurement of the energy and get $\frac{9\hbar^2}{2I}$ (call this measurement #1). You then make a subsequent measurement (#2) of the energy of the same system you measured the first time. What value or range of values could you obtain in measurement #2?
- e. Would your answer to part c) be any different if in between your two measurements of the energy (#1 and #2) you measured the particle's angular momentum? Why or why not? What possible values could you get for the angular momentum? Recall that the angular momentum operator for the particle on a ring is

$$\hat{L} = -i\hbar \frac{d}{d\theta}$$

- f. Let's say that after making a measurement of the energy and getting $\frac{9\hbar^2}{2I}$ you measure the angular position and get θ_0 for a result. If you then remeasure the position 10 seconds later, will you get the same result? Why or why not?
- g. After making these two measurements of the position (as stated in 3f.), you then measure the energy. What value or range of values could you get?

4. Evaluate the commutator $[\hat{A}, \hat{B}]$ where $\hat{A}$ and $\hat{B}$ are given below.

- a. $\hat{A} = \frac{d^2}{dx^2}$ $\hat{B} = x$
- b. $\hat{A} = \frac{d}{dx} - x$ $\hat{B} = \frac{d}{dx} + x$

5. Answer the following questions TRUE or FALSE and briefly describe the reasoning behind your answer. Be sure to read each question **carefully**.
- a. The wave function for a quantum mechanical system must be an eigenfunction of the Hamiltonian operator.
- b. Every physical observable in classical mechanics can be represented by a Hermitian operator in quantum mechanics.
- c. If the wave function for a system is not an eigenfunction of the operator $\hat{A}$, then a measurement of the observable corresponding to $\hat{A}$ might give a value that is not one of the eigenvalues of $\hat{A}$.
- d. The wave function for a quantum mechanical system is always equal to a function of time multiplied by a function of the coordinates.
- e. After making a measurement of the observable corresponding to $\hat{A}$ and getting the result a_n , the wave function becomes one of the eigenfunctions of $\hat{A}$, but you cannot be certain which one.
- f. One can sometimes measure two physical observables with infinite relative precision.