

EXAM Fonctions et Réactions Organiques III / CH-234

June 27, 2019

Duration of the exam: 120 Minutes

Answer the Questions only on the associated sheet of the question!

NO loose sheets are accepted!

NO PENCIL or RED pen is allowed!

Allowed aids: NONE

Name: _____

First Name: _____

Stud. ID No. _____

Place Number _____

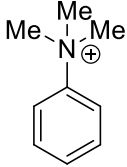
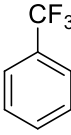
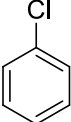
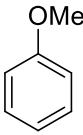
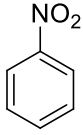
Signature _____

Exercise 1	10 Points	
Exercise 2	10 Points	
Exercise 3	10 Points	
Exercise 4	10 Points	
Exercise 5	10 Points	
Exercise 6	10 Points	
Exercise 7	10 Points	
Exercise 8	10 Points	
Exercise 9	10 Points	
Exercise 10	10 Points	
Total Points	100 Points	
GRADE		

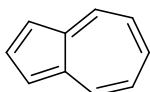
GOOD LUCK!

1. Characterize the properties of the substituents by ticking the appropriate answers! (10P)

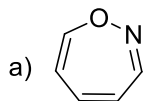
CAUTION: Wrong answers lead to a deduction of points. There is a minimum of 0 points for each subpart of the question. (2P max. for each subpart)

		σ -donor	σ -acceptor	π -donor	π -acceptor
a)		☐ yes ☐ no	☐ yes ☐ no	☐ yes ☐ no	☐ yes ☐ no
b)		☐ yes ☐ no	☐ yes ☐ no	☐ yes ☐ no	☐ yes ☐ no
c)		☐ yes ☐ no	☐ yes ☐ no	☐ yes ☐ no	☐ yes ☐ no
d)		☐ yes ☐ no	☐ yes ☐ no	☐ yes ☐ no	☐ yes ☐ no
e)		☐ yes ☐ no	☐ yes ☐ no	☐ yes ☐ no	☐ yes ☐ no

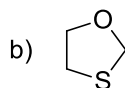
2a. Characterize the following molecules according to *Hückel* criteria by ticking the appropriate answer! (5P)

		aromatic	anti-aromatic	non-aromatic
a)		☐	☐	☐
b)		☐	☐	☐
c)		☐	☐	☐
d)		☐	☐	☐
e)		☐	☐	☐

2b. Name the asked molecule according to the Hantzsch-Widman system (2P)

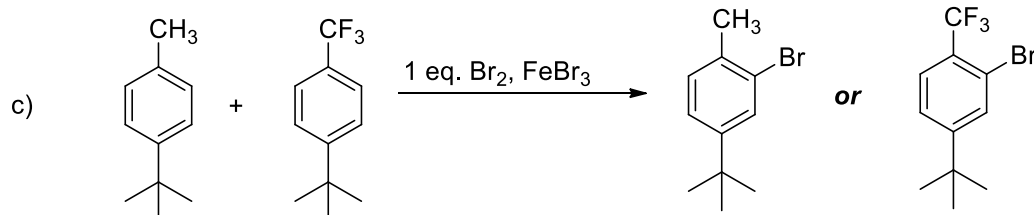
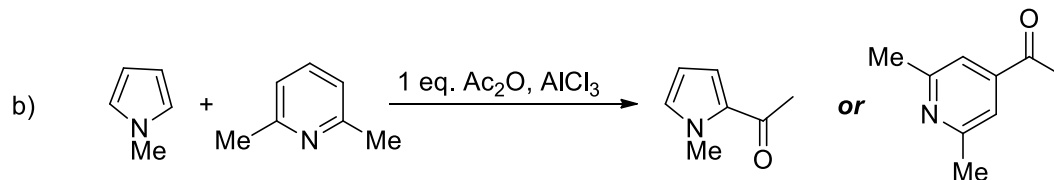
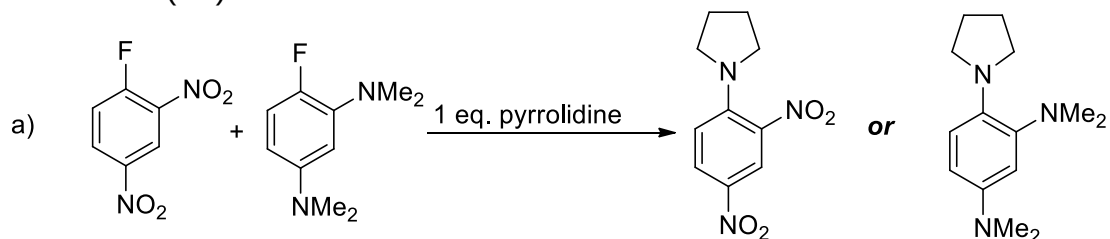


Name:

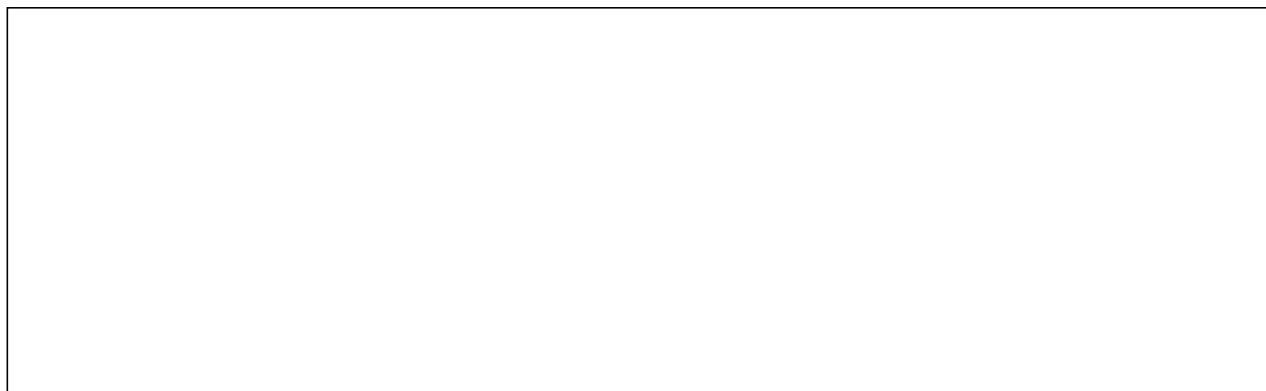


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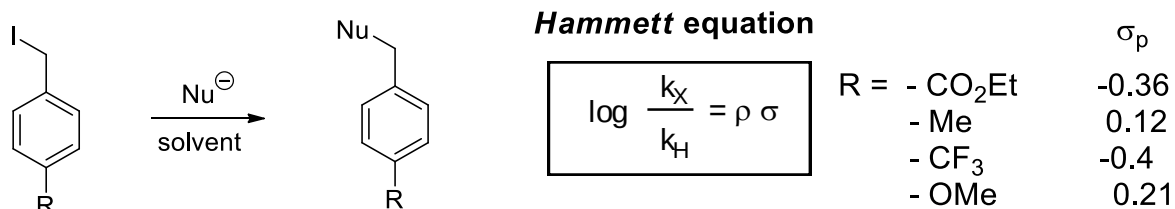
2c. Competing Reactants: An **equimolar** (1:1) mixture of two starting materials is reacted with just **1 equivalent** of the indicated reagent. **Encircle** the major product that is formed. (3P)



3a. Aniline is chlorinated by an electrophilic aromatic substitution reaction. Draw the reasonable mesomeric structures of the σ -complex intermediate for the para-chlorination path. (3P)



3b. The *Hammett*-Equation provides reaction mechanism insights. The shown values were measured for this substitution reaction. (3P)



These results indicate that the reaction proceeds **faster** with substrates having **donating substituents**. Please answer the following questions:

a) What partial charge has the transition state:

☐ positive ☐ neutral ☐ negative

b) What sign has then the value “ ρ ” in the equation?:

☐ plus ☐ minus

c) With which of the three given substituents R, the reaction proceeds fastest?

R = ☐ CN ☐ NMe₂ ☐ Et

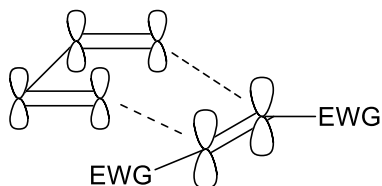
3c. The “unassigned/uncolored” frontier orbitals for a model Diels-Alder reaction is shown. Color the correct orbital lobes that are involved in the Diels-Alder process and name/characterize them. (4P)

Tick:

☐ orbital is filled
☐ orbital is empty

Orbital Name

1P



2P

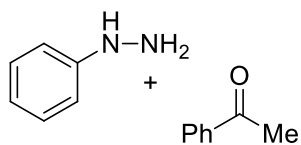
Tick:

☐ orbital is filled
☐ orbital is empty

Orbital Name

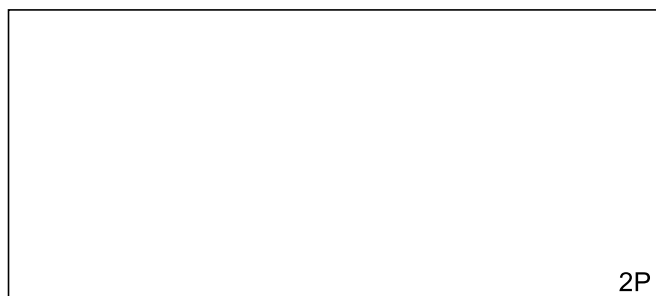
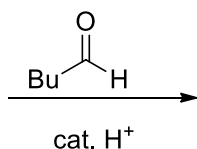
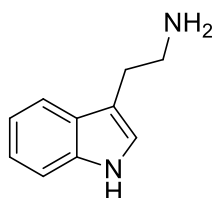
1P

4. Complete the following scheme by adding the missing structures as well as the reaction names when requested. (10P)



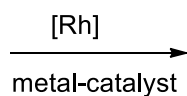
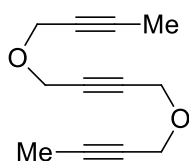
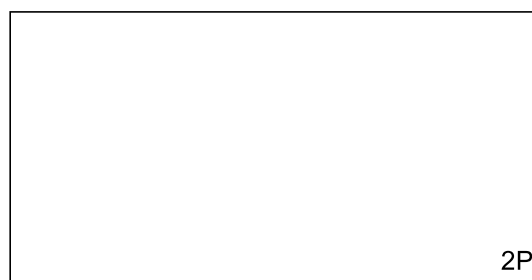
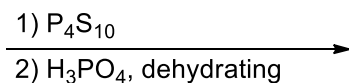
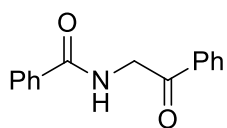
Reaction Name

1P

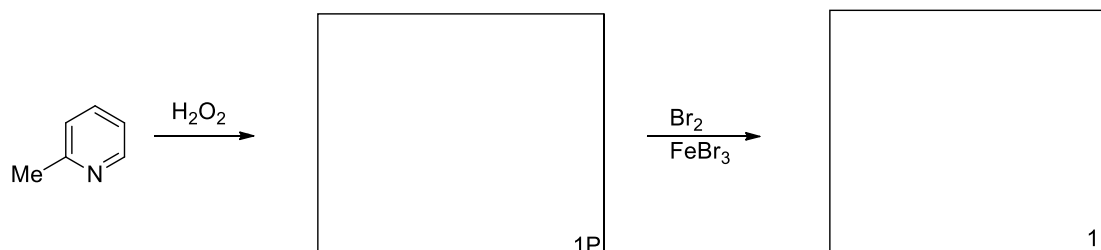
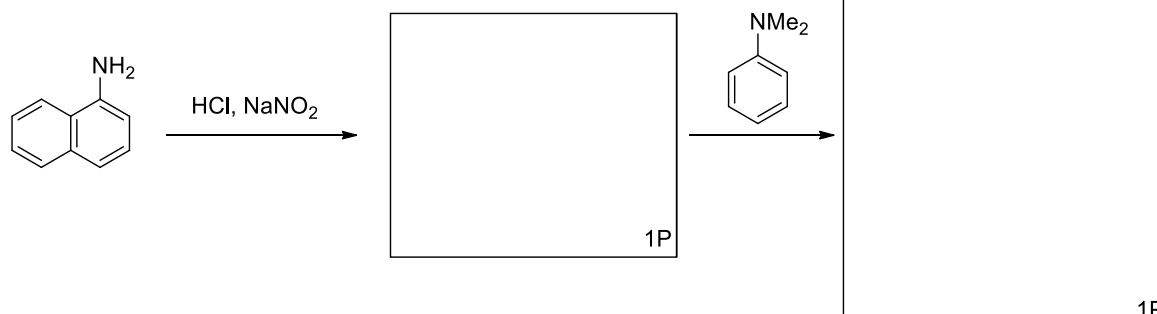
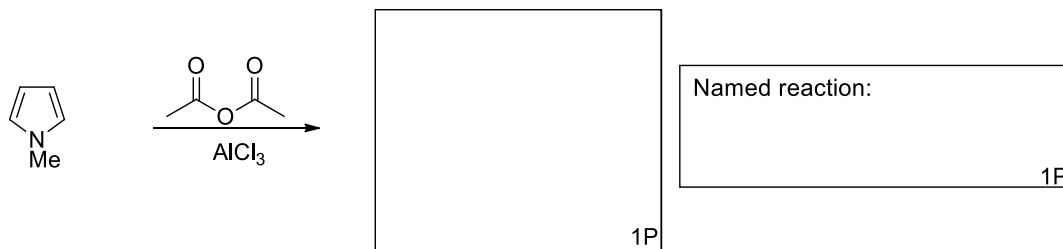


Reaction Name

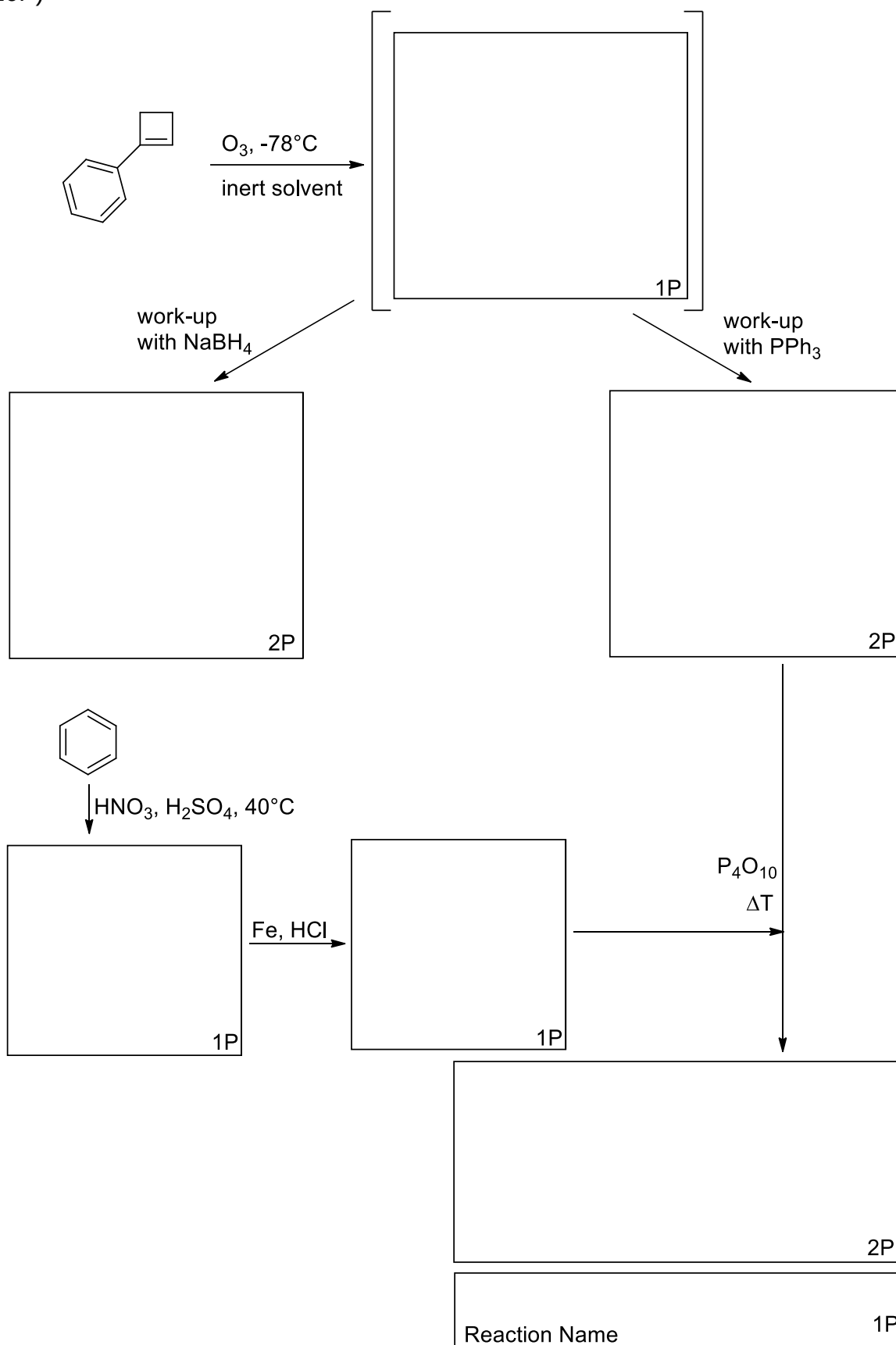
1P



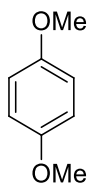
5. Complete the following schemes by adding the major product! If asked, give the name of the reaction. (10P)



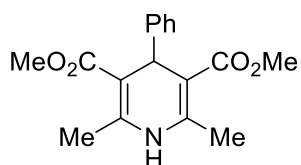
6. Complete the blanks by filling in the missing compounds and the reaction name.
(10P)



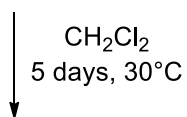
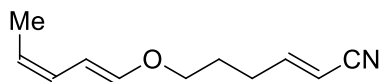
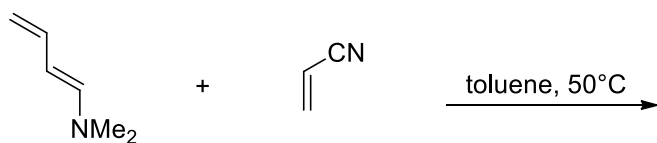
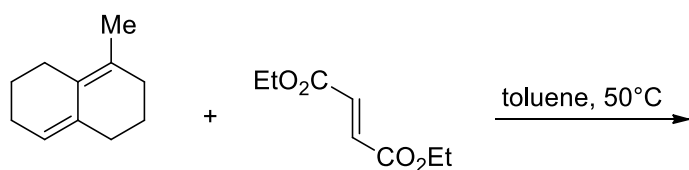
7a. 1,4-Dimethoxy benzene is submitted to a Vilsmeier-Haack formylation. Draw the required reagents and the product. Provide the appropriate details on the reaction mechanism by drawing the active reagent and relevant intermediates (4P).



7b. The Hantzsch synthesis gives access to the shown dihydropyridine. Draw the required starting materials and illustrate the reaction mechanism in appropriate detail by drawing the relevant intermediates and condensation steps! (6P)



8. Fill in the blanks! Pay especially attention to the **regio-** and **stereochemistry** of your products. **Clearly** mark the stereochemistry in your answers. (10P)



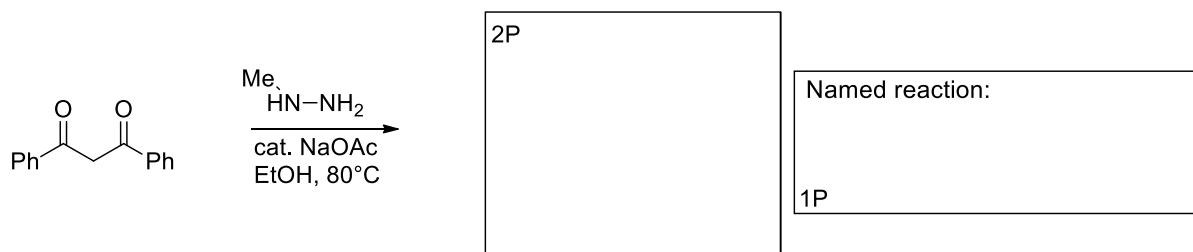
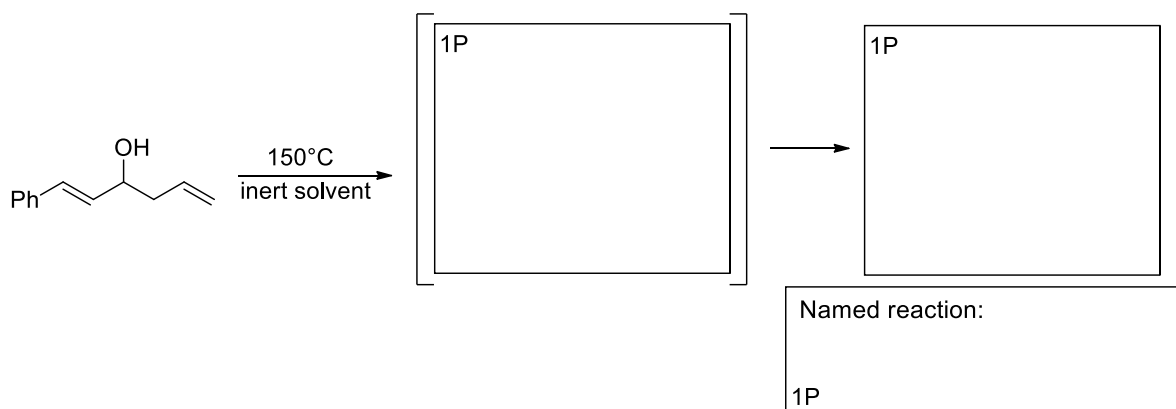
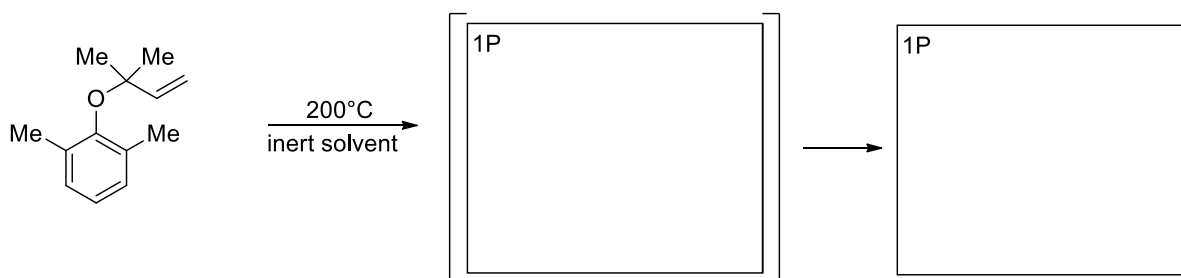
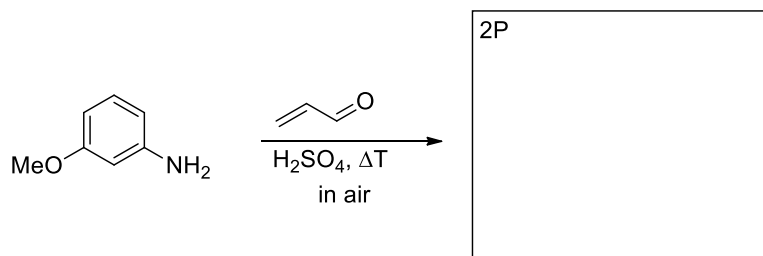
0.5 P

Major:-product

0.5 P

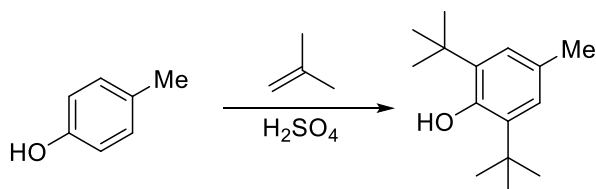
Minor:-product

9. Draw the main product of the shown reactions and provide their reaction name.
(10P)



10. 1) Do the starting materials **react** with each other under the given conditions?
 2) Are the shown products the **expected main products**?

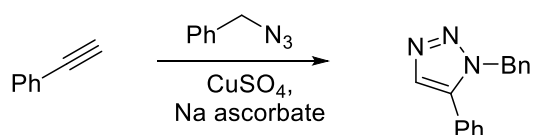
If not: draw the correct product in the box **or** explain in very **FEW** words why no reaction takes place (10P total)



Reaction? ☐ yes ☐ no

if yes,
correct main product? ☐ yes ☐ no

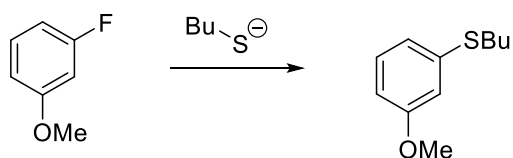
Fill out **only** when one **NO** before:
 Give a reason why there is no reaction
 or draw the correct reaction product



Reaction? ☐ yes ☐ no

if yes,
correct main product? ☐ yes ☐ no

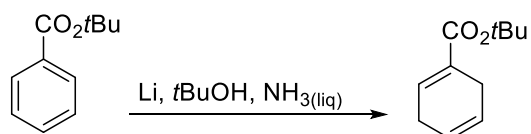
Fill out **only** when one **NO** before:
 Give a reason why there is no reaction
 or draw the correct reaction product



Reaction? ☐ yes ☐ no

if yes,
correct main product? ☐ yes ☐ no

Fill out **only** when one **NO** before:
 Give a reason why there is no reaction
 or draw the correct reaction product



Reaction? ☐ yes ☐ no

if yes,
correct main product? ☐ yes ☐ no

Fill out **only** when one **NO** before:
 Give a reason why there is no reaction
 or draw the correct reaction product